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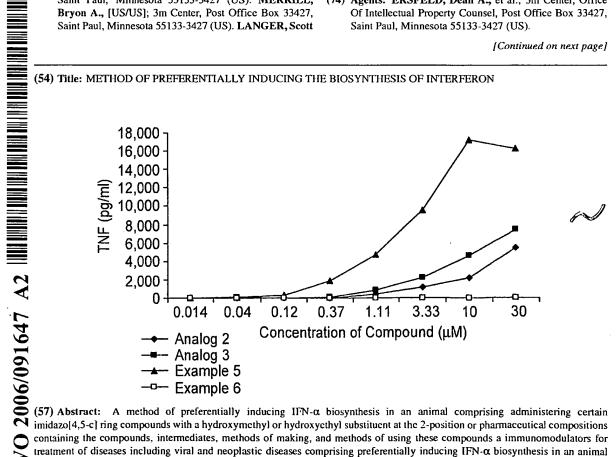
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containing the compounds, intermediates, methods of making, and methods of using these compounds a immunomodulators for treatment of diseases including viral and neoplastic diseases comprising preferentially inducing IFN-a biosynthesis in an animal are disclosed.

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# METHOD OF PREFERENTIALLY INDUCING THE BIOSYNTHESIS OF INTERFERON

## CROSS REFERENCE TO RELATED APPLICATIONS

The present invention claims priority to U.S. Provisional Application Serial Nos. 60/655452, 60/655508, 60/655380, and 60/655495, each of which was filed on February 23, 2005, and each of which is incorporated herein by reference.

## **BACKGROUND**

Certain compounds have been found to be useful as immune response modifiers (IRMs), rendering them useful in the treatment of a variety of disorders. However, there continues to be interest in and a need for compounds that have the ability to modulate the immune response, by induction of cytokine biosynthesis or other means.

## **SUMMARY**

The present invention provides a method of preferentially inducing the biosynthesis of interferon ( $\alpha$ ) (IFN- $\alpha$ ) in an animal comprising administering an effective amount of a compound of Formulas I, II, and/or III:

$$R_B$$
 $R_A$ 
 $R_1$ 
 $R_1$ 
 $R_1$ 
 $R_2$ 
 $R_1$ 
 $R_2$ 
 $R_3$ 
 $R_4$ 
 $R_1$ 

$$R_B$$
 $R_A$ 
 $R_1$ 
 $R_B$ 
 $R_A$ 
 $R_1$ 

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$$\begin{array}{c|c}
 & NH_2 \\
 & N \\
 & (CH_2)_n - O - G_2
\end{array}$$

III

wherein R<sub>A</sub>, R<sub>B</sub>, R<sub>1</sub>, G<sub>1</sub>, G<sub>2</sub>, and n are as defined below.

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It has now surprisingly been discovered that the amount of TNF-α induced by the 2-(hydroxyalkyl) substituted compounds of Formula I is substantially less than the amount of TNF-α induced by closely related analogs having an alkyl or alkyl ether substituent at the 2-position and that the compounds of Formula I, which can be administered as Formula II, Formula II, and/or Formula III, and/or a pharmaceutically acceptable salt thereof, can still retain the ability to induce the biosynthesis of IFN-a. See, for example, Figures 1 -8 below. The reduction in the amount of TNF-α induced is seen over a broad range of test concentrations. In some embodiments the amount of TNF-α induced by the compounds of Formulas I, II, and/or III is at least two-fold less than the amount of TNF-a induced by analogs having an alkyl or alkyl ether substituent at the 2-position. In other embodiments the amount of TNF-a induced by the compounds of Formulas I, II, and/or III is at least three-fold less than the amount of TNF-α induced by analogs having an alkyl or alkyl ether substituent at the 2-position. In still other embodiments the amount of TNF-α induced by the compounds of Formulas I, II, and/or III is at least four-fold less than the amount of TNF-α induced by analogs having an alkyl or alkyl ether substituent at the 2position.

The compounds or salts of Formulas I, II, and III are especially useful as immune response modifiers due to their ability to preferentially induce interferon- $\alpha$ , thus providing a benefit over compounds that also induce pro-inflammatory cytokines (e.g. TNF- $\alpha$ ) or that induce pro-inflammatory cytokines at higher levels.

A compound is said to preferentially induce IFN- $\alpha$  if, when tested according to the test methods described herein, the effective minimum concentration for IFN- $\alpha$  induction is less than the effective minimum concentration for TNF- $\alpha$  induction. In some embodiments, the effective minimum concentration for IFN- $\alpha$  induction is at least 3-fold less than the effective minimum concentration for TNF- $\alpha$  induction. In some embodiments, the effective minimum concentration for IFN- $\alpha$  induction is at least 6-fold

less than the effective minimum concentration for TNF- $\alpha$  induction. In other embodiments, the effective minimum concentration for IFN- $\alpha$  induction is at least 9-fold less than the effective minimum concentration for TNF- $\alpha$  induction. In some embodiments, when tested according to the test methods described herein, the amount TNF- $\alpha$  induced by compounds of Formulas I, II, and/or III is at or below the background level of TNF- $\alpha$  in the test method.

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The invention further provides a method of preferentially inducing the biosynthesis of IFN-α in an animal wherein an effective amount of the compound or salt of Formulas I, II, and/or III (or any one of the embodiments described herein) is administered as a pharmaceutical composition comprising a therapeutically effective amount of a compound or salt of Formulas I, II, and/or III (or any one of the embodiments described herein) and a pharmaceutically acceptable carrier.

The invention further provides a method of treating a viral infection or disease and/or treating a neoplastic disease in an animal comprising preferentially inducing the biosynthesis of IFN-α in the animal by administering an effective amount of a compound or salt of Formulas I, II, and/or III (or any one of the embodiments described herein) or a pharmaceutical composition containing an effective amount of a compound or salt of Formulas I, II, and/or III (or any one of the embodiments described herein) to the animal.

In addition, methods of synthesizing compounds of Formulas I, II, and III and intermediates useful in the synthesis of these compounds are provided.

As used herein, "a," "an," "the," "at least one," and "one or more" are used interchangeably.

The terms "comprises" and variations thereof do not have a limiting meaning where these terms appear in the description and claims.

The above summary of the present invention is not intended to describe each disclosed embodiment or every implementation of the present invention. The description that follows more particularly exemplifies illustrative embodiments. In several places throughout the description, guidance is provided through lists of examples, which examples can be used in various combinations. In each instance, the recited list serves only as a representative group and should not be interpreted as an exclusive list.

## BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 shows the IFN- $\alpha$  dose response curves (corresponding to values shown in Table 7 below) for Example 6, Analog 2, Analog 3, and Analog 5.

Figure 2 shows the TNF-α dose response curves (corresponding to values shown in Table 7 below) for Example 6, Analog 2, Analog 3, and Analog 5.

Figure 3 shows the IFN-α dose response curves (corresponding to values shown in Table 7 below) for Example 7, Analog 1, Analog 2, and Analog 4.

Figure 4 shows the TNF- $\alpha$  dose response curves (corresponding to values shown in Table 7 below) for Example 7, Analog 1, Analog 2, and Analog 4.

Figure 5 shows the IFN-α dose response curves (corresponding to values shown in Table 8 below) for Example 148, Example 149, Analog 6, and Analog 7.

Figure 6 shows the TNF-a dose response curves (corresponding to values shown in Table 8 below) for Example 148, Example 149, Analog 6, and Analog 7.

Figure 7 shows the IFN-α dose response curves (corresponding to values shown in Table 9 below) for Example 163, Analog 8, Analog 9, and Analog 10.

Figure 8 shows the TNF-α dose response curves (corresponding to values shown in Table 9 below) for Example 163, Analog 8, Analog 9, and Analog 10.

## DETAILED DESCRIPTION OF ILLUSTRATIVE EMBODIMENTS OF THE INVENTION

The present invention provides a method of preferentially inducing the biosynthesis of IFN-α in an animal comprising administering an effective amount of a compound of Formulas I, II, and/or III:

$$R_{B}$$
 $R_{A}$ 
 $R_{1}$ 
 $R_{1}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{2}$ 
 $R_{3}$ 
 $R_{1}$ 

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$$R_{B}$$
 $R_{A}$ 
 $R_{1}$ 
 $R_{1}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{3}$ 
 $R_{4}$ 
 $R_{1}$ 

II

$$\begin{array}{c|c}
 & N \\
 & N \\$$

Ш

wherein R<sub>A</sub>, R<sub>B</sub>, R<sub>1</sub>, G<sub>1</sub>, G<sub>2</sub>, and n are as defined below; and pharmaceutically acceptable salts thereof.

In one embodiment, the present invention provides a method of preferentially inducing the biosynthesis of IFN- $\alpha$  in an animal comprising administering an effective amount of a compound of the following Formula I:

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wherein:

n is 1 or 2;

 $R_{\text{A}}$  and  $R_{\text{B}}$  are each independently selected from the group consisting of:

15 hydrogen,

halogen,

alkyl,

alkenyl,

alkoxy,

20 alkylthio and

 $-N(R_9)_2$ ;

or when taken together, RA and RB form a fused aryl ring or heteroaryl ring

containing one heteroatom selected from the group consisting of N and S wherein the aryl or heteroaryl ring is unsubstituted or substituted by one or more R groups, or substituted by one R<sub>3</sub> group, or substituted by one R<sub>3</sub> group and one R group;

or when taken together, R<sub>A</sub> and R<sub>B</sub> form a fused 5 to 7 membered saturated ring, optionally containing one heteroatom selected from the group consisting of N and S, and unsubstituted or substituted by one or more R groups;

R is selected from the group consisting of:

halogen,

hydroxy,

10 alkyl,

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alkenyl,

haloalkyl,

alkoxy,

alkylthio, and

15  $-N(R_9)_2$ ;

 $R_1$  is selected from the group consisting of:

-R<sub>4</sub>,

-X-R<sub>4</sub>,

-X-Y-R4,

-X-Y-X-Y-R<sub>4</sub>, and

-X-R<sub>5</sub>;

R<sub>3</sub> is selected from the group consisting of:

-Z-R4,

-Z-X-R<sub>4</sub>,

 $-Z-X-Y-R_4$ 

-Z-X-Y-X-Y-R4, and

-Z-X-R<sub>5</sub>;

X is selected from the group consisting of alkylene, alkenylene, alkynylene, arylene, heteroarylene, and heterocyclylene wherein the alkylene, alkenylene, and alkynylene groups can be optionally interrupted or terminated by arylene, heteroarylene or heterocyclylene and optionally interrupted by one or more -O- groups;

Y is selected from the group consisting of:

$$-O-,$$

$$-S(O)_{0-2}-,$$

$$-S(O)_{2}-N(R_{8})-,$$

$$-C(R_{6})-,$$

$$-C(R_{6})-0-,$$

$$-O-C(R_{6})-,$$

$$-O-C(O)-O-,$$

$$-N(R_{8})-Q-,$$

$$-C(R_{6})-N(R_{8})-,$$

$$-C(R_{6})-N(OR_{9})-,$$

$$-O-N(R_{8})-Q-,$$

$$-O-N=C(R_{4})-,$$

$$-C(=N-O-R_{8})-,$$

$$-CH(-N(-O-R_{8})-Q-R_{4})-,$$

$$-N-C(R_{6})-N-W-$$

$$R_{7}$$

Z is a bond or -O-;

R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl, heteroaryloxyalkylenyl, alkylheteroarylenyl, and heterocyclyl wherein the alkyl, alkenyl,

alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl, heteroaryloxyalkylenyl, alkylheteroarylenyl, and heterocyclyl groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, hydroxyalkyl, haloalkyl, haloalkoxy, halogen, nitro, hydroxy, mercapto, cyano, aryl, aryloxy, arylalkyleneoxy, heteroaryl, heteroaryloxy, heteroarylalkyleneoxy, heterocyclyl, amino, alkylamino, dialkylamino, (dialkylamino)alkyleneoxy, and in the case of alkyl, alkenyl, alkynyl, and heterocyclyl, oxo;

R<sub>5</sub> is selected from the group consisting of

 $R_6$  is selected from the group consisting of =O and =S;

 $R_7$  is  $C_{2-7}$  alkylene;

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R<sub>8</sub> is selected from the group consisting of hydrogen, alkyl, alkoxyalkylenyl, hydroxyalkylenyl, arylalkylenyl, and heteroarylalkylenyl;

Ro is selected from the group consisting of hydrogen and alkyl;

R<sub>10</sub> is C<sub>3-8</sub> alkylene;

A is selected from the group consisting of -O-, -C(O)-, -CH<sub>2</sub>-, -S(O)<sub>0-2</sub>-, and -N(O-R<sub>4</sub>)-;

A' is selected from the group consisting of -O-, -S(O) $_{0-2}$ -, -N(-Q-R<sub>4</sub>)-, and -CH<sub>2</sub>-;

Q is selected from the group consisting of a bond,  $-C(R_6)$ -,  $-C(R_6)$ -,  $-S(O)_2$ -,

 $-C(R_6)-N(R_8)-W-$ ,  $-S(O)_2-N(R_8)-$ ,  $-C(R_6)-O-$ ,  $-C(R_6)-S-$ , and  $-C(R_6)-N(OR_9)-$ ;

V is selected from the group consisting of  $-C(R_6)$ -,  $-O-C(R_6)$ -,  $-N(R_8)-C(R_6)$ -, and  $-S(O)_2$ -;

W is selected from the group consisting of a bond, -C(O)-, and  $-S(O)_2$ -; and a and b are independently integers from 1 to 6 with the proviso that a + b is  $\leq 7$ ; or a pharmaceutically acceptable salt thereof to the animal.

In another embodiment, the present invention provides a method of preferentially inducing the biosynthesis of IFN- $\alpha$  in an animal comprising administering an effective amount of a compound of the following Formula II, which is a prodrug:

$$R_{B}$$
 $R_{A}$ 
 $R_{1}$ 
 $G_{1}$ 
 $CCH_{2})_{n}$ -OH

II

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wherein:

G<sub>1</sub> is selected from the group consisting of:

-C(O)-R',

α-aminoacyl,

10 α-aminoacyl-α-aminoacyl,

-C(O)-O-R',

-C(O)-N(R'')R',

-C(=NY')-R'

-CH(OH)-C(O)-OY',

-CH(OC<sub>1-4</sub> alkyl)Y<sub>0</sub>,

-CH<sub>2</sub>Y<sub>1</sub>, and

 $-CH(CH_3)Y_1;$ 

R' and R" are independently selected from the group consisting of  $C_{1-10}$  alkyl,  $C_{3-7}$  cycloalkyl, phenyl, benzyl, and 2-phenylethyl, each of which may be unsubstituted or substituted by one or more substituents independently selected from the group consisting of halogen, hydroxy, nitro, cyano, carboxy,  $C_{1-6}$  alkyl,  $C_{1-4}$  alkoxy, aryl, heteroaryl, aryl- $C_{1-4}$  alkylenyl, heteroaryl- $C_{1-4}$  alkylenyl, halo- $C_{1-4}$  alkylenyl, halo- $C_{1-4}$  alkoxy, -O-C(O)- $CH_3$ , -C(O)-O- $CH_3$ , -C(O)- $NH_2$ , -O- $CH_2$ -C(O)- $NH_2$ , and - $S(O)_2$ - $NH_2$ , with the proviso that R" can also be hydrogen;

 $\alpha$ -aminoacyl is an  $\alpha$ -aminoacyl group derived from an  $\alpha$ -amino acid selected from the group consisting of racemic, D-, and L-amino acids;

Y' is selected from the group consisting of hydrogen,  $C_{1-6}$  alkyl, and benzyl; Y<sub>0</sub> is selected from the group consisting of  $C_{1-6}$  alkyl, carboxy- $C_{1-6}$  alkylenyl,

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amino-C<sub>1-4</sub> alkylenyl, mono-N-C<sub>1-6</sub> alkylamino-C<sub>1-4</sub> alkylenyl, and
          di-N, N-C<sub>1-6</sub> alkylamino-C<sub>1-4</sub> alkylenyl;
                   Y<sub>1</sub> is selected from the group consisting of mono-N-C<sub>1-6</sub> alkylamino,
          di-N, N-C<sub>1-6</sub> alkylamino, morpholin-4-yl, piperidin-1-yl, pyrrolidin-1-yl, and
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          4-C<sub>1-4</sub> alkylpiperazin-1-yl;
                   n is 1 or 2;
                   R_{\text{A}} and R_{\text{B}} are each independently selected from the group consisting of:
                           hydrogen,
                           halogen,
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                           alkyl,
                           alkenyl,
                           alkoxy,
                           alkylthio and
                           -N(R_9)_2;
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                   or when taken together, RA and RB form a fused aryl ring or heteroaryl ring
          containing one heteroatom selected from the group consisting of N and S wherein the aryl
          or heteroaryl ring is unsubstituted or substituted by one or more R groups, or substituted
          by one R<sub>3</sub> group, or substituted by one R<sub>3</sub> group and one R group;
                  or when taken together, RA and RB form a fused 5 to 7 membered saturated
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          ring, optionally containing one heteroatom selected from the group consisting of N and S,
          and unsubstituted or substituted by one or more R groups;
                  R is selected from the group consisting of:
                          halogen,
                           hydroxy,
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                           alkyl,
                           alkenyl,
                           haloalkyl,
                           alkoxy,
                           alkylthio, and
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                           -N(R_9)_2;
                  R<sub>1</sub> is selected from the group consisting of:
                           -R<sub>4</sub>,
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-X-R<sub>4</sub>,
-X-Y-R<sub>4</sub>,
-X-Y-R<sub>4</sub>,
and
-X-R<sub>5</sub>;

R<sub>3</sub> is selected from the group consisting of:
-Z-R<sub>4</sub>,
-Z-X-R<sub>4</sub>,
-Z-X-Y-R<sub>4</sub>,
-Z-X-Y-R<sub>4</sub>,
and
-Z-X-Y-R<sub>4</sub>,
and
-Z-X-Y-R<sub>5</sub>;
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X is selected from the group consisting of alkylene, alkenylene, alkynylene, arylene, heteroarylene, and heterocyclylene wherein the alkylene, alkenylene, and alkynylene groups can be optionally interrupted or terminated by arylene, heteroarylene or heterocyclylene and optionally interrupted by one or more -O- groups;

15 Y is selected from the group consisting of:

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-O-,
                          -S(O)_{0-2}-,
                          -S(O)_2-N(R_8)-,
                          -C(R_6)-,
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                          -C(R_6)-O-,
                          -O-C(R_6)-,
                          -O-C(O)-O-,
                          -N(R_8)-Q_{-}
                          -C(R_6)-N(R_8)-,
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                          -O-C(R_6)-N(R_8)-,
                          -C(R_6)-N(OR_9)-,
                          -O-N(R_8)-Q-
                          -O-N=C(R_4)-,
                          -C(=N-O-R_8)-,
                          -CH(-N(-O-R<sub>8</sub>)-Q-R<sub>4</sub>)-,
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```

$$N-Q R_{10}$$
,
 $N-Q R_{10}$ 
,
 $N-Q R_{7}$ 
,
 $R_{7}$ 
,
 $R_{7}$ 
,
 $R_{7}$ 
,
 $R_{10}$ 
, and
 $R_{10}$ 

Z is a bond or -O-;

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0.

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0

R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl, heteroarylalkylenyl, heteroaryloxyalkylenyl, alkylheteroarylenyl, and heterocyclyl wherein the alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl, heteroaryloxyalkylenyl, alkylheteroarylenyl, and heterocyclyl groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, hydroxyalkyl, haloalkyl, haloalkoxy, halogen, nitro, hydroxy, mercapto, cyano, aryl, aryloxy, arylalkyleneoxy, heteroaryl, heteroaryloxy, heteroarylalkyleneoxy, heterocyclyl, amino, alkylamino, dialkylamino, (dialkylamino)alkyleneoxy, and in the case of alkyl, alkenyl, alkynyl, and heterocyclyl, oxo;

R<sub>5</sub> is selected from the group consisting of

R<sub>6</sub> is selected from the group consisting of =O and =S;

 $R_7$  is  $C_{2-7}$  alkylene;

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R<sub>8</sub> is selected from the group consisting of hydrogen, alkyl, alkoxyalkylenyl, hydroxyalkylenyl, arylalkylenyl, and heteroarylalkylenyl;

R<sub>9</sub> is selected from the group consisting of hydrogen and alkyl;

R<sub>10</sub> is C<sub>3-8</sub> alkylene;

A is selected from the group consisting of -O-, -C(O)-, -CH<sub>2</sub>-, -S(O)<sub>0-2</sub>-, and -N(Q-R<sub>4</sub>)-;

A' is selected from the group consisting of -O-, -S(O)<sub>0-2</sub>-, -N(-Q-R<sub>4</sub>)-, and -CH<sub>2</sub>-;

Q is selected from the group consisting of a bond,  $-C(R_6)$ -,  $-C(R_6)$ -,  $-C(R_6)$ -,  $-S(O)_2$ -,  $-C(R_6)$ - $N(R_8)$ -W-,  $-S(O)_2$ - $N(R_8)$ -,  $-C(R_6)$ -O-,  $-C(R_6)$ -S-, and  $-C(R_6)$ - $N(OR_9)$ -;

V is selected from the group consisting of  $-C(R_6)$ -,  $-O-C(R_6)$ -,  $-N(R_8)-C(R_6)$ -, and  $-S(O)_2$ -;

W is selected from the group consisting of a bond, -C(O)-, and  $-S(O)_2$ -; and a and b are independently integers from 1 to 6 with the proviso that a + b is  $\leq 7$ ; or a pharmaceutically acceptable salt thereof to the animal.

In another embodiment, the present invention provides a method of preferentially inducing the biosynthesis of IFN- $\alpha$  in an animal comprising administering an effective amount of a compound of the following Formula III, which is a prodrug:

$$R_B$$
 $R_A$ 
 $R_1$ 
 $N$ 
 $(CH_2)_n$ -O-G

III

wherein:

G<sub>2</sub> is selected from the group consisting of:

 $-X_2-C(O)-R'$ ,

α-aminoacyl,

α-aminoacyl-α-aminoacyl,

-X2-C(O)-O-R', and

-C(O)-N(R")R';

X<sub>2</sub> is selected from the group consisting of a bond; -CH<sub>2</sub>-O-; -CH(CH<sub>3</sub>)-O-;

-C(CH<sub>3</sub>)<sub>2</sub>-O-; and, in the case of - $X_2$ -C(O)-O-R', -CH<sub>2</sub>-NH-;

R' and R" are independently selected from the group consisting of C<sub>1-10</sub> alkyl, C<sub>3-7</sub> cycloalkyl, phenyl, benzyl, and 2-phenylethyl, each of which may be unsubstituted or substituted by one or more substituents independently selected from the group consisting of halogen, hydroxy, nitro, cyano, carboxy, C<sub>1-6</sub> alkyl, C<sub>1-4</sub> alkoxy, aryl, heteroaryl, aryl-C<sub>1-4</sub> alkylenyl, heteroaryl-C<sub>1-4</sub> alkylenyl, halo-C<sub>1-4</sub> alkylenyl, halo-C<sub>1-4</sub> alkoxy, -O-C(O)-CH<sub>3</sub>, -C(O)-O-CH<sub>3</sub>, -C(O)-NH<sub>2</sub>, -O-CH<sub>2</sub>-C(O)-NH<sub>2</sub>, -NH<sub>2</sub>, and -S(O)<sub>2</sub>-NH<sub>2</sub>, with the proviso that R" can also be hydrogen;

 $\alpha$ -aminoacyl is an  $\alpha$ -aminoacyl group derived from an  $\alpha$ -amino acid selected from the group consisting of racemic, D-, and L-amino acids;

n is 1 or 2;

R<sub>A</sub> and R<sub>B</sub> are each independently selected from the group consisting of:

hydrogen,

halogen,

alkyl,

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alkenyl,

alkoxy,

alkylthio and

 $-N(R_9)_2;$ 

or when taken together, R<sub>A</sub> and R<sub>B</sub> form a fused aryl ring or heteroaryl ring containing one heteroatom selected from the group consisting of N and S wherein the aryl or heteroaryl ring is unsubstituted or substituted by one or more R groups, or substituted by one R<sub>3</sub> group, or substituted by one R<sub>3</sub> group and one R group;

or when taken together,  $R_A$  and  $R_B$  form a fused 5 to 7 membered saturated ring, optionally containing one heteroatom selected from the group consisting of N and S, and unsubstituted or substituted by one or more R groups;

R is selected from the group consisting of:

halogen,

hydroxy,

alkyl,

alkenyl,

haloalkyl,

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alkoxy,
                            alkylthio, and
                            -N(R_9)_2;
                   R<sub>1</sub> is selected from the group consisting of:
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                            -R<sub>4</sub>,
                            -X-R<sub>4</sub>,
                            -X-Y-R<sub>4</sub>,
                            -X-Y-X-Y-R<sub>4</sub>, and
                            -X-R<sub>5</sub>;
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                   R<sub>3</sub> is selected from the group consisting of:
                            -Z-R<sub>4</sub>,
                            -Z-X-R<sub>4</sub>,
                            -Z-X-Y-R<sub>4</sub>,
                            -Z-X-Y-X-Y-R<sub>4</sub>, and
                            -Z-X-R<sub>5</sub>;
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                   X is selected from the group consisting of alkylene, alkenylene, alkynylene,
          arylene, heteroarylene, and heterocyclylene wherein the alkylene, alkenylene, and
          alkynylene groups can be optionally interrupted or terminated by arylene, heteroarylene or
          heterocyclylene and optionally interrupted by one or more -O- groups;
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                   Y is selected from the group consisting of:
                            -0-,
                            -S(O)<sub>0-2</sub>-,
                            -S(O)_2-N(R_8)-,
                            -C(R_6)-,
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                            -C(R_6)-O_{-}
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-N(R<sub>8</sub>)-Q-, -C(R<sub>6</sub>)-N(R<sub>8</sub>)-, -O-C(R<sub>6</sub>)-N(R<sub>8</sub>)-, -C(R<sub>6</sub>)-N(OR<sub>9</sub>)-, -O-N(R<sub>8</sub>)-Q-,

-O-C(R<sub>6</sub>)-, -O-C(O)-O-,

-O-N=C(R<sub>4</sub>)-,  
-C(=N-O-R<sub>8</sub>)-,  
-CH(-N(-O-R<sub>8</sub>)-Q-R<sub>4</sub>)-,  
-N-Q-  

$$R_{10}$$
,  
-N-C(R<sub>6</sub>)-N-W-  
 $R_7$ ,  
-V-N
 $R_7$ ,
, and

Z is a bond or -O-;

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R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl, heteroaryloxyalkylenyl, alkylheteroarylenyl, and heterocyclyl wherein the alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl, heteroaryloxyalkylenyl, alkylheteroarylenyl, and heterocyclyl groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, hydroxyalkyl, haloalkyl, haloalkoxy, halogen, nitro, hydroxy, mercapto, cyano, aryl, aryloxy, arylalkyleneoxy, heteroaryl, heteroaryloxy, heteroarylalkyleneoxy, heterocyclyl, amino, alkylamino, dialkylamino, (dialkylamino)alkyleneoxy, and in the case of alkyl, alkenyl, alkynyl, and heterocyclyl, oxo;

R<sub>5</sub> is selected from the group consisting of

 $R_6$  is selected from the group consisting of =0 and =S;

 $R_7$  is  $C_{2-7}$  alkylene;

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R<sub>8</sub> is selected from the group consisting of hydrogen, alkyl, alkoxyalkylenyl, hydroxyalkylenyl, arylalkylenyl, and heteroarylalkylenyl;

R<sub>9</sub> is selected from the group consisting of hydrogen and alkyl;

R<sub>10</sub> is C<sub>3-8</sub> alkylene;

A is selected from the group consisting of -O-, -C(O)-, -CH<sub>2</sub>-, -S(O)<sub>0-2</sub>-, and -N(Q-R<sub>4</sub>)-;

A' is selected from the group consisting of -O-, -S(O)<sub>0-2</sub>-, -N(-Q-R<sub>4</sub>)-, and -CH<sub>2</sub>-;

Q is selected from the group consisting of a bond,  $-C(R_6)$ -,  $-C(R_6)$ -,  $-C(R_6)$ -,  $-C(R_6)$ -N(R<sub>8</sub>)-W-,  $-S(O)_2$ -N(R<sub>8</sub>)-,  $-C(R_6)$ -O-,  $-C(R_6)$ -S-, and  $-C(R_6)$ -N(OR<sub>9</sub>)-;

V is selected from the group consisting of  $-C(R_6)$ -,  $-O-C(R_6)$ -,  $-N(R_8)-C(R_6)$ -, and  $-S(O)_2$ -;

W is selected from the group consisting of a bond, -C(O)-, and  $-S(O)_2$ -; and a and b are independently integers from 1 to 6 with the proviso that a + b is  $\leq 7$ ; or a pharmaceutically acceptable salt thereof to the animal.

As used herein "substantially less than the amount of TNF- $\alpha$ " means that there is at least a two-fold reduction in the maximal TNF- $\alpha$  response as determined using the test methods described herein.

As used herein, the terms "alkyl", "alkenyl", "alkynyl" and the prefix "alk-" are inclusive of both straight chain and branched chain groups and of cyclic groups, e.g., cycloalkyl and cycloalkenyl. Unless otherwise specified, these groups contain from 1 to 20 carbon atoms, with alkenyl groups containing from 2 to 20 carbon atoms, and alkynyl groups containing from 2 to 20 carbon atoms. In some embodiments, these groups have a total of up to 10 carbon atoms, up to 8 carbon atoms, up to 6 carbon atoms, or up to 4

carbon atoms. Cyclic groups can be monocyclic or polycyclic and preferably have from 3 to 10 ring carbon atoms. Exemplary cyclic groups include cyclopropyl, cyclopropylmethyl, cyclobutyl, cyclobutylmethyl, cyclopentyl, cyclopentylmethyl, cyclohexyl, cyclohexylmethyl, adamantyl, and substituted and unsubstituted bornyl, norbornyl, and norbornenyl.

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Unless otherwise specified, "alkylene", "alkenylene", and "alkynylene" are the divalent forms of the "alkyl", "alkenyl", and "alkynyl" groups defined above. The terms, "alkylenyl", "alkenylenyl", and "alkynylenyl" are use when "alkylene", "alkenylene", and "alkynylene", respectively, are substituted. For example, an arylalkylenyl group comprises an alkylene moiety to which an aryl group is attached.

The term "haloalkyl" is inclusive of groups that are substituted by one or more halogen atoms, including perfluorinated groups. This is also true of other groups that include the prefix "halo-." Examples of suitable haloalkyl groups are chloromethyl, trifluoromethyl, 2,2,2-trifluoroethyl, and the like.

The term "aryl" as used herein includes carbocyclic aromatic rings or ring systems. Examples of aryl groups include phenyl, naphthyl, biphenyl, fluorenyl and indenyl.

Unless otherwise indicated, the term "heteroatom" refers to the atoms O, S, or N.

The term "heteroaryl" includes aromatic rings or ring systems that contain at least one ring heteroatom (e.g., O, S, N). In some embodiments, the term "heteroaryl" includes a ring or ring system that contains 2-12 carbon atoms, 1-3 rings, 1-4 heteroatoms, and O, S, and N as the heteroatoms. Exemplary heteroaryl groups include furyl, thienyl, pyridyl, quinolinyl, isoquinolinyl, indolyl, isoindolyl, triazolyl, pyrrolyl, tetrazolyl, imidazolyl, pyrazolyl, oxazolyl, thiazolyl, benzofuranyl, benzothiophenyl, carbazolyl, benzoxazolyl, pyrimidinyl, benzimidazolyl, quinoxalinyl, benzothiazolyl, naphthyridinyl, isoxazolyl, isothiazolyl, purinyl, quinazolinyl, pyrazinyl, 1-oxidopyridyl, pyridazinyl, triazinyl, tetrazinyl, oxadiazolyl, thiadiazolyl, and so on.

The term "heterocyclyl" includes non-aromatic rings or ring systems that contain at least one ring heteroatom (e.g., O, S, N) and includes all of the fully saturated and partially unsaturated derivatives of the above mentioned heteroaryl groups. In some embodiments, the term "heterocyclyl" includes a ring or ring system that contains 2-12 carbon atoms, 1-3 rings, 1-4 heteroatoms, and O, S, and N as the heteroatoms. Exemplary heterocyclyl groups include pyrrolidinyl, tetrahydrofuranyl, morpholinyl, thiomorpholinyl, 1,1-

dioxothiomorpholinyl, piperidinyl, piperazinyl, thiazolidinyl, imidazolidinyl, isothiazolidinyl, tetrahydropyranyl, quinuclidinyl, homopiperidinyl (azepanyl), 1,4-oxazepanyl, homopiperazinyl (diazepanyl), 1,3-dioxolanyl, aziridinyl, azetidinyl, dihydroisoquinolin-(1*H*)-yl, octahydroisoquinolin-(1*H*)-yl, dihydroquinolin-(2*H*)-yl, octahydroquinolin-(2*H*)-yl, dihydro-1*H*-imidazolyl, 3-azabicyclo[3.2.2]non-3-yl, and the like.

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The term "heterocyclyl" includes bicylic and tricyclic heterocyclic ring systems. Such ring systems include fused and/or bridged rings and spiro rings. Fused rings can include, in addition to a saturated or partially saturated ring, an aromatic ring, for example, a benzene ring. Spiro rings include two rings joined by one spiro atom and three rings joined by two spiro atoms.

When "heterocyclyl" contains a nitrogen atom, the point of attachment of the heterocyclyl group may be the nitrogen atom.

The terms "arylene", "heteroarylene", and "heterocyclylene" are the divalent forms of the "aryl", "heteroaryl", and "heterocyclyl" groups defined above. The terms, "arylenyl", "heteroarylenyl", and "heterocyclylenyl" are used when "arylene", "heteroarylene", and "heterocyclylene", respectively, are substituted. For example, an alkylarylenyl group comprises an arylene moiety to which an alkyl group is attached.

When a group (or substituent or variable) is present more than once in any Formula described herein, each group (or substituent or variable) is independently selected, whether explicitly stated or not. For example, for the formula  $-N(R_8)-C(O)-N(R_8)$  each  $R_8$  group is independently selected. In another example, when  $R_1$  and  $R_3$  each contain an  $R_4$  group then each  $R_4$  group is independently selected. In a further example, when two Y groups are present and each Y group contains one or more  $R_8$  groups, then each Y group and each  $R_8$  group is independently selected.

The compounds described herein can be administered according to the methods of the present invention in any of the compounds' pharmaceutically acceptable forms, including isomers (e.g., diastereomers and enantiomers), salts, solvates, polymorphs, and the like. In particular, if a compound is optically active, the methods of the invention specifically include the use each of the compound's enantiomers as well as racemic mixtures of the enantiomers. It should be understood that the term "compound" includes

any or all of such forms, whether explicitly stated or not (although at times, "salts" are explicitly stated).

The term "prodrug" means a compound that can be transformed in vivo to yield an immune response modifying compound, including any of the salt, solvated, polymorphic, or isomeric forms described above. The prodrug, itself, may be an immune response modifying compound, including any of the salt, solvated, polymorphic, or isomeric forms described above. The transformation may occur by various mechanisms, such as through a chemical (e.g., solvolysis or hydrolysis, for example, in the blood) or enzymatic biotransformation. A discussion of the use of prodrugs is provided by T. Higuchi and W. Stella, "Pro-drugs as Novel Delivery Systems," Vol. 14 of the A. C. S. Symposium Series, and in Bioreversible Carriers in Drug Design, ed. Edward B. Roche, American Pharmaceutical Association and Pergamon Press, 1987.

For any of the compounds presented herein, each one of the following variables (e.g., X, Y, Z, R<sub>A</sub>, R<sub>B</sub>, R<sub>1</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, Q, G<sub>1</sub>, G<sub>2</sub>, and so on) in any of its embodiments can be combined with any one or more of the other variables in any of their embodiments and associated with any one of the formulas described herein, as would be understood by one of skill in the art. Each of the resulting combinations of variables describes a compound or compounds which can be administered according to any one of the methods of the present invention, and the resulting method is an embodiment of the present invention.

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For certain embodiments of any one of the above methods, n is 1. For certain embodiments of any one of the above methods, n is 2.

For certain embodiments of any one of the above methods, including any one of the above embodiments,  $R_A$  and  $R_B$  form a fused benzene ring which is unsubstituted or substituted by one or more R groups, or substituted by one  $R_3$  group, or substituted by one  $R_3$  group and one R group. For certain of these embodiments, the fused benzene ring is substituted by an  $R_3$  group at the 7-position.

For certain embodiments of any one of the above methods, including any one of the above embodiments except where  $R_A$  and  $R_B$  form the fused benzene ring,  $R_A$  and  $R_B$  form a fused pyridine ring which is unsubstituted or substituted by one or more R groups, or substituted by one  $R_3$  group, or substituted by one  $R_3$  group and one R group.

For certain embodiments of any one of the above methods, including any one of the above embodiments except where  $R_A$  and  $R_B$  form the fused benzene or pyridine ring,  $R_A$  and  $R_B$  form a fused 5 to 7 membered saturated ring, optionally containing one heteroatom selected from the group consisting of N and S, wherein the ring is unsubstituted or substituted by one or more R groups.

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For certain embodiments of any one of the above methods, including any one of the above embodiments except where  $R_A$  and  $R_B$  form the fused benzene, pyridine, or 5 to 7 membered saturated ring,  $R_A$  and  $R_B$  are each independently selected from the group consisting of hydrogen, halogen, alkyl, alkenyl, alkoxy, alkylthio, and  $-N(R_9)_2$ . For certain of these embodiments,  $R_A$  and  $R_B$  are each methyl.

For certain embodiments of any one of the above methods, including any one of the above embodiments,  $R_1$  is selected from the group consisting of  $-R_4$ ,  $-X-R_4$ ,  $-X-Y-R_4$ ,  $-X-Y-X^1-Y^1-R_4$ , and  $-X-R_5$ ; wherein X is alkylene that is optionally interrupted or terminated by heterocyclylene and optionally interrupted by one -O- group; Y is selected from the group consisting of -O-,  $-S(O)_2$ -,  $-S(O)_2$ -N( $R_8$ )-, -C(O)-, -C(O)-O-, -O-C(O)-,

-N-Q - - N-Q - - N-R<sub>7</sub> - N-Q - - N-R<sub>8</sub>)-Q - - N(R<sub>8</sub>)-Q-, -C(O)-N(R<sub>8</sub>)-, and xylene; 
$$Y^1$$
 is selected from the group consisting of alkylene and arylene;  $Y^1$  is selected from the group consisting of -S-, -C(O)-, -C(O)-O-, -C(O)-N(R<sub>8</sub>)-, -S(O)<sub>2</sub>-N(R<sub>8</sub>)-, and -N(R<sub>8</sub>)-C(O)-; R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, aryl, heterocyclyl, heteroaryl, heteroarylalkylenyl, alkynyl, arylalkylenyl, and arylalkenylenyl, wherein the alkyl, aryl, arylalkylenyl, heterocyclyl, heteroaryl, and heteroarylalkylenyl groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, haloalkyl, haloalkoxy, halogen, hydroxy, cyano, aryl, aryloxy, heteroaryl, heterocyclyl, amino, dialkylamino, and in the case of alkyl and heterocyclyl, oxo; R<sub>5</sub> is selected from the group consisting of:

$$-N-C(O) -N-S(O)_2 -N(R_8)-C(O)-N A (CH_2)_b R_7 , and ; R_6 is selected from the$$

group consisting of =O and =S;  $R_7$  is  $C_{2-7}$  alkylene;  $R_8$  is selected from the group consisting of hydrogen, alkyl, alkoxyalkylenyl, hydroxyalkylenyl, arylalkylenyl, and heteroarylalkylenyl;  $R_{10}$  is  $C_{3-8}$  alkylene; A is selected from the group consisting of -O-,

-C(O)-, and -N(R<sub>4</sub>)-; Q is selected from the group consisting of a bond, -C(R<sub>6</sub>)-, -S(O)<sub>2</sub>-, -C(R<sub>6</sub>)-N(R<sub>8</sub>)-W-, -S(O)<sub>2</sub>-N(R<sub>8</sub>)-, -C(O)-O-, and -C(O)-S-; W is selected from the group consisting of a bond and -C(O)-; and a and b are independently integers from 1 to 6 with the proviso that a + b is  $\leq 7$ .

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For certain embodiments of any one of the above methods, including any one of the above embodiments, R<sub>1</sub> is selected from the group consisting of C<sub>1-5</sub> alkyl, C<sub>2-5</sub> alkynyl, arylC<sub>1-4</sub> alkylenyl, cycloalkylC<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkyl-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl, aryl-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkylenyl, dihydroxyC<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkylenyl, dihydroxyC<sub>1-4</sub> alkylenyl, haloC<sub>1-4</sub> alkylenyl, aminoC<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkylenyl, C(O)-O-C<sub>1-4</sub> alkylenyl, C<sub>1-6</sub> alkyl-C(O)-NH-C<sub>1-4</sub> alkylenyl, aryl-C(O)-NH-C<sub>1-4</sub> alkylenyl wherein aryl is unsubstituted or substituted with one or two halogen groups, heteroaryl-C(O)-NH-C<sub>1-4</sub> alkylenyl, di(C<sub>1-4</sub> alkyl)amino-S(O)<sub>2</sub>-NH-C<sub>1-4</sub> alkylenyl, aryl-NH-C(O)-NH-C<sub>1-4</sub> alkylenyl, heteroaryl-NH-C(S)-NH-C<sub>1-4</sub> alkylenyl, di(C<sub>1-4</sub> alkyl)amino-C(O)-NH-C<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkylamino-C(O)-NH-C<sub>1-4</sub> alkylenyl, di(C<sub>1-4</sub> alkyl)amino-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkylamino-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl, heteroarylC<sub>1-4</sub> alkylenyl, heteroarylC<sub>1-4</sub> alkylenyl, heteroarylC<sub>1-4</sub> alkylenyl, amino-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl, heteroarylC<sub>1-4</sub> alkylenyl

consisting of aryl, heteroaryl, and alkyl, and heterocyclylC<sub>1-4</sub> alkylenyl wherein heterocyclyl is unsubstituted or substituted by one, two, or three substituents selected from the group consisting of alkyl, aryl, heteroaryl, and oxo.

wherein heteroaryl is unsubstituted or substituted by a substituent selected from the group

For certain embodiments of any one of the above methods, including any one of the above embodiments, R<sub>1</sub> is selected from the group consisting of methyl, ethyl, propyl, 2-methylpropyl, 2,2-dimethylpropyl, butyl, pent-4-ynyl, 2-phenylethyl, 2-hydroxy-2-methylpropyl, 2-fluoro-2-methylpropyl, 2,3-dihydroxypropyl, 4-hydroxybutyl, 2-amino-2-methylpropyl, 2-aminoethyl, 4-aminobutyl, 2-(methylsulfonyl)ethyl, 2-(propylsulfonyl)ethyl, 4-(methylsulfonyl)butyl, 2,2-dimethyl-3-(methylsulfonyl)propyl, 3-(phenylsulfonyl)propyl, 2-methyl-2-[2-(methylsulfonyl)ethoxy]propyl, 4-acetoxybutyl, 2-[(methylsulfonyl)amino]ethyl, 4-[(methylsulfonyl)amino]butyl, 2-methyl-2-[(methylsulfonyl)amino]propyl, 2-{[(1-methylethyl)sulfonyl]amino}ethyl, 2-(benzenesulfonylamino)ethyl, 2-(dimethylaminosulfonylamino)ethyl, 4-

(aminosulfonyl)butyl, 4-[(methylamino)sulfonyl]butyl, 4-[(dimethylamino)sulfonyl]butyl,

2-[(cyclohexylcarbonyl)amino]-2-methylpropyl, 2-[(cyclopropylcarbonyl)amino]ethyl, , 4-[(cyclopropylcarbonyl)amino]butyl, 2-[(cyclopropylcarbonyl)amino]-2-methylpropyl, 2methyl-2-{[(1-methylethyl)carbonyl]amino}propyl, 2-methyl-2-[(ethylcarbonyl)amino]propyl, 2-methyl-2-[(pyridin-3-ylcarbonyl)amino]propyl, 2-5 methyl-2-[(pyridin-4-ylcarbonyl)amino]propyl, 2-(acetylamino)-2-methylpropyl, 2-(benzoylamino)ethyl, 2-(benzoylamino)-2-methylpropyl, 2-[(4-fluorobenzoyl)amino]-2methylpropyl, 2-[(3,4-difluorobenzoyl)amino]-2-methylpropyl, 2-[(pyridin-3-ylcarbonyl)amino]ethyl, 2-{[(1-methylethyl)carbonyl]amino}ethyl, 4-{[(1methylethyl)carbonyl]amino}butyl, 2-methyl-2-({[(1-10 methylethyl)amino]carbonyl}amino)propyl, 2-({[(1methylethyl)amino]carbonyl}amino)ethyl, 4-(4-pyridin-2-ylpiperazin-1-yl)butyl, tetrahydro-2H-pyran-4-ylmethyl, 4-(1,1-dioxidoisothiazolidin-2-yl)butyl, (2,2-dimethyl-1,3-dioxolan-4-yl)methyl, 3-(3-methylisoxazol-5-yl)propyl, 3-(3-isopropylisoxazol-5yl)propyl, 3-(3-phenylisoxazol-5-yl)propyl, 3-(3-pyridin-3-ylisoxazol-5-yl)propyl, 4-(3.5.5-trimethyl-1,2,4-oxadiazol-4(5H)-yl)butyl, 4-(3-methyl-1-oxa-2,4-15 diazaspiro[4.4]non-2-en-4-yl)butyl, 2-{[(pyridin-3-ylamino)carbonothioyl]amino}ethyl, 2-

For certain embodiments of any one of the above methods, including any one of the above embodiments,  $R_1$  is selected from the group consisting of alkyl, aminoalkyl, dihydroxyalkyl, haloalkyl, and hydroxyalkyl, except where  $R_1$  as defined does not include this definition. For certain of these embodiments,  $R_1$  is selected from the group consisting of methyl, ethyl, n-propyl, n-butyl, 2-methylpropyl, 2-amino-2-methylpropyl, 3-amino-2,2-dimethylpropyl, 2,3-dihydroxypropyl, 2-fluoro-2-methylpropyl, and 2-hydroxy-2-methylpropyl. Alternatively, for certain of these embodiments,  $R_1$  is selected from the group consisting of (1-hydroxycyclobutyl)methyl, (1-hydroxycyclopentyl)methyl, and (1-hydroxycyclohexyl)methyl.

{[(dimethylamino)carbonyl]amino}ethyl, and 2-{[(phenylamino)carbonyl]amino}ethyl.

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For certain embodiments of any one of the above methods, including any one of the above embodiments,  $R_1$  is heterocyclylalkylenyl wherein heterocyclyl is unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, aryl, heteroaryl, hydroxy, and oxo, except where  $R_1$  as defined does not include this definition. For certain of these embodiments as well as any one of the above embodiments wherein  $R_1$  as defined includes heterocyclyl, heterocyclyl is selected

from the group consisting of 1,3-dioxolanyl, tetrahydropyranyl, tetrahydrofuranyl, pyrrolidinyl, piperidinyl, and morpholinyl, each of which is unsubstituted or substituted by one, two, or three substituents selected from the group consisting of alkyl, aryl, heteroaryl, and oxo. For certain of these embodiments wherein R<sub>1</sub> is heterocyclylalkylenyl, heterocyclyl is selected from the group consisting of 1,3-dioxolanyl, tetrahydropyranyl, tetrahydrofuranyl, pyrrolidinyl, piperidinyl, and morpholinyl, and alkylenyl is C<sub>14</sub> alkylenyl. For certain of these embodiments, R<sub>1</sub> is selected from the group consisting of tetrahydro-2*H*-pyran-4-ylmethyl and (2,2-dimethyl-1,3-dioxolan-4-yl)methyl. Alternatively, for certain of these embodiments, R<sub>1</sub> is (4-hydroxytetrahydro-2*H*-pyran-4-yl)methyl.

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For certain embodiments of any one of the above methods, including any one of the above embodiments,  $R_1$  is -X-Y-R<sub>4</sub>, except where  $R_1$  as defined does not include this definition, wherein X is  $C_{1-6}$  alkylene which may be interrupted by one -O- group; Y is selected from the group consisting of -N(R<sub>8</sub>)-C(O)-, -N(R<sub>8</sub>)-S(O)<sub>2</sub>-, -N(R<sub>8</sub>)-C(O)-N(R<sub>8</sub>)-, and -S(O)<sub>2</sub>- wherein  $R_8$  is selected from hydrogen and methyl; and  $R_4$  is selected from the group consisting of  $C_{1-6}$  alkyl, isoquinolinyl, N-methylimidazolyl, pyridinyl, quinolinyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of chloro, cyano, fluoro, hydroxy, and methyl. For certain of these embodiments,  $R_1$  is selected from the group consisting of 2-[(cyclopropylcarbonyl)amino]ethyl, 4-[(cyclopropylcarbonyl)amino]butyl, 2-[(cyclohexylcarbonyl)amino]-2-methylpropyl, 2-{[(1-methylethyl)carbonyl]amino}ethyl, 4-{[(1-methylethyl)carbonyl]amino}propyl, 2-[(methylsulfonyl)amino]ethyl, 4-[(methylsulfonyl)amino]butyl, 2-methyl-2-[(methylsulfonyl)amino]propyl, 2-methyl-2-({[(1-methylethyl)amino]carbonyl}amino)propyl, 2-methyl-2-[2-

25 (methylsulfonyl)ethoxy]propyl, and 2,2-dimethyl-3-(methylsulfonyl)propy.

For certain embodiments of any one of the above methods, including any one of the above embodiments,  $R_1$  is -X-Y-R<sub>4</sub>, except where  $R_1$  as defined does not include this definition, wherein X is  $C_{1-6}$  alkylene which may be interrupted by an -O- group; Y is selected from the group consisting of -N(R<sub>8</sub>)-C(O)-, -N(R<sub>8</sub>)-S(O)<sub>2</sub>-, -N(R<sub>8</sub>)-C(O)-N(R<sub>8</sub>)-,

 $-N(R_8)-S(O)_2-N(R_8)-$ ,  $-S(O)_2-$ , and

S(O)<sub>2</sub>-, R<sub>10</sub> is pentylene, R<sub>8</sub> is hydrogen or methyl; and R<sub>4</sub> is selected from the group consisting of C<sub>1-6</sub> alkyl, hydroxyC<sub>1-6</sub> alkyl, isoquinolinyl, N-methylimidazolyl, pyridinyl, quinolinyl, benzyl, 1-phenylethyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of chloro, cyano, fluoro, hydroxy, and methyl. For certain of these embodiments, X is C<sub>1-6</sub> alkylene, Y is selected from the group consisting of  $-N(R_8)-C(O)-$ ,  $-N(R_8)-S(O)_2-$ , and  $-N(R_8)-C(O)-N(R_8)-$ , and  $R_4$  is selected from the group consisting of C14 alkyl, hydroxyC14 alkyl, pyridinyl, benzyl, 1-phenylethyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of chloro, cyano, fluoro, hydroxy, and methyl. Alternatively, for certain of these embodiments, X is

C<sub>1-6</sub> alkylene, Y is

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 $(R_{10})^{N-Q}$  wherein Q is -C(O)-, -C(O)-NH-, or S(O)<sub>2</sub>-, and R<sub>10</sub> is

pentylene, and R<sub>4</sub> is C<sub>1-4</sub> alkyl. For certain of these embodiments where Y is

, X is methylene. Alternatively, for certain of these embodiments, Y is -NH-S(O)<sub>2</sub>-N(R<sub>8</sub>)-, R<sub>8</sub> is methyl, and R<sub>4</sub> is C<sub>1-4</sub> alkyl. For certain of these embodiments where Y is -NH-S(O)<sub>2</sub>-N( $R_8$ )-, X is  $C_{2-6}$  alkylene.

For certain embodiments of any one of the above methods, including any one of the above embodiments, R<sub>1</sub> is -X-R<sub>5</sub>, except where R<sub>1</sub> as defined does not include this

definition, wherein X is C<sub>1-6</sub> alkylene, and R<sub>5</sub> is

$$-N(R_8)-C(O)-N$$
 $(CH_2)_a$ 
 $A$ 
 $(CH_2)_b$ 
 $A$ 
 $(C$ 

$$-N(R_8)-C(O)-N$$
 $(CH_2)_b$ 
 $A$ 

and b are each 2. For certain of these embodiments, Q-R4 is methyl. Alternatively, for

certain of these embodiments,  $R_5$  is  $R_7$ . For certain of these embodiments,  $R_1$  is selected from the group consisting of 4-(1,1-dioxidoisothiazolidin-2-yl)butyl, 4-[(4-morpholinecarbonyl)amino]butyl, and 2-[(4-morpholinecarbonyl)amino]ethyl.

For certain embodiments of any one of the above methods, including any one of the above embodiments,  $R_1$  is -X- $R_5$ , except where  $R_1$  as defined does not include this

definition, wherein X is  $C_{1.4}$  alkylene, and  $R_5$  is

$$(CH_2)_a$$
 $(CH_2)_b$ 
 $(CH_2)_b$ 

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wherein R<sub>6</sub> is =0, R<sub>7</sub> is propylene, R<sub>10</sub> is pentylene, A is

-O-, and a and b are each 2. For certain of these embodiments, X is ethylene or butylene.

For certain embodiments of any one of the above methods, including any one of the above embodiments having an R<sub>3</sub> group, R<sub>3</sub> is selected from the group consisting of aryl, arylalkyleneoxy, heteroaryl, and heteroarylakyleneoxy, wherein aryl, arylalkyleneoxy, heteroaryl, and heteroarylalkylenoxy are unsubstituted or substituted with one or more substituents selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl. For certain of these embodiments, R<sub>3</sub> is phenyl, pyridin-2-yl, pyridin-3-yl, pyridin-4-yl, quinolin-3-yl, or thiazol-4-ylmethoxy, any of which may be unsubstituted or substituted by one or more substituents selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl. For certain of these embodiments, R<sub>3</sub> is selected from the group consisting of pyridin-3-yl, pyridin-4-yl, 6-fluoropyridin-3-yl, 5-(hydroxymethyl)pyridin-3-yl, 2-ethoxyphenyl, quinolin-3-yl, and thiazol-4-ylmethoxy.

For certain embodiments of any one of the above methods, including any one of the above embodiments having an R<sub>3</sub> group, R<sub>3</sub> is thien-3-yl, phenyl, pyridin-2-yl, pyridin-3-yl, pyridin-4-yl, or quinolin-3-yl any of which may be unsubstituted or substituted by one or more substituents selected from the group consisting of alkyl, alkoxy, halogen, cyano, hydroxy, and hydroxyalkyl, except where R<sub>3</sub> as defined does not include this

definition.

For certain embodiments of any one of the above methods, including any one of the above embodiments having an R<sub>3</sub> group, R<sub>3</sub> is -Z-X-Y-R<sub>4</sub>, except where R<sub>3</sub> as defined

does not include this definition, wherein Z is a bond, X is phenylene, Y is selected from the group consisting of -C(O)-, -C(O)- $N(R_8)$ -,  $-N(R_8)$ -C(O)-,  $-N(R_8)$ - $S(O)_2$ -, and  $-N(R_8)$ -C(O)- $N(R_8)$ - wherein  $R_8$  is selected from hydrogen and methyl; and  $R_4$  is selected from the group consisting of  $C_{1-6}$  alkyl, morpholin-4-yl, phenyl, and phenyl substituted by a substituent selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl. For certain of these embodiments,  $R_3$  is 2-(4-morpholinecarbonyl)phenyl.

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For certain embodiments of any one of the above methods, including any one of the above embodiments having an R3 group, except where R3 as defined does not include this definition, R<sub>3</sub> is -X-Y-R<sub>4</sub>, wherein X is phenylene, Y is selected from the group consisting of -C(O)-,  $-C(O)-N(R_8)$ -,  $-N(R_8)-C(O)$ -,  $-N(R_8)-S(O)_2$ -, and  $-N(R_8)-C(O)-N(R_8)$ - wherein  $R_8$  is selected from hydrogen and methyl; and  $R_4$  is selected from the group consisting of C<sub>1-6</sub> alkyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl; with the proviso that when Y is  $-C(O)-N(R_8)$  or  $-N(R_8)-C(O)-N(R_8)$  then  $R_4$  can also be hydrogen; and with the further proviso that when Y is -C(O)- or -N(R<sub>8</sub>)-C(O)- then R<sub>4</sub> can also be morpholin-4-yl, piperidin-1-yl, or pyrrolidin-1-yl. For certain of these embodiments, Y is -C(O)-NH-, and R<sub>4</sub> is hydrogen or C<sub>1-4</sub> alkyl. For certain of these embodiments, R<sub>4</sub> is hydrogen. Alternatively, for certain of these embodiments, Y is -NH-C(O)-, and R<sub>4</sub> is C<sub>1-4</sub> alkyl. Alternatively, for certain of these embodiments, Y is -C(O)-, and R<sub>4</sub> is morpholin-4-yl, piperidin-1-yl, or pyrrolidin-1-yl. For certain of these embodiments, R<sub>3</sub> is 3-(methylsulfonylamino)phenyl, 3-(pyrrolidin-1-ylcarbonyl)phenyl, or 3-(morpholin-4-ylcarbonyl)phenyl.

For certain embodiments of any one of the above methods, including any one of the above embodiments which includes an R group, R is not present.

For certain embodiments of any one of the above methods, including any one of the above embodiments which includes an R group and an R<sub>3</sub> group, neither R<sub>3</sub> nor R is present.

For certain embodiments of any one of the above methods, including any one of the above embodiments which includes an R group, R is selected from the group consisting of hydroxy and methoxy.

For certain embodiments of any one of the above methods, including any one of the above embodiments which includes an R<sub>3</sub> group, R<sub>3</sub> is not present.

For certain embodiments of any one of the above methods, including any one of the above embodiments, an effective amount of the compound or salt is administered as a pharmaceutical composition comprising a therapeutically effective amount of the compound or salt and a pharmaceutically acceptable carrier.

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For certain embodiments, the present invention provides a method of treating a viral disease in an animal in need thereof comprising preferentially inducing the biosynthesis of IFN- $\alpha$  in the animal according to any one of the above methods.

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For certain embodiments, the present invention provides a method of treating a neoplastic disease in an animal in need thereof comprising preferentially inducing the biosynthesis of IFN- $\alpha$  in the animal according to any one of the above methods.

For certain embodiments of any one of the above methods, including any one of the above embodiments, the compound or salt is administered systemically.

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For certain embodiments, R<sub>A</sub> and R<sub>B</sub> form a fused aryl ring or heteroaryl ring containing one heteroatom selected from the group consisting of N and S wherein the aryl or heteroaryl ring is unsubstituted or substituted by one or more R groups, or substituted by one R<sub>3</sub> group, or substituted by one R<sub>3</sub> group and one R group.

For certain embodiments,  $R_A$  and  $R_B$  form a fused aryl ring which is unsubstituted or substituted by one or more R groups, or substituted by one  $R_3$  group, or substituted by one  $R_3$  group and one R group.

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For certain embodiments  $R_A$  and  $R_B$  form a fused benzene ring which is unsubstituted or substituted by one or more R groups, or substituted by one  $R_3$  group, or substituted by one  $R_3$  group and one R group.

For certain embodiments R<sub>A</sub> and R<sub>B</sub> form a fused benzene ring which is unsubstituted.

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For certain embodiments  $R_A$  and  $R_B$  form a fused benzene ring which is substituted by one or more R groups, or substituted by one  $R_3$  group, or substituted by one  $R_3$  group and one R group.

For certain embodiments  $R_A$  and  $R_B$  form a fused benzene ring which is substituted by one  $R_3$  group. For certain of these embodiments, the  $R_3$  group is at the 7-position.

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For certain embodiments  $R_A$  and  $R_B$  form a fused benzene ring which is substituted by one or more R groups.

For certain embodiments, R<sub>A</sub> and R<sub>B</sub> form a fused heteroaryl ring containing one heteroatom selected from the group consisting of N and S wherein the heteroaryl ring is unsubstituted or substituted by one or more R groups, or substituted by one R<sub>3</sub> group, or substituted by one R<sub>3</sub> group and one R group.

For certain embodiments, R<sub>A</sub> and R<sub>B</sub> form a fused pyridine ring which is unsubstituted or substituted by one or more R groups, or substituted by one R<sub>3</sub> group, or substituted by one R<sub>3</sub> group and one R group.

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For certain embodiments, R<sub>A</sub> and R<sub>B</sub> form a fused pyridine ring which is unsubstituted.

For certain embodiments, R<sub>A</sub> and R<sub>B</sub> form a fused pyridine ring which is substituted by one or more R groups, or substituted by one R<sub>3</sub> group, or substituted by one R<sub>3</sub> group and one R group. For certain of these embodiments, the fused pyridine ring is

, wherein the highlighted bond is the position where the ring is fused. For certain or these embodiments,  $R_A$  and  $R_B$  form a fused pyridine ring which is substituted by one

 $R_3$  group. For certain of these embodiments, the fused pyridine ring is the highlighted bond is the position where the ring is fused. For certain of these embodiments, the  $R_3$  group is at the 7-position.

For certain embodiments,  $R_{\text{A}}$  and  $R_{\text{B}}$  form a fused pyridine ring which is substituted by one or more R groups.

For certain embodiments, R<sub>A</sub> and R<sub>B</sub> form a fused 5 to 7 membered saturated ring, optionally containing one heteroatom selected from the group consisting of N and S, and unsubstituted or substituted at a carbon atom by one or more R groups.

For certain embodiments, R<sub>A</sub> and R<sub>B</sub> form a fused 5 to 7 membered saturated ring, wherein the ring is unsubstituted or substituted by one or more R groups. For certain of these embodiments, the fused ring is a cyclohexene ring wherein the double bond is the position where the ring is fused. For certain of these embodiments, the fused cyclohexene ring is unsubstituted.

For certain embodiments,  $R_A$  and  $R_B$  form a fused 5 to 7 membered saturated ring, containing one nitrogen atom, and unsubstituted or substituted at a carbon atom by one or more R groups. For certain of these embodiments, the fused ring is a

tetrahydropyridine ring. For certain of these embodiments, the fused tetrahydropyridine

ring is , wherein the highlighted bond indicates the position where the ring is fused. For certain of these embodiments, the fused tetrahydropyridine ring is unsubstituted.

For certain embodiments,  $R_A$  and  $R_B$  are each independently selected from the group consisting of hydrogen, halogen, alkyl, alkenyl, alkoxy, alkylthio, and  $-N(R_9)_2$ . For certain of these embodiments,  $R_A$  and  $R_B$  are each independently alkyl. For certain of these embodiments,  $R_A$  and  $R_B$  are each methyl.

For certain embodiments, R is selected from the group consisting of halogen, hydroxy, alkyl, alkenyl, haloalkyl, alkoxy, alkylthio, and -N(R<sub>9</sub>)<sub>2</sub>.

For certain embodiments, R is selected from the group consisting of hydroxy and methoxy.

For certain embodiments, R is not present.

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For certain embodiments,  $R_1$  is selected from the group consisting of  $-R_4$ ,  $-X-R_4$ ,  $-X-Y-R_4$ ,  $-X-Y-X-Y-R_4$ , and  $-X-R_5$ .

For certain embodiments,  $R_1$  is selected from the group consisting of  $-R_4$ ,  $-X-R_4$ ,  $-X-Y-R_4$ ,  $-X-Y-X^1-Y^1-R_4$ , and  $-X-R_5$ ; wherein X is alkylene that is optionally interrupted or terminated by heterocyclylene and optionally interrupted by one -O- group; Y is selected from the group consisting of -O-,  $-S(O)_2$ -,  $-S(O)_2$ -N( $R_8$ )-, -C(O)-, -C(O)-O-,

O-C(O)-, -N(R<sub>8</sub>)-Q-, -C(O)-N(R<sub>8</sub>)-, and R<sub>7</sub> ; X<sup>1</sup> is selected from the group consisting of alkylene and arylene; Y<sup>1</sup> is selected from the group consisting of -S-, -C(O)-, -C(O)-O-, -C(O)-N(R<sub>8</sub>)-, -S(O)<sub>2</sub>-N(R<sub>8</sub>)-, and -N(R<sub>8</sub>)-C(O)-; R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, aryl, heterocyclyl, heteroaryl, heteroarylalkylenyl, arlylalkylenyl, and arylalkenylenyl, wherein the alkyl, aryl, arylalkylenyl, heterocyclyl, heteroaryl, and heteroarylalkylenyl groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, haloalkyl, haloalkoxy, halogen, hydroxy, cyano, aryl, aryloxy, heteroaryl, heterocyclyl, amino, dialkylamino, and in the case of alkyl and heterocyclyl, oxo; R<sub>5</sub> is selected from the group consisting of:

$$-N-C(O)$$
  $-N-S(O)_2$   $-N(R_8)-C(O)-N$   $A$   $(CH_2)_b$  ;  $R_6$  is selected from the

group consisting of =O and =S;  $R_7$  is  $C_{2-7}$  alkylene;  $R_8$  is selected from the group consisting of hydrogen, alkyl, alkoxyalkylenyl, hydroxyalkylenyl, arylalkylenyl, and heteroarylalkylenyl;  $R_{10}$  is  $C_{3-8}$  alkylene; A is selected from the group consisting of -O-, -C(O)-, and -N( $R_4$ )-; Q is selected from the group consisting of a bond, -C( $R_6$ )-, -S(O)<sub>2</sub>-, -C( $R_6$ )-N( $R_8$ )-W-, -S(O)<sub>2</sub>-N( $R_8$ )-, -C(O)-O-, and -C(O)-S-; W is selected from the group consisting of a bond and -C(O)-; and a and b are independently integers from 1 to 6 with the proviso that a + b is  $\leq 7$ .

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For certain embodiments, R<sub>1</sub> is selected from the group consisting of C<sub>1-5</sub> alkyl,

C<sub>2-5</sub> alkynyl, arylC<sub>1-4</sub> alkylenyl, cycloalkylC<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkyl-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl,

aryl-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkyl-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl-O-C<sub>1-4</sub> alkylenyl,

C<sub>1-4</sub> alkyl-S(O)<sub>2</sub>-NH-C<sub>1-4</sub> alkylenyl, hydroxyC<sub>1-4</sub> alkylenyl, dihydroxyC<sub>1-4</sub> alkylenyl,

haloC<sub>1-4</sub> alkylenyl, aminoC<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkyl-C(O)-O-C<sub>1-4</sub> alkylenyl,

C<sub>1-6</sub> alkyl-C(O)-NH-C<sub>1-4</sub> alkylenyl, aryl-C(O)-NH-C<sub>1-4</sub> alkylenyl wherein aryl is

unsubstituted or substituted with one or two halogen groups,

heteroaryl-C(O)-NH-C<sub>1-4</sub> alkylenyl, di(C<sub>1-4</sub> alkyl)amino-S(O)<sub>2</sub>-NH-C<sub>1-4</sub> alkylenyl,

aryl-S(O)<sub>2</sub>-NH-C<sub>1-4</sub> alkylenyl, aryl-NH-C(O)-NH-C<sub>1-4</sub> alkylenyl,

heteroaryl-NH-C(S)-NH-C<sub>1-4</sub> alkylenyl, di(C<sub>1-4</sub> alkyl)amino-C(O)-NH-C<sub>1-4</sub> alkylenyl,

C<sub>1-4</sub> alkylamino-C(O)-NH-C<sub>1-4</sub> alkylenyl, di(C<sub>1-4</sub> alkyl)amino-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl,

C<sub>1-4</sub> alkylamino-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl, amino-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl, heteroarylC<sub>1-4</sub> alkylenyl

C<sub>1-4</sub> alkylamino-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl, amino-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl, heteroarylC<sub>1-4</sub> alkylenyl wherein heteroaryl is unsubstituted or substituted by a substituent selected from the group consisting of aryl, heteroaryl, and alkyl, and heterocyclylC<sub>1-4</sub> alkylenyl wherein heterocyclyl is unsubstituted or substituted by one, two, or three substituents selected from the group consisting of alkyl, aryl, heteroaryl, and oxo.

For certain embodiments, R<sub>1</sub> is selected from the group consisting of methyl, ethyl, propyl, 2-methylpropyl, 2,2-dimethylpropyl, butyl, pent-4-ynyl, 2-phenylethyl, 2-hydroxy-2-methylpropyl, 2-fluoro-2-methylpropyl, 2,3-dihydroxypropyl, 4-hydroxybutyl, 2-amino-2-methylpropyl, 2-aminobutyl, 2-(methylsulfonyl)ethyl, 2-(propylsulfonyl)ethyl, 4-(methylsulfonyl)butyl, 2,2-dimethyl-3-(methylsulfonyl)propyl, 3-(phenylsulfonyl)propyl, 2-methyl-2-[2-(methylsulfonyl)ethoxy]propyl, 4-acetoxybutyl, 2-

[(methylsulfonyl)amino]ethyl, 4-[(methylsulfonyl)amino]butyl, 2-methyl-2-[(methylsulfonyl)amino]propyl, 2-{[(1-methylethyl)sulfonyl]amino}ethyl, 2-(benzenesulfonylamino)ethyl, 2-(dimethylaminosulfonylamino)ethyl, 4-(aminosulfonyl)butyl, 4-[(methylamino)sulfonyl]butyl, 4-[(dimethylamino)sulfonyl]butyl, 2-[(cyclohexylcarbonyl)amino]-2-methylpropyl, 2-[(cyclopropylcarbonyl)amino]ethyl, 4-5 [(cyclopropylcarbonyl)amino]butyl, 2-[(cyclopropylcarbonyl)amino]-2-methylpropyl, 2methyl-2-{[(1-methylethyl)carbonyl]amino}propyl, 2-methyl-2-[(ethylcarbonyl)amino]propyl, 2-methyl-2-[(pyridin-3-ylcarbonyl)amino]propyl, 2methyl-2-[(pyridin-4-ylcarbonyl)amino]propyl, 2-(acetylamino)-2-methylpropyl, 2-(benzoylamino)ethyl, 2-(benzoylamino)-2-methylpropyl, 2-[(4-fluorobenzoyl)amino]-2-10 methylpropyl, 2-[(3,4-difluorobenzoyl)amino]-2-methylpropyl, 2-[(pyridin-3-ylcarbonyl)amino]ethyl, 2-{[(1-methylethyl)carbonyl]amino}ethyl, 4-{[(1methylethyl)carbonyl]amino}butyl. 2-methyl-2-({[(1-methylethyl)amino]carbonyl}amino)propyl, 15 2-({[(1-methylethyl)amino]carbonyl}amino)ethyl, 4-(4-pyridin-2-ylpiperazin-1-yl)butyl, tetrahydro-2H-pyran-4-ylmethyl, 4-(1,1-dioxidoisothiazolidin-2-yl)butyl, (2,2-dimethyl-1,3-dioxolan-4-yl)methyl, 3-(3-methylisoxazol-5-yl)propyl, 3-(3-isopropylisoxazol-5yl)propyl, 3-(3-phenylisoxazol-5-yl)propyl, 3-(3-pyridin-3-ylisoxazol-5-yl)propyl, 4-(3,5,5-trimethyl-1,2,4-oxadiazol-4(5H)-yl)butyl, 4-(3-methyl-1-oxa-2,4-

 ${[(dimethylamino)carbonyl]amino}ethyl, and 2-{[(phenylamino)carbonyl]amino}ethyl.$  For certain embodiments,  $R_1$  is  $-R_4$ .

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For certain embodiments, R<sub>1</sub> is selected from the group consisting of alkyl, aminoalkyl, dihydroxyalkyl, haloalkyl, and hydroxyalkyl.

For certain embodiments, R<sub>1</sub> is selected from the group consisting of methyl, ethyl, n-propyl, n-butyl, 2-methylpropyl, 2-amino-2-methylpropyl, 3-amino-2,2-dimethylpropyl, 2,3-dihydroxypropyl, 2-fluoro-2-methylpropyl, and 2-hydroxy-2-methylpropyl.

diazaspiro[4.4]non-2-en-4-yl)butyl, 2-{[(pyridin-3-ylamino)carbonothioyl]amino}ethyl, 2-

For certain embodiments, R<sub>1</sub> is selected from the group consisting of (1-hydroxycyclobutyl)methyl, (1-hydroxycyclopentyl)methyl, and (1-hydroxycyclohexyl)methyl.

For certain embodiments, R<sub>1</sub> is heterocyclylalkylenyl wherein heterocyclyl is unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, aryl, heteroaryl, hydroxy, and oxo.

For certain embodiments, R<sub>1</sub> is heterocyclylalkylenyl wherein heterocyclyl is selected from the group consisting of 1,3-dioxolanyl, tetrahydropyranyl, tetrahydrofuranyl, pyrrolidinyl, piperidinyl, and morpholinyl, and alkylenyl is C<sub>1-4</sub> alkylenyl.

For certain embodiments, R<sub>1</sub> is selected from the group consisting of tetrahydro-2*H*-pyran-4-ylmethyl and (2,2-dimethyl-1,3-dioxolan-4-yl)methyl.

For certain embodiments, R<sub>1</sub> is (4-hydroxytetrahydro-2*H*-pyran-4-yl)methyl.

For certain embodiments, R<sub>1</sub> is -X-Y-R<sub>4</sub>.

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For certain embodiments,  $R_1$  is -X-Y- $R_4$  wherein X is  $C_{1-6}$  alkylene which may be interrupted by one -O- group; Y is selected from the group consisting of -N( $R_8$ )-C(O)-, -N( $R_8$ )-S(O)<sub>2</sub>-, -N( $R_8$ )-C(O)-N( $R_8$ )-, and -S(O)<sub>2</sub>- wherein  $R_8$  is selected from hydrogen and methyl; and  $R_4$  is selected from the group consisting of  $C_{1-6}$  alkyl, isoquinolinyl, N-methylimidazolyl, pyridinyl, quinolinyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of chloro, cyano, fluoro, hydroxy, and methyl.

For certain embodiments, R<sub>1</sub> is selected from the group consisting of 2[(cyclopropylcarbonyl)amino]ethyl, 4-[(cyclopropylcarbonyl)amino]butyl, 2[(cyclohexylcarbonyl)amino]-2-methylpropyl, 2-{[(1-methylethyl)carbonyl]amino}ethyl,
4-{[(1-methylethyl)carbonyl]amino}butyl, 2-methyl-2-{[(1methylethyl)carbonyl]amino}propyl, 2-[(methylsulfonyl)amino]ethyl, 4[(methylsulfonyl)amino]butyl, 2-methyl-2-[(methylsulfonyl)amino]propyl, 2-methyl-2({[(1-methylethyl)amino]carbonyl}amino)propyl, 2-methyl-2-[2(methylsulfonyl)ethoxy]propyl, and 2,2-dimethyl-3-(methylsulfonyl)propyl.

For certain embodiments,  $R_1$  is -X-Y- $R_4$  wherein X is  $C_{1-6}$  alkylene which may be interrupted by an -O- group; Y is selected from the group consisting of -N( $R_8$ )-C(O)-,

-N(R<sub>8</sub>)-S(O)<sub>2</sub>-, -N(R<sub>8</sub>)-C(O)-N(R<sub>8</sub>)-, -N(R<sub>8</sub>)-S(O)<sub>2</sub>-N(R<sub>8</sub>)-, -S(O)<sub>2</sub>-, and wherein Q is -C(O)-, -C(O)-NH-, or S(O)<sub>2</sub>-, R<sub>10</sub> is pentylene, R<sub>8</sub> is hydrogen or methyl; and R<sub>4</sub> is selected from the group consisting of C<sub>1-6</sub> alkyl, hydroxyC<sub>1-6</sub> alkyl, isoquinolinyl, N-methylimidazolyl, pyridinyl, quinolinyl, benzyl, 1-phenylethyl, phenyl, and phenyl

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substituted by a substituent selected from the group consisting of chloro, cyano, fluoro, hydroxy, and methyl.

For certain embodiments, R<sub>1</sub> is -X-Y-R<sub>4</sub> wherein X is C<sub>1-4</sub> alkylene; Y is

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; and R<sub>4</sub> is C<sub>1-4</sub> alkyl. For certain of these embodiments, R<sub>10</sub> is pentylene,

and Q is selected from the group consisting of -S(O)2-, -C(O)-, and -C(O)-NH-.

For certain embodiments, R<sub>1</sub> is -X-Y-R<sub>4</sub> wherein Y is -NH-S(O)<sub>2</sub>-N(R<sub>8</sub>)-, R<sub>8</sub> is methyl, and  $R_4$  is  $C_{1-4}$  alkyl.

For certain embodiments, R<sub>1</sub> is -X-R<sub>5</sub>.

For certain embodiments, R<sub>1</sub> is -X-R<sub>5</sub> wherein X is C<sub>1-6</sub> alkylene, and R<sub>5</sub> is

$$-N-S(O)_2 \qquad -N(R_8)-C(O)-N \qquad A \qquad (CH_2)_b \qquad A \qquad (CH_2)_b \qquad A$$

For certain embodiments, R<sub>1</sub> is -X-R<sub>5</sub> wherein X is C<sub>1-6</sub> alkylene, and R<sub>5</sub> is

$$-N(R_8)-C(O)-N$$

(CH<sub>2</sub>)<sub>a</sub>

A

wherein R<sub>8</sub> is hydrogen, A is -O-, -CH<sub>2</sub>-, or -N(Q-R<sub>4</sub>)-, and a

and b are each 2.

For certain embodiments, R<sub>1</sub> is -X-R<sub>5</sub> wherein X is C<sub>1-6</sub> alkylene, R<sub>5</sub> is  $-N-S(O)_2$   $(R_7)$ , and  $R_7$  is propylene.

For certain embodiments, R<sub>1</sub> is -X-R<sub>5</sub>, wherein X is C<sub>1-4</sub> alkylene, and R<sub>5</sub> is

$$-N - C(R_6) \qquad \qquad N - C(R_6) - N \qquad A \\ (CH_2)_b \qquad A \\ wherein R_6 is = O, R_7 is propylene, R_{10}$$

is pentylene, A is -O-, and a and b are each 2.

For certain embodiments, R<sub>1</sub> is selected from the group consisting of 4-(1,1dioxidoisothiazolidin-2-yl)butyl, 4-[(4-morpholinecarbonyl)amino]butyl, and 2-[(4morpholinecarbonyl)amino]ethyl.

For certain embodiments, R<sub>3</sub> is selected from the group consisting of -Z-R<sub>4</sub>, -Z-X-R<sub>4</sub>, -Z-X-Y-R<sub>4</sub>, -Z-X-Y-X-Y-R<sub>4</sub>, and -Z-X-R<sub>5</sub>.

For certain embodiments, R<sub>3</sub> is -Z-R<sub>4</sub>.

For certain embodiments, R<sub>3</sub> is selected from the group consisting of aryl, arylalkyleneoxy, heteroaryl, and heteroarylalkyleneoxy, wherein aryl, arylalkyleneoxy, heteroaryl, and heteroarylalkyleneoxy are unsubstituted or substituted with one or more substituents selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl.

For certain embodiments, R<sub>3</sub> is selected from the group consisting of aryl, arylalkyleneoxy, and heteroaryl, wherein aryl, arylalkyleneoxy, and heteroaryl are unsubstituted or substituted with one or more substituents selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl.

For certain embodiments, R<sub>3</sub> is thien-3-yl, phenyl, pyridin-2-yl, pyridin-3-yl, pyridin-4-yl, or quinolin-3-yl any of which may be unsubstituted or substituted by one or more substituents selected from the group consisting of alkyl, alkoxy, halogen, cyano, hydroxy, and hydroxyalkyl.

For certain embodiments, R<sub>3</sub> is phenyl, pyridin-2-yl, pyridin-3-yl, pyridin-4-yl, quinolin-3-yl, or thiazol-4-ylmethoxy, any of which may be unsubstituted or substituted by one or more substituents selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl.

For certain embodiments, R<sub>3</sub> is selected from the group consisting of pyridin-3-yl, pyridin-4-yl, 6-fluoropyridin-3-yl, 5-(hydroxymethyl)pyridin-3-yl, 2-ethoxyphenyl, quinolin-3-yl, or thiazol-4-ylmethoxy.

For certain embodiments, R<sub>3</sub> is -Z-X-Y-R<sub>4</sub>.

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For certain embodiments,  $R_3$  is -Z-X-Y-R<sub>4</sub> wherein X is phenylene, Y is selected from the group consisting of -C(O)-, -C(O)-N(R<sub>8</sub>)-, -N(R<sub>8</sub>)-C(O)-, -N(R<sub>8</sub>)-S(O)<sub>2</sub>-, and -N(R<sub>8</sub>)-C(O)-N(R<sub>8</sub>)- wherein R<sub>8</sub> is selected from hydrogen and methyl; Z is a bond; and R<sub>4</sub> is selected from the group consisting of C<sub>1-6</sub> alkyl, phenyl, morpholin-4-yl, and phenyl substituted by a substituent selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl.

For certain embodiments,  $R_3$  is  $-Z-X-Y-R_4$  wherein X is phenylene, Y is selected from the group consisting of  $-N(R_8)-C(O)$ ,  $-N(R_8)-S(O)_2$ , and  $-N(R_8)-C(O)-N(R_8)$  wherein  $R_8$  is selected from hydrogen and methyl; Z is a bond; and  $R_4$  is selected from the group consisting of  $C_{1-6}$  alkyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl.

For certain embodiments,  $R_3$  is -Z-X-Y-R<sub>4</sub>, wherein Z is a bond, X is phenylene, Y is selected from the group consisting of -C(O)-, -C(O)-N(R<sub>8</sub>)-, -N(R<sub>8</sub>)-C(O)-, -N(R<sub>8</sub>)-S(O)<sub>2</sub>-, and -N(R<sub>8</sub>)-C(O)-N(R<sub>8</sub>)- wherein R<sub>8</sub> is selected from hydrogen and methyl; and R<sub>4</sub> is selected from the group consisting of  $C_{1-6}$  alkyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl; with the proviso that when Y is -C(O)-N(R<sub>8</sub>)- or -N(R<sub>8</sub>)-C(O)-N(R<sub>8</sub>)- then R<sub>4</sub> can also be hydrogen; and with the further proviso that when Y is -C(O)- or -N(R<sub>8</sub>)-C(O)- then R<sub>4</sub> can also be morpholin-4-yl, piperidin-1-yl, or pyrrolidin-1-yl.

For certain embodiments,  $R_3$  is -Z-X-Y-R<sub>4</sub>, wherein Z is a bond, X is phenylene, Y is -C(O)-NH-, and R<sub>4</sub> is hydrogen or  $C_{1-4}$  alkyl.

For certain embodiments,  $R_3$  is -Z-X-Y- $R_4$ , wherein Z is a bond, X is phenylene, Y is -NH-C(O)-, and  $R_4$  is  $C_{1.4}$  alkyl.

For certain embodiments, R<sub>3</sub> is -Z-X-Y-R<sub>4</sub>, wherein Z is a bond, X is phenylene, Y is -C(O)-, and R<sub>4</sub> is morpholin-4-yl, piperidin-1-yl, or pyrrolidin-1-yl.

For certain embodiments, R<sub>3</sub> is 3-(methylsulfonylamino)phenyl, 3-(pyrrolidin-1-ylcarbonyl)phenyl, or 3-(morpholin-4-ylcarbonyl)phenyl.

For certain embodiments, R<sub>3</sub> is 2-(4-morpholinecarbonyl)phenyl.

For certain embodiments, R<sub>3</sub> is not present.

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For certain embodiments, neither R<sub>3</sub> nor R is present.

For certain embodiments, R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl, heteroaryloxyalkylenyl, alkylheteroarylenyl, and heterocyclyl wherein the alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroarylalkylenyl, heteroaryloxyalkylenyl, alkylheteroarylenyl, and heterocyclyl groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, hydroxyalkyl, haloalkyl, haloalkoxy, halogen, nitro, hydroxy, mercapto, cyano, aryl, aryloxy, arylalkyleneoxy, heteroaryl, heteroaryloxy, heteroarylalkyleneoxy, heterocyclyl, amino, alkylamino, dialkylamino, (dialkylamino)alkyleneoxy, and in the case of alkyl, alkenyl, alkynyl, and heterocyclyl, oxo.

For certain embodiments, R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, alkenyl, aryl, arylalkylenyl, heteroaryl, heteroarylalkylenyl, and heterocyclyl,

wherein the alkyl, alkenyl, aryl, arylalkylenyl, heteroaryl, heteroarylalkylenyl, and heterocyclyl groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, hydroxyalkyl, haloalkyl, haloalkoxy, halogen, nitro, hydroxy, mercapto, cyano, aryl, aryloxy, heteroaryl, heteroaryloxy, heterocyclyl, amino, alkylamino, dialkylamino, and, in the case of alkyl, alkenyl, and heterocyclyl, oxo.

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For certain embodiments, R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, alkenyl, aryl, arylalkylenyl, heteroaryl, and heteroarylalkylenyl, wherein the alkyl, alkenyl, aryl, arylalkylenyl, heteroaryl, and heteroarylalkylenyl, groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, hydroxyalkyl, haloalkyl, haloalkoxy, halogen, nitro, hydroxy, mercapto, cyano, aryl, aryloxy, heteroaryl, heteroaryloxy, heterocyclyl, amino, alkylamino, dialkylamino, and, in the case of alkyl and alkenyl, oxo.

For certain embodiments, R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, aryl, heterocyclyl, heteroaryl, heteroarylalkylenyl, alkynyl, arylalkylenyl, and arylalkenylenyl, wherein the alkyl, aryl, arylalkylenyl, heterocyclyl, heteroaryl, and heteroarylalkylenyl groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, haloalkyl, haloalkoxy, halogen, hydroxy, cyano, aryl, aryloxy, heteroaryl, heterocyclyl, amino, dialkylamino, and in the case of alkyl and heterocyclyl, oxo.

For certain embodiments,  $R_4$  is selected from the group consisting of  $C_{1-6}$  alkyl, hydroxy $C_{1-6}$  alkyl, isoquinolinyl, N-methylimidazolyl, pyridinyl, quinolinyl, benzyl, 1-phenylethyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of chloro, cyano, fluoro, hydroxy, and methyl.

For certain embodiments,  $R_4$  is selected from the group consisting of  $C_{1-6}$  alkyl, isoquinolinyl, N-methylimidazolyl, pyridinyl, quinolinyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of chloro, cyano, fluoro, hydroxy, and methyl.

For certain embodiments,  $R_4$  is selected from the group consisting of  $C_{1-6}$  alkyl, morpholin-4-yl, phenyl, and phenyl substituted by a substituent selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl.

For certain embodiments, R4 is selected from the group consisting of C1-6 alkyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl.

For certain embodiments, R4 is morpholin-4-yl, piperidin-1-yl, or pyrrolidin-1-yl.

For certain embodiments, R<sub>4</sub> is C<sub>1-6</sub> alkyl.

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For certain embodiments, R<sub>4</sub> is hydrogen or C<sub>1-4</sub> alkyl.

For certain embodiments, R<sub>4</sub> is C<sub>1.4</sub> alkyl.

For certain embodiments, R4 is hydrogen.

For certain embodiments, R<sub>5</sub> is selected from the group consisting of:

$$-N-C(R_{6}) -N-S(O)_{2} -N(R_{8})-C(R_{6})-N A (CH_{2})_{b} A$$

$$-N-C(R_{6})-N A (CH_{2})_{b} A$$

$$-N-C(R_{6})-N A (CH_{2})_{b} A$$

$$-N-C(R_{6})-N A (CH_{2})_{b} A$$

For certain embodiments, R<sub>5</sub> is selected from the group consisting of:

For certain embodiments, R<sub>5</sub> is

$$-N(R_8)-C(O)-N$$
 $(CH_2)_a$ 
 $A$ 
 $(CH_2)_b$  wherein  $R_8$ 

For certain embodiments, R<sub>5</sub> is

hydrogen, A is -O-, -CH<sub>2</sub>-, or -N(Q-R<sub>4</sub>)-, and a and b are each 2.

For certain embodiments, 
$$R_5$$
 is
$$\begin{array}{c} -N - S(O)_2 \\ R_7 \end{array}$$
, and  $R_7$  is propylene.

For certain embodiments, R<sub>5</sub> is

wherein  $R_6$  is =0,  $R_7$  is propylene,  $R_{10}$  is pentylene, A is -0-, and a and b are each 2.

For certain embodiments, R<sub>6</sub> is selected from the group consisting of =O and =S.

For certain embodiments,  $R_6$  is =0.

For certain embodiments,  $R_6$  is =S.

For certain embodiments,  $R_7$  is  $C_{2-7}$  alkylene.

For certain embodiments, R<sub>7</sub> is C<sub>2-4</sub> alkylene.

For certain embodiments, R<sub>8</sub> is selected from the group consisting of hydrogen, alkyl, alkoxyalkylenyl, hydroxyalkylenyl, arylalkylenyl, and heteroarylalkylenyl.

For certain embodiments,  $R_8$  is selected from the group consisting of hydrogen,  $C_{1-4}$  alkyl, and  $C_{1-4}$  alkoxy $C_{1-4}$  alkylenyl.

For certain embodiments, R<sub>8</sub> is hydrogen or C<sub>1-4</sub> alkyl.

For certain embodiments, R<sub>8</sub> is selected from hydrogen and methyl

For certain embodiments, R<sub>8</sub> is methyl.

For certain embodiments, R<sub>8</sub> is hydrogen.

For certain embodiments, R<sub>9</sub> is selected from the group consisting of hydrogen and alkyl.

For certain embodiments,  $R_{10}$  is  $C_{3-8}$  alkylene.

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For certain embodiments,  $R_{10}$  is  $C_{4-6}$  alkylene.

For certain embodiments, R<sub>10</sub> is pentylene.

For certain embodiments, A is selected from the group consisting of -O-, -C(O)-, -CH<sub>2</sub>-, -S(O)<sub>0-2</sub>-, and -N(Q-R<sub>4</sub>)-.

20 For certain embodiments, A is -O-, -CH<sub>2</sub>-, -S-, or -S(O)<sub>2</sub>-.

For certain embodiments, A is -O-, -CH<sub>2</sub>-, or -N(Q-R<sub>4</sub>)-.

For certain embodiments, A is -O- or -S(O)<sub>2</sub>-.

For certain embodiments, A is -O-.

For certain embodiments, A is -CH<sub>2</sub>-.

25 For certain embodiments, A is  $-N(Q-R_4)$ -.

For certain embodiments, A is -N(CH<sub>3</sub>)-.

For certain embodiments, A' is selected from the group consisting of -O-, -S(O)<sub>0-2</sub>-, -N(-Q-R<sub>4</sub>)-, and -CH<sub>2</sub>-.

For certain embodiments, including any one of the above embodiments of Formula II,  $G_1$  is selected from the group consisting of -C(O)-R',  $\alpha$ -aminoacyl,  $\alpha$ -aminoacyl, -C(O)-O-R', -C(O)-N(R")R', -C(=NY')-R', -CH(OH)-C(O)-OY',

-CH(OC<sub>1-4</sub> alkyl)Y<sub>0</sub>, -CH<sub>2</sub>Y<sub>1</sub>, and -CH(CH<sub>3</sub>)Y<sub>1</sub>; R' and R" are independently selected from the group consisting of C<sub>1-10</sub> alkyl, C<sub>3-7</sub> cycloalkyl, phenyl, benzyl, and 2-phenylethyl, each of which may be unsubstituted or substituted by one or more substituents independently selected from the group consisting of halogen, hydroxy, nitro, cyano, carboxy, C<sub>1-6</sub> alkyl, C<sub>1-4</sub> alkoxy, aryl, heteroaryl, aryl-C<sub>1-4</sub> alkylenyl, heteroaryl-C<sub>1-4</sub> alkylenyl, halo-C<sub>1-4</sub> alkoxy, -O-C(O)-CH<sub>3</sub>, -C(O)-O-CH<sub>3</sub>, -C(O)-NH<sub>2</sub>, -O-CH<sub>2</sub>-C(O)-NH<sub>2</sub>, and -S(O)<sub>2</sub>-NH<sub>2</sub>, with the proviso that R" can also be hydrogen; α-aminoacyl is an α-aminoacyl group derived from an α-amino acid selected from the group consisting of racemic, D-, and L-amino acids; Y' is selected from the group consisting of hydrogen, C<sub>1-6</sub> alkyl, and benzyl; Y<sub>0</sub> is selected from the group consisting of C<sub>1-6</sub> alkyl, carboxy-C<sub>1-6</sub> alkylenyl, amino-C<sub>1-4</sub> alkylenyl, mono-N-C<sub>1-6</sub> alkylamino-C<sub>1-4</sub> alkylenyl, and di-N,N-C<sub>1-6</sub> alkylamino-C<sub>1-4</sub> alkylenyl; and Y<sub>1</sub> is selected from the group consisting of mono-N-C<sub>1-6</sub> alkylamino, di-N,N-C<sub>1-6</sub> alkylamino, morpholin-4-yl, piperidin-1-yl, pyrrolidin-1-yl, and 4-C<sub>1-4</sub> alkylpiperazin-1-yl.

For certain embodiments, including any one of the above embodiments of Formula II,  $G_1$  is selected from the group consisting of -C(O)-R',  $\alpha$ -aminoacyl, and -C(O)-O-R'. For certain of these embodiments, R' contains one to ten carbon atoms. For certain of these embodiments,  $\alpha$ -aminoacyl is an  $\alpha$ - $C_{2-11}$  aminoacyl group derived from an  $\alpha$ -amino acid selected from the group consisting of racemic, D-, and L-amino acids containing a total of at least 2 carbon atoms and a total of up to 11 carbon atoms, and may also include one or more heteroatoms selected from the group consisting of O, O, and O.

For certain embodiments, including any one of the above embodiments of Formula III,  $G_2$  is selected from the group consisting of  $-X_2$ -C(O)-R',  $\alpha$ -aminoacyl,  $\alpha$ -amin

-O-CH<sub>2</sub>-C(O)-NH<sub>2</sub>, -NH<sub>2</sub>, and -S(O)<sub>2</sub>-NH<sub>2</sub>, with the proviso that R" can also be hydrogen; and  $\alpha$ -aminoacyl is an  $\alpha$ -aminoacyl group derived from an  $\alpha$ -amino acid selected from the group consisting of racemic, D-, and L-amino acids.

For certain embodiments, including any one of the above embodiments of Formula III, G<sub>2</sub> is selected from the group consisting of -C(O)-R' and α-aminoacyl, wherein R' is C<sub>1-6</sub> alkyl or phenyl which is unsubstituted or substituted by one or more substituents independently selected from the group consisting of halogen, hydroxy, nitro, cyano, carboxy, C<sub>1-6</sub> alkyl, C<sub>1-4</sub> alkoxy, aryl, heteroaryl, aryl-C<sub>1-4</sub> alkylenyl, heteroaryl-C<sub>1-4</sub> alkylenyl, halo-C<sub>1-4</sub> alkylenyl, halo-C<sub></sub>

 $-C(O)-O-CH_3$ ,  $-C(O)-NH_2$ ,  $-O-CH_2-C(O)-NH_2$ ,  $-NH_2$ , and  $-S(O)_2-NH_2$ .

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For certain embodiments, including any one of the above embodiments of Formula III,  $G_2$  is selected from the group consisting of  $\alpha$ -amino- $C_{2-5}$  alkanoyl,  $C_{2-6}$  alkanoyl,  $C_{1-6}$  alkoxycarbonyl, and  $C_{1-6}$  alkylcarbamoyl.

For certain embodiments, including any one of the above embodiments which include an  $\alpha$ -aminoacyl group,  $\alpha$ -aminoacyl is an  $\alpha$ -aminoacyl group derived from a naturally occurring  $\alpha$ -amino acid selected from the group consisting of racemic, D-, and L-amino acids.

For certain embodiments, including any one of the above embodiments which include an α-aminoacyl group, α-aminoacyl is an α-aminoacyl group derived from an α-amino acid found in proteins, wherein the the amino acid is selected from the group consisting of racemic, D-, and L-amino acids.

For certain embodiments, the hydrogen atom of the hydroxy group of Formula II (including any one of its embodiments) is replaced by  $G_2$ , wherein  $G_2$  is defined as in any one of the above embodiments of  $G_2$ .

For certain embodiments, Q is selected from the group consisting of a bond,  $-C(R_6)$ -,  $-C(R_6)$ - $C(R_6)$ -,  $-S(O)_2$ -,  $-C(R_6)$ - $N(R_8)$ -W-,  $-S(O)_2$ - $N(R_8)$ -,  $-C(R_6)$ -O-,  $-C(R_6)$ -S-, and  $-C(R_6)$ - $N(OR_9)$ -.

For certain embodiments, Q is selected from the group consisting of a bond,  $-C(R_6)$ -,  $-S(O)_2$ ,  $-C(R_6)$ - $N(R_8)$ -,  $-S(O)_2$ - $N(R_8)$ -,  $-C(R_6)$ -O-, and  $-C(R_6)$ -S-.

For certain embodiments, Q is selected from the group consisting of a bond,  $-C(R_6)$ -,  $-S(O)_2$ -, and  $-C(R_6)$ - $N(R_8)$ -.

For certain embodiments, Q is selected from the group consisting of -C(O)-,

-S(O)<sub>2</sub>-, and -C(O)-N(R<sub>8</sub>)-. In certain of these embodiments, R<sub>8</sub> is hydrogen or methyl. For certain embodiments, Q is selected from the group consisting of -S(O)<sub>2</sub>-, -C(O)-, and -C(O)-NH-.

For certain embodiments, Q is -C(O)-.

For certain embodiments, Q is -S(O)<sub>2</sub>-.

For certain embodiments, Q is  $-C(R_6)-N(R_8)-$ .

For certain embodiments, Q is -C(O)-N(R<sub>8</sub>)- wherein R<sub>8</sub> is hydrogen or methyl.

For certain embodiments, V is selected from the group consisting of -C(R<sub>6</sub>)-,

 $-O-C(R_6)$ -,  $-N(R_8)-C(R_6)$ -, and  $-S(O)_2$ -.

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For certain embodiments, V is  $-C(R_6)$ -.

For certain embodiments, W is selected from the group consisting of a bond, -C(O)-, and  $-S(O)_2$ -.

For certain embodiments, W is a bond.

For certain embodiments, X is selected from the group consisting of alkylene, alkenylene, alkynylene, arylene, heteroarylene, and heterocyclylene wherein the alkylene, alkenylene, and alkynylene groups can be optionally interrupted or terminated by arylene, heteroarylene or heterocyclylene and optionally interrupted by one or more -O- groups.

For certain embodiments, X is alkylene that is optionally interrupted or terminated by heterocyclylene and optionally interrupted by one -O- group.

For certain embodiments, X is C<sub>1-6</sub> alkylene which may be interrupted by one -O-group.

For certain embodiments, X is  $C_{1-6}$  alkylene.

For certain embodiments, X is  $C_{2-6}$  alkylene.

For certain embodiments, X is C<sub>1-4</sub> alkylene.

For certain embodiments, X is phenylene.

For certain embodiments, X is methylene.

For certain embodiments, X is ethylene.

For certain embodiments, X is butylene.

For certain embodiments, Y is selected from the group consisting of -O-, -S(O)<sub>0-2</sub>-,

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$$-S(O)_2-N(R_8)-$$
,  $-C(R_6)-$ ,  $-C(R_6)-O-$ ,  $-O-C(R_6)-$ ,  $-O-C(O)-O-$ ,  $-N(R_8)-Q-$ ,  $-C(R_6)-N(R_8)-$ ,  $-C(R_6)-N(OR_9)-$ ,  $-O-N(R_8)-Q-$ ,  $-O-N=C(R_4)-$ ,  $-C(=N-O-R_8)-$ ,

-CH(-N(-O-R<sub>8</sub>)-Q-R<sub>4</sub>)-, 
$$R_{10}$$
 , and  $R_{10}$  ,  $R$ 

For certain embodiments, Y is selected from the group consisting of -O-, -S(O)<sub>2</sub>-, -S(O)<sub>2</sub>-N(R<sub>8</sub>)-, -C(O)-, -C(O)-O-, -O-C(O)-, -N(R<sub>8</sub>)-Q-, -C(O)-N(R<sub>8</sub>)-,

$$R_{10}$$
 , and  $R_7$   $R_7$   $R_7$ 

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For certain embodiments, Y is selected from the group consisting of -O-,-C(R<sub>6</sub>)-,

-S(O)<sub>0-2</sub>-, -N(R<sub>8</sub>)-Q-, 
$$R_{10}$$
 ,  $R_{7}$  , and  $R_{7}$ 

For certain embodiments, Y is selected from the group consisting of -N(R<sub>8</sub>)-C(O)-,

For certain embodiments, Y is selected from the group consisting of -N( $R_8$ )-C(O)-, -N( $R_8$ )-S(O)<sub>2</sub>-, -N( $R_8$ )-C(O)-N( $R_8$ )-, and -S(O)<sub>2</sub>-. In certain of these embodiments,  $R_8$  is selected from hydrogen and methyl.

For certain embodiments, Y is selected from the group consisting of -N(R<sub>8</sub>)-C(O)-, -N(R<sub>8</sub>)-S(O)<sub>2</sub>-, and -N(R<sub>8</sub>)-C(O)-N(R<sub>8</sub>)- wherein R<sub>8</sub> is selected from hydrogen and methyl.

For certain embodiments, Y is selected from the group consisting of -C(O)-, -C(O)-N(R<sub>8</sub>)-,-N(R<sub>8</sub>)-C(O)-, -N(R<sub>8</sub>)-S(O)<sub>2</sub>-, and -N(R<sub>8</sub>)-C(O)-N(R<sub>8</sub>)-. In certain of these embodiments, R<sub>8</sub> is selected from hydrogen and methyl.

For certain embodiments, Y is  $R_{10}$ . For certain of these embodiments,  $R_{10}$  is pentylene, and Q is selected from the group consisting of -S(O)<sub>2</sub>-, -C(O)-, and -C(O)-NH-.

For certain embodiments, Y is -NH-S(O)<sub>2</sub>-N( $R_8$ )-. In certain of these embodiments,  $R_8$  is methyl.

For certain embodiments,  $Y^1$  is selected from the group consisting of -S-, -C(O)-, -C(O)-O-, -C(O)-N(R<sub>8</sub>)-, -S(O)<sub>2</sub>-N(R<sub>8</sub>)-, and -N(R<sub>8</sub>)-C(O)-.

For certain embodiments, Z is a bond or -O-.

For certain embodiments, Z is a bond.

10 Fro certain embodiments, Z is -O-.

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For certain embodiments, a and b are independently integers from 1 to 6 with the proviso that a + b is  $\leq 7$ .

For certain embodiments, a and b are each independently 1 to 3.

For certain embodiments, a and b are each 2.

For certain embodiments, a is 1, 2, or 3, and b is 2.

For certain embodiments, n is 1 or 2.

For certain embodiments, n is 1.

For certain embodiments, n is 2.

For certain embodiments, the compound, 1-[4-amino-2-hydroxymethyl-7-(thiazol-4-ylmethoxy)-1*H*-imidazo[4,5-*c*]quinolin-1-yl]-2-methylpropan-2-ol, pharmaceutically acceptable salts thereof, pharmaceutical compositions containing this compound or salt thereof in combination with a pharmaceutically acceptable carrier, and the use of this compound in the methods described herein are provided.

# 25 Preparation of the Compounds

Compounds of the invention may be synthesized by synthetic routes that include processes analogous to those well known in the chemical arts, particularly in light of the description contained herein. The starting materials are generally available from commercial sources such as Aldrich Chemicals (Milwaukee, Wisconsin, USA) or are readily prepared using methods well known to those skilled in the art (e.g., prepared by

methods generally described in Louis F. Fieser and Mary Fieser, Reagents for Organic Synthesis, v. 1-19, Wiley, New York, (1967-1999 ed.); Alan R. Katritsky, Otto Meth-Cohn, Charles W. Rees, Comprehensive Organic Functional Group Transformations, v. 1-6, Pergamon Press, Oxford, England, (1995); Barry M. Trost and Ian Fleming, Comprehensive Organic Synthesis, v. 1-8, Pergamon Press, Oxford, England, (1991); or Beilsteins Handbuch der organischen Chemie, 4, Aufl. Ed. Springer-Verlag, Berlin, Germany, including supplements (also available via the Beilstein online database)).

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For illustrative purposes, the reaction schemes depicted below provide potential routes for synthesizing the compounds of the present invention as well as key intermediates. For more detailed description of the individual reaction steps, see the EXAMPLES section below. Those skilled in the art will appreciate that other synthetic routes may be used to synthesize the compounds of the invention. Although specific starting materials and reagents are depicted in the reaction schemes and discussed below, other starting materials and reagents can be easily substituted to provide a variety of derivatives and/or reaction conditions. In addition, many of the compounds prepared by the methods described below can be further modified in light of this disclosure using conventional methods well known to those skilled in the art.

In the preparation of compounds of the invention it may sometimes be necessary to protect a particular functionality while reacting other functional groups on an intermediate. The need for such protection will vary depending on the nature of the particular functional group and the conditions of the reaction step. Suitable amino protecting groups include acetyl, trifluoroacetyl, tert-butoxycarbonyl (Boc), benzyloxycarbonyl, and 9-fluorenylmethoxycarbonyl (Fmoc). Suitable hydroxy protecting groups include acetyl and silyl groups such as the tert-butyl dimethylsilyl group. For a general description of protecting groups and their use, see T. W. Greene and P. G. M. Wuts, Protective Groups in Organic Synthesis, John Wiley & Sons, New York, USA, 1991.

Conventional methods and techniques of separation and purification can be used to isolate compounds of the invention, as well as various intermediates related thereto. Such techniques may include, for example, all types of chromatography (high performance liquid chromatography (HPLC), column chromatography using common absorbents such as silica gel, and thin layer chromatography), recrystallization, and differential (i.e., liquid-liquid) extraction techniques.

In some embodiments, compounds of the invention can be prepared according to Reaction Scheme I, wherein  $R_1$ , R, m, and n are as defined above and alkyl is methyl or ethyl.

In Reaction Scheme I an ether substituted 1*H*-imidazo[4,5-*c*]quinolin-4-amine of Formula X is cleaved to provide a hydroxyalkyl substituted 1*H*-imidazo[4,5-*c*]quinolin-4-amine of Formula I. The reaction is conveniently carried out by adding a solution of boron tribromide in a suitable solvent such as dichloromethane to a solution or suspension of a compound of Formula X in a suitable solvent such as dichloromethane at ambient or at a sub-ambient temperature, for example, at 0°C. The product or pharmaceutically acceptable salt thereof can be isolated using conventional methods.

Numerous compounds of Formula X are known; others can be prepared using known synthetic methods. See, for example, United States Patent Nos. 6,069,149; 6,331,539; 6,451,810; 6,541,485; 6,756,382; 6,677,349; 6,573,273; 6,664,264; 6,664,265; 6,677,347; 6,660,735; 6,683,088; and 6,667,312 and the references cited therein.

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#### Reaction Scheme I

$$(R)_m$$
 $(CH_2)_n$ -O-alkyl
 $(R)_m$ 
 $(CH_2)_n$ -O- $(CH_2)_$ 

In some embodiments, compounds of the invention can be prepared according to Reaction Scheme II, wherein R<sub>1</sub>, G<sub>1</sub>, and n are as defined above. Compounds of Formula I can be prepared according to the method described above. The amino group of a compound of Formula I can be converted by conventional methods to a functional group such as an amide, carbamate, urea, amidine, or another hydrolyzable group. A compound of this type can be made by the replacement of a hydrogen atom in an amino group with a group such as -C(O)-R', α-aminoacyl, α-aminoacyl-α-aminoacyl, -C(O)-O-R', -C(O)-N(R")R', -C(=NY')-R', -CH(OH)-C(O)-OY', -CH(OC<sub>1-4</sub> alkyl)Y<sub>0</sub>, -CH<sub>2</sub>Y<sub>1</sub>, and -CH(CH<sub>3</sub>)Y<sub>1</sub>; wherein R' and R" are independently selected from the group consisting of C<sub>1-10</sub> alkyl, C<sub>3-7</sub> cycloalkyl, phenyl, benzyl, and 2-phenylethyl, each of which may be unsubstituted or substituted by one or more substituents independently selected from the

group consisting of halogen, hydroxy, nitro, cyano, carboxy, C<sub>1-6</sub> alkyl, C<sub>1-4</sub> alkoxy, aryl, heteroaryl, aryl-C<sub>1-4</sub> alkylenyl, heteroaryl-C<sub>1-4</sub> alkylenyl, halo-C<sub>1-4</sub> alkylenyl, halo-C<sub>1-4</sub> alkoxy, -O-C(O)-CH<sub>3</sub>, -C(O)-O-CH<sub>3</sub>, -C(O)-NH<sub>2</sub>, -O-CH<sub>2</sub>-C(O)-NH<sub>2</sub>, -NH<sub>2</sub>, and  $-S(O)_2$ -NH<sub>2</sub>, with the proviso that R" can also be hydrogen; each  $\alpha$ -aminoacyl is an  $\alpha$ aminoacyl group derived from an a-amino acid selected from the group consisting of racemic, D-, and L-amino acids; Y' is selected from the group consisting of hydrogen,  $C_{1-6}$  alkyl, and benzyl;  $Y_0$  is selected from the group consisting of  $C_{1-6}$  alkyl, carboxy-C<sub>1-6</sub> alkylenyl, amino-C<sub>1-4</sub> alkylenyl, mono-N-C<sub>1-6</sub> alkylamino-C<sub>1-4</sub> alkylenyl, and di-N,N-C<sub>1-6</sub> alkylamino-C<sub>1-4</sub> alkylenyl; and Y<sub>1</sub> is selected from the group consisting of mono-N-C<sub>1-6</sub> alkylamino, di-N, N-C<sub>1-6</sub> alkylamino, morpholin-4-yl, piperidin-1-yl, pyrrolidin-1-yl, and 4-C<sub>1-4</sub> alkylpiperazin-1-yl. Particularly useful compounds of Formula II are amides derived from carboxylic acids containing one to ten carbon atoms, amides derived from amino acids, and carbamates containing one to ten carbon atoms. The reaction can be carried out, for example, by combining a compound of Formula I with a chloroformate or acid chloride, such as ethyl chloroformate or acetyl chloride, in the presence of a base such as triethylamine in a suitable solvent such as dichloromethane at ambient temperature.

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Alternatively, the hydroxy group on a compound of Formula I can be protected using a suitable silyl group such as *tert*-butyl dimethylsilyl using conventional methods. The G<sub>1</sub> group may then be installed using conventional methods followed by the removal of the hydroxy protecting group under acidic conditions to provide a compound of Formula II.

### Reaction Scheme II

$$(R)_m$$
 $N$ 
 $(CH_2)_nOH$ 
 $(R)_m$ 
 $(R)$ 

In some embodiments, compounds of the invention can be prepared according to Reaction Scheme III, wherein  $R_1$ ,  $G_2$ , and n are as defined above. Compounds of Formula I can be prepared according to the method described above. The hydrogen atom of the

alcohol group of a compound of Formula I can be replaced using conventional methods with a group such as X<sub>2</sub>-C(O)-R', α-aminoacyl, α-aminoacyl-α-aminoacyl, -X<sub>2</sub>-C(O)-O-R', and -C(O)-N(R")R'; wherein X2 is selected from the group consisting of a bond; -CH2-O-; -CH(CH<sub>3</sub>)-O-; -C(CH<sub>3</sub>)<sub>2</sub>-O-; and, in the case of -X<sub>2</sub>-C(O)-O-R', -CH<sub>2</sub>-NH-; R' and R" are independently selected from the group consisting of C<sub>1-10</sub> alkyl, C<sub>3-7</sub> cycloalkyl, phenyl, benzyl, and 2-phenylethyl, each of which may be unsubstituted or substituted by one or more substituents independently selected from the group consisting of halogen, hydroxy, nitro, cyano, carboxy, C1-6 alkyl, C1-4 alkoxy, aryl, heteroaryl, aryl-C1-4 alkylenyl,  $heteroaryl-C_{1-4}\,alkylenyl,\,halo-C_{1-4}\,alkylenyl,\,halo-C_{1-4}\,alkoxy,\,-O-C(O)-CH_3,\\$ -C(O)-O-CH<sub>3</sub>, -C(O)-NH<sub>2</sub>, -O-CH<sub>2</sub>-C(O)-NH<sub>2</sub>, -NH<sub>2</sub>, and -S(O)<sub>2</sub>-NH<sub>2</sub>, with the proviso that R" can also be hydrogen; and each α-aminoacyl is an α-aminoacyl group derived from an a-amino acid selected from the group consisting of racemic, D-, and L-amino acids. Particularly useful compounds of Formula III are esters made from carboxylic acids containing one to six carbon atoms, unsubstituted or substituted benzoic acid esters, or esters made from naturally occurring amino acids. For example, the reaction can be carried out by treating a compound of Formula I with a carboxylic acid or amino acid under Mitsunobu reaction conditions by adding triphenylphosphine and a carboxylic acid to a solution or suspension of a compound of Formula I in a suitable solvent such as tetrahydrofuran and then slowly adding diisopropyl azodicarboxylate. The reaction can be run at a sub-ambient temperature such as 0 °C.

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## Reaction Scheme III

$$NH_2$$
 $NH_2$ 
 $NH_2$ 

In some embodiments, compounds of the invention can also be prepared using the synthetic methods described in the EXAMPLES below.

Pharmaceutical Compositions and Biological Activity

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Pharmaceutical compositions of the invention contain a therapeutically effective amount of a compound or salt described above in combination with a pharmaceutically acceptable carrier.

The terms "a therapeutically effective amount" and "effective amount" mean an amount of the compound or salt sufficient to induce a therapeutic or prophylactic effect, such as cytokine induction, immunomodulation, antitumor activity, and/or antiviral activity. Cytokine induction can include preferentially inducing the biosynthesis of IFN-α. The exact amount of compound or salt used in a pharmaceutical composition of the invention will vary according to factors known to those of skill in the art, such as the physical and chemical nature of the compound or salt, the nature of the carrier, and the intended dosing regimen.

In some embodiments, the compositions of the invention will contain sufficient active ingredient or prodrug to provide a dose of about 100 nanograms per kilogram (ng/kg) to about 50 milligrams per kilogram (mg/kg), preferably about 10 micrograms per kilogram (µg/kg) to about 5 mg/kg, of the compound or salt to the subject.

In other embodiments, the compositions of the invention will contain sufficient active ingredient or prodrug to provide a dose of, for example, from about  $0.01 \text{ mg/m}^2$  to about  $5.0 \text{ mg/m}^2$ , computed according to the Dubois method, in which the body surface area of a subject (m²) is computed using the subject's body weight: m² = (wt kg $^{0.425}$  x height cm $^{0.725}$ ) x 0.007184, although in some embodiments the methods may be performed by administering a compound or salt or composition in a dose outside this range. In some of these embodiments, the method includes administering sufficient compound to provide a dose of from about  $0.1 \text{ mg/m}^2$  to about  $2.0 \text{ mg/m}^2$  to the subject, for example, a dose of from about  $0.4 \text{ mg/m}^2$  to about  $1.2 \text{ mg/m}^2$ .

A variety of dosage forms may be used, such as tablets, lozenges, capsules, parenteral formulations (e.g., intravenous formulations), syrups, creams, ointments, aerosol formulations, transdermal patches, transmucosal patches and the like. These dosage forms can be prepared with conventional pharmaceutically acceptable carriers and additives using conventional methods, which generally include the step of bringing the active ingredient into association with the carrier.

The compounds or salts of the invention can be administered as the single therapeutic agent in the treatment regimen, or the compounds or salts described herein may be administered in combination with one another or with other active agents, including additional immune response modifiers, antivirals, antibiotics, antibodies, proteins, peptides, oligonucleotides, etc.

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Compounds or salts of the invention have been shown to induce the production of certain cytokines in experiments performed according to the tests set forth below. These results indicate that the compounds or salts are useful for modulating the immune response in a number of different ways, rendering them useful in the treatment of a variety of disorders. The compounds or salts of the invention are especially useful as immune response modifiers due to their ability to preferentially induce interferon-α, thus providing a benefit over compounds that also induce pro-inflammatory cytokines (e.g. TNF- $\alpha$ ) or that induce pro-inflammatory cytokines at higher levels. While interferon- $\alpha$ and pro-inflammatory cytokines are beneficial in treating certain conditions, interferon-α preferentially induced is believed to be better tolerated by patients, because the significantly lower levels of pro-inflammatory cytokines can result in fewer or less severe adverse side effects experienced by patients. For example, if a subject is treated for a disease (e.g., hepatitis C, metastatic cancer) with a compound that induces significant levels of pro-inflammatory cytokines, while treating the disease, the compound may also cause side effects, such as severe and/or widespread inflammation, tissue destruction, or emesis, that render the subject unable or unwilling to receive the treatment. Alternatively, if a subject is treated with a compound that preferentially induces interferon-α then the compound may treat the disease with less risk of adverse side effects from proinflammatory cytokines such as TNF-a. Therefore, by maintaining the ability to treat a condition and reducing adverse side effects, compounds that preferentially induce IFN-a provide an advantage over compounds that would also induce pro-inflammatory cytokines, such as TNF-α, at higher levels.

The ability of the compounds or salts of the invention to preferentially induce the biosynthesis of IFN- $\alpha$  may be particularly advantageous when administered systemically, since adverse side effects, including for example widespread inflammation, may be reduced or even eliminated. Compounds of the invention may be administered

systemically in a number of ways, including but not limited to oral and intravenous administration.

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Cytokines whose biosynthesis may be induced by compounds or salts of the invention include IFN-α, IP-10, MCP-1, and a variety of other cytokines. In some instances, cytokines such as TNF-α, IL-12 may be induced, albeit at significantly reduced levels. Among other effects, these and other cytokines can inhibit virus production and tumor cell growth, making the compounds or salts useful in the treatment of viral diseases and neoplastic diseases. Accordingly, the invention provides a method of inducing cytokine biosynthesis in an animal comprising administering an effective amount of a compound or salt of the invention to the animal. The animal to which the compound or salt is administered for induction of cytokine biosynthesis may have a disease as described *infra*, for example a viral disease or a neoplastic disease, and administration of the compound or salt may provide therapeutic treatment. Alternatively, the compound or salt may be administered to the animal prior to the animal acquiring the disease so that administration of the compound or salt may provide a prophylactic treatment.

In addition to the ability to induce the production of cytokines, compounds or salts of the invention can affect other aspects of the innate immune response. For example, the compounds or salts may cause maturation of dendritic cells or proliferation and differentiation of B-lymphocytes.

Whether for prophylaxis or therapeutic treatment of a disease, and whether for effecting innate or acquired immunity, the compound or salt or composition may be administered alone or in combination with one or more active components as in, for example, a vaccine adjuvant. When administered with other components, the compound or salt or composition and other component or components may be administered separately; together but independently such as in a solution; or together and associated with one another such as (a) covalently linked or (b) non-covalently associated, e.g., in a colloidal suspension.

Conditions for which compounds or salts or compositions identified herein may be used as treatments include, but are not limited to:

(a) viral diseases such as, for example, diseases resulting from infection by an adenovirus, a herpesvirus (e.g., HSV-I, HSV-II, CMV, or VZV), a poxvirus (e.g., an orthopoxvirus such as variola or vaccinia, or molluscum contagiosum), a picornavirus

(e.g., rhinovirus or enterovirus), an orthomyxovirus (e.g., influenzavirus), a paramyxovirus (e.g., parainfluenzavirus, mumps virus, measles virus, and respiratory syncytial virus (RSV)), a coronavirus (e.g., SARS), a papovavirus (e.g., papillomaviruses, such as those that cause genital warts, common warts, or plantar warts), a hepadnavirus (e.g., hepatitis B virus), a flavivirus (e.g., hepatitis C virus or Dengue virus), or a retrovirus (e.g., a lentivirus such as HIV);

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- (b) bacterial diseases such as, for example, diseases resulting from infection by bacteria of, for example, the genus Escherichia, Enterobacter, Salmonella, Staphylococcus, Shigella, Listeria, Aerobacter, Helicobacter, Klebsiella, Proteus, Pseudomonas, Streptococcus, Chlamydia, Mycoplasma, Pneumococcus, Neisseria, Clostridium, Bacillus, Corynebacterium, Mycobacterium, Campylobacter, Vibrio, Serratia, Providencia, Chromobacterium, Brucella, Yersinia, Haemophilus, or Bordetella;
- (c) other infectious diseases, such as chlamydia, fungal diseases including but not limited to candidiasis, aspergillosis, histoplasmosis, cryptococcal meningitis, or parasitic diseases including but not limited to malaria, pneumocystis carnii pneumonia, leishmaniasis, cryptosporidiosis, toxoplasmosis, and trypanosome infection;
- (d) neoplastic diseases, such as intraepithelial neoplasias, cervical dysplasia, actinic keratosis, basal cell carcinoma, squamous cell carcinoma, renal cell carcinoma, Kaposi's sarcoma, melanoma, leukemias including but not limited to acute myeloid leukemia, acute lymphocytic leukemia, chronic myeloid leukemia, chronic lymphocytic leukemia, multiple myeloma, Hodgkin's lymphoma, non-Hodgkin's lymphoma, cutaneous T-cell lymphoma, B-cell lymphoma, and hairy cell leukemia, and other cancers;
- (e) T<sub>H</sub>2-mediated, atopic diseases, such as atopic dermatitis or eczema, eosinophilia, asthma, allergy, allergic rhinitis, and Ommen's syndrome;
- (f) certain autoimmune diseases such as systemic lupus erythematosus, essential thrombocythaemia, multiple sclerosis, discoid lupus, alopecia areata; and
- (g) diseases associated with wound repair such as, for example, inhibition of keloid formation and other types of scarring (e.g., enhancing wound healing, including chronic wounds).

Additionally, a compound or salt identified herein may be useful as a vaccine adjuvant for use in conjunction with any material that raises either humoral and/or cell mediated immune response, such as, for example, live viral, bacterial, or parasitic

immunogens; inactivated viral, tumor-derived, protozoal, organism-derived, fungal, or bacterial immunogens; toxoids; toxins; self-antigens; polysaccharides; proteins; glycoproteins; peptides; cellular vaccines; DNA vaccines; autologous vaccines; recombinant proteins; and the like, for use in connection with, for example, BCG, cholera, plague, typhoid, hepatitis A, hepatitis B, hepatitis C, influenza A, influenza B, parainfluenza, polio, rabies, measles, mumps, rubella, yellow fever, tetanus, diphtheria, hemophilus influenza b, tuberculosis, meningococcal and pneumococcal vaccines, adenovirus, HIV, chicken pox, cytomegalovirus, dengue, feline leukemia, fowl plague, HSV-1 and HSV-2, hog cholera, Japanese encephalitis, respiratory syncytial virus, rotavirus, papilloma virus, yellow fever, and Alzheimer's Disease.

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Compounds or salts identified herein may be particularly helpful in individuals having compromised immune function. For example, compounds or salts may be used for treating the opportunistic infections and tumors that occur after suppression of cell mediated immunity in, for example, transplant patients, cancer patients and HIV patients.

Thus, one or more of the above diseases or types of diseases, for example, a viral disease or a neoplastic disease may be treated in an animal in need thereof (having the disease) by administering a therapeutically effective amount of a compound or salt of the invention to the animal.

An animal may also be vaccinated by administering an effective amount of a compound or salt described herein, as a vaccine adjuvant. In one embodiment, there is provided a method of vaccinating an animal comprising administering an effective amount of a compound or salt described herein to the animal as a vaccine adjuvant.

An amount of a compound or salt effective to induce cytokine biosynthesis is an amount sufficient to cause one or more cell types, such as dendritic cells and B-cells to produce an amount of one or more cytokines such as, for example, IFN- $\alpha$ , IP-10, and MCP-1 that is increased (induced) over a background level of such cytokines. The precise amount will vary according to factors known in the art but is expected to be a dose of about 100 ng/kg to about 50 mg/kg, preferably about 10 µg/kg to about 5 mg/kg. In other embodiments, the amount is expected to be a dose of, for example, from about 0.01 mg/m² to about 5.0 mg/m², (computed according to the Dubois method as described above) although in some embodiments the induction of cytokine biosynthesis may be performed by administering a compound or salt in a dose outside this range. In some of these

embodiments, the method includes administering sufficient compound or salt or composition to provide a dose of from about 0.1 mg/m<sup>2</sup> to about 2.0 mg/ m<sup>2</sup> to the subject, for example, a dose of from about 0.4 mg/m<sup>2</sup> to about 1.2 mg/m<sup>2</sup>.

The invention provides a method of treating a disease which is responsive to the induction of cytokine biosynthesis, particularly the preferential induction of IFN-a, including a method of treating a viral infection in an animal and a method of treating a neoplastic disease in an animal, comprising administering an effective amount of a compound or salt or composition of the invention to the animal. An amount effective to treat or inhibit a viral infection is an amount that will cause a reduction in one or more of the manifestations of viral infection, such as viral lesions, viral load, rate of virus production, and mortality as compared to untreated control animals. The precise amount that is effective for such treatment will vary according to factors known in the art but is expected to be a dose of about 100 ng/kg to about 50 mg/kg, preferably about 10 µg/kg to about 5 mg/kg. An amount of a compound or salt effective to treat a neoplastic condition is an amount that will cause a reduction in tumor size or in the number of tumor foci. Again, the precise amount will vary according to factors known in the art but is expected to be a dose of about 100 ng/kg to about 50 mg/kg, preferably about 10 µg/kg to about 5 mg/kg. In other embodiments, the amount is expected to be a dose of, for example, from about 0.01 mg/m<sup>2</sup> to about 5.0 mg/m<sup>2</sup>, (computed according to the Dubois method as described above) although in some embodiments either of these methods may be performed by administering a compound or salt in a dose outside this range. In some of these embodiments, the method includes administering sufficient compound or salt to provide a dose of from about 0.1 mg/m<sup>2</sup> to about 2.0 mg/ m<sup>2</sup> to the subject, for example, a dose of from about 0.4 mg/m<sup>2</sup> to about 1.2 mg/m<sup>2</sup>.

In addition to the formulations and uses described specifically herein, other formulations, uses, and administration devices suitable for compounds of the present invention are described in, for example, International Publication Nos. WO 03/077944 and WO 02/036592, U.S. Patent No. 6,245,776, and U.S. Publication Nos. 2003/0139364, 2003/185835, 2004/0258698, 2004/0265351, 2004/076633, and 2005/0009858.

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Objects and advantages of this invention are further illustrated by the following examples, but the particular materials and amounts thereof recited in these examples, as well as other conditions and details, should not be construed to unduly limit this invention.

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#### **EXAMPLES**

In the examples below normal high performance flash chromatography (prep HPLC) was carried out using a COMBIFLASH system (an automated high-performance flash purification product available from Teledyne Isco, Inc., Lincoln, Nebraska, USA) or a HORIZON HPFC system (an automated high-performance flash purification product available from Biotage, Inc, Charlottesville, Virginia, USA). The eluent used for each purification is given in the example. In some chromatographic separations, the solvent mixture 80/18/2 v/v/v chloroform/methanol/concentrated ammonium hydroxide (CMA) was used as the polar component of the eluent. In these separations, CMA was mixed with chloroform in the indicated ratio.

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Example 1

N-{3-[4-Amino-2-(2-hydroxyethyl)-1H-imidazo[4,5-c]quinolin-1-yl]propyl}-4-methylbenzenesulfonamide

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Boron tribromide (5.50 mL of 1 M in dichloromethane) was added dropwise to a chilled (0 °C) suspension of N-{3-[4-amino-2-(2-methoxyethyl)-1H-imidazo[4,5-c]quinolin-1-yl]propyl}-4-methylbenzenesulfonamide (1.0 g, 2.2 mmol; U.S. Patent No. 6,677,349, Example 253) in dichloromethane (20 mL). The reaction mixture was stirred at 0 °C for 3 hours. The reaction mixture was quenched with methanol. Hydrochloric acid (about 10 mL of 6 N) was added and the mixture was stirred at 50 °C overnight. The mixture was diluted with water (50 mL) and ethyl acetate (100 mL) and then brought to

neutral pH with solid sodium hydroxide. The layers were separated and the aqueous was extracted with ethyl acetate (x2). The combined organics were dried over magnesium sulfate, filtered, and then concentrated under reduced pressure to provide a yellow solid. This material was purified by prep HPLC (COMBIFLASH system eluting first with a gradient of 0 to 5% methanol in dichloromethane containing 1% ammonium hydroxide and then with a gradient of 5 to 10% methanol in dichloromethane containing 1% ammonium hydroxide) to provide a white solid. This material was suspended in hot acetonitrile, allowed to cool, and then the solvent was decanted. The resulting material was dried under vacuum to provide about 200 mg of *N*-{3-[4-amino-2-(2-hydroxyethyl)-1*H*-imidazo[4,5-*c*]quinolin-1-yl]propyl}-4-methylbenzenesulfonamide as a white solid, m.p.231-232 °C. Anal. calcd for C<sub>22</sub>H<sub>25</sub>N<sub>5</sub>O<sub>3</sub>S•0.20 CH<sub>4</sub>O: %C, 59.79; %H, 5.85; %N, 15.70. Found: %C, 59.44; %H, 5.89; %N, 15.52.

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## Example 2

15  $N-\{3-[4-Amino-2-(2-hydroxyethyl)-1H-imidazo[4,5-c]quinolin-1-yl]propyl\}$  isoquinoline-3-carboxamide

Boron tribromide (5.50 mL of 1 M in dichloromethane) was added dropwise to a chilled (0 °C) suspension of N-{3-[4-amino-2-(2-methoxyethyl)-1H-imidazo[4,5-c]quinolin-1-yl]propyl}isoquinoline-3-carboxamide (1.0 g, 2.2 mmol; U.S. Patent No. 6,756,382, Example 192) in dichloromethane (20 mL). The reaction mixture was stirred at 0 °C for 45 minutes and then allowed to warm to ambient temperature. After 5 hours the reaction mixture was concentrated under reduced pressure and the residue was allowed to stand over the weekend. The residue was diluted with methanol (20 mL) and then heated to 50 °C. Hydrochloric acid (about 10 mL of 6 N) was added and the mixture was stirred for about 2.5-hours. The mixture was made basic with aqueous sodium hydroxide and

then extracted with ethyl acetate (x2). The combined extracts were dried over magnesium sulfate, filtered, and then concentrated under reduced pressure to provide a yellow solid. This material was purified by prep HPLC (COMBIFLASH system eluting first with a gradient of 0 to 5% methanol in dichloromethane containing 1% ammonium hydroxide and then with a gradient of 5 to 10% methanol in dichloromethane containing 1% ammonium hydroxide) to provide a white solid. This material was suspended in hot acetonitrile, allowed to cool, and then the solvent was decanted. The resulting material was dried under vacuum to provide about 400 mg of *N*-{3-[4-amino-2-(2-hydroxyethyl)-1*H*-imidazo[4,5-*c*]quinolin-1-yl]propyl}isoquinoline-3-carboxamide as a white solid, mp 245-246 °C. Anal calcd for C<sub>25</sub>H<sub>24</sub>N<sub>6</sub>O<sub>2</sub>: %C, 67.73; %H, 5.59; %N, 18.80; Found: %C, 67.38; %H, 5.54; %N, 18.84.

Example 3  $N-\{4-[4-Amino-2-(2-hydroxyethyl)-1H-imidazo[4,5-c]quinolin-1-yl]butyl\}$ methanesulfonamide

Part A

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3-Methoxypropionyl chloride (15.4 g, 126 mmol) was added dropwise over a period of 20 minutes to a chilled (ice bath) solution of *tert*-butyl *N*-{4-[(3-aminoquinolin-4-yl)amino]butyl}carbamate (38 g, 115 mmol, U.S. Patent No. 6,541,485, Example 2, Part B) in pyridine. The reaction mixture was stirred for 4 hours and then allowed to stand at ambient temperature over the weekend. Pyridine hydrochloride (3.9 g, 34 mmol) was added and the reaction mixture was heated at reflux overnight. The reaction mixture was concentrated under reduced pressure and the residue was diluted with dichloromethane (250 mL) and aqueous sodium bicarbonate (250 mL). The layers were separated. The separatory funnel was rinsed with a small amount of methanol to remove a residue coating the walls. The combined organics were concentrated under reduced pressure. The residue was purified by prep HPLC (COMBIFLASH system eluting first with a gradient of 0 to

5% methanol in dichloromethane containing 1% ammonium hydroxide and then with a gradient of 5 to 10% methanol in dichloromethane containing 1% ammonium hydroxide) to provide 18 g of *tert*-butyl *N*-{4-[2-(2-methoxyethyl)-1*H*-imidazo[4,5-*c*]quinolin-1-yl]butyl}carbamate.

## 5 Part B

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3-Chloroperoxybenzoic acid (20 g of 77%) was added in a single portion to a solution of the material from Part A (18 g, 45.2 mmol) in dichloroethane (170 mL). After 2 hours concentrated ammonium hydroxide (150 mL) was added and the reaction mixture was stirred until the phases were mixed well. *Para*-Toluenesulfonyl chloride (10.6 g, 54 mmol) was added in a single portion along with a small amount of dichloroethane. The reaction mixture was stirred overnight at ambient temperature and then diluted with water and dichloromethane. The layers were separated and the aqueous layer was extracted with dichloromethane (x2). The combined organics were dried over magnesium sulfate, filtered, and then concentrated under reduced pressure to provide 23 g of crude *tert*-butyl *N*-{4-[4-amino-2-(2-methoxyethyl)-1*H*-imidazo[4,5-c]quinolin-1-yl]butyl}carbamate as a red tar.

### Part C

The material from Part B was combined with a solution of hydrochloric acid in dioxane (325 mL of 4 M) and stirred at ambient temperature for 3 hours. The reaction mixture was concentrated under reduced pressure. The residue was dissolved in methanol (30 mL) and 6 M sodium hydroxide was added with stirring to about pH 9. Attempts to extract with dichloromethane and ethyl acetate were not successful. The organic and aqueous layers were concentrated under reduced pressure and combined to provide a dark orange solid. This material was purified by prep HPLC (COMBIFLASH system eluting first with a gradient of 0 to 8% methanol in dichloromethane containing 1% ammonium hydroxide and then with a gradient of 9 to 35% methanol in dichloromethane containing 1% ammonium hydroxide) to provide 10.65 g of 1-(4-aminobutyl)-2-(2-methoxyethyl)-1H-imidazo[4,5-c]quinolin-4-amine as an orange solid.

Part D

Triethylamine (10.5 mL, 75.0 mmol) was added to a mixture of a portion (4.7 g, 15 mmol) of the material from Part C in pyridine (50 mL). The reaction mixture was stirred for several minutes and then methanesulfonyl chloride (1.27 mL, 16.5 mmol) was added

dropwise. The reaction mixture was stirred at ambient temperature for 2 hours and then at 50 °C for 2 hours. More methanesulfonyl chloride (0.5 eq) was added and the reaction mixture was stirred at 50 °C for 2 hours. Another portion of methanesulfonyl chloride (0.25 eq) was added and the reaction mixture was stirred at ambient temperature overnight. The reaction mixture was diluted with dichloromethane and water. The layers were separated and the aqueous layer was extracted with dichloromethane (x3). The combined organics were dried over magnesium sulfate, filtered, and then concentrated under reduced pressure to provide 5 g of crude N-{4-[4-amino-2-(2-methoxyethyl)-1H-imidazo[4,5-c]quinolin-1-yl]butyl}methanesulfonamide as a red oil.

10 Part E

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Boron tribromide (22.4 mL of 1 M in dichloromethane) was added slowly to a chilled (ice bath) mixture of a portion of the material from Part D (3.5 g, about 8.9 mmol) and dichloromethane (50 mL). After the addition was complete the ice bath was removed and the reaction mixture was allowed to stir at ambient temperature for 3 hours. The reaction mixture was concentrated under reduced pressure. The residue was dissolved in methanol and then combined with hydrochloric acid (50 mL of 6 M). The mixture was stirred at 50 °C for 2 hours and then concentrated under reduced pressure. The residue was combined with ammonia in methanol (about 50 mL of 7 M) to neutralize the acid and then concentrated. This procedure was repeated 3 times. The crude product was purified by prep HPLC (COMBIFLASH system eluting with a gradient of 0 to 10% methanol in dichloromethane containing 1% ammonium hydroxide). The product was stirred with hot acetonitrile, allowed to stand overnight, and then isolated by filtration, washed with acetonitrile, and dried in a vacuum oven to provide 1.1 g of N-{4-[4-amino-2-(2hydroxyethyl)-1*H*-imidazo[4,5-c]quinolin-1-yl]butyl}methanesulfonamide, mp 206-208 °C. Anal calcd for C<sub>17</sub>H<sub>23</sub>N<sub>5</sub>O<sub>3</sub>S: %C, 54.09; %H, 6.14; %N, 18.55. Found: %C, 53.83; %H, 6.29; %N, 18.29.

### Example 4

1-(2-Amino-2-methylpropyl)-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-4-amine

Part A

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Under a nitrogen atmosphere, triethylamine (6.6 mL, 47 mmol) was added slowly to a solution of 2,4-dichloro-3-nitroquinoline (10.0 g, 41.1 mmol) in anhydrous 1-methyl-2-pyrrolidinone (40 mL). The reaction mixture was cooled to 0 °C with an ice bath. A solution of 1,2-diamino-2-methylpropane (4.1 g, 47.3 mmol) in anhydrous 1-methyl-2-pyrrolidinone (5 mL) was added dropwise over a period of 15 minutes while maintaining the temperature of the reaction mixture below 4 °C. After the addition was completed the ice bath was removed and the reaction mixture was allowed to stir at ambient temperature for 4 hours. The reaction mixture was slowly poured into vigorously stirred warm water (300 mL). The resulting suspension was stirred for 1 hour and then cooled to 13 °C by adding ice. The solid was isolated by filtration and then washed with cold water until the filtrate was clear to provide 12.1 g of  $N^1$ -(2-chloro-3-nitroquinolin-4-yl)-2-methylpropane-1,2-diamine as a damp yellow solid.

Part B

A solution of sodium hydroxide (1.8 g of solid sodium hydroxide dissolved in 45 mL of water) was added slowly to a solution of the material from Part A (41.1 mmol) in tetrahydrofuran (96 mL). A solution of di-tert-butyl dicarbonate (10.8 g, 49.4 mmol) in tetrahydrofuran (30 mL) was added dropwise over a period of 15 minutes. The reaction solution was stirred at ambient temperature. After 6 hours 10% sodium hydroxide (2 mL) and additional di-tert-butyl dicarbonate (1.5 g) were added and the reaction solution was stirred at ambient temperature overnight. The layers were separated and the tetrahydrofuran was removed under reduced pressure to provide a mixture. The mixture was diluted with water (200 mL) and then extracted with dichloromethane (2 x 100 mL). The organics were combined, washed sequentially with aqueous sodium carbonate (2 x 150 mL) and brine (100 mL), dried over sodium sulfate and magnesium sulfate, filtered, and then concentrated under reduced pressure. The residue was triturated with heptane (75 mL) for 15 minutes at 65 °C and then filtered while hot. The isolated solids were

washed with heptane (20 mL) to provide 13.2 g of *tert*-butyl *N*-{2-[(2-chloro-3-nitroquinolin-4-yl)amino]-1,1-dimethylethyl}carbamate as a yellow powdery solid. Part C

A Parr vessel was charged with 5% Pt/C (0.5 g) and acetonitrile (10 mL). A solution of the material from Part B in acetonitrile (450 mL) was added. The vessel was placed on a Parr shaker under hydrogen pressure (40 psi, 2.8 x 10<sup>5</sup> Pa) for 5 hours. The reaction mixture was filtered through a layer of CELITE filter aid to remove the catalyst. The filtrate was carried on to the next step.

Part D

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The solution of *tert*-butyl *N*-{2-[(3-amino-2-chloroquinolin-4-yl)amino]-1,1-dimethylethyl} carbamate in acetonitrile from Part C was cooled to 5 °C using an ice bath. A solution of acetoxyacetyl chloride (4.8 g, 35.1 mmol) in acetonitrile (20 mL) was added dropwise at a rate such that the temperature of the reaction mixture was maintained at 5 °C. After the addition was complete the ice bath was removed and the reaction mixture was allowed to stir at ambient temperature for 5 hours. The reaction mixture was concentrated under reduced pressure to provide 16.7 g of *N*-{2-[(3-acetoxyacetylamino-2-chloroquinolin-4-yl)amino]-1,1-dimethylethyl} carbamate hydrochloride as a yellow powder.

Part E

A mixture of the material from Part D (15.7 g) and ammonia in methanol (235 mL of 7 N) was divided into equal portions and placed in pressure vessels. The vessels were sealed, heated at 160 °C for 20 hrs, and then allowed to cool to ambient temperature overnight. The reaction mixtures were filtered. The isolated solids were washed with water and dried in a vacuum oven at 60 °C overnight to provide 6.0 g of a tan powder. A portion (1 g) was treated with activated charcoal and recrystallized from ethanol (75 mL) to provide 0.5 g of 1-(2-amino-2-methylpropyl)-2-hydroxymethyl-1*H*-imidazo[4,5-c]quinolin-4-amine as a white granular solid, mp 248-250 °C. Anal calcd for C<sub>15</sub>H<sub>19</sub>N<sub>5</sub>O: %C, 63.14; %H, 6.71; %N, 24.54. Found: %C, 63.13; %H, 6.81; %N, 24.64.

# Example 5

N-[2-(4-Amino-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1-yl)-1,1-dimethylethyl]cyclohexanecarboxamide

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A solution of 1-(2-amino-2-methylpropyl)-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-4-amine (2.0 g, 7.0 mmol) in 1-methyl-2-pyrrolidinone (30 mL) was cooled to -20 °C. Triethylamine (1.1 mL, 7.7 mmol) was added in a single portion. A chilled (-5 °C) solution of cyclohexanecarbonyl chloride (1.03 g, 7.0 mmol) in 1-methyl-2-pyrrolidinone (2 mL) was added dropwise over a period of 20 minutes while maintaining the reaction mixture at -20 °C. The reaction mixture was stirred at ambient temperature overnight. Additional cyclohexanecarbonyl chloride (0.1 g) was added and the reaction mixture stirred for 2 hours. The reaction mixture was poured into water with vigorous stirring. The resulting precipitate was isolated by filtration to provide 1.7 g of an ivory powder. Analysis by high performance liquid chromatography and NMR indicated that the powder was a mixture of the desired product and an ester formed from the reaction of the hydroxy group of the desired product with cyclohexanecarbonyl chloride.

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The powder was dissolved in ethanol (25 mL), combined with a solution of sodium hydroxide (0.21 g) in water (25 mL), and then heated at 50 °C for 3 hours. The ethanol was removed under reduced pressure and the solids were isolated by filtration to provide 1.2 g of a light tan powder. The powder was dissolved in a mixture of acetonitrile (100 mL), water (2 mL) and ethanol (25 mL). The solution was allowed to stand overnight and was then concentrated to a volume of 5 mL to provide a white paste. The paste was triturated with warm (70 °C) acetonitrile (50 mL) for 30 minutes, heated to reflux, and then allowed to cool to ambient temperature. The resulting solid was isolated by filtration to provide 1.05 g of *N*-[2-(4-amino-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1-yl)-1,1-dimethylethyl]cyclohexanecarboxamide as a light yellow powder, mp 248-250 °C. Anal calcd for C<sub>22</sub>H<sub>29</sub>N<sub>5</sub>O<sub>2</sub>: %C, 66.81; %H, 7.39; %N, 17.71; Found: %C, 66.56; %H, 7.60; %N, 17.82.

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# Example 6

 $N-\{2-[4-Amino-2-(2-hydroxyethyl)-1H-imidazo[4,5-c]quinolin-1-yl]-1,1-dimethylethyl\}$  methanesulfonamide

Part A

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Triethylamine (39.3 mL, 0.282 mol) was added to a chilled (ice bath) solution of  $N^{1}$ -(2-chloro-3-nitroquinolin-4-yl)-2-methylpropane-1,2-diamine (41.42 g, 0.141 mol) in dichloromethane (about 500 mL). Under a nitrogen atmosphere a solution of methanesulfonic anhydride in (29.47 g, 0.169 mol) in dichloromethane (100 mL) was added via a cannula to the reaction mixture over a period of 45 minutes. After the addition was complete the ice bath was removed and the reaction mixture was allowed to stir at ambient temperature overnight. The reaction mixture was washed sequentially with saturated aqueous sodium bicarbonate (x2) and brine, dried over a mixture of sodium sulfate and magnesium sulfate, filtered, and then concentrated under reduced pressure to provide 46.22 g of an orange solid. This material was recrystallized from toluene (about 1 L), isolated by filtration, rinsed with cold toluene, and dried under high vacuum at 60 °C to provide 33.09 g of N-{2-[(2-chloro-3-nitroquinolin-4-yl)amino]-1,1-dimethylethyl} methanesulfonamide.

20 Part B

A hydrogenation vessel was charged with 5% Pt/C (4.14 g) and a solution of N-{2-[(2-chloro-3-nitroquinolin-4-yl)amino]-1,1-dimethylethyl}methanesulfonamide (54.59 g, 0.147 mol) in acetonitrile (1800 mL). The vessel was placed under hydrogen pressure (48 psi, 3.3 x  $10^5$  Pa) overnight. An additional portion (4.25 g) of catalyst was added and the vessel was placed under hydrogen pressure (48 psi, 3.3 x  $10^5$  Pa) for 4 hours. The reaction mixture was filtered through a layer of CELITE filter aid and the filter cake was rinsed with fresh acetonitrile until the washes were clear.

Part C

Under a nitrogen atmosphere, 3-methoxypropionyl chloride (17.6 mL, 0.162 mol) was added dropwise to the solution of N-{2-[(3-amino-2-chloroquinolin-4-yl)amino]-1,1dimethylethyl}methanesulfonamide (0.147 mol) in acetonitrile (2.2 L) from Part B. The reaction mixture was allowed to stir at ambient temperature over the weekend. The resulting precipitate was isolated by filtration, rinsed with a small amount of acetonitrile, and then dried under high vacuum at 60 °C to provide 55.84 g of N-{2-chloro-4-[2-(methanesulfonylamino)-2-methylpropyl]quinolin-3-yl}-3-methoxypropionamide. Part D

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A Parr bomb was charged with 25.0 g of N-{2-chloro-4-[2-(methanesulfonylamino)-2-methylpropyl]aminoquinolin-3-yl}-3-methoxypropionamide and ammonia in methanol (300 mL of 7 N). A second vessel was charged with 30.21 g of N-{2-chloro-4-[2-(methanesulfonylamino)-2-methylpropyl]quinolin-3-yl}-3methoxypropionamide and ammonia in methanol (400 mL of 7 N). Both vessels were sealed and then heated at 170 °C for 14 hours. The reaction mixtures were combined and the solvent was removed under reduced pressure. The residue was partitioned between dichloromethane and saturated aqueous sodium bicarbonate. The organic layer was washed sequentially with saturated aqueous sodium bicarbonate and brine, dried over sodium sulfate, filtered, and then concentrated under reduced pressure to provide 38.16 g of  $N-\{2-[4-amino-2-(2-methoxyethyl)-1H-imidazo[4,5-c]quinolin-1-yl]-1,1$ dimethylethyl}methanesulfonamide as an off white foam. Part E

Under a nitrogen atmosphere, boron tribromide (3.5 mL of 1 M in dichloromethane) was added dropwise to a chilled (0 °C) solution of N-{2-[4-amino-2-(2methoxyethyl)-1H-imidazo[4,5-c]quinolin-1-yl]-1,1-dimethylethyl}methanesulfonamide (0.55 g, 1.40 mmol) in dichloromethane (20 mL). The reaction was allowed to warm to ambient temperature overnight. The reaction was quenched with methanol (10 mL) and the solvent was removed under reduced pressure. The residue was dissolved in hydrochloric acid (6 N), stirred at 50 °C for about 2.5 hours, and then allowed to cool to ambient temperature. The reaction mixture was adjusted to pH 11 with ammonium hydroxide and then extracted with dichloromethane (x 10). The combined organics were washed with brine, dried over sodium sulfate, filtered, and then concentrated under

reduced pressure to provide 0.47 g of a white solid. This material was purified by prep HPLC (HORIZON HPFC system, eluting with a gradient of 30-50% CMA in chloroform for 15 column volumes followed by 50% CMA in chloroform for 5 column volumes) and then dried under high vacuum to provide 250 mg of N-{2-[4-amino-2-(2-hydroxyethyl)-1H-imidazo[4,5-c]quinolin-1-yl]-1,1-dimethylethyl}methanesulfonamide as white solid, m.p. 209 - 212°C. <sup>1</sup>H NMR (500 MHz, DMSO- $d_6$ )  $\delta$  8.30 (d, J= 8.2 Hz, 1H), 7.60 (d, J= 8.2 Hz, 1H), 7.39 (m, 1H), 7.27 (s, 1H), 7.21 (m, 1H), 6.49 (s, 2H), 4.84 (t, J= 5.4 Hz, 2H), 4.82 (br s, 1H), 3.88 (m, 2H), 3.18 (br s, 2H), 3.00 (s, 3H), 1.27 (br s, 6H); <sup>13</sup>C NMR (125 MHz, DMSO- $d_6$ )  $\delta$  153.6, 152.0, 145.4, 133.5, 126.9, 126.8, 126.5, 121.3, 120.8, 115.6, 60.5, 57.9, 54.1, 44.8, 31.4, 25.8; MS (ESI) m/z 378 (M + H)<sup>+</sup>; Anal. calcd for  $C_{17}H_{23}N_5O_3S$ : %C, 54.09; %H, 6.14; %N, 18.55. Found: %C, 53.76; %H, 6.02; %N, 18.32.

# Example 7

N-[2-(4-Amino-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1-yl)-1,1-dimethylethyl]methanesulfonamide

### Part A

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A pressure vessel was charged with a solution of of N-{2-[(2-chloro-3-nitroquinolin-4-yl)amino]-1,1-dimethylethyl}methanesulfonamide (5 g, 13 mmol) in acetonitrile (150 mL). Catalyst was added (0.5 g of 5% Pt/C) and the vessel was placed under hydrogen pressure (50 psi, 3.4 X 10<sup>5</sup> Pa) for 2 hours. The reaction mixture was filtered through a layer of CELITE filter aid.

Part B

The solution of N-{2-[(3-amino-2-chloroquinolin-4-yl)amino]-1,1-dimethylethyl}methanesulfonamide in acetonitrile from Part A was chilled in an ice bath. Acetoxyacetyl chloride (1.5 mL, 14 mmol) was added over a period of 5 minutes. The reaction mixture was allowed to stir for 3 hours. A precipitate was isolated by filtration

and rinsed with acetonitrile to provide crude  $N-\{2-\text{chloro-}4-[2-\text{(methanesulfonylamino)-}2-\text{methylpropyl}\}$  acetoxyacetamide hydrochloride.

A solution of sodium hydroxide (0.8 g) in water (15 mL) was added to a suspension of the material from Part B in ethanol (60 mL) until all of the solid dissolved. The reaction mixture was heated at 60 °C overnight and then concentrated under reduced pressure. The residue was dissolved in water (50 mL), sodium chloride (10 g) was added, and the mixture was extracted with chloroform (3 x 300 mL). The extracts were concentrated under reduced pressure to provide about 4 g of crude *N*-[2-(4-chloro-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1-yl)-1,1-dimethylethyl]methanesulfonamide. Part D

The material from Part C was combined with a solution of ammonia in methanol (50 mL of 7 N) and heated at 150 °C for 10 hours. The reaction mixture was allowed to cool to ambient temperature. A precipitate was isolated by filtration, rinsed with methanol (20 mL), slurried with water (50 mL), isolated by filtration, washed with water (20 mL), and dried to provide 2.7 g of a brown crystalline solid. This material was combined with methanol (50 mL), heated at 50 °C overnight, and then isolated by filtration to provide 2.3 g of N-[2-(4-amino-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1-yl)-1,1-dimethylethyl]methanesulfonamide, mp 262-265 °C. Anal. calcd for C<sub>16</sub>H<sub>21</sub>N<sub>5</sub>O<sub>3</sub>S: %C, 52.88; %H, 5.82; %N, 19.27. Found: %C, 52.64; %H, 5.95; %N, 19.50.

## Examples 8 - 72

Part A

Part C

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A reagent (1.1 eq) from Table 1 below was added to a test tube containing a solution of 1-(4-aminobutyl)-2-(2-methoxyethyl)-1H-imidazo[4,5-c]quinolin-4-amine (73 mg) in N,N-dimethylacetamide (1 mL) containing N,N-diisopropylethylamine (2 eq). The test tube was placed on a shaker overnight. The solvent was removed by vacuum centrifugation. The reaction mixtures were separated by solid-supported liquid-liquid extraction according to the following procedure. Each sample was dissolved in chloroform (1 mL) then loaded onto diatomaceous earth that had been equilibrated with de-ionized water (600  $\mu$ L) for about 20 minutes. After 10 minutes chloroform (500  $\mu$ L) was added to elute the product from the diatomaceous earth into a well of a collection

plate. After an additional 10 minutes the process was repeated with additional chloroform (500 μL). The solvent was then removed by vacuum centrifugation.

The residue (in a test tube) was combined with dichloromethane (1 mL) and the

mixture was sonicated to dissolve the solids. The solution was cooled (0 °C) and then

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Part B

combined with boron tribromide (400 µL of 1 M in heptane). The mixture was shaken for 5 minutes, placed in an ice bath for 30 minutes, and then shaken overnight. The solvents were removed by vacuum centrifugation. The residue was diluted with methanol (1 mL) and hydrochloric acid (500 µL of 6 N). The mixture was shaken for 30 minutes and then the solvents were removed by vacuum centrifugation. The compounds were purified by preparative high performance liquid chromatography (prep HPLC) using a Waters FractionLynx automated purification system. The prep HPLC fractions were analyzed using a Waters LC/TOF-MS, and the appropriate fractions were centrifuge evaporated to provide the trifluoroacetate salt of the desired compound. Reversed phase preparative liquid chromatography was performed with non-linear gradient elution from 5-95% B where A is 0.05% trifluoroacetic acid/water and B is 0.05% trifluoroacetic

acid/acetonitrile. Fractions were collected by mass-selective triggering. Table 1 below shows the reagent used for each example, the structure of the resulting compound, and the

observed accurate mass for the isolated trifluoroacetate salt.

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Table 1				
NH <sub>2</sub> N OH				
Example	Reagent	R	Measured Mass (M+H)	
8	None	Н	300.1840	
9	Cyclopropanecarbonyl chloride		368.2063	

10	Isobutyryl chloride	H³C CH³	370.2224
11	Pivaloyl chloride	H <sub>3</sub> C CH <sub>3</sub>	384.2390
12	Benzoyl chloride	To the second se	404.2103
13	Phenyl chloroformate	To C	420.2056
14	3-Cyanobenzoyl chloride	O	429.2031
15	Hydrocinnamoyl chloride		432.2377
16	Isonicotinoyl chloride hydrochloride	o N	405.2071
17	Nicotinoyl chloride hydrochloride	0 2	405.2058
18	Methanesulfonyl chloride	,O S,-CH <sub>3</sub> O	378.1592
19	Ethanesulfonyl chloride	O,O S, CH₃	392.1729
20	1-Propanesulfonyl chloride	O.O. CH,	406.1899

21	Isopropylsulfonyl chloride	ONO CH <sub>3</sub>	406.1888
22	Dimethylsulfamoyl chloride	O O H <sub>3</sub> C,N-CH <sub>3</sub>	407.1853
23	1-Butanesulfonyl chloride	Q.O S,O CH₃	420.2050
24	Benzenesulfonyl chloride	0.0	440.1741
25	1-Methylimidazole-4-sulfonyl chloride	O.O S N CH <sub>3</sub>	444.1806
26	3-Methylbenzenesulfonyl chloride	O.O S	454.1895
27	alpha-Toluenesulfonyl chloride	0,0	454.1923
28	o-Toluenesulfonyl chloride	H <sub>3</sub> C	454.1944
29	p-Toluenesulfonyl chloride	O.O S CH₃	454.1907
30	2-Fluorobenzenesulfonyl chloride	P S	458.1664

31	3-Fluorobenzenesulfonyl chloride	O O	458.1652
32	4-Fluorobenzenesulfonyl chloride	O,O F	458.1639
33	3-Cyanobenzenesulfonyl chloride	O.O.	465.1678
34	4-Cyanobenzenesulfonyl chloride	0,0	465.1668
35	beta-Styrene sulfonyl chloride	0.0	466.1895
36	2,5-Dimethylbenzenesulfonyl chloride	H <sub>3</sub> C CH <sub>3</sub>	468.2063
37	3,5-Dimethylbenzenesulfonyl chloride	O, O S'C H <sub>3</sub> C	468.2046
38	2-Chlorobenzenesulfonyl chloride	O.O CI	474.1351

39	3-Chlorobenzenesulfonyl chloride	O, O S	474.1385
40	4-Chlorobenzenesulfonyl chloride	0,0	474.1390
41	1-Naphthalenesulfonyl chloride	0,0	490.1891
42	2-Naphthalenesulfonyl chloride	0,0	490.1885
43	2- (Trifluoromethyl)benzenesulfonyl chloride	F F	508.1592
44	3- (Trifluoromethyl)benzenesulfonyl chloride	O. S	508.1612
45	4- (Trifluoromethyl)benzenesulfonyl chloride	O O F F	508.1640
46	2,3-Dichlorobenzenesulfonyl chloride	O,O S	508.0967

47	2,4-Dichlorobenzenesulfonyl chloride	O.O. CI	508.0979
48	2,5-Dichlorobenzenesulfonyl chloride	O,O S,O CI	508.0987
49	2,6-Dichlorobenzenesulfonyl chloride	O.O.CI	508.0968
50	3,4-Dichlorobenzenesulfonyl chloride	CI CI	508.0961
51	3,5-Dichlorobenzenesulfonyl chloride	O, O	508.0985
52	Methyl isocyanate	N-CH₃	357.2073
53	Ethyl isocyanate	N CH <sub>3</sub>	371.2203
54	Isopropyl isocyanate	O N CH <sub>3</sub>	385.2347
55	n-Propyl isocyanate	N CH3	385.2349

		<del></del>	<del></del>
56	n-Butyl isocyanate	CH <sub>3</sub>	399.2494
57	sec-Butyl isocyanate	O N CH <sub>3</sub>	399.2517
58	Cyclopentyl isocyanate	N \	411.2516
59	Cyclopropylmethyl isothiocyanate	HZ	413.2133
60	Phenyl isocyanate	HZ	419.2226
61	Cyclohexyl isocyanate	HZ O	425.2701
62	Benzyl isocyanate	HZ	433.2374
63	m-Tolyl isocyanate	O CH <sub>3</sub>	433.2344
64	Benzoyl isocyanate	P N N N N N N N N N N N N N N N N N N N	447.2126

65	2-Phenyl ethylisocyanate		447.2512
66	4-Chlorophenyl isocyanate	N CI	453.1797
67	trans-2-Phenylcyclopropyl isocyanate	N Control of the cont	459.2518
68	N,N-Dimethylcarbamoyl chloride	H <sub>3</sub> C <sup>,N</sup> -CH <sub>3</sub>	371.2185
69	1-Pyrrolidinecarbonyl chloride	° × >	397.2382
70	1-Piperidinecarbonyl chloride	N N	411.2526
71	4-Morpholinylcarbonyl chloride	N O	413.2330
72	N-Methyl-N-phenylcarbamoyl chloride	H <sub>3</sub> CN	433.2364

Examples 73 - 110

# Part A

Tert-Butyl 3-[4-amino-2-(2-methoxyethyl)-1H-imidazo[4,5-c]quinolin-1yl]propylcarbamate (5 g, U.S. Patent No. 6,573,273, example 148) and hydrochloric acid in dioxane (100 mL of 4 M) were combined and stirred for 4 hours at ambient temperature. The reaction mixture was concentrated under reduced pressure. The residue was dissolved in methanol (30 mL). The pH was adjusted to pH 8 with 6 M sodium hydroxide. The solution was diluted with dichloromethane, ethyl acetate, triethylamine,

and brine. The organic layer was concentrated under reduced pressure to provide an orange solid. This material was purified by prep HPLC (COMBIFLASH system eluting first with a gradient of 0 to 10% methanol in dichloromethane containing 1% ammonium hydroxide and then with a gradient of 9 to 30% methanol in dichloromethane containing 1% ammonium hydroxide) to provide 1.58 g of 1-(3-aminopropyl)-2-(2-methoxyethyl)-1*H*-imidazo[4,5-*c*]quinolin-4-amine as a yellow solid.

#### Part B

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A reagent (1.1 eq) from Table 2 below was added to a test tube containing a solution of 1-(3-aminopropyl)-2-(2-methoxyethyl)-1H-imidazo[4,5-c]quinolin-4-amine (30 mg) in chloroform (1 mL) containing N,N-diisopropylethylamine (1.5 eq). The test tube was placed on a shaker overnight. The reaction mixtures were separated by solid-supported liquid-liquid extraction according to the following procedure. Each reaction mixture was loaded onto diatomaceous earth that had been equilibrated with de-ionized water (600  $\mu$ L) for about 20 minutes. After 10 minutes chloroform (500  $\mu$ L) was added to elute the product from the diatomaceous earth into a well of a collection plate. After an additional 10 minutes the process was repeated with additional chloroform (500  $\mu$ L). The solvent was then removed by vacuum centrifugation.

### Part C

The ether was cleaved and the resulting product was purified using the method of Part B in Examples 8-72. Table 2 below shows the reagent used for each example, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

	Table 2			
	NH <sub>2</sub>	N OH H N R		
Example	Reagent	R	Measured Mass (M+H)	
73	None	\ H	286.1689	

		<del></del>	
74	Propionyl chloride	O CH <sub>3</sub>	342.1956
75	Cyclopropanecarbonyl chloride	0	354.1946
76	Butyryl chloride	О СН3	356.2122
77	Isobutyryl chloride	O CH <sub>3</sub>	356.2119
78	Cyclobutanecarbonyl chloride	<i>→</i>	368.2120
79	3-Chlorobenzoyl chloride	0	424.1570
80	4-Chlorobenzoyl chloride	o CI	424.1583
81	Nicotinoyl chloride hydrochloride	o N	391.1913
82	trans-2-Phenyl-1- cyclopropanecarbonyl chloride	o Link	430.2257
83	Methanesulfonyl chloride	O <sup>z</sup> S=O CH <sub>3</sub>	364.1479
84	Ethanesulfonyl chloride	O CH <sub>3</sub>	378.1639
85	1-Propanesulfonyl chloride	O=S- CH <sub>3</sub>	392.1783
86	Isopropylsulfonyl chloride	O=S CH₃ O CH₃	392.1788

87	Dimethylsulfamoyl chloride	O=S-NCH <sub>3</sub> CH <sub>3</sub>	393.1715
88	1-Butanesulfonyl chloride	O;S O'CH <sub>3</sub>	406.1946
89	Benzenesulfonyl chloride	0:S	426.1633
90	2,2,2- Trifluoroethanesulfonyl chloride	0=S 0 F F	432.1355
91	3- Methylbenzenesulfonyl chloride	O-S CH <sub>3</sub>	440.1774
92	alpha-Toluenesulfonyl chloride	o-s	440.1762
93	p- Toluenesulfonyl chloride	O=S O'CH <sub>3</sub>	440.1790
94	3- Fluorobenzenesulfonyl chloride	O:S F	444.1523
95	4- Fluorobenzenesulfonyl chloride	0-S 0'-S	444.1545
96	3- Cyanobenzenesulfonyl chloride	O-S	451.1554
97	4- Cyanobenzenesulfonyl chloride	0;S	451.1582
98 .	Ethyl isocyanate	O H CH3	357.2050

99	Isopropyl isocyanate	H <sub>3</sub> C CH <sub>3</sub>	371.2234
100	n-Butyl isocyanate	OH3	385.2364
101	Cyclopentyl isocyanate	O H	397.2359
102	Cyclopropylmethyl isothiocyanate	SH	399.1979
103	Phenyl isocyanate		405.2040
104	Cyclohexyl isocyanate	0	411.2526
105	Benzyl isocyanate	0	419.2239
106	trans-2- Phenylcyclopropyl isocyanate	HZ.	445.2388
107	1-Piperidinecarbonyl chloride	9 N	397.2384
108	4-Morpholinylcarbonyl chloride	0 N C	399.2173

109	4-Methyl-1- piperazinecarbonyl chloride	ON-CH <sub>3</sub>	412.2485
110	N-Methyl-N- phenylcarbamoyl chloride	O NCH3	419.2229

Examples 111 - 140

Boron tribromide (400  $\mu$ L of 1 M in heptane) was added to a tube containing a chilled (0 °C) solution of a compound of Formula Xa (about 25 mg) in dichloromethane (1 mL). The tube was vortexed, maintained at 0 °C for 0.5 hour, and then shaken overnight at ambient temperature. The reaction mixture was diluted with methanol (1 mL) and hydrochloric acid (250  $\mu$ L of 6 N), vortexed, and then the solvents were removed by vacuum centrifugation. The compounds were purified by prep HPLC as described in Examples 8 – 72. Table 3 shows the structure of the starting material, a reference for the starting material, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

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		Measured Mass (M+H)	455.2222	458.1657
Table 3	NH2 N N N OH N N N N N N N N N N N N N N N N N N N	. R <sub>1</sub>	O ZI	O.S. N. H.
		Reference Formula Xa	U.S. Patent No. 6,756,382 Example 57	U.S. Patent No. 6,331,539 Example 121
		Example	111	112

378.1599	300.1853	413.2301	455.2198
O.SCH	NH <sub>2</sub>	O ZI	O ZI
U.S. Patent No. 6,331,539 Example 111	Example 3 Part C	U.S. Patent No. 6,541,485 Example 121	U.S. Patent No. 6,756,382 Example 182
113	114	115	116

456.2161	475.2829	434.2253	286.1683
O ZI	O ZI ZI	HO.SN	NH <sub>2</sub>
U.S. Patent No. 6,756,382 Example 183	U.S. Patent No. 6,573,273 Example 145	U.S. Patent No. 6,677,349 Example 243	Example 73 Part A
117	118	119	120

460.2737	364.1446	411.2505	
ZI	N N O-S:O	IZO	H <sub>3</sub> C CH <sub>3</sub>
U.S. Patent No. 6,756,382 Example 187	U.S. Patent No. 6,677,349 Example 247	U.S. Patent No. 6,573,273 Example 158	U.S. Patent No. 6,756,382 Example 190
121	122	123	124

377.1655	385.2358	440.1720	399.2145	314.1980
O-is O-is	IZ	H N S=0 S=0 CH <sub>3</sub>	IN O	H <sub>3</sub> C CH <sub>3</sub>
U.S. Patent No. 6,664,264 Example 16	U.S. Patent No. 6,573,273 Example 162	U.S. Patent No. 6,677,349 Example 253	U.S. Patent No. 6,573,273 Example 163	U.S. Patent No. 6,677,349#
125	126	127	128	129

433.2321	392.1757	390.1929	408.1714	
H <sub>3</sub> C CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub> H <sub>3</sub> C S <sub>2</sub> cO	IZO	O+S, O	
U.S. Patent No. 6,573,273 Example 169	U.S. Patent No. 6,677,349 Example 256	U.S. Patent No. 6,756,382 Example 196	U.S. Patent No. 6,683,088 Example 3	
130	131	132	133	

434.2197	440.2672	350.1316	343.1884	356.2078
O CH3	CH <sub>3</sub>	HO.S-N	N N N N N N N N N N N N N N N N N N N	HO OH,
U.S. Patent No. 6,664,265 Example 8	U.S. Patent No. 6,664,265 Example 73	U.S. Patent No. 6,677,349#	U.S. Patent No. 6,573,273#	U.S. Patent No. 6,451,810#
134	135	136	137	138

378.1595	554.4064	tic routes.
H3C N-S-CH3	HZ O	Although not specifically exemplified the compound can be readily prepared using the disclosed synthetic routes.
U.S. Patent No. 6,677,349#	U.S. Patent Publication 2004/0091491 IRM3	not specifically exe
139	140	"Although n

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### Example 141

N-{3-[4-Amino-2-(2-hydroxyethyl)-1*H*-imidazo[4,5-*c*]quinolin-1-yl]propyl}-2-methylpropionamide

## 5 Part A

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1-(3-Aminopropyl)-2-(2-methoxyethyl)-1*H*-imidazo[4,5-*c*]quinolin-4-amine dihydrochloride (6 g, 16 mmol) was combined with triethylamine (11.2 mL, 80 mmol) and pyridine (100 mL). Isobutyryl chloride (1.9 g, 18 mmol) was added dropwise and the reaction mixture was stirred at ambient temperature for 1 hour. The reaction mixture was combined with saturated aqueous sodium bicarbonate and extracted with dichloromethane (3 x 200 mL). The combined organics were dried over magnesium sulfate, filtered through a layer of CELITE filter aid, and then concentrated under reduced pressure to provide 6.2 g of crude *N*-{3-[4-amino-2-(2-methoxyethyl)-1*H*-imidazo[4,5-*c*]quinolin-1-yl]propyl}-2-methylpropionamide as a brown solid.

### Part B

The material from Part A was combined with dichloromethane (40 mL), stirred until homogeneous, and then chilled in an ice bath. Boron tribromide (40 mL of 1 M in dichloromethane) was slowly added. The ice bath was removed and the reaction mixture was stirred overnight at ambient temperature. The reaction mixture was concentrated under reduced pressure. The residue was combined with methanol (50 mL) and hydrochloric acid (50 mL of 6 N) and heated at 50 °C for 2 hours. The solution was adjusted to pH 9 with sodium hydroxide (6 M) and then extracted first with ethyl acetate (3 x 100 mL) and then with dichloromethane. The organics were dried over magnesium sulfate, filtered through a layer of CELITE filter aid, and then concentrated under reduced pressure. The residue was purified by prep HPLC (HORIZON HPFC system, eluting with a gradient of 0-10% methanol in dichloromethane), recrystallized from acetonitrile, and then dried in a vacuum oven to provide 208 mg of N-{3-[4-amino-2-(2-hydroxyethyl)-1*H*-

imidazo[4,5-c]quinolin-1-yl]propyl}-2-methylpropionamide as an off-white solid, mp 196-198'°C. Anal. calcd for C<sub>19</sub>H<sub>25</sub>N<sub>5</sub>O<sub>2</sub>: %C, 64.20; %H, 7.09; %N, 19.70; Found: %C, 63.99; %H, 7.28; %N, 19.63.

# Example 142

1-[2-(4-Amino-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1-yl)-1,1-dimethylethyl]-3-(1-methylethyl)urea

Part A

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Under a nitrogen atmosphere, a solution of 1,2-diamino-2-methylpropane (52.20 mL, 503.3 mmol), triethylamine (131.8 mL, 958.8 mmol), and dichloromethane (1.0 L) was chilled in an ice water bath. 4-Chloro-3-nitroquinoline (100.0 g, 479.4 mmol) was added in portions over a period of 5 minutes. The reaction mixture was stirred at 0 °C for 2 hours and then allowed to slowly warm to ambient temperature. After 16 hours the reaction mixture was concentrated under reduced pressure. The residue was triturated with water (500 mL) for 1 hour. The resulting solid was isolated by filtration and dried overnight in a vacuum desiccator to provide 124.6 g of  $N^{1}$ -(3-nitroquinolin-1-yl)-2-methylpropane-1,2-diamine as a yellow crystalline solid.

Part B

Under a nitrogen atmosphere, a suspension of N<sup>1</sup>-(3-nitroquinolin-1-yl)-2-methylpropane-1,2-diamine (60.0 g, 231 mmol) in dichloromethane (1.0 L) was chilled in an ice bath. Isopropyl isocyanate (23.8 mL, 242 mmol) was added dropwise over a period of 10 minutes. The reaction was allowed to slowly warm to room temperature. After 17 hours additional isopropyl isocyanate (about 2 mL) was added. After an additional 3 hours more isopropyl isocyanate (1 mL) was added. After 2 more hours the reaction mixture was concentrated under reduced pressure to provide 79.8 g of 1-{1,1-dimethyl-2-[(3-nitroquinolin-1-yl)amino]ethyl}-3-(1-methylethyl)urea as a bright yellow solid.

Part C

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A pressure vessel was charged with the material from Part B, 5% Pt/C (4.24 g), and acetonitrile (1.5 L). The mixture was placed under hydrogen pressure for 20 hours and then filtered through a layer of CELITE filter aid. The filter cake was rinsed with additional acetonitrile. The filtrate was concentrated under reduced pressure. The residue was dissolved in toluene (750 mL) and then concentrated under reduced pressure to remove residual water. The toluene concentration was repeated. The residue was dissolved in dichloromethane (about 1 L), concentrated under reduced pressure, and then dried under high vacuum to provide 66.4 g of 1-{1,1-dimethyl-2-[(3-aminoquinolin-1-yl)amino]ethyl}-3-(1-methylethyl)urea as an orange foam.

Part D

Under a nitrogen atmosphere, a solution of 1-{1,1-dimethyl-2-[(3-aminoquinolin-1-yl)amino]ethyl}-3-(1-methylethyl)urea (66.0 g, 209 mmol) and triethylamine (32.1 mL, 230 mmol) in dichloromethane (1.0 L) was chilled in an ice bath. Ethoxyacetyl chloride (23.6 mL, 291 mmol) was added dropwise over a period of 10 minutes. The reaction mixture was allowed to slowly warm to ambient temperature overnight. The reaction mixture was concentrated under reduced pressure. The residue was combined with 1-butanol (800 mL) and triethylamine (87 mL, 627 mmol) and heated at 140 °C for 3 hours. The reaction mixture was cooled to ambient temperature and then concentrated under reduced pressure to provide a light brown foam. This material was purified by column chromatography (silica gel, eluting with 98/2/0.5 chloroform/methanol/ammonium hydroxide) to provide 29.36 g of 1-[2-(2-ethoxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1-yl)-1,1-dimethylethyl]-3-(1-methylethyl)urea as a light yellow foam.

3-Chloroperoxybenzoic acid (26.33 g of 60%, 91.56 mmol) was added in portions over a period of 5 minutes to a chilled solution of the material from Part D in chloroform (350 mL). The reaction mixture was allowed to slowly warm to ambient temperature. After 2 hours the reaction mixture was chilled in an ice bath and ammonium hydroxide (100 mL) was added with vigorous stirring to homogenize. *Para*-toluenesulfonyl chloride (15.27 g, 80.12 mmol) was added in portions over a period of 10 minutes. The ice bath was removed and the reaction mixture was stirred for 30 minutes. The reaction mixture was diluted with water (100 mL) and chloroform (250 mL). The layers were separated.

The organic layer was washed with 10% sodium carbonate (200 mL) and water (200 mL). The combined aqueous was back extracted with chloroform (100 mL). The combined organics were washed with brine (200 mL), dried over magnesium sulfate, filtered, and then concentrated under reduced pressure to provide a light brown foam. The foam was purified by column chromatography (silica gel, eluting with 95/5 chloroform/methanol) and then recrystallized from acetonitrile to provide 3.75 g of 1-[2-(4-amino-2-ethoxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1-yl)-1,1-dimethylethyl]-3-(1-methylethyl)urea as an off white solid.

Part F

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Under a nitrogen atmosphere, a suspension of 1-[2-(4-amino-2-ethoxymethyl-1Himidazo[4,5-c]quinolin-1-yl)-1,1-dimethylethyl]-3-(1-methylethyl)urea (1.19 g, 2.99) mmol) in dichloromethane (30 mL) was chilled in an ice bath. Boron tribromide (7.47 mL of 1 M in dichloromethane) was added. The reaction mixture was allowed to warm slowly to ambient temperature and then stirred for 18 hours. Additional boron tribromide (2 eq) was added. After 2 hours the reaction mixture was diluted with acetonitrile (10 mL) and the reaction mixture was stirred overnight. The reaction mixture was diluted with dichloromethane (10 mL) and acetonitrile (10 mL), stirred for an additional 16 hours, quenched with methanol (25 mL), and then concentrated under reduced pressure to provide an orange foam. The foam was dissolved in hydrochloric acid (25 mL of 6 N) and heated at 50 °C for 2 hours. The solution was neutralized with 50% sodium hydroxide. The resulting gummy precipitate was extracted with chloroform (3 x 15 mL). The combined organics were washed with brine (15 mL), dried over magnesium sulfate, filtered, and then concentrated under reduced pressure to provide an off white solid. This material was purified by prep HPLC (HORIZON HPFC system, eluting with a gradient of 15-50% CMA in chloroform) and then recrystallized from acetonitrile to provide 335 g of 1-[2-(4-amino-2-hydroxymethyl-1*H*-imidazo[4,5-c]quinolin-1-yl)-1,1-dimethylethyl]-3-(1-methylethyl)urea as a white crystalline solid, mp 196–199 °C; <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  8.38 (d, J = 8.0 Hz, 1 H), 7.59 (d, J = 7.5 Hz, 1 H), 7.43-7.38 (m, 1 H), 7.24-7.19 (m, 1 H), 6.54 (s, 2 H), 5.72 (s, 1 H), 5.63 (d, J = 7.6 Hz, 1 H), 5.46 (t, J = 5.7 Hz, 1 H), 5.01 (s, 2 H), 4.78 (s, 2 H), 3.78-3.67 (m, 1 H), 1.17 (bs, 6 H), 1.05 (d, J = 6.9 Hz, 6 H); <sup>13</sup>C NMR (75 MHz, DMSO-d<sub>6</sub>) δ 157.2, 154.2, 152.3, 145.6, 134.3, 126.8, 126.7, 121.5, 120.9, 115.8, 56.5, 54.2, 52.1, 26.4, 23.6; MS (APCI) m/z 371 (M + H)<sup>+</sup>; Anal.

Calcd for  $C_{19}H_{26}N_6O_2\cdot 0.3H_2O$ : %C, 60.72; %H, 7.13; %N, 22.36; Found: %C, 60.44; %H, 7.42; %N, 22.52.

# Example 143

N-[2-(4-Amino-2-hydroxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]cyclohexanecarboxamide

Part A

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1,2-Diamino-2-methylpropane (8.4 mL, 80.0 mmol) was added to a chilled (0 °C) solution of 4-chloro-3-nitro[1,5]naphthyridine (15.2 g, 72.7 mmol) and triethylamine (20.2 mL, 145 mmol) in dichloromethane (350 mL). The reaction mixture was stirred overnight and then concentrated under reduced pressure. The residue was combined with water (300 mL) and heated at reflux with stirring for 1 hour. The reaction mixture was cooled and filtered. The isolated solid was washed with water and then dried under high vacuum to provide 18.5 g of  $N^I$ -(3-nitro[1,5]naphthyridin-4-yl)-2-methylpropane-1,2-diamine as a bright yellow powder.

Part B

Under a nitrogen atmosphere, a solution of sodium hydroxide (3.12 g, 78.0 mmol) in water (50 mL) was added to a solution of the material from Part A (18.5 g, 70.9 mmol) in tetrahydrofuran (200 mL). A solution of di-tert-butyl dicarbonate (17.0 g, 78.0 mmol) in tetrahydrofuran (100 mL) was added dropwise over a period of 30 minutes. Two (2) days later additional di-tert-butyl dicarbonate (2.0 g) was added. The reaction mixture was stirred for another 8 hours and then concentrated under reduced pressure. The residue was dissolved in ethyl acetate (250 mL), washed sequentially with water (x2) and brine, dried over sodium sulfate, filtered, and then concentrated under reduced pressure. The residue was dissolved in warm 1/1 ethyl acetate/hexanes. The solution was allowed to slowly cool. The resulting precipitate was isolated by filtration and washed with hexanes to provide 17.7 g of tert-butyl N-{2-[(3-nitro[1,5]naphthyridin-4-yl)amino]-1,1-dimethylethyl}carbamate as a bright yellow crystalline solid.

### Part C

A Parr vessel was charged with a solution of *tert*-butyl *N*-{2-[(3-nitro[1,5]naphthyridin-4-yl)amino]-1,1-dimethylethyl}carbamate (12.62 g, 34.9 mmol) in acetonitrile (100 mL) and 5% Pt/C (2.00 g). The vessel was placed under hydrogen pressure (50 psi, 3.4 X 10<sup>5</sup> Pa) until hydrogen uptake ceased. The reaction mixture was filtered through a layer of CELITE filter aid and the filter cake was rinsed with acetonitrile. The filtrate was concentrated under reduced pressure to provide 11.07 g of *tert*-butyl *N*-{2-[(3-amino[1,5]naphthyridin-4-yl)amino]-1,1-dimethylethyl]carbamate as a bright yellow foam.

## 10 Part D

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Under a nitrogen atmosphere, a solution of the material from Part C (11.07 g, 33.4 mmol) in dichloromethane (330 mL) was cooled to 0 °C. Triethylamine (5.11 mL, 36.7 mmol) and ethoxyacetyl chloride (3.70 mL, 36.7 mmol) were added sequentially. The reaction mixture was stirred overnight while warming to ambient temperature and then concentrated under reduced pressure. The residue was dissolved in ethanol (300 mL). Triethylamine (16 mL) was added and the solution was heated at reflux under a nitrogen atmosphere over the weekend. The reaction mixture was allowed to cool to ambient temperature and then concentrated under reduced pressure. The residue was dissolved in dichloromethane (250 mL), washed sequentially with water and brine, dried over magnesium sulfate, filtered, and then concentrated under reduced pressure. The residue was purified by flash chromatography (6 x 12 cm silica gel column eluting with ethyl acetate) to provide 11.5 g of a purple foam. This material was purified by flash chromatography (eluting with 2.5 % methanol in chloroform) to provide 10.07 g of tert-butyl N-[2-(2-ethoxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]carbamate a purple foam.

### Part E

3-Chloroperoxybenzoic acid (7.50 g of 57-86%) was added to a solution of the material from Part D in dichloromethane (250 mL). After 2.5 hours, additional 3-chloroperoxybenzoic acid (250 mg) was added and the reaction mixture was stirred for 1.5 hours. The reaction mixture was washed sequentially with 1% sodium carbonate (4 x 75 mL), water, and brine, dried over sodium sulfate, filtered, and then concentrated under

reduced pressure to provide 10.32 g of tert-butyl N-[2-(2-ethoxymethyl-5N-oxide-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]carbamate as a purple foam. Part F

Concentrated ammonium hydroxide (20 mL) was added to a solution of the material from Part E (10.32 g, 24.9 mmol) in dichloromethane (200 mL). Toluenesulfonyl chloride (5.02 g, 26.3 mmol) was added in small portions over a period of 2 minutes. The reaction mixture was stirred for 2 hours and then diluted with water. The layers were separated. The organic layer was washed sequentially with 1% sodium carbonate (x3), water, and brine, dried over sodium sulfate, filtered, and then concentrated under reduced pressure. The residue was purified by flash chromatography (6 x 15 cm column of silica gel, eluting with 10% CMA in chloroform) to provide about 8 g of a purple foam. The foam was dissolved in ethanol, combined with activated charcoal (2 g), heated at reflux for 15 minutes, filtered, and then concentrated under reduced pressure to provide 7.59 g of tert-butyl N-[2-(4-amino-2-ethoxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]carbamate as a violet foam.

Part G

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A solution of hydrochloric acid in ethanol (17 mL of 4.3 M) was added to a solution of the material from Part F in ethanol (100 mL). The reaction mixture was heated at 90 °C for 2 hours, allowed to cool, and then concentrated under reduced pressure. The residue was dissolved in water (100 mL) and extracted with chloroform (2 x 25 mL). The extracts were discarded. The aqueous was made basic with concentrated ammonium hydroxide and then extracted with chloroform (4 x 50 mL). The combined extracts were dried over sodium sulfate, filtered, and then concentrated under reduced pressure. The residue was crystallized from ethyl acetate/hexanes (about 100 mL). The solid was isolated by filtration, rinsed with cold 20% ethyl acetate in hexanes, and dried under vacuum. A second crop was obtained and combined with the first crop to provide 3.82 g of 1-(2-amino-2-methylpropyl)-2-ethoxymethyl-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-4-amine as a gray crystalline solid.

Part H

Under a nitrogen atmosphere, a solution of 1-(2-amino-2-methylpropyl)-2-ethoxymethyl-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-4-amine (1.552 g, 4.94 mmol) in dichloromethane (50 mL) was cooled to 0 °C. Triethylamine (1.38 mL, 9.92 mmol) and

cyclohexylcarbonyl chloride (661 μL, 4.94 mmol) were added sequentially. Two (2) days later the reaction mixture was cooled and additional cyclohexylcarbonyl chloride (40 μL) was added. The reaction mixture was stirred overnight and then diluted with saturated sodium bicarbonate and dichloromethane (50 mL). The layers were separated. The organic layer was washed sequentially with water (x2) and brine, dried over sodium sulfate, filtered, and then concentrated under reduced pressure. The residue was purified by flash chromatography (4 x 13 cm silica gel column, eluting with 3% methanol in chloroform). The purified material was dissolved in refluxing propyl acetate (80 mL) with the aid of methanol, the methanol was boiled off, and the solution was allowed to slowly cool. The resulting precipitate was isolated by filtration, rinsed with cold propyl acetate, and dried under high vacuum at 70 °C to provide 1.37 g of *N*-[2-(4-amino-2-ethoxymethyl-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]cyclohexanecarboxamide as a colorless crystalline solid, mp 210-211 °C. Anal. calcd for C<sub>23</sub>H<sub>32</sub>N<sub>6</sub>O<sub>2</sub>: %C, 65.07; %H, 7.60; %N, 19.80; Found: %C, 64.93; %H, 7.76; %N, 19.97.

Part I

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Boron tribromide (1.24 mL of 1 M in dichloromethane) was added dropwise to a chilled (ice bath) suspension of N-[2-(4-amino-2-ethoxymethyl-1H-imidazo[4,5c][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]cyclohexanecarboxamide (500 mg, 1.18 mmol) in dichloromethane (15 mL). The reaction mixture was allowed to slowly warm to ambient temperature and then stirred over the weekend. Additional boron tribromide (1 mL) was added and the reaction mixture was stirred for 24 hours. The reaction was quenched with methanol (10 mL) and then concentrated under reduced pressure. The residue was combined with hydrochloric acid (15 mL of 6 M), heated to 50 °C, and stirred for 2 hours. The resulting solution was cooled to ambient temperature and then neutralized (pH 7) with 10% sodium hydroxide. The resulting gummy precipitate was extracted with chloroform (3 x 15 mL). The combined extracts were washed with brine (15 mL), dried over sodium sulfate, filtered, and then concentrated under reduced pressure to provide an off white solid. This material was purified by prep HPLC (HORIZON HPFC system, eluting with a gradient of 10-50% CMA in chloroform) to provide a white solid. The solid was triturated with hot acetonitrile, allowed to cool, isolated by filtration, and dried under vacuum to provide 233 mg of N-[2-(4-amino-2-hydroxymethyl-1H-

imidazo[4,5-c][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]cyclohexanecarboxamide as a fine white solid, mp 230-232 °C; <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ , 350 K)  $\delta$  8.53 (dd, J = 4.3, 1.6 Hz, 1 H), 7.95 (dd, J = 8.4, 1.5 Hz, 1 H), 7.87 (s, 1 H), 7.47 (dd, J = 8.4, 4.4 Hz, 1 H), 6.55 (s, 2 H), 5.31 (s, 1 H), 5.15 (s, 2 H), 4.79 (d, J = 5.4 Hz, 2 H), 1.90-1.80 (m, 1 H), 1.67-1.43 (m, 5 H), 1.31 (s, 6 H), 1.24-1.02 (m, 5 H); <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ )  $\delta$  175.9, 154.6, 152.8, 142.8, 140.8, 134.2, 133.5, 133.3, 129.3, 122.5, 56.4, 55.0, 52.3, 44.9, 29.4, 25.7, 25.6, 24.9; MS (ESI) m/z 397 (M + H)<sup>+</sup>; Anal. Calcd for C<sub>21</sub>H<sub>28</sub>N<sub>6</sub>O<sub>2</sub>: C, 63.62; H, 7.12; N, 21.20; Found: C, 63.77; H, 7.34; N, 21.50.

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# Example 144

N-[2-(4-Amino-2-hydroxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]methanesulfonamide

Part A

Under a nitrogen atmosphere, a solution of 1-(2-amino-2-methylpropyl)-2-ethoxymethyl-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-4-amine (1.588 g, 5.06 mmol) in dichloromethane (50 mL) was cooled to 0 °C. Triethylamine (1.41 mL, 10.12 mmol) and methanesulfonyl chloride (392 µL, 5.06 mmol) were added sequentially. The reaction mixture was allowed to slowly warm to ambient temperature overnight. Additional methanesulfonyl chloride (40 µL) was added and the reaction mixture was stirred at ambient temperature for an additional 5 hours. The reaction mixture was diluted with aqueous saturated sodium bicarbonate and the layers were separated. The organic layer was washed sequentially with water and brine, dried over sodium sulfate, filtered, and then concentrated under reduced pressure. The residue was purified by flash chromatography (4 x 15 cm silica gel column, eluting with a gradient of 5-7.5% methanol in chloroform). The purified material was dissolved in refluxing propyl acetate (80 mL) with the aid of methanol, the methanol was boiled off, and the solution was allowed to slowly cool. The resulting precipitate was isolated by filtration, rinsed with cold propyl acetate, and dried under high vacuum at 70 °C to provide 1.35 g of *N*-[2-(4-amino-2-ethoxoxymethyl-1*H*-

imidazo[4,5-c][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]methanesulfonamide as colorless needles, mp 209-210 °C. Anal. calcd for  $C_{17}H_{24}N_6O_3S$ : %C, 52.02; %H, 6.16; %N, 21.41; Found: %C, 52.09; %H, 6.35; %N, 21.60.

Part B

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Boron tribromide (1.34 mL of 1 M in dichloromethane) was added dropwise to a chilled (ice bath) suspension of N-[2-(4-amino-2-ethoxoxymethyl-1H-imidazo[4,5c][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]methanesulfonamide (500 mg, 1.27 mmol) in dichloromethane (15 mL). The reaction mixture was allowed to slowly warm to ambient temperature and then stirred over the weekend. Additional boron tribromide (1.5 mL) was added and the reaction mixture was stirred for 4 hours. Additional boron tribromide (1.5 mL) was added and the reaction mixture was stirred overnight. The reaction was quenched with methanol (15 mL) and then concentrated under reduced pressure. The residue was combined with hydrochloric acid (15 mL of 6 M), heated to 50 °C, and stirred for 2 hours. The resulting solution was cooled to ambient temperature and then neutralized (pH 7) with 10% sodium hydroxide. The resulting precipitate was isolated by filtration and rinsed with water to provide a white solid. This material was purified by prep HPLC (HORIZON HPFC system, eluting with a gradient of 10-50% CMA in chloroform) to provide a white solid. This material was recrystallized from acetonitrile and dried in a vacuum oven to provide 103 mg of N-[2-(4-amino-2-hydroxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)-1.1-dimethylethyl]methanesulfonamide as a white crystalline solid, mp 268-271 °C; <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  8.50 (dd, J = 4.4, 1.6 Hz, 1 H), 7.95 (dd, J = 8.4, 1.5 Hz, 1 H), 7.90 (s, 1 H), 7.48 (dd, J = 8.4, 4.4 Hz, 1 H), 6.91 (s, 2 H), 5.62 (t, J = 5.9 Hz, 1 H), 5.10 (bs, 2 H), 4.92 (s, 2 H), 2.87 (s, 3 H), 1.35 (s, 6 H);  ${}^{13}$ C NMR (75 MHz, DMSO- $d_6$ ) 8 154.2, 152.3, 142.3, 140.3, 133.4, 133.1, 132.9, 128.8, 122.1, 57.2, 56.4, 54.3, 44.1, 25.1; MS (APCI) m/z 365 (M + H)<sup>+</sup>; Anal. Calcd for C<sub>15</sub>H<sub>20</sub>N<sub>6</sub>O<sub>3</sub>S: C, 49.44; H, 5.53; N, 23.06; Found: C, 49.48; H, 5.40; N, 23.31.

## Example 145

 $N-\{4-\{4-Amino-2-(2-hydroxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]$ butyl $\}$ methanesulfonamide

### 5 Part A

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3-Methoxypropionyl chloride (2.7 g, 22 mmol) was added dropwise to a chilled (ice bath) solution of *tert*-butyl *N*-{4-[(3-amino[1,5]naphthyridin-4-yl)amino]butyl}carbamate 6.7 g, 20 mmol, U.S. Patent No.6,194,425, Example 42) in anhydrous pyridine (75 mL). The reaction mixture was heated at 120 °C overnight. The reaction was repeated on the same scale. The reaction mixtures were combined and concentrated under reduced pressure to provide 28 g of crude *tert*-butyl *N*-((4-{([3-(3-methoxypropionyl)amino[1,5]naphthyridin-4-yl]amino}butyl))carbamate as a red oil. Part B

The crude material from Part A was dissolved in pyridine (150 mL). Pyridine hydrochloride (2.1 g) was added and the reaction mixture was heated at reflux overnight. The reaction mixture was concentrated under reduced pressure. The residue was diluted with dichloromethane and washed with brine. The aqueous layer was extracted with dichloromethane (x4). The combined organics were concentrated under reduced pressure. The residue was purified by prep HPLC (COMBIFLASH system eluting with a gradient of 0-7% methanol in dichloromethane containing 1% ammonium hydroxide) to provide 9.72 g of *tert*-butyl *N*-{4-[2-(2-methoxyethyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-1-yl]buytl} carbamate as a brown glassy solid.

3-Chloroperoxybenzoic acid (7.8 g of 77%) was added in a single portion to a solution of *tert*-butyl *N*-{4-[2-(2-methoxyethyl)-1*H*-imidazo[4,5-c][1,5]naphthyridin-1-yl]buytl}carbamate (7 g) in dichloroethane (100 mL). The reaction mixture was stirred at ambient temperature for 3 hours. Concentrated ammonium hydroxide (100 mL) was

added and the reaction mixture was stirred until a suspension formed. *Para*toluenesulfonyl chloride (3.6 g) was added in a single portion. The reaction mixture was
stirred at ambient temperature for 2 hours and then diluted with dichloromethane and
brine. The organic layer was separated, washed with brine (x2), dried over magnesium
sulfate, filtered through a layer of CELITE filter aid, and then concentrated under reduced
pressure to provide 8.83 g of crude *tert*-butyl *N*-{4-[4-amino-2-(2-methoxyethyl)-1*H*imidazo[4,5-c][1,5]naphthyridin-1-yl]buytl}carbamate as a brown solid.
Part D

The material from Part C was diluted with a small amount of dichloromethane and then hydrochloric acid in dioxane (126 mL of 4 M) was slowly added. The reaction mixture was stirred at ambient temperature overnight and then concentrated under reduced pressure. The residue was purified by prep HPLC (COMBIFLASH system eluting with a gradient of 0-7% methanol in dichloromethane containing 1% ammonium hydroxide) to provide 8 g of crude 1-(4-aminobutyl)-2-(2-methoxyethyl)-1*H*-imidazo[4,5-c][1,5]naphthyridin-4-amine.

Part E

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Triethylamine (3.9 mL) was added to a solution of a portion (1.8 g) of the material from Part D in pyridine (20 mL). Methanesulfonyl chloride (485  $\mu$ L) was added dropwise. The reaction mixture was stirred at ambient temperature for 2 hours, quenched with water (25 mL), and the stirred overnight. The reaction mixture was concentrated under reduced pressure and then diluted with dichloromethane. The organic layer was washed with brine (x2) and then concentrated under reduced pressure. The residue was purified by prep HPLC (COMBIFLASH system eluting with a gradient of 0-5% methanol in dichloromethane containing 1% ammonium hydroxide for 5 minutes and then holding at 5%) to provide 400 mg of N-{4-{4-amino-2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]butyl}methanesulfonamide.

Boron tribromide (2.55 mL of 1 M in dichloromethane) was slowly added to a chilled mixture of the material from Part E in dichloromethane (10 mL). The reaction mixture was stirred at ambient temperature overnight and then concentrated under reduced pressure. The residue was dissolved in methanol, combined with hydrochloric acid (50 mL of 6 M), heated at 50 °C for 2 hours, and concentrated under reduced pressure. The

residue was combined with a solution of ammonia in methanol (about 50 mL of 7 M) and then concentrated again. This procedure was repeated 3 times. The residue from the final concentration was purified by prep HPLC (COMBIFLASH system eluting with a gradient of 0-10% methanol in dichloromethane containing 1% ammonium hydroxide for 10 minutes). The combined fractions were concentrated and then distributed onto solid phase extraction cartridges. The cartridges were eluted with ammonia in methanol (7 M). The resulting material was triturated with hot acetonitrile, cooled, isolated, and then dried in a vacuum oven to provide 111 mg of N-{4-{4-amino-2-(2-hydroxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]butyl}methanesulfonamide, mp 194-195 °C. Anal. calcd for  $C_{16}H_{22}N_6O_3S$ : %C, 50.78; %H, 5.86; %N, 22.21; Found: %C, 50.83; %H, 6.12; %N, 21.70.

## Example 146

N-[2-(4-Amino-2-hydroxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)ethyl]methanesulfonamide

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Part A

Methoxyacetyl chloride (5.9 g, 54 mmol) was added dropwise to a chilled (ice bath) solution of *tert*-butyl *N*-{2-[(3-amino[1,5]naphthyridin-4-yl)amino]ethyl} carbamate (15.0 g, 49.5 mmol, U.S. Patent No.6,194,425, Example 87) in anhydrous pyridine (100 mL). The reaction mixture was heated at reflux until analysis by liquid chromatography/mass spectroscopy (LCMS) indicated that the reaction was complete. The reaction mixture was concentrated under reduced pressure. The residue was diluted with ethanol (100 mL), combined with potassium carbonate solution (200 mL of 2 M), and heated at reflux for 4 hours. The reaction mixture was cooled and then concentrated under reduced pressure. The residue was partitioned between water and dichloromethane. The aqueous layer was extracted with dichloromethane. The combined organics were dried over magnesium sulfate, filtered, and then concentrated under reduced pressure to provide

14 g of tert-butyl N-[2-(2-methoxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)ethyl]carbamate.

## Part B

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Using the method of Example 145 Part C, the material from Part A was oxidized and then aminated to provide 17 g of crude *tert*-butyl *N*-[2-(4-amino-2-methoxymethyl-1*H*-imidazo[4,5-c][1,5]naphthyridin-1-yl)ethyl]carbamate as a sticky amber solid. Part C

The material from Part B was dissolved in a mixture of dichloromethane (20 mL) and methanol (5 mL). Hydrochloric acid in dioxane (28 mL of 4 M) was added. More dichloromethane was added to facilitate stirring. The reaction mixture was stirred at ambient temperature overnight and then concentrated under reduced pressure to provide crude 1-(2-aminoethyl)-2-methoxymethyl)-1*H*-imidazo[4,5-c][1,5]naphthyridin-4-amine as an orange solid.

### Part D

Triethylamine (35.6 mL) was added to a mixture of the material from Part C and pyridine (100 mL). The reaction mixture was cooled in an ice bath and then methanesulfonyl chloride (4.3 mL) was added dropwise. The reaction mixture was stirred at ambient temperature for 1 hour. Twice, more methanesulfonyl chloride (0.43 mL) was added and the reaction mixture was stirred at ambient temperature for 2 hours. The reaction mixture was partitioned between water and dichloromethane. The aqueous layer was extracted with dichloromethane (x2). The combined organics were dried over magnesium sulfate, filtered through a layer of CELITE filter aid, and then concentrated under reduced pressure to provide 14 g of N-[2-(4-amino-2-methoxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)ethyl]methanesulfonamide.

### Part E

Boron tribromide (71.4 mL of 1 M in dichloromethane) was slowly added to a chilled (ice bath) mixture of the material from Part D in dichloromethane (50 mL). The reaction mixture was stirred at ambient temperature for 2 hours. Additional boron tribromide (0.5 eq) was added and the reaction mixture was stirred at ambient temperature overnight. The reaction mixture was concentrated under reduced pressure. The residue was dissolved in methanol, combined with hydrochloric acid (50 mL of 6 M), heated at 50 °C for 2 hours, and concentrated under reduced pressure. The residue was combined with

a solution of ammonia in methanol (about 40 mL of 7 M) and then concentrated again. This procedure was repeated 3 times. The residue from the final concentration was purified by prep HPLC (COMBIFLASH system eluting with a gradient of 0-5% methanol in dichloromethane containing 1% ammonium hydroxide with a 10 minute ramp and a 20 minute hold, then with gradient of 6-10% methanol in dichloromethane containing 1% ammonium hydroxide with a 10 minute ramp and a 20 minute hold, and finally with gradient of 11-20% methanol in dichloromethane containing 1% ammonium hydroxide with a 10 minute ramp and a 20 minute hold) to provide 2.4 g of a brown solid. A small portion of this material was combined with hot acetonitrile containing a small amount of methanol, cooled, and then isolated by filtration. This procedure was carried out 3 times. After the final isolation the material was rinsed with ether and dried in a vacuum oven to provide 75 mg of *N*-[2-(4-amino-2-hydroxymethyl-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-1-yl)ethyl]methanesulfonamide as a beige solid, mp 239-242 °C. Anal. calcd for C<sub>13</sub>H<sub>16</sub>N<sub>6</sub>O<sub>3</sub>S: %C, 46.42; %H, 4.79; %N, 24.98; Found: %C, 46.35; %H, 4.70; %N, 24.70.

## Example 147

 $N-\{2-[4-Amino-2-(2-hydroxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]ethyl\}$  methanesulfonamide

20 Part A

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Using the general method of Example 146 Part A, tert-butyl N-{2-[(3-amino[1,5]naphthyridin-4-yl)amino]ethyl}carbamate (17.0 g, 56.1 mmol) was reacted with 3-methoxypropionyl chloride (7.5 g, 61.7 mmol) to provide 9.0 g of crude product. Analysis by LCMS indicated that the crude product was about a 1:1 mixture of tert-butyl N-{2-[2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]ethyl}carbamate and 1-(2-aminoethyl)-2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridine.

# Part B

Triethylamine (13.8 mL) was added to a mixture of the material from Part A and dichloromethane (70 mL). The resulting solution was chilled in an ice bath. Di-tert-butyl dicarbonate (8.6 g) was added. The reaction mixture was stirred at ambient temperature for 2 hours and then quenched with water. The layers were separated. The organic layer was washed with sodium carbonate, dried over magnesium sulfate, filtered through a layer of CELITE filter aid, and then concentrated under reduced pressure to provide 11 g of tert-butyl N-{2-[2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]ethyl} carbamate as a tan solid.

## 10 Part C

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3-Chloroperoxybenzoic acid (13.2 g of 77%) was added in a single portion to a solution of the material from Part B (11 g, 29.6 mmol) in dichloroethane (50 mL). The reaction mixture was stirred at ambient temperature for 1.5 hours, then diluted with dichloromethane and washed with aqueous ammonium hydroxide (25 mL of concentrated ammonium hydroxide in 250 mL of water). The aqueous layer was extracted with dichloromethane. The combined organics were concentrated under reduced pressure. The residue was dissolved in dichloroethane (100 mL). Concentrated ammonium hydroxide (70 mL) was added and the reaction mixture was stirred until a suspension formed. Para-Toluenesulfonyl chloride (6.2 g, 32.5 mmol) was added in a single portion. The reaction mixture was stirred at ambient temperature for 1.5 hours, then diluted with aqueous sodium bicarbonate and extracted with dichloromethane (x3). The combined organics were dried over magnesium sulfate, filtered through a layer of CELITE filter aid, and then concentrated under reduced pressure. The residue was purified by prep HPLC (COMBIFLASH system eluting with a gradient of 0-5% methanol in dichloromethane containing 1% ammonium hydroxide over 6 minutes and then holding at 5%) to provide 3.5 g of tert-butyl N-{2-[4-amino-2-(2-methoxyethyl)-1H-imidazo[4,5c][1,5]naphthyridin-1-yl]ethyl}carbamate as an orange solid. Part D

A solution of hydrochloric acid in dioxane (58 mL of 4 M) was added to a solution of of tert-butyl N-{2-[4-amino-2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]ethyl}carbamate (3 g) in a small amount of dichloromethane/methanol. The reaction mixture was stirred overnight at ambient temperature and then concentrated under reduced

pressure to provide 3.7 g of crude 1-(2-aminoethyl)-2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridine-4-amine hydrochloride.

Part E

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Using the general method of Example 146 Part D, a portion (1.1 g) of the material from Part D was reacted with methanesulfonyl chloride (322  $\mu$ L) to provide 1.0 g of N-{2-[4-amino-2-(2-methoxyethyl)-1*H*-imidazo[4,5-c][1,5]naphthyridin-1-yl]ethyl}methanesulfonamide as a red solid.

Part F

Boron tribromide (7 mL of 1 M in dichloromethane) was slowly added to a chilled (ice bath) mixture of the material from Part E in dichloromethane (25 mL). The reaction mixture was stirred at ambient temperature overnight and then concentrated under reduced pressure. The residue was dissolved in methanol, combined with hydrochloric acid (50 mL of 6 M), heated at 50 °C for 2 hours, and concentrated under reduced pressure. The residue was combined with a solution of ammonia in methanol (about 30 mL of 7 M) and then concentrated again. This procedure was repeated 3 times. The residue from the final concentration was purified by prep HPLC (COMBIFLASH system eluting with a gradient of 0-10% methanol in dichloromethane containing 1% ammonium hydroxide). The residue was combined with hot acetonitrile, cooled, and the acetonitrile was decanted off. This procedure was carried out 3 times. The material was isolated by filtration, rinsed with ether and dried in a vacuum oven to provide 950 mg of *N*-{2-[4-amino-2-(2-hydroxyethyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-1-yl]ethyl}methanesulfonamide, mp 136-138 °C. Anal. calcd for C<sub>14</sub>H<sub>18</sub>N<sub>6</sub>O<sub>3</sub>S: %C, 47.99; %H, 5.18; %N, 23.98; Found: %C, 47.69; %H, 5.36; %N, 23.77.

### Example 148

2-ol

25 1-(4-Amino-2-hydroxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)-2-methylpropan-

## Part A

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Under a nitrogen atmosphere, 1-amino-2-methylpropan-2-ol (25.5 g, 0.28 mol) was added over a period of 30 minutes to a solution of 4-chloro-3-nitro[1,5]naphthyridine (54.5 g, 0.26 mol) in dichloromethane (1 L). A water bath was used to control the exotherm and maintain the temperature of the reaction at or below 27 °C. The reaction mixture was stirred at ambient temperature overnight. The resulting precipitate (crop 1) was isolated by filtration. The filtrate was concentrated under reduced pressure to provide crop 2. The two crops were slurried separately with de-ionized water for 2 hours and then isolated by filtration. Crop 1: 40.53 g of 2-methyl-2-[(3-nitro[1,5]naphthyridin-4-yl)amino]propan-2-ol as a yellow solid. Crop 2: tan solid. Crop 2 was dissolved in dichloromethane and loaded onto an alumina column. The column was eluted first with 1% methanol in dichloromethane and then with acetone. The combined eluents were concentrated under reduced pressure. The residue was recrystallized from ethanol (10 mL/g) to provide 6.95 g of 2-methyl-1-[(3-nitro[1,5]naphthyridin-4-yl)amino]propan-2-ol.

Part B

A Parr vessel was charged with 2-methyl-1-[(3-nitro[1,5]naphthyridin-4-yl)amino]propan-2-ol (44.12 g, 0.17 mol), 5% Pt/C (4.4 g) and isopropyl alcohol (890 mL). The vessel was placed under hydrogen pressure (35 psi, 2.4 x 10<sup>5</sup> Pa) until hydrogen uptake ceased. The reaction mixture was filtered through a layer of filter aid. The filter cake was rinsed with additional isopropyl alcohol. The filtrate was concentrated under reduced pressure to provide 1-[(3-amino[1,5]naphthyridin-4-yl)amino]-2-methylpropan-2-ol as a thick oil.

Part C

Under a nitrogen atmosphere, ethoxyacetyl chloride (19.1 g, 0.156 mol) was added over a period of 12 minutes to a mixture of 1-[(3-amino[1,5]naphthyridin-4-yl)amino]-2-methylpropan-2-ol (28.95 g, 0.125 mol) in pyridine (300 mL). The reaction mixture was stirred at ambient temperature for 4 hours and then at reflux for 4 hours. The reaction mixture was allowed to cool to ambient temperature overnight and then concentrated under high vacuum. The residue was dissolved in 5% potassium carbonate (200 mL) and then extracted with dichloromethane (200 mL). The extract was filtered to remove some insoluble material, dried over magnesium sulfate, filtered, and then concentrated under high vacuum. The residue was dissolved in dichloromethane (150 mL) and eluted through

a short column of alumina. The eluent was concentrated under reduced pressure and air dried to provide 31.9 g of 1-(2-ethoxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)-2methylpropan-2-ol.

Part D

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A flask containing a solution of 1-(2-ethoxymethyl-1H-imidazo[4,5c][1,5]naphthyridin-1-yl)-2-methylpropan-2-ol (29.94 g, 83 mmol) in dichloromethane (300 mL) was covered with aluminum foil. 3-Chloroperoxybenzoic acid (28.65 g of 50%) was added in portions over a period of 50 minutes. The reaction mixture was stirred for an additional 40 minutes, then diluted with 5% aqueous potassium carbonate and stirred. The organic layer was separated, washed with brine (100 mL), dried over magnesium sulfate, filtered, and then concentrated under reduced pressure to provide a yellow paste. This material was combined with ether (100 mL) and stirred overnight. The resulting solid was isolated by filtration to provide 11.84 g of 1-(2-ethoxymethyl-5-oxo-1H-imidazo[4,5c][1,5]naphthyridin-1-yl)-2-methylpropan-2-ol. The aqueous potassium carbonate layer was partially concentrated, saturated with additional potassium carbonate, and then extracted with dichloromethane. The extract was dried over magnesium sulfate, filtered, and then concentrated under reduced pressure to provide 15.23 g of a dark oil. The oil was combined with ether (100 mL) and stirred overnight. The resulting solid was isolated by filtration to provide 11.51 g of 1-(2-ethoxymethyl-5-oxo-1H-imidazo[4,5c][1,5]naphthyridin-1-yl)-2-methylpropan-2-ol.

Part E

Concentrated ammonium hydroxide (241 mL) was added to a solution of 1-(2ethoxymethyl-5-oxo-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-1-yl)-2-methylpropan-2-ol (23.35 g, 74 mmol) in dichloromethane (300 mL). A solution of para-toluenesulfonyl chloride (15.52 g, 81 mmol) in dichloromethane (50 mL) was added with rapid stirring over a period of 25 minutes. The reaction mixture was stirred overnight. Concentrated ammonium hydroxide (25 mL) and a solution of para-toluenesulfonyl chloride (2 g) in dichloromethane (10 mL) was added and the reaction mixture was stirred for 5 hours. The organic phase was separated, washed with a solution of potassium carbonate (16 g) in water (300 mL), dried over magnesium sulfate, filtered, and concentrated under reduced pressure to provide 30.17 g of crude product. This material was combined with acetonitrile (300 mL), stirred, heated to reflux, and then allowed to cool with stirring to

ambient temperature. The resulting solid was isolated by filtration and then dried at 75 °C under vacuum to provide 14.4 g of a solid. This material was recrystallized from ethyl acetate (17.5 mL/g), isolated by filtration, and then dried under vacuum at 75 °C for 22 hours to provide 12.29 g of 1-(4-amino-2-ethoxymethyl-1*H*-imidazo[4,5-

c][1,5]naphthyridin-1-yl)-2-methylpropan-2-ol as an off white solid, mp 157-159 °C. Anal. calcd for  $C_{16}H_{21}N_5O_2$ : %C, 60.94; %H, 6.71; %N, 22.21; Found: %C, 61.06; %H, 6.67; %N, 22.37.

Part F

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A solution of boron tribromide in dichloromethane (11.8 mL of 1 M) was added to a chilled (0 °C) suspension of 1-(4-amino-2-ethoxymethyl-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-1-yl)-2-methylpropan-2-ol hydrobromide (1.24 g, 3.93 mmol) in dichloromethane (30 mL). The reaction mixture was allowed to come to ambient temperature with stirring for 16 hours. Methanol (15 mL) and hydrochloric acid (10 mL of 6 N) were added and the reaction mixture was heated at reflux for 2.5 hours. The reaction mixture was made basic with sodium hydroxide and the layers were separated. The aqueous layer was extracted with ethyl acetate (3 x 100 mL). The combined extracts were washed sequentially with water and brine, dried over magnesium sulfate, and then concentrated under reduced pressure to provide a white solid. This material was crystallized from ethyl acetate and then dried under vacuum at 95 °C for 16 hours to provide 0.55 g of 1-(4-amino-2-hydroxymethyl-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-1-yl)-2-methylpropan-2-ol as a white powder, mp 235-237 °C. Anal. calcd for C<sub>14</sub>H<sub>17</sub>N<sub>5</sub>O<sub>2</sub>: %C, 58.52; %H, 5,96; %N, 24.37; Found: %C, 58.40; %H, 5.82; %N, 24.45.

### Example 149

25 l-[4-Amino-2-(2-hydroxyethyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-1-yl]-2-methylpropan-2-ol

#### Part A

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A mixture of triethyl orthoformate (10 mL, 60.1 mmol) and 2,2-dimethyl-[1,3]-dioxane-4,6-dione (40.9 g, 0.23 mol) (Meldrum's acid) was heated at 92 °C for 90 minutes and then cooled to 70 °C over one hour. 3-Amino-5-bromopyridine (40.9 g, 0.20 mol) was slowly added over 10 minutes with an ethanol rinse while maintaining the reaction temperature between 60 and 70 °C. The reaction was then heated for an additional 20 minutes and allowed to cool to room temperature. The reaction mixture was filtered and washed with ethanol (150 mL) yielding a tan solid. The solid was dried under vacuum for 2 hours to yield 59.14 g of 5-{[(5-bromopyridin-3-yl)imino]methyl}-2,2-dimethyl-1,3-dioxane-4,6-dione as a light yellow crystalline solid, mp 200-202 °C.

#### Part B

5-{[(5-Bromopyridin-3-yl)imino]methyl}-2,2-dimethyl-1,3-dioxane-4,6-dione (59 g, 0.18 mol) was slowly added to DOWTHERM A heat transfer fluid (2000 mL) over a period of 5 minutes at 235-238 °C. Following addition, the reaction was maintained for an additional 5 minutes and then allowed to cool to 40 °C. A brown precipitate formed, which was filtered and washed with hexanes (150 mL). The brown solid was suspended in an ethanol/water mixture (90:10, 1500 mL), heated to a boil for 30 minutes, isolated by filtration, and washed with ethanol (200 mL) to yield 30.8 g of 7-bromo[1,5]naphthyridin-4-ol as a dark brown powder.

#### Part C

A mixture of 7-bromo[1,5]naphthyridin-4-ol (33 g, 0.147 mol) and furning nitric acid (350 mL) was heated at reflux (90 °C internal reaction vessel temperature) for 3 hours. The reaction mixture was cooled to 50 °C, poured over 1 L of ice and neutralized to pH 2-3 with a solution of 50% aqueous sodium hydroxide. The resulting precipitate was filtered, washed with water, and dried over vacuum for 3 days to yield 25.1 g of 7-bromo-3-nitro[1,5]naphthyridin-4-ol as a yellow crystalline solid.

#### Part D

Phosphorous oxychloride (16.76 g, 10.19 mL, 109.3 mmol) was added slowly dropwise to a suspension of 7-bromo-3-nitro[1,5]naphthyridin-4-ol (21.09 g, 78.1 mmol) in *N,N*-dimethylformamide (250 mL) (DMF) at ambient temperature and maintained overnight. The reaction mixture was then added to ice water (400 mL) with stirring. A solid precipitate formed, which was isolated by vacuum filtration and washed with water.

The material was dried under high vacuum at ambient temperature overnight to yield 20.79 g of 7-bromo-4-chloro-3-nitro[1,5]naphthyridine as a tan solid.

Part E

Triethylamine (35.95 mL, 257.9 mmol) was added to a suspension of 7-bromo-4-chloro-3-nitro[1,5]naphthyridine (49.6 g, 172 mmol) in dichloromethane (500 mL). 1-Amino-2-methylpropan-2-ol (16.86 g, 189 mmol) was added dropwise. The reaction mixture was stirred at ambient temperature for 16 hours and then concentrated under reduced pressure. The residue was triturated with water and stirred for 1 hour. The precipitated solid was isolated by filtration, washed with water, and dried. This material was suspended in diethyl ether (400 mL), sonicated, isolated by filtration, and then dried in a vacuum oven at 40 °C for 16 hours to provide 58.1 g of 1-[(7-bromo-3-nitro[1,5]naphthyridin-4-yl)amino]-2-methylpropan-2-ol as a yellow solid, mp 189-190 °C.

Part F

Part H

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A Parr vessel was charged with 5% Pt/C (5.8 g) and a suspension of 1-[(7-bromo-3-nitro[1,5]naphthyridin-4-yl)amino]-2-methylpropan-2-ol (58.00 g) in acetonitrile (800 mL) and methanol (400 mL). The vessel was placed under hydrogen pressure (30 psi, 2.1 X 10<sup>5</sup> Pa) for 8 hours. The reaction mixture was filtered through a layer of CELITE filter aid. The filtrate was concentrated under reduced pressure to provide 52.70 g of 1-[(3-amino-7-bromo[1,5]naphthyridin-4-yl)amino]-2-methylpropan-2-ol as a yellow foam. Part G

3-Methoxypropionyl chloride (24.90 g, 203 mmol) was added over a period of 5 minutes to a mixture of 1-[(3-amino-7-bromo[1,5]naphthyridin-4-yl)amino]-2-methylpropan-2-ol (52.70 g, 169 mmol), chloroform (100 mL), and acetonitrile (530 mL). The reaction mixture was stirred at ambient temperature overnight. The precipitated solid was isolated by filtration, washed well with acetonitrile, and then dried to provide 60.10 g of *N*-{7-bromo-4-[(2-hydroxy-2-methylpropyl)amino][1,5]naphthyridin-3-yl}-3-methoxypropionamide hydrochloride as a brown solid, mp 206-208 °C.

A mixture of N-{7-bromo-4-[(2-hydroxy-2-methylpropyl)amino][1,5]naphthyridin-3-yl}-3-methoxypropionamide hydrochloride (60.00 g, 138 mmol), potassium carbonate (60 g), water (300 mL), and ethanol (900 mL) was heated at reflux for 16 hours and then

concentrated under reduced pressure. The precipitated solid was isolated by filtration, washed sequentially with water and methanol, and dried to provide a brown solid. This material was dissolved in a 3/1 mixture of chloroform/methanol and decolorized with activated charcoal to provide 38.5 g of 1-[7-bromo-2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]-2-methylpropan-2-ol as a white solid, mp 125 °C. Anal. calcd for  $C_{16}H_{19}BrN_4O_2$ : %C, 50.67; %H, 5.05; %N, 14.77; Found: %C, 50.86; %H 4.94; %N, 15.01.

Part I

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3-Chloroperoxybenzoic acid (34.77 g of 75%, 151 mmol) was added to a solution of 1-[7-bromo-2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]-2methylpropan-2-ol (38.2 g, 101 mmol) in dichloromethane (450 mL) and the reaction mixture was stirred for 3 hours. The reaction mixture was diluted with dichloromethane (200 mL), washed sequentially with 4% aqueous sodium carbonate (2 x 150 mL) and brine (1 x 150 mL), and concentrated under reduced pressure to provide the N-oxide derivative. The N-oxide derivative was combined with dichloromethane (450 mL) and concentrated ammonium hydroxide (200 mL) and the mixture was cooled in an ice bath. Para-Toluenesulfonyl chloride (24 g) was added in portions. After the addition was complete the ice bath was removed and the reaction mixture was stirred at ambient temperature for 16 hours. The reaction mixture was diluted with dichloromethane (200 mL). Suspended solids were isolated by filtration, washed with water, and dried to provide 7.60 g of 1-[4amino-7-bromo-2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]-2methylpropan-2-ol as an off white solid, mp 210-211 °C. Anal. calcd for C<sub>16</sub>H<sub>20</sub>BrN<sub>5</sub>O<sub>2</sub>: %C, 48.74; %H, 5.11; %N, 17.76; Found: %C, 48.63; %H, 5.10; %N, 17.80. Part J

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A Parr vessel was charged with 10 % Pd/C (0.6 g) and a suspension of 1-[4-amino-7-bromo-2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]-2-methylpropan-2-ol (4.0 g) in acetonitrile (150 mL) and methanol (50 mL). The vessel was placed under hydrogen pressure (50 psi, 3.4 X  $10^5$  Pa) for 3 hours. The reaction mixture was diluted with 1/1 chloroform/methanol (100 mL), filtered through a layer of CELITE filter aid, and concentrated under reduced pressure. The residue was triturated with acetonitrile to provide 3.55 g of 1-[4-amino-2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]-2-methylpropan-2-ol hydrobromide as a white powder, mp 234-235 °C. Anal. calcd

for C<sub>16</sub>H<sub>22</sub>BrN<sub>5</sub>O<sub>2</sub>: %C, 48.49; %H, 5.60; %N, 17.67; Found: %C, 48.64; %H, 5.69; %N, 17.62.

Part K

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A solution of boron tribromide in dichloromethane (22.71 mL of 1 M) was added to a chilled (0 °C) suspension of 1-[4-amino-2-(2-methoxyethyl)-1H-imidazo[4,5c][1,5]naphthyridin-1-yl]-2-methylpropan-2-ol hydrobromide (3.00 g, 7.57 mmol) in dichloromethane (100 mL). The reaction mixture was allowed to come to ambient temperature with stirring for 16 hours. Methanol (30 mL) and hydrochloric acid (30 mL of 6 N) were added and the reaction mixture was heated at reflux for 2.5 hours. The reaction mixture was made basic with sodium hydroxide and the layers were separated. The aqueous layer was extracted with dichloromethane (100 mL). The extract was washed sequentially with water and brine, dried over magnesium sulfate, and then concentrated under reduced pressure to provide a pink solid. This material was crystallized from acetonitrile to provide 0.68 g of 1-[4-amino-2-(2-hydroxyethyl)-1Himidazo[4,5-c][1,5]naphthyridin-1-yl]-2-methylpropan-2-ol. The aqueous layer was combined with the water and brine washings and allowed to stand overnight. A precipitate was isolated by filtration, washed with water, and dried under vacuum at 95 °C for 3 hours to provide 1.16 g of 1-[4-amino-2-(2-hydroxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl]-2-methylpropan-2-ol as a pink crystalline solid, mp 194-195 °C. Anal. calcd for  $C_{15}H_{19}N_5O_2$ : %C, 59.79; %H, 6.36; %N, 23.24; Found: %C, 59.51; %H, 6.59; %N, 23.34.

## Examples 150 - 155

Preparation of N-[2-(4-amino-2-ethoxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]-2-methylpropionamide

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Triethylamine (556  $\mu$ L, 4.00 mmol) and isobutyryl chloride (230  $\mu$ L, 2.20 mmol) were added sequentially to a chilled (0 °C) solution of 1-(2-amino-2-methylpropyl)-2-ethoxymethyl-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-4-amine (628 mg, 2.00 mmol) in

dichloromethane (20 mL). The reaction mixture was allowed to warm slowly to ambient temperature overnight. The reaction mixture was quenched with aqueous saturated sodium bicarbonate and diluted with dichloromethane (50 mL). The organic layer was separated, washed sequentially with water and brine, dried over sodium sulfate, filtered, and then concentrated under reduced pressure to an amber foam. This material was dissolved in hot propyl acetate (10 mL) and then allowed to cool overnight. Hexanes were added and the now cloudy solution was heated until clear and then allowed stand until crystals formed. The solvent was removed by pipette. The crystals were rinsed with cold propyl acetate/hexanes and then dried under high vacuum at 70 °C to provide 464 mg of *N*-[2-(4-amino-2-ethoxymethyl-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]-2-methylpropionamide as an off white crystalline solid, mp 154.5-155.5 °C. Anal. calcd for C<sub>20</sub>H<sub>28</sub>N<sub>6</sub>O<sub>2</sub>: %C, 62.48; %H, 7.34; %N, 21.86; Found: %C, 62.14; %H, 7.62; %N, 21.71.

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Preparation of 1-[2-(4-amino-2-ethoxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]-3-(1-methylethyl)urea

Under a nitrogen atmosphere, isopropyl isocyanate (206 μL, 2.10 mmol) was added to a chilled (0 °C) solution of 1-(2-amino-2-methylpropyl)-2-ethoxymethyl-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-4-amine (628 mg, 2.00 mmol) in dichloromethane (20 mL). The reaction mixture was allowed to warm slowly to ambient temperature overnight. The resulting precipitate was isolated by filtration and then dried under vacuum at 70 °C to provide 669 mg of 1-[2-(4-amino-2-ethoxymethyl-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-1-yl)-1,1-dimethylethyl]-3-(1-methylethyl)urea as white powder, mp 172.5-173.5 °C. Anal. calcd for C<sub>20</sub>H<sub>29</sub>N<sub>7</sub>O<sub>2</sub>: %C, 60.13; %H, 7.32; %N, 24.54; Found: %C, 59.88; %H, 7.55; %N, 24.51.

A solution of boron tribromide in dichloromethane (about 4 eq of 1 M) was added to a tube containing a chilled (0 °C) solution of a compound of Formula Xb (25 mg, 1 eq) in dichloromethane (1 mL). The tube was vortexed, maintained at 0 °C for 0.5 hour, and then shaken overnight at ambient temperature. The reaction mixture was diluted with methanol (1 mL) and hydrochloric acid (500 µL of 6 N), vortexed, and then the solvents were removed by vacuum centrifugation. The compounds were purified by preparative high performance liquid chromatography (prep HPLC) using a Waters FractionLynx automated purification system. The prep HPLC fractions were analyzed using a Waters LC/TOF-MS, and the appropriate fractions were centrifuge evaporated to provide the trifluoroacetate salt of the desired compound. Reversed phase preparative liquid chromatography was performed with non-linear gradient elution from 5-95% B where A is 0.05% trifluoroacetic acid/water and B is 0.05% trifluoroacetic acid/acetonitrile. Fractions were collected by mass-selective triggering. Table 4 shows the structure of the starting material, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

Table 4				
NH <sub>2</sub> N OH N N OH N R <sub>1</sub>				
Xb	<u> </u>	la		
Example	R <sub>1</sub>	Measured Mass (M+H)		
150	CH <sub>3</sub> OH	288.1440		
151	CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> O	365.1378		
152	CH <sub>3</sub> N N H <sub>3</sub> C H	397.2348		

153	CH <sub>3</sub> CH <sub>3</sub> NH <sub>2</sub>	287.1607
154	CH <sub>3</sub> CH <sub>3</sub> N CH <sub>3</sub>	357.2055
155	CH <sub>3</sub> C CH <sub>3</sub> C CH <sub>3</sub>	372.2157

Examples 156 – 161

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A solution of boron tribromide in heptane (400  $\mu$ L of 1 M) was added to a tube containing a chilled (0 °C) solution of a compound of Formula Xc (about 25 mg) in dichloromethane (1 mL). The tube was vortexed, maintained at 0 °C for 0.5 hour, and then shaken overnight at ambient temperature. The reaction mixture was diluted with methanol (1 mL) and hydrochloric acid (250  $\mu$ L of 6 N), vortexed, and then the solvents were removed by vacuum centrifugation. The compounds were purified by preparative high performance liquid chromatography (prep HPLC) using a Waters FractionLynx automated purification system. The prep HPLC fractions were analyzed using a Waters LC/TOF-MS, and the appropriate fractions were centrifuge evaporated to provide the trifluoroacetate salt of the desired compound. Reversed phase preparative liquid chromatography was performed with non-linear gradient elution from 5-95% B where A is 0.05% trifluoroacetic acid/water and B is 0.05% trifluoroacetic acid/acetonitrile. Fractions were collected by mass-selective triggering. Table 5 shows the structure of the starting material, a reference for the starting material, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

Table 5

156	U.S. Patent Publication 2004/0147543 Example 206	N N		430.2227
157	U.S. Patent Publication 2004/0147543 Example 136	CH <sub>3</sub> CCH <sub>3</sub> OH		377.1985
158	U.S. Patent Publication 2004/0147543 Example 145	CH₃ CH₃	N N	362.2008
159	U.S. Patent Publication 2004/0147543 Example 146	CH <sub>3</sub>	но	392.2104
160	U.S. Patent Publication 2004/0147543 Example 183	N N		431.2209
161	U.S. Patent Publication 2004/0147543 Example 184	N O	N	431.2220

Examples 162 - 186

## Part A

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1-(4-Amino-7-bromo-2-ethoxymethyl-1H-imidazo[4,5-c]quinolin-1-yl)-2-methylpropan-2-ol (2 g, U.S. Patent Publication 2004/0147543 Example 125) was dissolved in 7:3 volume:volume chloroform:methanol (100 mL). Aliquots (2 mL, 1.0 eq.) were added to test tubes and the solvent was removed by vacuum centrifugation. A tube was charged with a boronic acid (1.1 eq) from the table below. n-Propanol (1.6 mL) was added to each tube, the tube was purged with nitrogen, and then sonicated until the contents were well mixed. Each tube was then charged sequentially with 150  $\mu$ L of a solution of palladium (II) acetate in toluene (60 mg of palladium (II) acetate dissolved in 15 mL of toluene), 600  $\mu$ L of 2 M aqueous sodium carbonate solution, 113  $\mu$ L of water, and 53  $\mu$ L of a 15 mole % solution of triphenylphosphine in n-propanol. The tubes were purged with nitrogen and then heated at 80 °C overnight.

The reaction mixtures were purified by solid phase extraction. Sufficient hydrochloric acid (1 N) was added to each reaction mixture to adjust the pH to <5. Each reaction mixture was loaded onto a cartridge (Waters Oasis Samples Extraction Cartridges MCX 6cc). Methanol (5 mL) was added to each cartridge. The cartridge was placed in a clean test tube. The cartridge was eluted with two successive 5 mL portions of 1 N ammonia in methanol. The solvent was removed by vacuum centrifugation.

Part B

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Dichloromethane (1 mL) was added to each tube, the tube was sonicated to dissolve the solids, and then the tube was chilled to 0 °C in an ice bath. A solution of boron tribromide in heptane (600  $\mu$ L of 1 M) was added to each tube. The tube was vortexed, maintained at 0 °C for 0.5 hour, and then shaken overnight at ambient temperature. The solvents were removed by vacuum centrifugation. Methanol (1 mL) and hydrochloric acid (1 mL of 6 N) were added to each tube, the tubes were vortexed, and then the solvents were removed by vacuum centrifugation. The compounds were purified as described above for Examples 156 – 161. Table 6 shows the boronic acid, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

Table 6					
	NH <sub>2</sub> N OH N CH <sub>3</sub> CCH <sub>3</sub>				
Example Reagent R <sub>3</sub> Measured Mass (M+H)					
162	Phenylboronic acid		363.1847		
163	163 Pyridine-3-boronic acid 364.1779				
164	3-Methylphenylboronic acid	CH <sub>3</sub>	377.2001		

165	4-Methylphenylboronic acid	H³C	377.1979
166	o-Tolylboronic acid	CH₃	377.1990
167	(2-Hydroxyphenyl)boronic acid	ОН	379.1776
168	3-Hydroxyphenylboronic acid	ОН	379.1755
169	3,5-Dimethylphenylboronic acid	H <sub>3</sub> C CH <sub>3</sub>	391.2130
170	4-(Hydroxymethyl)phenylboronic acid	ОН	393.1935
171	3-Chlorophenylboronic acid	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	397.1432
172	2-Chlorophenylboronic acid	CI	397.1447
173	4-Chlorophenylboronic acid	cı	397.1431
174	2,4-Difluorophenylboronic acid	F	399.1642
175	Benzo[b]furan-2-boronic acid		403.1812
176	(3-Aminocarbonylphenyl)boronic acid	O NH <sub>2</sub>	406.1889
177	4-(N,N-Dimethylamino)phenylboronic acid	H <sub>3</sub> C <sub>·N</sub> ĊH <sub>3</sub>	406.2255

178	(3-Aminomethylphenyl)boronic acid hydrochloride	H <sub>2</sub> N	392.2108
179	3,4-Dichlorophenylboronic acid	CI	431.1061
180	4-(Ethylsulfonyl)phenylboronic acid	H₃C O∵S	455.1771
181	3- (Methylsulfonylamino)phenylboronic acid	O;s NH H₃C Ö	456.1727
182	3-(Pyrrolidine-1- carbonyl)phenylboronic acid	CNO	460.2364
183	4-( Pyrrolidine-1- carbonyl)phenylboronic acid		460.2395
184	3-(Butylaminocarbonyl)phenylboronic acid	H <sub>3</sub> C	462.2488
185	3- (Isobutylaminocarbonyl)phenylboroni acid	C HNO CH <sub>3</sub>	462.2527

4'-(4,4,5,5-Tetramethyl-1,3,2-dioxaborolan-2-yl)acetanilinde

4'20.2022

Example 187

 $1-[4-Amino-2-hydroxymethyl-7-(thiazol-4ylmethoxy)-1\\ \\ H-imidazo[4,5-c] \\ quinolin-1-yl]-2-methylpropan-2-ol$ 

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Under a nitrogen atmosphere, a solution of 1-[4-amino-2-ethoxymethyl-7-(thiazol-4ylmethoxy)-1*H*-imidazo[4,5-*c*]quinolin-1-yl]-2-methylpropan-2-ol (400 mg, 0.94 mmol, which can be prepared as described in International Application No. PCT/US04/28021 Example 137) in dichloromethane (50 mL) was cooled to 0 °C in an ice bath. A solution of boron tribromide in dichloromethane (3.76 mL of 1.0 M) was added slowly. The reaction mixture was allowed to slowly warm to ambient temperature overnight. The reaction mixture was diluted with hydrochloric acid (20 mL of 6 N) and stirred for 30 minutes. The layers were separated. The organic layer was washed with 6 N hydrochloric acid (3 x 20 mL) and then discarded. The aqueous layer was made basic by the addition of solid potassium carbonate. A precipitate was isolated by filtration, dissolved in hot chloroform, and then purified by prep HPLC (HORIZON HPFC system eluting with 0 -10% CMA in chloroform over 192 mL and then with 10-40% CMA in chloroform over 1400 mL) to provide a solid. This material was crystallized from acetonitrile to provide 181 mg of 1-[4-amino-2-hydroxymethyl-7-(thiazol-4ylmethoxy)-1H-imidazo[4,5c]quinolin-1-yl]-2-methylpropan-2-ol as a white solid, mp 260-262 °C. Anal. calcd for C<sub>19</sub>H<sub>21</sub>N<sub>5</sub>O<sub>3</sub>S: %C, 56.81; %H, 5.41; %N, 17.26; Found: %C, 56.82; %H, 5.54; %N, 17.23.

#### Example 188

[4-Amino-7-pyridin-3-yl-1-(tetrahydro-2H-pyran-4-ylmethyl)-1H-imidazo[4,5-c]quinolin-2-yl]methanol

Part A

To a mixture of 1-tetrahydro-2*H*-pyran-4-ylmethanamine HCl (19 g, 120 mmol), dichloromethane (626 mL), and triethyl amine (43.7 mL, 313 mmol) was added 4-chloro-3-nitroquinoline at 0 °C. The resulting bright yellow solution was stirred at ambient temperature for 18 hours. The reaction was then concentrated under reduced pressure. The resulting solid was stirred in water (100 mL) and filtered to give 43 g of 7-bromo-3-nitro-*N*-(tetrahydro-2*H*-pyran-4-ylmethyl)quinolin-4-amine as a yellow powder.

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7-Bromo-3-nitro-N-(tetrahydro-2H-pyran-4-ylmethyl)quinolin-4-amine (20 g, 55 mmol) was dissolved in a mixture of acetonitrile (500 mL) and isopropyl alcohol (50 mL) and the solution was placed in a pressure bottle. Platinum on carbon (5%, 2 g) was then added and the reaction mixture was shaken under  $H_2$  at 48 PSI (3.3 x  $10^5$  Pa). After 2 hours, the reaction mixture was filtered through a pad of CELITE filter agent. The pad was rinsed with acetonitrile and the combined filtrates were concentrated under reduced pressure to give 7-bromo- $N^4$ -(tetrahydro-2H-pyran-4-ylmethyl)quinoline-3,4-diamine which was carried forward without further purification assuming quantitative yield. Part C

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Chloroacetyl chloride (5.2 mL, 65 mmol) was added to 7-bromo-N<sup>4</sup>-(tetrahydro-2*H*-pyran-4-ylmethyl)quinoline-3,4-diamine (55 mmol) dissolved in 273 mL of dichloromethane at 0 °C. A solid formed after adding half of the chloroacetyl chloride at which point additional dichloromethane (100 mL) was added. The reaction was stirred for 1 hour at ambient temperature. The yellow suspension was quenched first with aqueous saturated sodium bicarbonate followed by 50% aqueous sodium hydroxide until a pH of 14 was reached. Filtration provided 10 g of N-{7-bromo-4-[(tetrahydro-2*H*-pyran-4-ylmethyl)amino]quinolin-3-yl}-2-chloroacetamide as a tan solid. The filtrate was placed in a separatory funnel and the layers were separated. The aqueous layer was extracted with additional dichloromethane. The combined organic extracts were combined, dried over

sodium sulfate, filtered, and concentrated under reduced pressure to afford additional *N*-{7-bromo-4-[(tetrahydro-2*H*-pyran-4-ylmethyl)amino]quinolin-3-yl}-2-chloroacetamide as a yellow oil. The yellow oil was carried forward without further purification assuming a 50% yield (27.3 mmol). The oil was combined with ethanol (100 mL) and triethylamine (7.5 mL, 54 mmol). The resulting yellow solution was refluxed for 2 hours. The reaction was cooled to ambient temperature and the solvent was removed under reduced pressure to provide 7-bromo-2-(chloromethyl)-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinoline as a brown oil that was used without further purification assuming quantitative yield.

#### 10 Part D

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Potassium acetate (5.3 g, 55 mmol) was added to 7-bromo-2-(chloromethyl)-1- (tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinoline (27.3 mmol) dissolved in dimethylformamide (100 mL). The resulting suspension was stirred at 90 °C for 1 hour. The reaction was cooled to ambient temperature and water (200 mL) was added. The aqueous layer was extracted with chloroform. The combined organic extracts were combined, dried over sodium sulfate, filtered, and concentrated under reduced pressure to afford an orange oily solid. Chromatography (SiO<sub>2</sub>, 0-30% 80/18/2 v/v/v CHCl<sub>3</sub>/CH<sub>3</sub>OH/concentrated NH<sub>4</sub>OH (CMA)/CHCl<sub>3</sub>) gave material that was stirred in acetonitrile and filtered to provide 2.3 g of [7-bromo-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinolin-2-yl]methyl acetate as a tan solid.

3-Chloroperoxybenozic acid (2.4 g, 50% pure, 7.0 mmol) was added to a mixture of [7-bromo-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinolin-2-yl]methyl acetate (2.3 g, 5.4 mmol) and chloroform (27 mL) at ambient temperature. The reaction was stirred at this temperature for 18 hours. Saturated aqueous sodium bicarbonate (50 mL) and water (50 mL) were then added to the reaction and the layers were separated. The aqueous layer was extracted with additional dichloromethane. The organic layers were combined, dried over sodium sulfate, and concentrated under reduced pressure to a dark oil. This oil was dissolved in methanol (27 mL) and to this solution was added 15 M ammonium hydroxide (3.6 mL, 54 mmol) and benzene sulfonyl chloride (2.9 mL, 23 mmol). The resulting reaction mixture was stirred at ambient temperature for 2 hours before adding additional 15 M ammonium hydroxide (3.6 mL, 54 mmol) and benzene

sulfonyl chloride (2.9 mL, 23 mmol). The reaction was stirred 18 hours. The reaction was then concentrated under reduced pressure and diluted with saturated aqueous sodium bicarbonate and chloroform. A suspension resulted that was filtered to afford a solid that was stirred with saturated aqueous sodium bicarbonate and filtered to give 1.1 g of [4-amino-7-bromo-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinolin-2-yl]methanol as a white solid.

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Part F

To a mixture of [4-amino-7-bromo-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinolin-2-yl]methanol (500 mg, 1.28 mmol), 3-pyridyl boronic acid (233 mg, 1.90 mmol), potassium carbonate (579 mg, 4.20 mmol), dimethoxyethane (5 mL), and water (2.5 mL) under a nitrogen atmosphere was added Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> (18 mg, 0.026 mmol). The resulting suspension was refluxed for 2 hours. The reaction was cooled to ambient temperature. The reaction mixture was diluted with chloroform and placed directly onto a silica gel column. Chromatography (SiO<sub>2</sub>, 0-40% CMA/CHCl<sub>3</sub>) gave material that was stirred in methanol and filtered to provide 263 mg of [4-amino-7-pyridin-3-yl-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinolin-2-yl]methanol as tan crystals, m.p. 260-262 °C. MS (APCI) *m/z* 500.3 (M + H)<sup>+</sup>; Anal. calcd for C<sub>22</sub>H<sub>23</sub>N<sub>5</sub>O<sub>2</sub>: C, 67.85; H, 5.95; N, 17.98. Found: C, 67.49; H, 5.87; N, 17.83.

Examples 189 – 207

The compounds in the table below were prepared according to the following general procedure. The ether analog was dissolved or suspended in a solvent such as dichloromethane and the reaction mixture was stirred at 0 °C or at ambient temperature. Boron tribromide (2.5-10 equivalents, 1 M solution in dichloromethane) was added dropwise to the reaction mixture. The reaction was stirred at ambient temperature for 4 h – 6 days after which it was quenched by the careful addition of methanol or water and the solvent was removed under reduced pressure. The product was isolated by a procedure similar to that described below. The residue was combined with 2-6 M hydrochloric acid, heated to 50°C, and stirred for 1-2 hours. The resulting solution was cooled (ice bath) and then free-based (pH 9) with the addition of 2-6 M aqueous sodium hydroxide. The desired material was extracted from the aqueous using an organic solvent such as dichloromethane, ethyl acetate, or chloroform. The organic layer was separated, dried

(MgSO<sub>4</sub>), filtered, and the solvent was evaporated under reduced pressure to afford the crude product. The final compound was isolated by prep HPLC (ISCO Combiflash Separation System or Analogix Purification System).

Example	Structure	Analytical Data
189	OH OH	Off-white needles, mp 180-182 °C. Anal. calcd for $C_{21}H_{23}N_5O_3$ •2.60 $H_2O$ : C, 57.29; H, 6.46; N, 15.91. Found: C, 57.32; H, 6.15; N, 15.73; MS (APCI) $m/z$ 394 (M+H) <sup>+</sup> .
190	NH <sub>2</sub> OH	Off-white needles, mp 196-198 °C. Anal. calcd for C <sub>23</sub> H <sub>26</sub> N <sub>6</sub> O <sub>3</sub> S: C, 59.21; H, 5.62; N, 18.01. Found: C, 59.16; H, 5.84; N, 17.98; MS (APCI) m/z 467 (M+H) <sup>+</sup> .
191	NH <sub>2</sub> N OH	Off-white needles, mp 154-157 °C. Anal. calcd for C <sub>26</sub> H <sub>30</sub> N <sub>6</sub> O <sub>2</sub> •0.25H <sub>2</sub> O: C, 67.44; H, 6.64; N, 18.15. Found: C, 67.48; H, 6.55; N, 18.00; MS (APCI) m/z 459 (M+H) <sup>+</sup> .
192	NH <sub>2</sub> N OH	Off-white needles, mp 182-184 °C. Anal. calcd for $C_{26}H_{31}N_{7}O_{2}$ : C, 65.94; H, 6.60; N, 20.70. Found: C, 65.70; H, 6.49; N, 20.39); MS (APCI) $m/z$ 474 (M+H) <sup>+</sup> .
193	NH <sub>2</sub> N OH	Beige needles, mp 111-114 °C. Anal. calcd for C <sub>20</sub> H <sub>20</sub> FN <sub>5</sub> O <sub>2</sub> •2.0 H <sub>2</sub> O: C, 57.55; H, 5.79; N,16.78. Found: C,57.33; H, 5.57; N, 16.76 MS (APCI) m/z 382 (M+H) <sup>+</sup>
194	NH <sub>2</sub> OH	Off-white solid, mp 188-190 °C Anal. calcd for C <sub>21</sub> H <sub>24</sub> N <sub>6</sub> O <sub>3</sub> S•1.70H <sub>2</sub> O C: 53.53, H: 5.86, N: 17.84. Found: C: 53.23, %H: 5.62, N: 17.81. MS (APCI) m/z 459 (M+H) <sup>+</sup>
195	NH <sub>2</sub> NOH	Green solid, mp 206-209 °C Anal. calcd for C <sub>24</sub> H <sub>29</sub> N <sub>7</sub> O <sub>2</sub> *0.27H <sub>2</sub> O C: 63.72, H: 6.58, N: 21.67. Found: C: 63.97, H: 6.26, N: 21.64. MS (APCI) m/z 448 (M+H) <sup>+</sup>

	NH,	Off-white solid, mp 211-212 °C
196	N OH	Anal. calcd for C <sub>24</sub> H <sub>28</sub> N <sub>6</sub> O <sub>2</sub> •0.25H <sub>2</sub> O
1	11>	C: 65.96, H: 6.57, N: 19.23.Found: C: 65.52
ŀ		H: 6.38, N: 19.38
<b>{</b>		MS (APCI) m/z 433 (M+H) <sup>+</sup>
1		IND (AT OI) HUB 135 (AT TA)
	HN.	
197	NH <sub>2</sub>	Yellow solid, mp 225-227 °C
1,77	N OH	Anal. calcd for C <sub>26</sub> H <sub>31</sub> N <sub>7</sub> O <sub>2</sub> •0.38H <sub>2</sub> O
Ì	L'AN	C: 65.00, H: 6.66, N: 20.41.
		Found: C: 65.26, H: 6.53, N: 20.42.
		MS (APCI) m/z 474 (M+H) <sup>+</sup>
į	N H	
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	Ö	NA: 1:1 241 242 9C
198	N OH	White solid, mp 241-242 °C
1	<b>1 1 1 1 1 1 1 1 1 1</b>	Anal. calcd for C <sub>26</sub> H3 <sub>0</sub> N <sub>6</sub> O <sub>2</sub>
		C: 68.10, H: 6.59, N: 18.33.
		Found: C: 67.85, H: 6.48, N: 18.32.
		MS (APCI) $m/z$ 459 (M+H) <sup>+</sup>
	TIN T	]
	O NH <sub>2</sub>	White solid, mp 225-227 °C
199	N OH	Anal. calcd for C <sub>24</sub> H <sub>28</sub> N <sub>6</sub> O <sub>2</sub> •0.38H <sub>2</sub> O
		C: 65.61, H: 6.60, N: 19.13.
		Found: C: 65.19, H: 6.74, N: 18.96.
		MS (APCI) m/z 433 (M+H) <sup>+</sup>
		WIS (AT CI) 1102 433 (111-11)
	HN.	
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200	NH <sub>2</sub>	White solid, mp >300 °C.
200	N OH	Anal. calcd for C <sub>24</sub> H <sub>28</sub> N <sub>6</sub> O <sub>4</sub> S•HBr •0.2H <sub>2</sub> O:
	OH CHI	C, 49.61; H, 5.10; N, 14.46. Found: C,
		49.26; H, 4.84; N, 14.29
-	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	MS (APCI) m/z 497 (M+H) <sup>+</sup>
	N N N N N N N N N N N N N N N N N N N	Ten solid mp >300 °C
201	NH <sub>2</sub>	Tan solid, mp >300 °C. Anal. calcd for C <sub>27</sub> H <sub>32</sub> N <sub>6</sub> O <sub>3</sub> •HBr: C, 56.94;
201		H, 5.84; N, 14.76. Found: C, 56.66; H,
[	OH CON	
		5.69; N, 14.63.
		MS (APCI) m/z 489 (M+H) <sup>+</sup>
	ŅH <sub>2</sub>	Off-white solid, mp >300 °C.
202	N OH	Anal. calcd for $C_{27}H_{33}N_7O_3$ •HBr: C, 55.14;
		H, 5.90; N, 16.67. Found: C, 54.86; H,
	OH CON	5.60; N, 16.64.
1		MS (APCI) m/z 504 (M+H) <sup>+</sup>
}	NH NH	IND (AT CI) IND SON (INT III)
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203	NH <sub>2</sub> N OH	Off white needles, mp 218-221 °C Anal. calcd for C <sub>26</sub> H <sub>29</sub> N <sub>5</sub> O <sub>2</sub> •1.25 H <sub>2</sub> O: C, 67.00; H, 6.81; N, 15.03. Found: C, 67.04; H, 6.78, N, 14.90. MS (APCI) m/z 444 (M+H) <sup>+</sup>
204	OH OH	Off white solid, mp >250 °C Anal. calcd for $C_{25}H_{27}N_5O_3 \cdot 0.75 H_2O$ : C, 65.41; H, 6.26; N, 15.26. Found: C, 65.48; H, 6.40; N, 15.07. MS (APCI) $m/z$ 446 (M+H) <sup>+</sup>
205	OH OH	Off-white solid, mp 166-170 °C Anal. calcd for $C_{24}H_{27}N_5O_2 \cdot 0.9 H_2O$ : C, 66.46; H, 6.69; N, 16.15. Found: C, 66.09; H, 6.73; N, 15.97. MS (APCI) $m/z$ 418 (M + H) <sup>+</sup>
206	NH.	Off-white solid, mp 260-264 °C Anal. calcd for C <sub>29</sub> H <sub>33</sub> N <sub>5</sub> O <sub>3</sub> •0.6 H <sub>2</sub> O•1.0 HCl: C, 63.69; H, 6.49; N, 12.81. Found: C, 63.37; H, 6.23; N, 12.62. MS (APCI) m/z 500 (M + H) <sup>+</sup>
207	NH, NOH	Off-white needles, mp 141-143 °C Anal. calcd for $C_{20}H_{21}N_5O_2 \cdot 1.00CH_4O \cdot 1.0$ $H_2O : C, 61.15 H, 6.35 N, 16.98$ . Found: C, 61.15 H, 6.06 N, 17.34. MS (APCI) $m/z$ 364 (M + H) <sup>+</sup>

Examples 208 - 318

## Part A

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A solution of 1-(4-aminobutyl)-2-ethoxymethyl-7-(pyridin-3-yl)-1H-imidazo[4,5-c]quinoline-4-amine (43 mg, 0.10 mmol, 1 eq, U.S. Patent Application Publication 2004/0147543, Example 372) and triethylamine (5 eq) in chloroform (1 mL) was added to a tube containing a reagent (1.1 eq) from the table below. The reaction mixture was vortexed overnight and then purified by solid-supported liquid-liquid extraction according to the following procedure. The reaction mixture was loaded onto diatomaceous earth that had been equilibrated with 1 N sodium hydroxide (600  $\mu$ L) for about 20 minutes. After 10 minutes chloroform (300  $\mu$ L) was added to elute the product from the diatomaceous earth into a well of a collection plate. After an additional 10 minutes the process was repeated

with additional chloroform (500  $\mu$ L). The solvent was then removed by vacuum centrifugation.

## Part B

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The material from Part A was dissolved in dichloromethane (600  $\mu$ L) and the solution was cooled to 0 °C. Boron tribromide (400  $\mu$ L of 1 M in dichloromethane) was added, the reaction mixture was vortexed, chilled for 15 minutes, and then vortexed at ambient temperature overnight. The solvent was removed by vacuum centrifugation. Methanol (300  $\mu$ L) and 6 N hydrochloric acid (300  $\mu$ L) were added and the reaction mixture was vortexed for 10 minutes. The solvent was removed by vacuum centrifugation. The compounds were purified as described above for Examples 156 – 161. The table below shows the reagent used for each example, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

NH <sub>2</sub> OH			
Example	Reagent	R	Measured Mass (M+H)
208	None	_H	363.1964
209	Propionyl chloride	O CH₃	419.2168
210	Cyclopropanecarbonyl chloride	o	431.2213
211	Butyryl chloride	O CH₃	433.2345
212	Isobutyryl chloride	H <sub>3</sub> C CH <sub>3</sub>	433.2346

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213	Methoxyacetyl chloride	ОН	421.1982
214	Cyclobutanecarbonyl chloride		445.2338
215	Isovaleryl chloride	CH³	447.2536
216	Cyclohexanecarbonyl chloride	°	473.2679
217	Phenylacetyl chloride		481.2368
218	4-Cyanobenzoyl chloride		492.2143
219	3-Methoxybenzoyl chloride	ОН	483.2121
220	p-Anisoyl chloride	ОН	483.2115
221	2-Chlorobenzoyl chloride	CI	501.1813
222	3-Chlorobenzoyl chloride	CI	501.1812

223	Nicotinoyl chloride hydrochloride		468.2122
224	Picolinoyl chloride hydrochloride	ON	468.2124
225	1-Propanesulfonyl chloride	O,O CH <sub>3</sub>	469.2039
226	Dimethylsulfamoyl chloride	O.O N-CH <sub>3</sub>	470.1961
227	1-Butanesulfonyl chloride	Q,O S CH <sub>3</sub>	483.2160
228	3- Methylbenzenesulfonyl chloride	H <sub>3</sub> C	517.2044
229	o-Toluenesulfonyl chloride	H <sub>3</sub> C	517.2071
300	p-Toluenesulfonyl chloride	O,O S	517.2020
301	2-Fluorobenzenesulfonyl chloride	O O F	521.1786

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302	3-Cyanobenzenesulfonyl chloride	0.00	528.1805
303	3- Methoxybenzenesulfonyl chloride	O.O. S	519.1829
304	4- Methoxybenzenesulfonyl chloride	О	519.1799
305	3-Pyridinesulfonyl chloride hydrochloride	0.0	504.1852
306	Ethyl isocyanate	O N H CH <sub>3</sub>	434.2307
307	Isopropyl isocyanate	O N CH <sub>3</sub>	448.2498
308	n-Propyl isocyanate	N CH3	448.2448
309	Cyclopentyl isocyanate	PO N	474.2629
310	Phenyl isocyanate	H N	482.2338
311	Cyclohexyl isocyanate	H N N	488.2759

312	2-Fluorophenyl isocyanate	N F	500.2209
313	3-Fluorophenyl isocyanate	O NH F	500.2206
314	4-Fluorophenyl isocyanate	O N H	500.2209
315	(R)-(+)-alpha- Methylbenzyl isocyanate	O CH <sub>3</sub>	510.2580
316	(S)-(-)-alpha- Methylbenzyl isocyanate	O CH <sub>3</sub>	510.2588
317	1-Piperidinecarbonyl chloride	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	474.2606
318	4-Methyl-1- piperazinecarbonyl chloride	CH3	489.2725

Examples 319 – 345

The compounds in the table below were prepared and purified according to the general method of Examples 162-186 using N-{4-[4-amino-7-bromo-2-ethoxymethyl-1H-imidazo[4,5-c]quinolin-1-yl]butyl}methanesulfonamide (U.S. Patent Application Publication 2004/0147543, Example 612) in lieu of 1-(4-amino-7-bromo-2-ethoxymethyl-1H-imidazo[4,5-c]quinolin-1-yl)-2-methylpropan-2-ol. Prior to purification by solid phase extraction, the reaction mixture for Example 345 was combined with water (500  $\mu$ L),

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glacial acetic acid (500  $\mu$ L), and tetrahydrofuran (500  $\mu$ L) and then heated at 60 °C for 2 hours. The table below shows the boronic acid, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

	NH <sub>2</sub> N OH N N S CH <sub>3</sub>						
Example	Reagent	R	Measured Mass (M+H)				
319	Phenylboronic acid		440.1745				
320	Pyridine-3-boronic acid		441.1745				
321	Pyridine-4-boronic acid	N	441.1679				
322	Thiophene-3-boronic acid	s	446.1307				
323	2-Fluorophenylboronic acid	€ F	458.1668				
324	3-Fluorophenylboronic acid	F	458.1671				
325	4-Fluorophenylboronic acid	F	458.1674				
326	4-Cyanophenylboronic acid	N	465.1684				
327	3-(Hydroxymethyl)phenylboronic acid	но	470.1882				

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328	4-(Hydroxymethyl)phenylboronic acid	ОН	470.1909
329	3-Chlorophenylboronic acid	CI	474.1408
330	2-Chlorophenylboronic acid	CI	474.1366
331	4-Chlorophenylboronic acid	CI	474.1384
332	(2-Aminocarbonylphenyl)boronic acid	NH <sub>2</sub>	483.1796
333	(3-Aminocarbonylphenyl)boronic acid	O NH <sub>2</sub>	483.1812
334	(2-Acetylaminophenyl)boronic acid	NH H <sub>3</sub> C O	497.1938
335	[3-(3-Hydroxypropyl)phenyl]boronic acid	но	498.2136
336	3,4-Dichlorophenylboronic acid	CI	508.0989
337	3-( <i>N</i> - Isopropylaminocarbonyl)phenylboronic acid	HN O H <sub>3</sub> C CH <sub>3</sub>	525.2331

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338	3-(N- Propylaminocarbonyl)phenylboronic acid	HN O	525.2284
339	3-(Methylsulfonylamino)phenylboronic acid	O. 9 NH H <sub>3</sub> C O	533.1659
340	3-(Pyrrolidine-1- carbonyl)phenylboronic acid	CNO	537.2320
341	4-( Pyrrolidine-1- carbonyl)phenylboronic acid		537.2271
342	3- (Isobutylaminocarbonyl)phenylboronic acid	HN O H <sub>3</sub> C CH <sub>3</sub>	539.2418
343	4- (Isobutylaminocarbonyl)phenylboronic acid	ONH H <sub>3</sub> C CH <sub>3</sub>	539.2429
344	3-(Piperidine-1- carbonyl)phenylboronic acid	ON O	551.2483
345	5-tert-butyldimethylsilanyloxy- methyl)pyridine-3-boronic acid	но	471.1819

# Examples 346 - 362

The compounds in the table below were prepared according to the following method. A test tube containing a solution of the corresponding ether analog (ethoxymethyl or methoxyethyl) in dichloromethane (1 mL) was cooled to 0 °C in an ice bath. Boron tribromide (4 eq of 1 M in dichloromethane) was added. The tube was vortexed, maintained at 0 °C for 0.5 hr, and then stirred at ambient temperature for 9 hours. Methanol (1 mL) and 6 N hydrochloric acid (500 µL) were added and the tube was vortexed for 5 minutes. The solvent was removed by vacuum centrifugation. The compounds were purified as described above for Examples 156-161. The table below shows a reference for the starting ether, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

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		Measured Mass (M+H)	347.1904		372.1819	440.1755		348 1910	0101.010	458.2540		470.1832	
		R3					<b>&gt;</b>		Z		<b>&gt;</b>	Ž,	
N N N N N N N N N N N N N N N N N N N		K2	₽ P	HO	7.	Н		₽,		H <sub>O</sub>		HO	
T Z	ž Ž		n f	CH <sub>3</sub>	, CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub> O N N N N N N N N N N N N N N N N N N N	0	£ 5	D'H	N CH3		CH3 OOH3	
	Reference for ether starting material	5 11401141	Example 102	Example 111		Example 201		Example 113		Example 194		Example 139	
	Example	346	2	347		348	370	243	350	Occ.		351	

466.1897	502.2554	460.2326	476.2285	469.2024	388.2130
	Z	ON O			Z
₹	Ю	НО	Ю	. HO	HO /
0:00		CH3 OH	CH3 OH HO	0.0 H-S.	9
Example 152	Example 180	Example 129	Example 130	Example 376	Example 438
352	353	354	355	356	357

467.1852	366.1574	433.2374	482.1815	476.2383
Z	N	\	H <sub>3</sub> C.S. NH	
НО	но	но	но	. Ho
0-S, N	HO	CH <sub>3</sub> CH <sub>3</sub> H <sub>3</sub> C CH <sub>3</sub>		O ZI
Example 492	Example 488	Example 422	Example 480	<b>*</b>
358	359	360	361	362

\*Although not specifically exemplified, the compound is readily prepared using the disclosed synthetic methods. All references are to U.S. Patent Application Publication 2004/0147543.

# Example 363

[4-Amino-7-[3-(pyrrolidin-1-ylcarbonyl)phenyl]-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinolin-2-yl]methanol

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To a mixture of [4-amino-7-bromo-1-(tetrahydro-2H-pyran-4-ylmethyl)-1H-imidazo[4,5-c]quinolin-2-yl]methanol (400 mg, 1.00 mmol), 3-pyrrolidinylcarbonyl phenyl boronic acid (328 mg, 1.50 mmol), potassium carbonate (455 mg, 3.30 mmol), dimethoxyethane (4 mL), and water (2 mL) under a nitrogen atmosphere was added Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> (14 mg, 0.02 mmol). The resulting suspension was refluxed for 18 hours. The reaction was cooled to ambient temperature. The reaction mixture was diluted with water and extracted with chloroform. The organic layer was dried over sodium sulfate, filtered and concentrated under reduced pressure. Chromatography (SiO<sub>2</sub>, 0-40% CMA/CHCl<sub>3</sub>) gave material that was stirred in acetonitrile and filtered to provide 100 mg of [4-amino-7-[3-(pyrrolidin-1-ylcarbonyl)phenyl]-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinolin-2-yl]methanol as a white powder, m.p. 281-284 °C. MS (APCI) *m/z* 486.3 (M + H)<sup>+</sup>; Anal. calcd for C<sub>28</sub>H<sub>31</sub>N<sub>5</sub>O<sub>3</sub>: C, 69.26; H, 6.43; N, 14.42. Found: C, 68.99; H, 6.16; N, 14.46.

## Example 364

4-Amino-1-[2,2-dimethyl-3-(methylsulfonyl)propyl]-1*H*-imidazo[4,5-*c*]quinolin-2-yl}methanol

To a suspension of 1-[2,2-dimethyl-3-(methylsulfonyl)propyl]-2-(ethoxymethyl)-1H-imidazo[4,5-c]quinolin-4-amine (0.4 g, 1.02 mmol) in dichloromethane (5 mL) was added boron tribromide (5.1 mL, 1M solution in dichloromethane). An exotherm was observed upon addition and the mixture turned light purple. After stirring at ambient temperature for 20 hours, the remaining starting material was consumed by adding boron tribromide (2.5 mL, 1M solution in dichloromethane). The reaction was quenched with aqueous hydrochloric acid (1N, 20 mL) to afford a homogeneous mixture. The layers were separated and the aqueous layer washed with dichloromethane (20 mL). The pH of the aqueous layer was adjusted to 12 by addition of aqueous sodium hydroxide (50%) at which time a solid precipitated out of solution. The solid was stirred for 18 hours, collected by filtration and washed with water. The crude product was purified by chromatography over silica gel (eluting with CMA) to afford a white powder. The powder was triturated with methanol (20 mL). The resulting solid was isolated by filtration, washed with methanol and dried for 4 hours at 65 °C to provide 150 mg of {4-amino-1-[2,2-dimethyl-3-(methylsulfonyl)propyl]-1*H*-imidazo[4,5-*c*]quinolin-2-yl}methanol as a white powder, mp 230-232 °C.

Anal. Calcd for  $C_{17}H_{22}N_4O_3S$ : %C, 56.33; %H, 6.12; %N, 15.46. Found: %C, 56.33; %H, 6.31; %N, 15.27.

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## Example 365

 $N-\{2-[4-amino-2-(hydroxymethyl)-1H-imidazo[4,5-c]quinolin-1-yl]ethyl\}-N-isopropylurea$ 

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A stirring solution of N-{2-[4-amino-2-(ethoxymethyl)-1H-imidazo[4,5-c]quinolin-1-yl]ethyl}-N-isopropylurea (400 mg, 1.1 mmol) in dichloromethane (50 mL) was sealed with a septum and purged with nitrogen gas. The solution was cooled in an ice/water bath and a 1.0 M solution of boron tribromide in dichloromethane (2.2 mL) was

added via syringe. The resulting mixture was stirred for 2 hours while warming to ambient temperature. The mixture was cooled back to 0 °C in an ice/water bath and the second portion of boron tribromide (1.0 M, 5.5 mL) was added. The reaction was stirred for 18 hours while warming to ambient temperature. Aqueous hydrochloric acid (6N, 10 ml) was added and the mixture was stirred for 1 hour. The layers were separated and the aqueous fraction was neutralized by the slow addition of solid sodium hydroxide until the pH reached 14. A fine precipitate formed. The aqueous mixture was extracted with chloroform (2x 50 mL) and filtered. The resulting solid (filter cake) was combined with the organic extracts, methanol (50 mL), and silica gel (5 g). The mixture was concentrated under reduced pressure. The crude product absorbed on silica was purified by chromatography using a HORIZON HPFC system (silica cartridge, eluting with 0-35% CMA in chloroform over 2.6 L) followed by recrystallization from acetonitrile to provide 170 mg of *N*-{2-[4-amino-2-(hydroxymethyl)-1*H*-imidazo[4,5-*c*]quinolin-1-yl]ethyl}-*N*-isopropylurea as an off-white solid, mp >240 °C.

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<sup>1</sup>H NMR (500 MHz, DMSO- $d_6$ )  $\delta$  8.30 (d, J = 7.9 Hz, 1H), 7.61 (dd, J = 8.3, 0.9 Hz, 1H), 7.43 (m, 1H), 7.24 (m, 1H), 6.53 (br s, 2H), 5.99 (t, J = 5.8 Hz, 1H), 5.82 (d, J = 7.8 Hz, 1H), 5.67 (d, J = 5.8 Hz, 1H), 4.75 (d, J = 5.8 Hz, 2H), 4.66 (t, J = 6.7 Hz, 2H), 3.69 (m, 1H), 3.48 (q, J = 6.4 Hz, 2H), 1.01 (d, J = 6.5 Hz, 6H); MS (APCI) m/z 343 (M + H)<sup>+</sup>;

20 Anal. Calcd. for  $C_{17}H_{22}N_6O_2$ : %C, 59.63; %H, 6.48; %N, 24.54. Found: %C, 59.64; %H, 6.59; %N, 24.58.

Example 366

*N*-{4-[4-amino-2-(hydroxymethyl)-1*H*-imidazo[4,5-*c*]quinolin-1-yl]butyl}cyclopentanecarboxamide

Boron tribromide (2.5 equivalents, 14.6 mL of 1 M solution in dichloromethane) was added dropwise to a cooled (ice bath) suspension of N-{4-[4-amino-2-(ethoxymethyl)-1H-imidazo[4,5-c]quinolin-1-yl]butyl}cyclopentanecarboxamide (2.4 g, 5.8 mmol) in dichloromethane (25 mL). The reaction mixture was allowed to slowly warm to ambient temperature and then stirred for 6 days. Additional boron tribromide (5 equivalents, 29 mmol, 29 mL) was added and the reaction was stirred at ambient until starting material was consumed. The reaction was quenched slowly with methanol (100 mL) and then concentrated under reduced pressure. The residue was combined with 6 M hydrochloric acid (100 mL), heated to 50°C, and stirred for 2 hours. The resulting solution was cooled (ice bath) and then free-based (pH 9) with the addition of 6 M aqueous sodium hydroxide. A brown gummy solid formed in the basic aqueous solution. The aqueous liquid was decanted from the solid and acetonitrile was added (30 mL). A white precipitate formed and was isolated by filtration. The white precipitate was then triturated with hot acetonitrile, allowed to cool, isolated by filtration, washed with ether, and dried under vacuum to provide N-{4-[4-amino-2-(hydroxymethyl)-1H-imidazo[4,5-c]quinolin-1yl]butyl}cyclopentanecarboxamide (0.48 g) as a fine white solid, mp 183-186°C; MS (ESI) m/z 382 (M+H)<sup>+</sup>; Anal. Calcd for  $C_{21}H_{27}N_5O_2$ : C, 65.35; H, 7.18; N, 18.14; Found C, 65.06; H, 6.90; N, 18.13.

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Example 367

N-[4-(4-amino-2-hydroxymethyl-1H-imidazo[4,5-c]quinolin-1-yl)butyl]isobutyramide

Boron tribromide (2.5 equivalents, 15.6 mL of 1 M solution in dichloromethane) was added dropwise to a cooled (ice bath) suspension of N-[4-(4-amino-2-ethoxymethyl-1H-imidazo[4,5-c]quinolin-1-yl)butyl]isobutyramide (2.4 g, 6.2 mmol) in dichloromethane (25 mL). The reaction mixture was allowed to slowly warm to ambient temperature and then stirred for 1 day. Additional boron tribromide (5 equivalents, 31

mmol, 31 mL) was added to the mixture. The reaction was quenched slowly with methanol (100 mL) and then concentrated under reduced pressure. The residue was combined with 6 M hydrochloric acid (100 mL), heated to 50°C, and stirred for 2 hours. The resulting solution was cooled (ice bath) and then free-based (pH 9) with the addition of 6 M sodium hydroxide. A brown gummy solid formed in the basic aqueous solution. The resulting solid was extracted with dichloromethane (6 x 50 mL). The combined extracts were washed with brine (100 mL), dried with magnesium sulfate, filtered, and then concentrated under reduced pressure. This material was purified by prep HPLC (Analogix Separation System, Biotage Si 40+M column, eluted with a gradient of 0-20% methanol in dichloromethane with 1% ammonium hydroxide) to provide a light brown solid. The solid was triturated with hot acetonitrile, allowed to cool, isolated by filtration, washed with ether, and dried under vacuum to provide N-[4-(4-amino-2-hydroxymethyl-1H-imidazo[4,5-c]quinolin-1-yl)butyl]isobutyramide (0.049 g) as a white solid, mp 222-224°C; MS (ESI) m/z 356 (M+H)<sup>+</sup>; Anal. Calcd for C<sub>19</sub>H<sub>25</sub>N<sub>3</sub>O<sub>2</sub>•0.25HBr•0.10H<sub>2</sub>O: C, 60.46; H, 6.80; N, 18.55; Found C, 60.26; H, 6.64; N, 18.43.

# Example 368 N-[4-(4-amino-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1yl)butyl]methanesulfonamide

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Boron tribromide (2.5 equivalents, 20 mL of 1 M solution in dichloromethane) was added dropwise to a cooled (ice bath) suspension of N-[4-(4-amino-2-ethoxymethyl-1H-imidazo[4,5-c]quinolin-1-yl)butyl]methanesulfonamide (3g, 7.92 mmol) in dichloromethane (20 mL). The reaction mixture was allowed to slowly warm to ambient temperature and then stirred for 4 hours. Additional boron tribromide (2 mL) was added and the mixture was stirred for 3 hours. The reaction was quenched slowly with methanol (20 mL) and then concentrated under reduced pressure. The residue was combined with 6

M hydrochloric acid (50mL), heated to 50°C, and stirred for 2 hours. The resulting solution was concentrated under reduced pressure to a slurry that cooled (ice bath) and then free-based with the addition of 7 M ammonia in methanol (40 mL). The mixture was concentrated under reduced pressure and the addition of 7 M ammonia in methanol (40mL) was repeated 2 more times. The concentrated brown sludge like material was purified by prep HPLC (ISCO Combiflash Separation System, Biotage Si 40+M column, eluted with a gradient of methanol in dichloromethane with 1% ammonium hydroxide) to provide a light brown solid. The solid was triturated with hot acetonitrile, allowed to cool, isolated by filtration, washed with ether, and dried under vacuum to provide *N*-[4-(4-amino-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1-yl)butyl]methanesulfonamide (0.1 g) as a fine beige solid, mp 216-219°C; MS (ESI) *m/z* 364 (M+H)<sup>+</sup>; Anal. Calcd for C<sub>16</sub>H<sub>21</sub>N<sub>5</sub>O3S: C, 52.88; H, 5.82; N, 19.27; Found C, 52.62; H, 5.71; N, 19.02.

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## Example 369

(4-Amino-1-{4-[(methylsulfonyl)amino]butyl}-1*H*-imidazo[4,5-*c*]quinolin-2-yl)methyl *N*[(benzyloxy)carbonyl]-L-valinate

To a stirred suspension of N-[4-(4-amino-2-hydroxymethyl-1H-imidazo[4,5-c]quinolin-1-yl)butyl]methanesulfonamide (2.1 g, 5.8 mmol) in THF was added triphenylphosphine (1.5 equivalents, 8.7 mmol, 2.2 g) followed by CBZ-L-valine (1.5 equivalents, 8.7 mmol, 2.3 g). The suspension was stirred for 5 min after which it was cooled in an ice-bath. To this cooled reaction mixture diisopropyl azodicarboxylate (DIAD,1.8 equivalents, 10.4 mmol, 2.0 mL) was added and the reaction was warmed to room temperature and stirred overnight. The solvent was evaporated under reduced pressure and the crude solid was purified by prep HPLC (ISCO Combiflash Separation

System, Biotage Si 40+M column, eluted with a gradient of 0-8% methanol in dichloromethane with 1% ammonium hydroxide) to provide a solid. The solid was heated in diethyl ether and filtered to afford (4-amino-1-{4-[(methylsulfonyl)amino]butyl}-1*H*-imidazo[4,5-*c*]quinolin-2-yl)methyl *N*-[(benzyloxy)carbonyl]-L-valinate (2 g) as a beige solid, mp 99-100°C; MS (ESI) *m/z* 597 (M+H)<sup>+</sup>; Anal. Calcd for C<sub>29</sub>H<sub>36</sub>N<sub>6</sub>O<sub>6</sub>S: C, 58.37; H, 6.08; N, 14.08; Found C, 57.98; H, 6.31; N, 13.82.

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## Example 370

(4-Amino-1- $\{4-[(methylsulfonyl)amino]butyl\}-1$ *H*-imidazo[4,5-c]quinolin-2-yl)methyl L-valinate

To a hydrogenation bottle was added (4-amino-1-{4-[(methylsulfonyl)amino]butyl}-1*H*-imidazo[4,5-*c*]quinolin-2-yl)methyl *N*-[(benzyloxy)carbonyl]-L-valinate (1.5 g, 2.5 mmol) followed by a mixture of methanol (30 mL), THF (15 mL) and water (5 mL) and conc HCl (5 mL). To this was added Pd/C (90 mg) and the reaction was hydrogenated at 40 psi (2.8 X 10<sup>5</sup> Pa) overnight. To the reaction mixture was added conc. HCl (5 mL) and Pd/C (90 mg) and the reaction was hydrogenated at 40 psi (2.8 X 10<sup>5</sup> Pa) for 18 hours. The reaction was filtered through CELITE filter aid and the filtrate was evaporated to afford a clear oil. The product was isolated by prep HPLC (ISCO Combiflash Separation System, Biotage Si 40+M column, eluted with a gradient of 0-8% methanol in dichloromethane with 1% ammonium hydroxide) to provide (4-amino-1-{4-[(methylsulfonyl)amino]butyl}-1*H*-imidazo[4,5-*c*]quinolin-2-yl)methyl L-valinate (0.495 g) as an off white solid, mp 161-163°C; MS (ESI) *m/z* 463 (M+H)<sup>‡</sup>; Anal. Calcd for C<sub>21</sub>H<sub>30</sub>N<sub>6</sub>O<sub>4</sub>S: C, 54.53; H, 6.54; N, 18.17; Found C, 53.96; H, 6.62; N, 17.85, delta C = 0.57.

# Example 371

[4-Amino-1-(tetrahydro-2H-pyran-4-ylmethyl)-1H-imidazo[4,5-c]quinolin-2-yl]methanol

## 5 Part A

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Under a nitrogen atmosphere THF (90 mL) and triethylamine (17.5 mL, 125.6 mmol) were added sequentially to a mixture of crude 4-chloro-3-nitroquinoline (13.10 g, 62.81 mmol) and 1-tetrahydro-2*H*-pyran-4-ylmethylamine hydrochloride (10.0 g, 65.95 mmol). The reaction mixture was placed in an oil bath at 45 °C for 1 hour and then concentrated under reduced pressure. The residue was diluted with THF (30 mL) and water (200 mL). The THF was removed under reduced pressure. A solid was isolated by filtration and dried to provide 16.10 g of 3-nitro-*N*-(tetrahydro-2*H*-pyran-4-ylmethyl)quinolin-4-amine as a light yellow solid.

Part B

A mixture of 3-nitro-N-(tetrahydro-2H-pyran-4-ylmethyl)quinolin-4-amine (2.50 g), 10% palladium on carbon (0.25 g), and ethanol (40 mL) was placed under hydrogen pressure on a Parr apparatus. When the reaction was complete, the mixture was filtered through a layer of CELITE filter agent. The filter cake was washed with ethanol. The filtrate was concentrated under reduced pressure to provide 2.23 g of  $N^4$ -(tetrahydro-2H-pyran-4-ylmethyl)quinoline-3,4-diamine as a yellowish-orange oil.

Part C

Chloroacetyl chloride (12 mL, 151 mmol) was dissolved in dichloromethane (30 mL) and added via addition funnel, over 20 minutes, to a stirring solution of  $N^4$ - (tetrahydro-2H-pyran-4-ylmethyl)quinoline-3,4-diamine (35.3g, 137 mmol) in dichloromethane (300 mL). The resulting solution was stirred at ambient temperature under nitrogen for 24 hours at which point the solution was heated to 40 °C for an additional 24 hours. The mixture was cooled to ambient temperature, diluted with dichloromethane (150 mL) and transferred to a separatory funnel. The organic layer was washed with water (2 x 200 mL) and brine (2 x 200 mL), dried over magnesium sulfate,

filtered and concentrated under reduced pressure to provide 38.3 g of 2-(chloromethyl)-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-c]quinoline as a light brown solid. Part D

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3-Chloroperoxybenzoic acid (mCPBA) (3.8 g of 77% pure material, 14.2 mmol) was added to a stirring solution of 2-(chloromethyl)-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinoline (3.0g, 9.50 mmol) in dichloromethane (60 mL). After 15.5 hours, ammonium hydroxide (12 mL) and then *p*-toluenesulfonyl chloride (2.2g, 11.4 mmol) were added to the stirring solution and the biphasic mixture was stirred at ambient temperature for 3 hours. The reaction was diluted with water (50 mL) and then transferred to a separatory funnel. The aqueous layer was extracted with dichloromethane (3 x 100 mL) and the combined organic fractions dried over magnesium sulfate, filtered and concentrated under reduced pressure. The residue was purified by column chromatography using a HORIZON HPFC system (silica cartridge, eluting with 3 – 20% methanol in dichloromethane) to provide 1.6 g of 2-(chloromethyl)-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinolin-4-amine as a yellow solid. Part E

Potassium acetate (0.41 g, 4.16 mmol) and potassium iodide (0.28g, 1.66 mmol) were added to a stirring solution of 2-(chloromethyl)-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinolin-4-amine (0.55 g, 1.66 mmol) and the resulting suspension was heated to 50 °C. After 17 hours, the suspension was cooled to ambient temperature and concentrated under reduced pressure. The residue was suspended in methanol (10 mL) and water (5 mL) and lithium hydroxide monohydrate (0.35 g, 8.31 mmol) was added in one portion. The resulting solution was stirred at ambient temperature 18 hours and concentrated under reduced pressure. The residue was diluted with water (20 mL) and neutralized with hydrochloric acid (6 N in water). The aqueous layer was extracted with dichloromethane (2 x 50 mL) and ethyl acetate (50 mL). The combined organic fractions were concentrated to a yellow solid which was crystallized from acetonitrile. The crystals were isolated by filtration and dried in a vacuum oven at 65 °C to provide 0.20 g of [4-amino-1-(tetrahydro-2*H*-pyran-4-ylmethyl)-1*H*-imidazo[4,5-*c*]quinolin-2-yl]methanol as an off-white solid, mp 239-241 °C.

Anal. calcd for  $C_{17}H_{20}N_4O_2 \cdot 0.2H_2O$ : C, 64.62; H, 6.51; N, 17.73. Found: C, 64.45; H, 6.69; N, 17.62.

## Examples 372 – 450

Part A

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A solution of 1-(4-aminobutyl)-2-methoxymethyl-1*H*-imidazo[4,5-*c*]quinoline-4-amine (30 mg, 1 eq, prepared according to the general method of Example 3 using methoxyacetyl chloride in lieu of 3-methoxypropionyl chloride) and *N*,*N*-diisopropylethylamine (2 eq) in *N*,*N*-dimethylacetamide (1 mL) was added to a tube containing a reagent (1.1 eq) from the table below. The reaction mixture was vortexed overnight and then quenched with water (100 μL). The solvents were removed by vacuum centrifugation. The residue was purified by solid-supported liquid-liquid extraction according to the following procedure. The sample was dissolved in chloroform (1 mL) then loaded onto diatomaceous earth that had been equilibrated with 1 M sodium hydroxide (600 μL) for about 20 minutes. After 10 minutes chloroform (500 μL) was added to elute the product from the diatomaceous earth into a well of a collection plate. After an additional 10 minutes the process was repeated with additional chloroform (500 μL). The solvent was then removed by vacuum centrifugation.

The residue (in a test tube) was combined with dichloromethane (500  $\mu$ L) and the tube was vortexed to dissolve the solids. The solution was cooled (0 °C) and then combined with boron tribromide (400  $\mu$ L of 1 M in dichloromethane). The mixture was vortexed for 5 minutes, chilled for 30 minutes, and then vortexed at ambient temperature for 64 hours. Additional dichloromethane (500  $\mu$ L) and boron tribromide (400  $\mu$ L of 1 M in dichloromethane) were added and the mixture was vortexed overnight. The solvent was then removed by vacuum centrifugation. The residue was diluted with methanol (500  $\mu$ L) and hydrochloric acid (500  $\mu$ L of 6 N). The solvents were removed by vacuum centrifugation. The compounds were purified according to the method described in Examples 8 – 72. The table below shows the reagent used for each example, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

NH <sub>2</sub> N OH			
	Ť	N-R H	
Example	Reagent	R	Measured Mass (M+H)
372	None	<b>/</b> H	286.1658
373	Cyclopropanecarbonyl chloride	o A	354.1907
374	Methoxyacetyl chloride	ОН	344.1699
375	Cyclobutanecarbonyl chloride		368.2050
376	Isovaleryl chloride	O CH₃ CH₃	370.2206
377	Pentanoyl chloride	CH <sub>3</sub>	370.2208
378	Benzoyl chloride	°	390.1909
379	Cyclohexanecarbonyl chloride	°	396.2412
380	Cyclopentylacetyl chloride	P° ()	396.2411
381	m-Toluoyl chloride	О СН,	404.2069

382	o- Toluoyl chloride	CH <sup>3</sup>	404.2072
383	p- Toluoyl chloride	CH,	404.2108
384	Phenylacetyl chloride		404.2056
385	Dimethylaminoacetyl chloride hydrochloride	O V. CH <sub>3</sub> CH <sub>3</sub>	371.2157
386	2-Fluorobenzoyl chloride	OF	408.1819
387	3-Fluorobenzoyl chloride	F	408.1811
388	4-Fluorobenzoyl chloride	O F	408.1819
389	3-Cyanobenzoyl chloride	O	415.1847
390	Hydrocinnamoyl chloride		418.2200

391	2-Methoxybenzoyl chloride	ООН	406.1880
392	3-Methoxybenzoyl chloride	О	406.1876
293	p-Anisoyl chloride	ОН	406.1860
394	3-Chlorobenzoyl chloride	CI	424.1517
395	4-Chlorobenzoyl chloride	°	424.1525
396	Isonicotinoyl chloride hydrochloride	O	391.1874
397	Nicotinoyl chloride hydrochloride	o N	391.1895
398	Picolinoyl chloride hydrochloride	ON	391.1846
399	trans-2-Phenyl-1- cyclopropanecarbonyl chloride	o <sub>m</sub>	430.2213
400	Methanesulfonyl chloride	_S,CH₃ O	364.1421

	7. 10 11111	O,O	270 1505
401	Ethanesulfonyl chloride	CH <sup>3</sup>	378.1595
402	1-Propanesulfonyl	o, S	392.1753
	chloride	CH <sub>3</sub>	
403	Dimethylsulfamoyl chloride	H³C, N-CH³	393.1685
		0.0	
404	1-Butanesulfonyl chloride	CH <sup>3</sup>	406.1881
		0,8	
405	Benzenesulfonyl chloride		426.1591
		0,0 -s,	
406	1-Methylimidazole-4- sulfonyl chloride	N CH <sub>3</sub>	430.1668
407	2-Thiophenesulfonyl	0,50	432.1135
407	chloride	s	432.1133
		0.0 - \$	
408	3-Methylbenzenesulfonyl chloride		440.1728
		H <sub>3</sub> C	
409	o-Toluenesulfonyl	0,0 '\$	440.1758
707	chloride	H₃C	710.1730
		0,0	
410	p-Toluenesulfonyl chloride		440.1766
		CH₃	

411	2-Fluorobenzenesulfonyl chloride	Q.O F	444.1479
412	3-Fluorobenzenesulfonyl chloride	Q.O S	444.1517
413	4-Fluorobenzenesulfonyl chloride	Q.O. F	444.1496
414	3-Cyanobenzenesulfonyl chloride	0.0	451.1568
415	4-Cyanobenzenesulfonyl chloride	0.0	451.1579
416	beta-Styrenesulfonyl chloride	0,0	452.1725
417	3- Methoxybenzenesulfonyl chloride	O O S	442.1534
418	4- Methoxybenzenesulfonyl chloride	Q,o S	442.1557

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419	2-Chlorobenzenesulfonyl chloride	Q,O CI	460.1173
420	3-Chlorobenzenesulfonyl chloride	O. O.	460.1242
421	4-Chlorobenzenesulfonyl chloride	O.O.	460.1191
422	3-Pyridinesulfonyl chloride hydrochloride	0.0 's	427.1530
423	3,4- Dimethoxybenzenesulfon yl chloride	O.O HO OH	458.1452
424	3,4- Dichlorobenzenesulfonyl chloride	0,0	494.0806
425	Methyl isocyanate	N-ch₃	343.1862
426	Ethyl isocyanate	O N CH <sub>3</sub>	357.2018
427	Isopropyl isocyanate	O N CH <sub>3</sub> CH <sub>3</sub>	371.2181

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428	n-Propyl isocyanate	N-CH <sub>3</sub>	371.2187
429	n-Butyl isocyanate	O NH CH₃	385.2314
430	Cyclopentyl isocyanate	0 ZH	397.2312
431	Pentyl isocyanate	O N H CH <sub>3</sub>	399.2512
432	Phenyl isocyanate	H X Y	405.2047
433	Cyclohexyl isocyanate	±2 %	411.2473
434	2-Fluorophenyl isocyanate	TZ 0	423.1959
435	3-Fluorophenyl isocyanate	0 ZT 0	423.1924
436	4-Cyanophenyl isocyanate	N N	430.1979
437	(R)-(+)-alpha- Methylbenzyl isocyanate	N''', CH <sub>3</sub>	433.2370

438	(S)-(-)- alpha- Methylbenzyl isocyanate	O CH <sub>3</sub>	433.2327
439	2-Phenylethylisocyanate	N. C.	433.2333
440	2-Methoxyphenyl isocyanate	O NH HO	421.2006
441	4-Methoxyphenyl isocyanate	N OH	421.1958
442	2-Chlorophenyl isocyanate	N CI	439.1650
443	4-Chlorophenyl isocyanate	N CI	439.1656
444	trans-2- Phenylcyclopropyl isocyanate	NH CONTRACTOR OF THE PARTY OF T	445.2328
445	N,N-Dimethylcarbamoyl chloride	H <sub>3</sub> C,N-CH <sub>3</sub>	357.2005
446	1-Pyrrolidinecarbonyl chloride	\$ N	383.2168
447	1-Piperidinecarbonyl chloride	0	397.2329

448	4-Morpholinylcarbonyl chloride	0	399.2112
449	4-Methyl-1- Piperazinecarbonyl chloride	CH <sup>3</sup>	412.2439
450	N-Methyl-N- phenylcarbamoyl chloride	H <sub>3</sub> CN	419.2167

Examples 451 - 466

### Part A

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A solution of 1-(2-amino-2-methylpropyl)-2-methoxymethyl-1H-imidazo[4,5-c]quinoline-4-amine (31 mg, 1 eq, prepared according to the general method of Example 3 using methoxyacetyl chloride in lieu of 3-methoxypropionyl chloride and tert-butyl N-{2-[(3-aminoquinolin-4-yl)amino]-1,1-dimethylethyl}carbamate in lieu of tert-butyl N-{4-[(3-aminoquinolin-4-yl)amino]butyl}carbamate) and N,N-diisopropylethylamine (2 eq) in N,N-dimethylacetamide (1 mL) was placed in a test tube. A reagent (1.1 eq) from the table below was added and the reaction mixture was vortexed overnight. The reaction was quenched with concentrated ammonium hydroxide (100  $\mu$ L) and the solvents were removed by vacuum centrifugation.

### Part B

The residue (in a test tube) was combined with dichloromethane (1 mL) and the tube was vortexed to dissolve the solids. The solution was cooled (0 °C) and then combined with boron tribromide (400  $\mu$ L of 1 M in dichloromethane). The reaction was maintained at about 0 °C for 20 minutes. Methanol (1 mL) and hydrochloric acid (500  $\mu$ L of 6 N) were added and the tube was vortexed for about 30 minutes. The solvents were removed by vacuum centrifugation. The compounds were purified according to the method described in Examples 8 – 72. The table below shows the reagent used for each example, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

NH <sub>2</sub> N OH N CH <sub>3</sub> CH <sub>3</sub> N-R				
Example	Reagent	R	Measured Mass (M+H)	
451	None	`н	286.1687	
452	Cyclopropanecarbonyl chloride		354.1936	
453	Butyryl chloride	CH₃	356.2094	
454	Isobutyryl chloride	H <sub>3</sub> C CH <sub>3</sub>	356.2119	
455	Cyclopentanecarbonyl chloride		382.2259	
456	Benzoyl chloride	ro	390.1908	
457	Nicotinoyl chloride hydrochloride		391.1844	
458	Methanesulfonyl chloride	S-CH <sub>3</sub>	364.1414	
459	Benzenesulfonyl chloride	0,0	426.1617	
460	2,2,2- Trifluoroethanesulfon yl chloride	O O F F F	432.1339	

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461	3- Fluorobenzenesulfony I chloride	O O	444.1523
462	n-Propyl isocyanate	O N H CH <sub>3</sub>	371.2215
463	Cyclopentyl isocyanate		397.2327
464	Phenyl isocyanate	HZ O	405.2063
465	Cyclohexyl isocyanate	HZ O	411.2515
466	3-Fluorophenyl isocyanate	O N F	423.1955

Examples 467 – 478

## Part A

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To a round-bottomed flask containing 1-(4-aminobutyl)-2-methoxymethyl-1*H*-imidazo[4,5-*c*]quinolin-4-amine (10.0 g, 33.4 mmol) was added methanol (160 mL) followed by acetic acid (40 mL). The reaction was stirred for 5 minutes and pyridine 3-carboxaldehyde (5.4 g, 50.1 mmol) was added and the reaction was stirred overnight at ambient temperature. Sodium cyanoborohydride (1 M in THF, 33.4 mL, 33.4 mmol) was added to the resultant imine in portions over 10 minutes. After 45 minutes the solvent was evaporated to afford an oil. To the oil was added saturated aqueous sodium bicarbonate (200 mL) and the aqueous layer was washed with ethyl acetate (200 mL) and dichloromethane (200 mL). The product was extracted from the aqueous with 20% methanol (2 x 100 mL) in dichloromethane. The organic layers were combined and the solvent evaporated to afford crude 2-methoxymethyl-1-{4-[(pyridin-3-ylmethyl)amino]butyl}-1*H*-imidazo[4,5-*c*]quinolin-4-amine (about 2 g). The aqueous

layer was again extracted with 20% dimethylformamide (2 x 100 mL) in dichloromethane. The organic layers were combined and the solvent evaporated to afford crude 2-methoxymethyl-1-{4-[(pyridin-3-ylmethyl)amino]butyl}-1*H*-imidazo[4,5-*c*]quinolin-4-amine (about 2 g).

### Part B

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A solution of 2-methoxymethyl-1-{4-[(pyridin-3-ylmethyl)amino]butyl}-1H-imidazo[4,5-c]quinolin-4-amine (40 mg, 1 eq) and N,N-diisopropylethylamine (2 eq) in N,N-dimethylacetamide (1 mL) was added to a tube containing a reagent (1.1 eq) from the table below. The reaction mixture was vortexed for 4 hours and then quenched with water (50  $\mu$ L). The solvents were removed by vacuum centrifugation. The residue was purified by solid-supported liquid-liquid extraction according to the following procedure. The sample was dissolved in chloroform (1 mL) then loaded onto diatomaceous earth that had been equilibrated with 1 M sodium hydroxide (600  $\mu$ L) for about 20 minutes. After 10 minutes chloroform (500  $\mu$ L) was added to elute the product from the diatomaceous earth into a well of a collection plate. After an additional 10 minutes the process was repeated with additional chloroform (500  $\mu$ L). The solvent was then removed by vacuum centrifugation.

### Part C

The residue (in a test tube) was combined with dichloromethane (500  $\mu$ L) and the tube was vortexed to dissolve the solids. The solution was cooled (0 °C) and then combined with boron tribromide (400  $\mu$ L of 1 M in dichloromethane). The mixture was vortexed for 10 minutes, chilled for 30 minutes, and then vortexed at ambient temperature overnight. The solvent was then removed by vacuum centrifugation. The residue was diluted with methanol (500  $\mu$ L) and hydrochloric acid (500  $\mu$ L of 6 N) and the mixture was vortexed for about 30 minutes. The solvents were removed by vacuum centrifugation. The compounds were purified according to the method described in Examples 8 – 72. The table below shows the reagent used for each example, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

NH <sub>2</sub> N OH			
Example	Reagent	R	Measured Mass (M+H)
467	None	н <sup>′</sup>	377.2087
468	Isobutyryl chloride	O H <sub>3</sub> C CH <sub>3</sub>	447.2468
469	Cyclohexanecarbonyl chloride	0	487.2783
470	Phenylacetyl chloride		495.2465
471	4-Fluorobenzoyl chloride	0 <del> </del> F	499.2272
472	3-Methoxybenzoyl chloride	O HO	497.2263
473	1-Methylimidazole-4- sulfonyl chloride	O.S.N. CH3	521.2071
474	2,2,2- Trifluoroethanesulfonyl chloride	0≥S 0 F F F	523.1717

475	alpha-Toluenesulfonyl chloride	0;s′ 0'	531.2134
476	3- Methoxybenzenesulfon yl chloride	HO	533.1941
477	Isopropyl isocyanate	O H₃C CH₃	462.2611
478	3-Fluorophenyl isocyanate	O NH	514.2357

Examples 479 - 543

The compounds in the table below were prepared and purified according to the methods of Parts B and C of Examples 467 – 478 using 1-(4-benzylaminobutyl)-2-ethoxymethy-1*H*-imidazo[4,5-*c*]quinolin-4-amine in lieu of 2-methoxymethyl-1-{4-[(pyridin-3-ylmethyl)amino]butyl}-1*H*-imidazo[4,5-*c*]quinolin-4-amine. 1-(4-Benzylaminobutyl)-2-ethoxymethy-1*H*-imidazo[4,5-*c*]quinolin-4-amine was prepared according to the general method of Part A of Examples 467 – 478 using benzaldehyde in lieu of pyridine 3-carboxaldehyde and 1-(4-aminobutyl)-2-ethoxymethyl-1*H*-imidazo[4,5-*c*]quinolin-4-amine in lieu of 1-(4-aminobutyl)-2-methoxymethyl-1*H*-imidazo[4,5-*c*]quinolin-4-amine. The table below shows the reagent used for each example, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

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	NH <sub>2</sub> N OH				
Example	Reagent	R R	Measured Mass (M+H)		
479	Cyclobutanecarbonyl chloride	0	458.2550		
480	DL-2-Methylbutyryl chloride	O CH <sub>3</sub>	460.2707		
481	Isovaleryl chloride	O H <sub>3</sub> C CH <sub>3</sub>	460.2714		
482	Pentanoyl chloride	O H <sub>3</sub> C	460.2730		
483	Pivaloyl chloride	O CH <sub>3</sub>	460.2714		
484	Cyclopentanecarbonyl chloride	0=	472.2712		
485	tert-Butylacetyl chloride	O H <sub>3</sub> C H <sub>3</sub> C CH <sub>3</sub>	474.2879		
486	Benzoyl chloride	0=	480.2398		

487	Thiophene-2-carbonyl chloride	0 \$	486.1971
488	Cyclohexanecarbonyl chloride	0=	486.2893
489	Cyclopentylacetyl chloride		486.2818
490	m-Toluoyl chloride	O H <sub>3</sub> C	494.2577
491	o-Toluoyl chloride	H <sub>3</sub> C	494.2531
492	<i>p</i> -Toluoyl chloride	O CH <sub>3</sub>	494.2527
493	3-Fluorobenzoyl chloride	O	498.2307
494	4-Fluorobenzoyl chloride	0 <del> </del> F	498.2326
495	3-Cyanobenzoyl chloride	O N	505.2378

496	4-Cyanobenzoyl chloride	O	505.2387
497	Hydrocinnamoyl chloride		508.2715
498	2-Methoxybenzoyl chloride	HO	496.2311
499	3-Methoxybenzoyl chloride	HO	496.2314
500	<i>p</i> -Anisoyl chloride	ОН	496.2365
501	3-Chlorobenzoyl chloride	CI	514.2026
502	4-Chlorobenzoyl chloride	O CI	514.2041
503	Picolinoyl chloride hydrochloride		481.2361

504	trans-2-Phenyl-1- cyclopropanecarbonyl chloride	0	520.2695
505	4-Dimethylaminobenzoyl chloride	H <sub>3</sub> C <sup>N-CH<sub>3</sub></sup>	523.2802
506	1-Propanesulfonyl chloride	O≈s O CH₃	482.2232
507	Dimethylsulfamoyl chloride	O O=S H <sub>3</sub> C,N-CH <sub>3</sub>	483.2196
508	2-Thiophenesulfonyl chloride		522.1613
509	alpha-Toluenesulfonyl chloride	0.5-	530.2239
510	o-Toluenesulfonyl chloride	O=S H <sub>3</sub> C	530.2197
511	4-Fluorobenzenesulfonyl chloride	O F	534.2028
512	3,5-Dimethylisoxazole-4- sulfonyl chloride	O S CH <sub>3</sub>	535.2106

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513	2-Cyanobenzenesulfonyl chloride	O=S	541.1968
514	3-Cyanobenzenesulfonyl chloride	O = S	541.2035
515	<i>beta-</i> Styrene sulfonyl chloride	o si	542.2234
516	3- Methoxybenzenesulfonyl chloride	O. S. HO	532.2052
517	4- Methoxybenzenesulfonyl chloride	O-S OH	532.2037
518	3-Pyridine sulfonyl chloride hydrochloride	0.5 N=	517.2015
519	2,5- Dimethoxybenzenesulfon yl chloride	O O O O O O O O O O O O O O O O O O O	548.1964
520	2,3- Dichlorobenzenesulfonyl chloride	Ozs CI	584.1294

521	3,5- Dichlorobenzenesulfonyl chloride	O = S	584.1282
522	Methyl isocyanate	O NH H₃C	433.2361
523	Ethyl isocyanate	O NH CH₃	447.2538
524	Isopropyl isocyanate	O NH H₃C ← CH₃	461.2663
525	n-Propyl isocyanate	O NH H₃C	461.2691
526	n-Butyl isocyanate	O NH CH <sub>3</sub>	475.2860
527	sec-Butyl isocyanate	O H <sub>3</sub> C H <sub>3</sub> C	475.2849
528	Pentyl isocyanate	O NH	489.3005

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529	Phenyl isocyanate	HX	495.2511
530	Cyclohexyl isocyanate	O ZZ H	501.2978
531	Benzyl isocyanate	O NH	509.2675
532	3-Fluorophenyl isocyanate	O NH F	513.2467
533	4-Fluorophenyl isocyanate	P N H	513.2388
534	Cycloheptyl isocyanate	ON H	515.3081
535	Cyclohexanemethyl isocyanate	HZ	515.3163
536	4-Cyanophenyl isocyanate	N H	520.2483
537	3,4-Dimethylphenyl isocyanate	H <sub>3</sub> C N H	523.2786
538	(S)-(-)-alpha- Methylbenzyl isocyanate	O ZH	523.2786

539	2-Methylbenzyl isocyanate	H <sub>3</sub> C H	523.2860
540	N,N-Dimethylcarbamoyl chloride	O H₃C· <sup>N-</sup> CH₃	447.2511
541	Diethylcarbamyl chloride	O N CH₃	475.2828
542	1-Piperidinecarbonyl chloride	0 <del> </del> N	487.2839
543	N-(4-Chlorobutyl)-N- methylcarbamyl chloride	O N-CH₃	523.2588

Examples 544 – 550

The compounds in the table below were prepared according to the general method of Examples 111 - 140. The table shows a reference for the ether starting material, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

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		NH <sub>2</sub> N N R <sub>1</sub>		
Exampl e	Reference (ether)	R <sub>I</sub>	R <sub>2</sub>	Measured Mass (M+H)

544	U.S. Patent No. 6,667,312*	H <sub>3</sub> C-S=0	ОН	335.1158
545	U.S. Patent No. 6,677,349*	N-S <sup>≥</sup> O	он	336.1098
546	U.S. Patent No. 6,677,349*	H O.S O CH₃	он	364.1454
547	U.S. Patent No. 6,677,347 Example 57	H³C, O HS 0: O	он	380.1391
548	U.S. Patent No. 6,756,382*	N CI	ОН	444.0999
549	U.S. Patent No. 6,683,088 Example 1	H, S; O	ОН	394.1588
550	U.S. Patent No. 6,677,349 Example 242	H O S	/-ОН	496.2401

\*Although not specifically exemplified, the compound is readily prepared using the disclosed synthetic methods.

# Example 551

[4-Amino-1-(2-fluoro-2-methylpropyl)-1H-imidazo[4,5-c][1,5]naphthyridin-2-yl]methanol

Part A

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A solution of 1-amino-2-methylpropan-2-ol (23.4 g, 263 mmol) dissolved in 150 mL of THF was treated with 150 mL of 1.8 M aqueous NaOH solution and the mixture was placed in an ice bath. A solution of di-tert-butyl dicarbonate (57.3 g, 263 mmol) in 150 mL THF was then added drop-wise over 45 min. The mixture was allowed to warm to ambient temperature overnight. The THF was removed under reduced pressure and the remaining aqueous solution was treated with 1 M H<sub>2</sub>SO<sub>4</sub> until the pH reached 3. The mixture was extracted with 200 mL of EtOAc. The organic portion was washed with H<sub>2</sub>O and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated under reduced pressure give tert-butyl 2-hydroxy-2-methylpropylcarbamate (50.4 g) as a colorless syrup which solidified on standing.

Part B

A stirred solution of *tert*-butyl 2-hydroxy-2-methylpropylcarbamate (7.81 g, 41.3 mmol) dissolved in 300 mL of anhydrous CH<sub>2</sub>Cl<sub>2</sub> was cooled to -78 °C under an atmosphere of N<sub>2</sub>. The reaction mixture was treated with diethylaminosulfur trifloride (DAST) (6.2 mL, 47 mmol) and allowed to warm to ambient temperature overnight. The reaction mixture was treated with saturated NaHCO<sub>3</sub> solution and the layers were separated. The organic portion was washed successively with saturated NaHCO<sub>3</sub> solution, H<sub>2</sub>O and brine. The organic portion was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated under reduced pressure. Chromatography (SiO<sub>2</sub>, 10% EtOAc/hexanes) gave 6.27 g of *tert*-butyl 2-fluoro-2-methylpropylcarbamate as an amber oil which solidified on standing.

### Part C

tert-Butyl 2-fluoro-2-methylpropylcarbamate (6.27 g, 32.8 mmol) was treated with 45 mL of 3.0 M HCl in ethanol and the mixture was heated to 90 °C for 2 hours. The reaction mixture was then concentrated under reduced pressure to give 4.02 g of 2-fluoro-2-methylpropan-1-amine hydrochloride as a white solid.

#### Part D

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2-Fluoro-2-methylpropan-1-amine hydrochloride (4.02 g, 31.4 mmol) was dissolved in 80 mL of dry CH<sub>2</sub>Cl<sub>2</sub>. Triethylamine (13.1 mL, 94.2 mmol) and 4-chloro-3-nitro[1,5]naphthyridine (6.56 g, 31.4 mmol) were then added and the reaction was stirred under N<sub>2</sub> for 2 days. The reaction mixture was then concentrated under reduced pressure to give a dark-yellow solid. The solid was treated with 200 mL of H<sub>2</sub>O and the mixture was heated to reflux with rapid stirring. The mixture was cooled and the yellow solid was isolated by filtration. The material was washed with H<sub>2</sub>O and the dried under vacuum to give N-(2-fluoro-2-methylpropyl)-3-nitro[1,5]naphthyridin-4-amine (8.36 g) as a yellow powder.

# Part E

A solution of N-(2-fluoro-2-methylpropyl)-3-nitro[1,5]naphthyridin-4-amine (2.64 g, 10.0 mmol) dissolved in 80 mL of acetonitrile was placed in a pressure bottle. Platinum on carbon (5%, 500 mg) was then added and the reaction mixture was shaken under  $H_2$  at 50 PSI (3.4 x  $10^5$  Pa). After 5 hours, the reaction mixture was filtered through a pad of CELITE filter agent. The pad was rinsed with acetonitrile and the combined filtrates were concentrated under reduced pressure to give 2.12 g of  $N^4$ -(2-fluoro-2-methylpropyl)[1,5]naphthyridine-3,4-diamine as a brown foam.

Part F

N<sup>4</sup>-(2-Fluoro-2-methylpropyl)[1,5]naphthyridine-3,4-diamine (2.12 g, 9.06 mmol) was dissolved in 90 mL of anhydrous CH<sub>2</sub>Cl<sub>2</sub> and the stirred solution was cooled to 0 °C under N<sub>2</sub>. Triethylamine (1.39 mL, 10.0 mmol) and acetoxyacetyl chloride (1.07 mL, 10.0 mmol) were then added and the reaction mixture was stirred overnight. The reaction mixture was concentrated under reduced pressure and the resulting material was dissolved in 90 mL of ethanol and treated with 5 mL of triethylamine. The mixture was heated at reflux for 4 days. The reaction mixture was then cooled and concentrated under reduced pressure to give a purple solid. The purple solid was partitioned between CH<sub>2</sub>Cl<sub>2</sub> (75 mL)

and H<sub>2</sub>O (75 mL). The layers were separated and the aqueous portion was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 x 20 mL). The combined organic portions were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated under reduced pressure to give a purple solid. The resulting material was dissolved in 50 mL of methanol and treated with 1 mL of saturated aqueous K<sub>2</sub>CO<sub>3</sub> solution. After 1 hour, the mixture was treated with 3.5% NaH<sub>2</sub>PO<sub>4</sub> solution and the methanol was removed by evaporation under reduced pressure. A brown solid precipitated out of the aqueous solution and was isolated by filtration. The brown solid was rinsed with H<sub>2</sub>O and then dried to give 1.81 g of [1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-c][1,5]naphthyridin-2-yl]methanol.

## 10 Part G

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A solution of [1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-2-yl]methanol (1.53 g, 5.58 mmol) dissolved in 50 mL of CH<sub>2</sub>Cl<sub>2</sub> was treated with triethylamine (1.55 mL, 11.2 mmol), acetic anhydride (663 μL, 6.70 mmol), and 10 mg of 4-(dimethylamino)pyridine (DMAP). After stirring for 2 hours, the reaction mixture was treated with saturated NaHCO<sub>3</sub> solution and the layers were separated. The organic portion was washed successively with 3.5% NaH<sub>2</sub>PO<sub>4</sub> solution, H<sub>2</sub>O and brine. The organic portion was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated under reduced pressure. Chromatography (SiO<sub>2</sub>, 40-60% acetone/hexanes) gave 1.59 g of [1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-2-yl]methyl acetate as an off-white powder.

## Part H

[1-(2-Fluoro-2-methylpropyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-2-yl]methyl acetate (1.59 g, 5.03 mmol) was dissolved in 50 mL of CH<sub>2</sub>Cl<sub>2</sub> and treated with 3-chloroperoxybenzoic acid (1.52 g, 57-86% purity). After stirring for 2 hours, the reaction mixture was treated with 25 mL of CH<sub>2</sub>Cl<sub>2</sub> and 20 mL of 5% Na<sub>2</sub>CO<sub>3</sub> solution and the layers were separated. The organic layer was washed with H<sub>2</sub>O (20 mL) and brine (20 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure to give 1.67 g of [1-(2-fluoro-2-methylpropyl)-5-oxido-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-2-yl]methyl acetate as an off-white solid.

### 30 Part I

[1-(2-Fluoro-2-methylpropyl)-5-oxido-1H-imidazo[4,5-c][1,5]naphthyridin-2-yl]methyl acetate (1.67 g, 5.03 mmol) was dissolved in 50 mL of CH<sub>2</sub>Cl<sub>2</sub> and treated with

5 mL of concentrated aqueous NH<sub>4</sub>OH solution. The mixture was stirred rapidly and then p-toluenesulfonyl chloride (1.05 g, 5.53 mmol) was carefully added. Rapid stirring was continued for 1 hour. The reaction mixture was then treated with 20 mL of H<sub>2</sub>O. The layers were separated and the organic portion was washed successively with 5% Na<sub>2</sub>CO<sub>3</sub> solution, H<sub>2</sub>O and brine. The organic portion was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure. Chromatography (SiO<sub>2</sub>, 2.5% methanol/CHCl<sub>3</sub>) gave 1.13 g of [4-amino-1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-c][1,5]naphthyridin-2-yl]methyl acetate as a light-yellow solid.

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A solution of [4-amino-1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-c][1,5]naphthyridin-2-yl]methyl acetate (1.13 g, 3.41 mmol) dissolved in 10 mL of methanol was treated with 10 mL of a 7% solution of ammonia in methanol. The mixture was stirred for 2 hours and then concentrated under reduced pressure. The resulting solid was treated with H<sub>2</sub>O and the mixture was heated to reflux for 15 minutes. The mixture was cooled and the resulting light-yellow solid was isolated by filtration. The light-yellow solid was then treated with 20 mL of CH<sub>2</sub>Cl<sub>2</sub> and the mixture was stirred rapidly for several minutes. The mixture was filtered and the resulting white solid was washed with several portions of cold CH<sub>2</sub>Cl<sub>2</sub> and dried with suction. Crystallization from ethanol/H<sub>2</sub>O gave 477 mg of [4-amino-1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-

c][1,5]naphthyridin-2-yl]methanol as fluffy cream colored crystals, mp 240 – 241 °C. 

<sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  8.50 (dd, J = 1.5, 4.3 Hz, 1H), 7.93 (dd, J = 1.5, 8.4 Hz, 1H), 7.46 (dd, J = 4.3, 8.4 Hz, 1H), 6.92 (s, 2H), 5.62 (t, J = 5.8 Hz, 1H), 5.33 (br s, 2H), 4.83 (d, J = 4.7 Hz, 2H), 1.33 (d, J = 20.3 Hz, 6H); <sup>13</sup>C NMR (125 MHz, DMSO- $d_6$ )  $\delta$  154.3, 152.7, 143.1, 140.8, 134.2, 133.4, 133.2, 128.7, 122.5, 96.8 (d, J = 170 Hz), 56.7 (d, J = 9.5 Hz), 52.7 (d, J = 21.4 Hz), 24.5; MS (ESI) m/z 290 (M + H)<sup>+</sup>; Anal. calcd for  $C_{14}H_{16}FN_5O$ : C, 58.12; H, 5.57; N, 24.21. Found: C, 58.19; H, 5.54; N, 24.16.

## Example 552

2-[4-Amino-1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-2-yl]ethanol

Part A

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 $N^4$ -(2-Fluoro-2-methylpropyl)[1,5]naphthyridine-3,4-diamine (2.34 g, 10.0 mmol) was dissolved in 80 mL of anhydrous CH<sub>2</sub>Cl<sub>2</sub> and the stirred solution was cooled to 0 °C under N<sub>2</sub>. Triethylamine (2.78 mL, 10.0 mmol) and 3-(benzyloxy)propanoyl chloride, prepared by the method of Li, J. Med. Chem., 42, pp. 706-721, (2.13 g, 10.0 mmol), were then added and the reaction mixture was stirred overnight. The reaction mixture was concentrated under reduced pressure. The resulting material was dissolved in 80 mL of ethanol and combined with 5 mL of triethylamine and the mixture was heated to reflux for 4 days. The reaction mixture was then cooled and concentrated under reduced pressure. The resulting solid was partitioned between CH<sub>2</sub>Cl<sub>2</sub> (75 mL) and H<sub>2</sub>O (75 mL). The layers were separated and the organic portion was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure to give a solid. Chromatography (SiO<sub>2</sub>, 1-2% CMA/CHCl<sub>3</sub>) gave 0.83 g of uncyclized amide (3-(benzyloxy)-N-{4-[(2-fluoro-2methylpropyl)amino][1,5]naphthyridin-3-yl}propanamide) and the desired 2-[2-(benzyloxy)ethyl]-1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridine. Additional chromatography (10% methanol/CHCl<sub>3</sub>) of the desired material gave 1.39 g of a light-orange syrup. The isolated amide was converted to the desired imidazole by dissolving the material in 10 mL of 7% ammonia in methanol. The mixture was placed in a stainless-steel pressure vessel and the vessel was sealed and heated to 150 °C overnight. The reaction mixture was cooled and concentrated under reduced pressure. Chromatography (SiO<sub>2</sub>, 2% CMA/CHCl<sub>3</sub>) gave 0.50 g of the desired product which was combined with the first batch of material for the next reaction. Part B

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2-[2-(Benzyloxy)ethyl]-1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5c][1,5]naphthyridine (1.89 g, 5.0 mmol) was dissolved in 50 mL of CH<sub>2</sub>Cl<sub>2</sub> and treated with 3-chloroperoxybenzoic acid (1.50 g, 57-86% purity). After stirring for 2 hours, the reaction mixture was treated with 50 mL of 2% Na<sub>2</sub>CO<sub>3</sub> solution and the layers were separated. The aqueous portion was extracted with an additional 25 mL of CH<sub>2</sub>Cl<sub>2</sub>. The

combined organic layers were washed successively with 2% Na<sub>2</sub>CO<sub>3</sub>, H<sub>2</sub>O and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure to give 1.97 g of 2-[2-(benzyloxy)ethyl]-1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridine 5-oxide as an off-white solid.

### 5 Part C

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2-[2-(Benzyloxy)ethyl]-1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridine 5-oxide (1.97 g, 5.00 mmol) was dissolved in 50 mL of CH<sub>2</sub>Cl<sub>2</sub> and treated with 5 mL of concentrated aqueous NH<sub>4</sub>OH solution. The mixture was stirred rapidly and then *p*-toluenesulfonyl chloride (1.00 g, 5.33 mmol) was carefully added. Rapid stirring was continued for 1 hour. The reaction mixture was then treated with 20 mL of H<sub>2</sub>O. The layers were separated and the organic portion was washed successively with 5% Na<sub>2</sub>CO<sub>3</sub> solution, H<sub>2</sub>O and brine. The organic portion was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure. Chromatography (SiO<sub>2</sub>, 10% CMA/CHCl<sub>3</sub>) gave 0.90 g of 2-[2-(benzyloxy)ethyl]-1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-4-amine as a yellow solid.

A solution of 2-[2-(benzyloxy)ethyl]-1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-4-amine (0.78 g, 1.98 mmol) dissolved in 20 mL of methanol was treated with 10% palladium on carbon (200 mg) and 0.68 mL of 3 M HCl in ethanol. The mixture was shaken under H<sub>2</sub> at 50 PSI (3.4 x 10<sup>5</sup> Pa) overnight. Additional 10% palladium on carbon (200 mg) and 3 M HCl in ethanol (0.33 mL) were added and shaking was continued for 24 hours. The reaction mixture was filtered through a pad of CELITE filter agent. The pad was rinsed with methanol and the combined filtrates were concentrated under reduced pressure. The resulting material was treated with 20 mL of H<sub>2</sub>O and 2 mL of concentrated NH<sub>4</sub>OH solution and extracted into CHCl<sub>3</sub> (3 x 25 mL). The combined organic layers were concentrated under reduced pressure. Chromatography (SiO<sub>2</sub>, 15-30% CMA/CHCl<sub>3</sub>) gave a white powder. Crystallization from ethanol/H<sub>2</sub>O gave 276 mg of 2-[4-amino-1-(2-fluoro-2-methylpropyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-2-yl]ethanol as white needles, mp 224 – 225 °C.

<sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ , 354 K)  $\delta$  8.470 (dd, J = 1.3, 4.0 Hz, 1H), 7.91 (d, J = 8.4 Hz, 1H), 7.40 (dd, J = 4.1, 8.3 Hz, 1H), 6.46 (s, 2H), 5.25 (d, J = 22.7 Hz, 2H), 4.57 (s, 1H), 3.91 (d, J = 5.4 Hz, 2H), 3.14 (t, J = 6.4 Hz, 2H), 1.33 (d, J = 21.7 Hz, 6H); <sup>13</sup>C NMR

(125 MHz, DMSO- $d_6$ )  $\delta$  154.0, 152.3, 143.0, 140.4, 134.1, 133.1, 132.6, 129.0, 122.2, 96.7 (d, J = 170 Hz), 60.2, 52.5 (d, J = 20.9 Hz), 30.6 (d, J = 6.6 Hz), 24.4; MS (ESI) m/z 304 (M + H)<sup>+</sup>; Anal. calcd for C<sub>15</sub>H<sub>18</sub>FN<sub>5</sub>O: C, 59.39; H, 5.98; N, 23.09. Found: C, 59.57; H, 5.75; N, 23.07.

## Examples 553 - 593

Part A

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A solution of 1-(2-aminoethyl)-2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-4-amine (57 mg, 0.1 mmol, 1 eq, prepared according to the general method of Example 146 using methoxypropionyl chloride in lieu of methoxyacetyl chloride) and N,N-diisopropylethylamine (87  $\mu$ L) in N, N-dimethylacetamide (1 mL) was added to a tube containing a reagent (1.1 eq) from the table below. The reaction mixture was vortexed overnight, the reaction was quenched with water (2 drops), and the solvent was removed by vacuum centrifugation. The reaction mixture was purified by solid-supported liquid-liquid extraction according to the following procedure. The sample was dissolved in chloroform (1 mL) then loaded onto diatomaceous earth that had been equilibrated with 2 M sodium carbonate solution (600  $\mu$ L) for about 20 minutes. After 10 minutes chloroform (500  $\mu$ L) was added to elute the product from the diatomaceous earth into a well of a collection plate. After an additional 10 minutes the process was repeated with additional chloroform (500  $\mu$ L). The solvent was then removed by vacuum centrifugation.

Part B

The material from Part A was dissolved in dichloromethane (1 mL) and the solution was cooled to 0 °C. Boron tribromide (400  $\mu$ L of 1 M in dichloromethane) was added and the reaction mixture was vortexed overnight. Methanol (1 mL) and 6 N hydrochloric acid (500  $\mu$ L) were added and the reaction mixture was vortexed for 15 minutes. The solvent was removed by vacuum. The compounds were purified as described for Examples 150 – 155. The table below shows the reagent used for each example, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

	NH <sub>2</sub> N OH					
Example	Reagent	R	Measured Mass (M+H)			
553	None	_H	273.1479			
554	Cyclopropanecarbonyl chloride	0	341.1730			
555	Isobutyryl chloride	H³C CH³	343.1909			
556	Cyclobutanecarbonyl chloride		355.1909			
557	Cyclopentanecarbonyl chloride		369.2062			
558	Benzoyl chloride		377.1747			
559	Cyclohexanecarbonyl chloride	( ) ( )	383.2206			
560	3-Cyanobenzoyl chloride	O	402.1702			
561	4-Cyanobenzoyl chloride		402.1700			

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562	Cinnamoyl chloride		403.1890
563	Hydrocinnamoyl chloride		405.2044
564	2-Methoxybenzoyl chloride	ООН	393.1672
565	3-Methoxybenzoyl chloride	ОН	393.1689
566	p-Anisoyl chloride	O D E	393.1678
567	2-Chlorobenzoyl chloride	CI	411.1306
568	3-Chlorobenzoyl chloride	CI	411.1369
569	4-Chlorobenzoyl chloride	CI	411.1368
570	Isonicotinoyl chloride hydrochloride	O	378.1698

571	Nicotinoyl chloride hydrochloride	O N	378.1676
572	Methanesulfonyl chloride	−ş,- <sub>CH3</sub>	351.1256
573	Ethanesulfonyl chloride	O.O CH <sub>3</sub>	365.1386
574	1-Propanesulfonyl chloride	O,O S CH₃	379.1534
575	Dimethylsulfamoyl chloride	O,O S,N-CH <sub>3</sub>	380.1512
576	Benzenesulfonyl chloride	0.0	413.1436
577	1-Methylimidazole-4- sulfonyl chloride	O.O S N NCH <sub>3</sub>	417.1462
578	2,2,2- Trifluoroethanesulfonyl chloride	O F F F	419.1139
579	alpha-Toluenesulfonyl chloride		427.1569
580	3-Cyanobenzenesulfonyl chloride	0.0	438.1380

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581	3- Methoxybenzenesulfonyl chloride	HO	429.1349
582	2-Chlorobenzenesulfonyl chloride	O,O CI	447.0996
583	4-Chlorobenzenesulfonyl chloride	O, O	447.1031
584	Isopropyl isocyanate	O CH <sub>3</sub> H CH <sub>3</sub>	358.1994
585	Phenyl isocyanate	NH NH	392.1794
586	Cyclohexyl isocyanate	HZ O	398.2305
587	(R)-(+)-alpha- Methylbenzyl isocyanate	NCH3	420.2178
588	(S)-(-)-alpha- Methylbenzyl isocyanate	O N CH <sub>3</sub>	420.2149
589	2-Chlorophenyl isocyanate	O N CI	426.1453

590	4-Chlorophenyl isocyanate	N CI	426.1460
591	N,N-Dimethylcarbamoyl chloride	H <sub>3</sub> C. <sup>N-</sup> CH <sub>3</sub>	344.1856
592	1-Piperidinecarbonyl chloride	\( \sqrt{N} \)	384.2137
593	4-Morpholinylcarbonyl chloride	0 N	386.1976

Examples 594 – 632

The compounds in the table below were prepared and purified according to the methods described in Examples 553 - 593 using 1-(2-aminoethyl)-2-methoxymethyl-1H-imidazo[4,5-c][1,5]naphthyridin-4-amine in lieu of 1-(2-aminoethyl)-2-(2-methoxyethyl)-1H-imidazo[4,5-c][1,5]naphthyridin-4-amine. The table below shows the reagent used for each example, the structure of the resulting compound, and the observed accurate mass for the isolated trifluoroacetate salt.

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		N OH N-R	
Example	Reagent	R	Measured Mass (M+H)
594	Cyclopropanecarbonyl chloride	o	327.1581
595	Isobutyryl chloride	H <sub>3</sub> C CH <sub>3</sub>	329.1709

596	Cyclopentanecarbonyl chloride	To Control of the Con	355.1859
597	Benzoyl chloride	°	363.1563
598	Cyclohexanecarbonyl chloride	°	369,2019
599	3-Cyanobenzoyl chloride	O	388.1517
600	Hydrocinnamoyl chloride		391.1868
601	3-Methoxybenzoyl chloride	ОН	379.1512
602	p-Anisoyl chloride	ОН	379.1526
603	2-Chlorobenzoyl chloride	CI	397.1193
604	3-Chlorobenzoyl chloride	CI	397.1198
605	Isonicotinoyl chloride hydrochloride	O	364.1515

606	Nicotinoyl chloride hydrochloride	o	364.1535
607	Picolinoyl chloride hydrochloride	0 2	364.1512
608	trans-2-Phenyl-1- cyclopropanecarbonyl chloride	Qui	403.1852
609	Methanesulfonyl chloride	,O S,-CH <sub>3</sub> ,	337.1070
610	Ethanesulfonyl chloride	O CH <sub>3</sub>	351.1212
611	1-Propanesulfonyl chloride	O CH3	365.1386
612	Isopropylsulfonyl chloride	H <sub>3</sub> C CH <sub>3</sub>	365.1433
613	Dimethylsulfamoyl chloride	O,O S,N-CH <sub>3</sub>	366.1355
614	Benzenesulfonyl chloride	0,0	399.1214
615	1-Methylimidazole-4- sulfonyl chloride	O,O S N NCH <sub>3</sub>	403.1311
616	2,2,2- Trifluoroethanesulfonyl chloride	O S F F F	405.0953

617	3- Cyanobenzenesulfonyl chloride	O O	424.1229
618	2- Chlorobenzenesulfonyl chloride	0.0 CI	433.0872
619	3- Chlorobenzenesulfonyl chloride	0,0 /s	433.0867
620	4- Chlorobenzenesulfonyl chloride	0.00	433.0853
621	Methyl isocyanate	O N-CH3	316.1528
622	Ethyl isocyanate	O N CH₃	330.1660
623	Isopropyl isocyanate	O N CH₃ H CH₃	344.1819
624	n-Propyl isocyanate	N ← CH₃	344.1809
625	Cyclopentyl isocyanate	H \	370.1994
626	Cyclohexyl isocyanate	N (	384.2152

627	3-Chlorophenyl isocyanate		412.1300
628	4-Chlorophenyl isocyanate	N CI	412.1273
629	N,N- Dimethylcarbamoyl chloride	H <sub>3</sub> C,N-CH <sub>3</sub>	330.1686
630	1-Piperidinecarbonyl chloride		370.1979
631	4-Morpholinylcarbonyl chloride	0 N O	372.1811
632	4-Methyl-1- piperazinecarbonyl chloride	O N CH <sub>3</sub>	385.2098

Example 633
[4-Amino-1-(2-methylpropyl)-1*H*-imidazo[4,5-*c*][1,5]naphthyridin-2-yl]methanol

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To a chilled solution (ice bath) of 2-(ethoxymethyl)-1-(2-methylpropyl)-1*H*-imidazo[4,5-c][1,5]naphthyridin-4-amine (2.0 g, 6.69 mmol, prepared according to the general methods of Example 148 using 2-methylpropan-1-amine in lieu of 1-amino-2-methylpropan-2-ol) in dichloromethane (50 mL) was added boron tribromide (20 mL, 1M solution in dichloromethane). The mixture turned light purple and was stirred at ambient temperature for 44 hours. The reaction was quenched with methanol (20 mL) and aqueous

hydrochloric acid (6N, 10 mL). After stirring for 4 hours, the pH was adjusted to 10 by the addition of aqueous sodium hydroxide (50%). Dichloromethane (50 mL) was added with stirring and the layers were separated. The aqueous fraction was extracted with chloroform (2x 250 mL). The combined organic fractions were concentrated to provide 1.4 g of [4-amino-1-(2-methylpropyl)-1*H*-imidazo[4,5-c][1,5]naphthyridin-2-yl]methanol as a white powder, mp 226-228 °C.

<sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  8.52-8.51 (dd, J = 1.6, 4.3 Hz, 1H), 7.93-7.89 (dd, J = 1.6, 8.4 Hz, 1H), 7.46-7.42 (dd, J = 4.3, 8.4 Hz, 1H), 6.83 (s, 2H), 5.69-5.65 (t, J = 5.8 Hz, 1H), 4.79-4.77 (d, J = 5.8 Hz, 2H), 4.74-4.71 (d, J = 7.6 Hz, 2H), 2.44-2.39 (m, 1H), 0.91-0.88 (d, J = 6.7 Hz, 6H);

Anal. calcd for  $C_{14}H_{17}N_5O$ : C, 61.98; H, 6.31; N, 25.81. Found: C, 61.26; H, 6.07; N, 25.75.

### Example 634

2-(4-Amino-1-isobutyl-1*H*-imidazo[4,5-c]quinolin-2-yl)ethyl acetate

Part A

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To a stirred suspension of  $N^4$ -isobutylquinoline-3,4-diamine (13.0 g, 60.46 mmol) in toluene was added pyridine hydrochloride (2.1 g, 8.14 mmol) followed by 3-chloropropionyl chloride (1.1 equivalents). The creamy suspension was stirred for 4 hours at ambient temperature and the solvent was then evaporated under reduced pressure. The tan solid obtained was dissolved in chloroform and the solution was transferred to a separatory funnel. The organic layer was washed with water (1x) and brine (2x). The organic layer was separated, dried (MgSO<sub>4</sub>), filtered, and the solvent was evaporated under reduced pressure to afford a tan solid (21 g). A portion of the tan solid (7 g) was taken up in acetic acid (110 mL) and the reaction was stirred at ambient temperature for 4 hours. The reaction was cooled in an ice-bath and 6M NaOH (300 mL) was added in portions to afford a creamy suspension. The reaction mixture was transferred to a separatory funnel and the product was extracted with chloroform (150 mL x 2). The

organic layers were combined, dried (MgSO<sub>4</sub>), and filtered; and the filtrate was evaporated under reduced pressure to afford 2-(1-isobutyl-1*H*-imidazo[4,5-*c*]quinolin-2-yl)ethyl acetate as a brown oil (10 g) which was taken forward to the next step.

Part B

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To a stirred solution of 2-(1-isobutyl-1*H*-imidazo[4,5-*c*]quinolin-2-yl)ethyl acetate (7.41 g, 22.9 mmol) in chloroform was added 3-chloroperoxybenzoic acid (77%, 10.3 g, 49.9 mmol) and the reaction was stirred at ambient temperature for 4 hours. The reaction was then transferred to a separatory funnel and washed with brine (2 x). The organic layers were combined, dried (MgSO<sub>4</sub>), and filtered; and the filtrate was evaporated under reduced pressure to afford the N-oxide (15 g) as a brown solid. The brown solid was dissolved in chloroform, cooled in an ice-bath, and trichloroacetyl isocyanate (6.4 g mL, 34.3 mmol) was added in a dropwise manner. The reaction was stirred for 1 hour after which an additional 1.5 equivalents of trichloroacetyl isocyanate was added and the reaction was stirred at ambient temperature overnight. The reaction was concentrated under reduced pressure and the residue was suspended in ethanol. Potassium ethoxide was added to this suspension and the reaction was stirred for 1 hour. The reaction was concentrated under reduced pressure, taken up in dichloromethane (250 mL) and transferred to a separatory funnel. The organic layer was washed with water (250 mL), separated from the aqueous, dried (MgSO<sub>4</sub>), and filtered; and the filtrate was evaporated under reduced pressure to afford a brown solid. The product was isolated by prep HPLC (ISCO Combiflash Separation System, Biotage Si 40+M column, eluted with a gradient of 0-10% methanol in dichloromethane with 1% ammonium hydroxide) to provide a solid (about 7g). A portion of this solid was recrystallized from acetonitrile to afford 2-(4amino-1-isobutyl-1H-imidazo[4,5-c]quinolin-2-yl)ethyl acetate as a white solid (0.109 g), mp 187-189°C; MS (ESI) m/z 327 (M+H); Anal. Calcd for C<sub>18</sub>H<sub>22</sub>N<sub>4</sub>O<sub>2</sub>•0.40H<sub>2</sub>O: C, 64.81; H. 6.89; N. 16.79; Found C, 64.54; H, 6.46; N, 16.90.

#### Example 635

[4-Amino-1-(2-hydroxy-2-methylpropyl)-1*H*-imidazo[4,5-c]quinolin-2-yl]methyl acetate

To a round-bottomed flask with stir bar was added 1-(4-amino-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1-yl)-2-methylpropan-2-ol (1 g, 3.5 mmol) followed by dichloroethane (20 mL) and pyridine (3 mL). To the stirred suspension was added acetyl chloride (0.27 mL, 1.1 equivalents) and the reaction was stirred at ambient temperature for 30 min. The solvent was evaporated under reduced pressure to afford a solid. The product was isolated by two purifications by prep HPLC (ISCO Combiflash Separation System, Biotage Si 40+M column, eluted with a gradient of 0-7% methanol in dichloromethane with 1% ammonium hydroxide) to provide [4-amino-1-(2-hydroxy-2-methylpropyl)-1*H*-imidazo[4,5-*c*]quinolin-2-yl]methyl acetate as an off white solid (130 mg), mp 203-205°C; MS (ESI) *m/z* 329 (M+H); Anal. Calcd for C<sub>17</sub>H<sub>20</sub>N<sub>4</sub>O<sub>3</sub>\*0.25CH<sub>4</sub>O: C, 61.59; H, 6.29; N, 16.66; Found C, 61.24; H, 6.22; N, 16.97.

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#### Example 636

[4-Amino-1-(2-hydroxy-2-methylpropyl)-1*H*-imidazo[4,5-*c*]quinolin-2-yl]methyl L-valinate

[4-Amino-1-(2-hydroxy-2-methylpropyl)-1*H*-imidazo[4,5-*c*]quinolin-2-yl]methyl L-valinate was prepared according to the general method used to prepare (4-amino-1-{4-20 {(methylsulfonyl)amino]butyl}-1*H*-imidazo[4,5-*c*]quinolin-2-yl)methyl L-valinate using 1-(4-amino-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1-yl)-2-methylpropan-2-ol in lieu of *N*-[4-(4-amino-2-hydroxymethyl-1*H*-imidazo[4,5-*c*]quinolin-1-yl)butyl]methanesulfonamide. The product was provided as off-white needles, mp 190-

192°C; MS (ESI) m/z 386 (M+H); Anal. Calcd for C<sub>20</sub>H<sub>27</sub>N<sub>5</sub>O<sub>3</sub>: C, 62.32; H, 7.06; N, 18.17; Found C, 62.08; H, 7.11; N, 17.96.

# 5 Exemplary Compounds Useful in Practicing Methods of the Invention

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Certain exemplary compounds, including some of those described above in the Examples, have the following Formulas Ia, Ib, Ic, Id, Ie, If, Ig, Ih, Ii, Ij, Ik, Im, In, Io, or Ip and the following substituents n and R<sub>1</sub> wherein each line of the table is matched to Formula Ia, Ib, Ic, Id, Ie, If, Ig, Ih, Ii, Ij, Ik, Im, In, Io, or Ip to represent a specific compound which is useful in practicing methods of the invention.

$$\begin{array}{c|c} & NH_2 \\ & N \\$$

$$\begin{array}{c|c}
 & NH_2 \\
 & N \\
 &$$

n	$R_1$
1	2-[(cyclohexylcarbonyl)amino]-2-methylpropyl
1	2-[(cyclopropylcarbonyl)amino]ethyl
1	4-[(cyclopropylcarbonyl)amino]butyl
1	2,3-dihydroxypropyl
1	2,2-dimethyl-3-(methylsulfonyl)propyl
1	2-fluoro-2-methylpropyl
1	2-hydroxy-2-methylpropyl
1	2-methylpropyl
1	2-methyl-2-({[(1-methylethyl)amino]carbonyl}amino)propyl
1	2-{[(1-methylethyl)carbonyl]amino}ethyl
1	4-{[(1-methylethyl)carbonyl]amino}butyl
1	2-methyl-2-[(methylsulfonyl)amino]propyl
1	4-[(methylsulfonyl)amino]butyl
1	2-[(methylsulfonyl)amino]ethyl
1	4-[(4-morpholinecarbonyl)amino]butyl

1	2-[(4-morpholinecarbonyl)amino]ethyl
1	tetrahydro-2 <i>H</i> -pyran-4-ylmethyl
1	(4-hydroxytetrahydro-2 <i>H</i> -pyran-4-yl)methyl
1	(1-hydroxycyclobutyl)methyl
1	(1-hydroxycyclopentyl)methyl
1	(1-hydroxycyclohexyl)methyl
2	2-[(cyclohexylcarbonyl)amino]-2-methylpropyl
2	2-[(cyclopropylcarbonyl)amino]ethyl
2	4-[(cyclopropylcarbonyl)amino]butyl
2	2,3-dihydroxypropyl
2	2,2-dimethyl-3-(methylsulfonyl)propyl
2	2-fluoro-2-methylpropyl
2	2-hydroxy-2-methylpropyl
2	2-methylpropyl
2	2-methyl-2-({[(1-methylethyl)amino]carbonyl}amino)propyl
2	2-{[(1-methylethyl)carbonyl]amino}ethyl
2	4-{[(1-methylethyl)carbonyl]amino}butyl
2	2-methyl-2-[(methylsulfonyl)amino]propyl
2	4-[(methylsulfonyl)amino]butyl
2	2-[(methylsulfonyl)amino]ethyl
2	4-[(4-morpholinecarbonyl)amino]butyl
2	2-[(4-morpholinecarbonyl)amino]ethyl
2	tetrahydro-2H-pyran-4-ylmethyl
2	(4-hydroxytetrahydro-2 <i>H</i> -pyran-4-yl)methyl
2	(1-hydroxycyclobutyl)methyl
2	(1-hydroxycyclopentyl)methyl
2	(1-hydroxycyclohexyl)methyl

## CYTOKINE INDUCTION IN HUMAN CELLS

An in vitro human blood cell system is used to assess cytokine induction. Activity is based on the measurement of interferon (α) and tumor necrosis factor (α) (IFN-α and TNF-α, respectively) secreted into culture media as described by Testerman et. al. in "Cytokine Induction by the Immunomodulators Imiquimod and S-27609", Journal of Leukocyte Biology, 58, 365-372 (September, 1995).

# Blood Cell Preparation for Culture

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Whole blood from healthy human donors is collected by venipuncture into vacutainer tubes or syringes containing EDTA. Peripheral blood mononuclear cells (PBMC) are separated from whole blood by density gradient centrifugation using HISTOPAQUE-1077 (Sigma, St. Louis, MO) or Ficoll-Paque Plus (Amersham Biosciences Piscataway, NJ). Blood is diluted 1:1 with Dulbecco's Phosphate Buffered Saline (DPBS) or Hank's Balanced Salts Solution (HBSS). Alternately, whole blood is placed in Accuspin (Sigma) or LeucoSep (Greiner Bio-One, Inc., Longwood, FL) centrifuge frit tubes containing density gradient medium. The PBMC layer is collected and washed twice with DPBS or HBSS and re-suspended at 4 x 10<sup>6</sup> cells/mL in RPMI complete. The PBMC suspension is added to 96 well flat bottom sterile tissue culture plates containing an equal volume of RPMI complete media containing test compound.

Compound Preparation

The compounds are solubilized in dimethyl sulfoxide (DMSO). The DMSO concentration should not exceed a final concentration of 1% for addition to the culture wells. The compounds are generally tested at concentrations ranging from 30-0.014  $\mu$ M. Controls include cell samples with media only, cell samples with DMSO only (no compound), and cell samples with reference compound.

### Incubation

The solution of test compound is added at 60  $\mu$ M to the first well containing RPMI complete and serial 3 fold dilutions are made in the wells. The PBMC suspension is then added to the wells in an equal volume, bringing the test compound concentrations to the desired range (usually 30-0.014  $\mu$ M). The final concentration of PBMC suspension is 2 x

10<sup>6</sup> cells/mL. The plates are covered with sterile plastic lids, mixed gently and then incubated for 18 to 24 hours at 37°C in a 5% carbon dioxide atmosphere.

## Separation

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Following incubation the plates are centrifuged for 10 minutes at 1000 rpm (approximately 200 x g) at 4°C. The cell-free culture supernatant is removed and transferred to sterile polypropylene tubes. Samples are maintained at -30 to -70°C until analysis. The samples are analyzed for IFN- $\alpha$  by ELISA and for TNF- $\alpha$  by IGEN/BioVeris Assay.

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Interferon (a) and Tumor Necrosis Factor (a) Analysis

IFN-α concentration is determined with a human multi-subtype colorimetric sandwich ELISA (Catalog Number 41105) from PBL Biomedical Laboratories, Piscataway, NJ. Results are expressed in pg/mL.

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The TNF-α concentration is determined by ORIGEN M-Series Immunoassay and read on an IGEN M-8 analyzer from BioVeris Corporation, formerly known as IGEN International, Gaithersburg, MD. The immunoassay uses a human TNF-α capture and detection antibody pair (Catalog Numbers AHC3419 and AHC3712) from Biosource International, Camarillo, CA. Results are expressed in pg/mL.

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Assay Data and Analysis

In total, the data output of the assay consists of concentration values of TNF- $\alpha$  and IFN- $\alpha$  (y-axis) as a function of compound concentration (x-axis).

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Analysis of the data has two steps. First, the greater of the mean DMSO (DMSO control wells) or the experimental background (usually 20 pg/mL for IFN- $\alpha$  and 40 pg/mL for TNF- $\alpha$ ) is subtracted from each reading. If any negative values result from background subtraction, the reading is reported as " \* ", and is noted as not reliably detectable. In subsequent calculations and statistics, " \* ", is treated as a zero. Second, all background subtracted values are multiplied by a single adjustment ratio to decrease experiment to experiment variability. The adjustment ratio is the area of the reference compound in the new experiment divided by the expected area of the reference compound based on the past 61 experiments (unadjusted readings). This results in the scaling of the

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reading (y-axis) for the new data without changing the shape of the dose-response curve. The reference compound used is 2-[4-amino-2-ethoxymethyl-6,7,8,9-tetrahydro- $\alpha$ , $\alpha$ -dimethyl-1H-imidazo[4,5-c]quinolin-1-yl]ethanol hydrate (U.S. Patent No. 5,352,784; Example 91) and the expected area is the sum of the median dose values from the past 61 experiments.

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The minimum effective concentration is calculated based on the background-subtracted, reference-adjusted results for a given experiment and compound. The minimum effective concentration (μmolar) is the lowest of the tested compound concentrations that induces a response over a fixed cytokine concentration for the tested cytokine (usually 20 pg/mL for IFN-α and 40 pg/mL for TNF-α). The maximal response (pg/mL) is the maximal response attained in the dose response curve.

Compounds used in the methods of the invention and close analogs were tested for their ability to induce cytokine biosynthesis using the test method described above. The analogs used are shown in the table below.

Analog	Chemical Name	Reference
1	N-[2-(4-Amino-2-methyl-1H-imidazo[4,5-c]quinolin-1-	U.S. Patent 6,677,349#
	yl)-1,1-dimethylethyl]methanesulfonamide	
2	N-[2-(4-Amino-2-ethyl-1H-imidazo[4,5-c]quinolin-1-	U.S. Patent 6,677,349#
	yl)-1,1-dimethylethyl]methanesulfonamide	
3	N-[2-(4-Amino-2-propyl-1H-imidazo[4,5-c]quinolin-1-	U.S. Patent 6,677,349#
	yl)-1,1-dimethylethyl]methanesulfonamide	
4	N-[2-(4-Amino-2-ethoxymethyl-1H-imidazo[4,5-	U.S. Patent 6,677,349
<u> </u>	c]quinolin-1-yl)-1,1-dimethylethyl]methanesulfonamide	Example 268
5	$N-\{2-[4-Amino-2-(2-methoxyethyl)-1H-imidazo[4,5-$	Example 6 Part D
	c]quinolin-1-yl]-1,1-	_
	dimethylethyl}methanesulfonamide	į.

"This compound is not specifically exemplified but can be readily prepare using the synthetic methods disclosed in the cited reference

The compounds of Examples 6 and 7 and several closely related analogs were tested using the test method described above. The IFN-α dose response curves for Example 6, Analog 2, Analog 3 and Analog 5 are shown in Figure 1. The TNF-α dose response curves for Example 6, Analog 2, Analog 3 and Analog 5 are shown in Figure 2. The IFN-α dose response curves for Example 7, Analog 1, Analog 2 and Analog 4 are shown in Figure 3. The TNF-α dose response curves for Example 7, Analog 1, Analog 2 and Analog 4 are shown in Figure 4. The minimum effective concentration for the

induction of IFN- $\alpha$ , minimum effective concentration for the induction of TNF- $\alpha$ , the maximal response for IFN- $\alpha$ , and the maximal response for TNF- $\alpha$  are shown in Table 7 below where # is the number of separate experiments in which the compound was tested. When a compound was tested in more than one experiment the values shown are the median values.

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Table 7

Table /						
NH <sub>2</sub> N R <sub>2</sub> N N <sub>2</sub> N N <sub>3</sub> N N <sub>4</sub> N N <sub>5</sub> N N N <sub>5</sub> N N N <sub>5</sub> N N N <sub>5</sub> N N N <sub>5</sub> N N N N N N N N N N N N N N N N N N N						
Compound	R <sub>2</sub>	1	Effective ation (µM)		Response /mL) TNF	#
Example 7	-CH₂OH	3.330	30.00	2250	121	5
Example 6	-(CH <sub>2</sub> ) <sub>2</sub> OH	1.11	>30	7521	-	3
Analog 1	-CH <sub>3</sub>	0.370	3.330	1846	1518	7
Analog 2	-CH₂CH₃	0.120	1.110	831	3670	4
Analog 3	-(CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	0.120	0.370	832	7245	9
Analog 4	-CH <sub>2</sub> OCH <sub>2</sub> CH <sub>3</sub>	0.040	0.370	889	10125	22
Analog 5	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.12	825	12518	6

Further compounds used in the methods of the invention and close analogs were tested for their ability to induce cytokine biosynthesis using the test method described above. The analogs used are shown in the table below.

Analog	Chemical Name	Reference
6	1-(4-amino-2-ethoxymethyl-1 <i>H</i> -imidazo[4,5-	Example 148 Part E
	c][1,5]naphthyridin-1-yl)-2-methylpropan-2-ol	
7	1-[4-amino-2-(2-methoxyethyl)-1H-imidazo[4,5-	Example 149 Part J
	c][1,5]naphthyridin-1-yl]-2-methylpropan-2-ol	_

The compounds of Examples 148 and 149 and several closely related analogs were tested using the test method described above. The IFN-α dose response curves for Example 148, Example 149, Analog 6, and Analog 7 are shown in Figure 5. The TNF-α

dose response curves for Example 148, Example 149, Analog 6, and Analog 7 are shown in Figure 6. The minimum effective concentration for the induction of IFN- $\alpha$ , minimum effective concentration for the induction of TNF- $\alpha$ , the maximal response for IFN- $\alpha$ , and the maximal response for TNF- $\alpha$  are shown in Table 8 below where # is the number of separate experiments in which the compound was tested. When a compound was tested in more than one experiment the values shown are the median values.

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Table 8

NH <sub>2</sub> N R <sub>2</sub> OH						
Compound	R <sub>2</sub>	Minimum Concentra			Response mL)	#
		IFN	TNF	IFN	TNF	
Example 148	-CH <sub>2</sub> OH	1.11	10.0	3038	684	2
Example 149	-(CH <sub>2</sub> ) <sub>2</sub> OH	3.33	30.0	1849	342	1
Analog 6	-CH <sub>2</sub> OCH <sub>2</sub> CH <sub>3</sub>	0.12	1.11	658	4921	1
Analog 7	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.04	0.37	4143	7762	1

A further compound used in the methods of the invention and close analogs were tested for their ability to induce cytokine biosynthesis using the test method described above. The analogs used are shown in the table below.

Analog	Chemical Name	Reference
8	1-(4-amino-2-ethyl-7-pyridin-3-yl-1 <i>H</i> -imidazo[4,5- <i>c</i> ]quinolin-1-yl)-2-methylpropan-2-ol	U.S. Patent Publication 2004/0147543
		Example 142
9	1-(4-amino-2-propyl-7-pyridin-3-yl-1 <i>H</i> -imidazo[4,5- <i>c</i> ]quinolin-1-yl)-2-methylpropan-2-ol	U.S. Patent Publication 2004/0147543 Example 418
10	1-(4-amino-2-ethoxymethyl-7-pyridin-3-yl-1 <i>H</i> -imidazo[4,5- <i>c</i> ]quinolin-1-yl)-2-methylpropan-2-ol	U.S. Patent Publication 2004/0147543 Example 126

The compound of Example 163 and several closely related analogs were tested using the test method described above. The IFN-α dose response curves are shown in

Figure 7. The TNF- $\alpha$  dose response curves are shown in Figure 8. The minimum effective concentration for the induction of IFN- $\alpha$ , minimum effective concentration for the induction of TNF- $\alpha$ , the maximal response for IFN- $\alpha$ , and the maximal response for TNF- $\alpha$  are shown in Table 9 below where # is the number of separate experiments in which the compound was tested. When a compound was tested in more than one experiment the values shown are the median values.

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Table 9

NH <sub>2</sub> N R <sub>2</sub> CH <sub>3</sub> CH <sub>3</sub> OH						
Compound	R <sub>2</sub>	Minimum Concentra IFN			Response mL) TNF	#
Example 163	-CH₂OH	1.11	>30	2251	*	1
Analog 8	-CH <sub>2</sub> CH <sub>3</sub>	0.12	0.37	1118	3234	4
Analog 9	-(CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	0.04	0.37	597	3951	1
Analog 10	-CH <sub>2</sub> OCH <sub>2</sub> CH <sub>3</sub>	0.04	0.12	840	0.12	5

<sup>\*</sup>Below experimental background level of 40 pg/mL.

Compounds of the invention and close analogs were tested for their ability to induce cytokine biosynthesis using the test method described above. The minimum effective concentration for the induction of IFN- $\alpha$ , minimum effective concentration for the induction of TNF- $\alpha$ , the maximal response for IFN- $\alpha$ , and the maximal response for TNF- $\alpha$  are shown in Table 10 below where # is the number of separate experiments in which the compound was tested. When a compound was tested in more than one experiment the values shown are the median values.

	:	#		m	9		7	4
		Response nL)	TAT	51	3571	3951	7867	3445
		Maximal Response (pg/mL)	IFN	2886	1652	597	840	829
		Effective tion (μΜ)	TNF	10	0.37	0.37	0.12	1.11
	.5.	Minimum Effective Concentration (μΜ)	IFN T'NF	0.37	0.12	0.04	0.04	0.37
Table 10	Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z	ጺ						\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
	<u>"</u>	R		CH3 OH GCH3	CH3 OH S	СН, НО НО НО НО Н	CH, OH	CH <sub>3</sub> OH OH
		.R		НО	CH <sub>3</sub>	CH <sub>3</sub>	CH <sub>3</sub>	0-cH <sub>3</sub>
		Compound		Example 163	Analog 8	Analog 9	Analog 10	Analog 11

∞	3	4	4	2	2	7
8386	*	8000	18284	10087	16296	16482
1065	4357	1771	6308	2084	5868	1079
0.014	>30	3.33	0.12	1:11	0.04	0.12
0.014	0.37	0.12	0.014	0.014	0.014	0.014
	H Z	HO-	H Z	H-Z	HO	HO
CH <sub>3</sub>	CH, CH,	CH, OH	CH3 HO OH H3	CH3 HO	PO HO	ДО НО
, CH <sub>3</sub>	Ъ	0-CH <sub>3</sub>	CH,	CH <sub>3</sub>	CH <sub>3</sub>	G,
Analog 12	Example 189	Analog 13	Analog 14	Analog 15	Analog 16	Analog 17

	2	∞	2	
*	1449	619	1054	*
696	2979	1686	1157	*
>30	0.37	1.11	0.37	>30
1.11	0.12	0.12	0.12	>30
Z.	Z-	Z	Z-	
OH3 OH3	N O CH <sub>3</sub> CH <sub>3</sub>	CH <sub>3</sub> CH <sub>3</sub>	CH <sub>3</sub> CH <sub>3</sub>	
F6 \	E. C.	CH <sub>3</sub>	CH <sub>3</sub>	но—
Example 191	Analog 18	Analog 19	Analog 20	Example 156

2	-			-		m
201	62	29	2296	2238	42	3151
1880	1665	1274	260	440	1180	1199
1.11	1.11	3.33	0.014	0.12	3.33	0.04
0.12	0.37	0.37	0.014	0.014	0.37	0.014
					\	Z-
	Z o	CH3 OH 3	. но Но Но	CH3 OH	CH <sub>3</sub>	CH <sub>3</sub>
CH <sub>3</sub>	CH <sub>3</sub>	но/	0—/	CH <sub>3</sub>	но/	, CH <sub>3</sub>
Analog 21	Analog 22	Example 157	Analog 23	Analog 24	Example 158	Analog 25

		2	m	
647	349	9563	3885	<u>8</u>
591	1891	1332	1263	5089
0.12	10	0.04	0.37	30
0.014	0.12	0.014	0.04	0.37
	I Z	T Z	T Z	Z_
CH <sub>3</sub>	ห็อ ห้อ	<sup>°</sup> HO	ะ รับ	O TE
CH <sub>3</sub>	HOL	, CH <sub>3</sub>	CH3	<del>Б</del>
Analog 26	Example 159	Analog 27	Analog 28	Example 195

2		-
1059	5284	*
936	531	3516
=======================================	0.37	>30
0.04	0.37	0.12
	Z-	Z-
O HO SE	O H H H	O NH OCH 3
<del>1</del> 5	CH3	НО
Analog 29	Analog 30	Example 196

		<del></del>
7	7	H
166	1647	009
965	862	4373
1.11	0.37	10
0.12	0.12	0.04
	Z-	Z
O THE TOTAL STATE OF THE TOTAL S	O HN CH3	O HZ
CH <sub>3</sub>	CH3	<sub>-</sub> -
Analog 31	Analog 32	Example 197

2	-	-
1618	9019	410
925	649	2745
1.11	0.37	3.33
0.014	0.014	0.12
	Z-	Z_
O TZ	O TZ	O ZI
CH.	-CH <sub>3</sub>	₹ }
Analog 33	Analog 34	Example 198

Analog 35	Analog 36	Example 199	Analog 37
CH3	CH <sub>3</sub>	Đ,	£ S
O ZEI	O ZII	N O CH <sub>3</sub>	HN CH3 CH3
Z	\	Z	
0.04	0.12	0.37	0.12
0.37	0.37	10	=
696	521	5880	1194
1366	2222	217	728
2		-	2

2	-	ED.	-	m
096	*	380	*	943
1610	109	753	*	6211
0.37	>30	1.11	×30	3.33
0.12	30	0.12	>30	0.37
	\			
N O CH <sub>3</sub> CH <sub>3</sub>				
CH <sub>3</sub>	НО	CH	HO C	CH <sub>3</sub>
Analog 38	Example 160 -	Analog 39	Example 161	Analog 40

				_	_
*	10184	7423	4456	352	3955
87	541	1681	650	12641	740
>30	0.12	0.37	0.12	10	0.04
30	0.014	0.37	0.12	0.37	0.04
				·	
					\
_£_	£	\\.	, F	<b>₹</b>	
					. :
£27.	2707 E.H.O.H.O.H.O.H.O.H.O.H.O.H.O.H.O.H.O.H.	£.7.9	2 <u>7</u> 2	2,79 2,24 1,04 1,04 1,04 1,04 1,04 1,04 1,04 1,0	7 7 0 1 1
1		]		1	1
	ਸ਼ੁੰ ਹ		ਸ਼ੂੰ ਨੂੰ		O, H,
HO_	0,	₽́	0	<del>Б</del> (	0
					]
υ υ	14	<u>a</u>	42	e e	. 43
Example 164	Analog 41	Example 165	Analog 42	Example 168	Analog 43
ш =		Ψ -	▼	Щ —	٩

	<b>T</b>	<u> </u>	1	Ţ
			2	-
*	3128	*	2865	*
*	1382	1087	1062	1266
>30	1.11	>30	1.11	>30
>30	0.04	3.33	0.014	1.11
H N	H N	H <sub>O</sub>	H	
CH <sub>3</sub> CH <sub>3</sub>	CH <sub>3</sub> CH <sub>3</sub>	9	5	5
Ъ	CH <sub>3</sub>	HO	CH <sub>3</sub>	ъ Г
Example 201	Analog 44	Example 205	Analog 45	Example 206

1054			
815			
0.37			
0.014			
	<u> </u>	0	
CH <sub>3</sub>	)		
Analog 46	<del></del>		

\*Below experimental background level
All analogs are either specifically exemplified in or are readily prepared using the synthetic methods disclosed in U.S. Patent Application Publication 2004/0147543

Compounds of the invention and close analogs were tested for their ability to induce cytokine biosynthesis using the test method described above. The minimum effective concentration for the induction of IFN- $\alpha$ , minimum effective concentration for the induction of TNF- $\alpha$ , the maximal response for IFN- $\alpha$ , and the maximal response for TNF- $\alpha$  are shown in Table 11 below where # is the number of separate experiments in which the compound was tested. When a compound was tested in more than one experiment the values shown are the median values.

5

		#		9	4	6	9	11	33	∞		3	4	13	79	22	10
		Response	TNF	154	*	1518	9675	9780	10665	13908	7151	#	*	526	619	850	1439
		Maximal Response (pg/mL)	正	1670	6527	1846	9601	832	1138	1308	1638	7220	2340	7293	2712	2184	2581
		Effective on (µM)	岂	30	30	3.33	1.11	0.37	0.37	0.12	3.33	>30	>30	10	3.33	1.11	1.11
	- <sub>2</sub>	Minimum Effective Concentration (µM)	IFN	3.33	1.11	0.37	0.12	0.12	0.04	0.014	0.37	0.37	0.37	0.12	0.04	0.12	0.04
Table 11	Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z	R <sub>2</sub>		-CH <sub>2</sub> OH	-(CH <sub>2</sub> ) <sub>2</sub> OH	-CH <sub>3</sub>	-CH <sub>2</sub> CH <sub>3</sub>	-CH2CH2CH3	-CH2OCH2CH3	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	-CH2OCH3	-CH <sub>2</sub> OH	-(CH <sub>2</sub> ) <sub>2</sub> OH	-CH3	-CH2CH3	-CH2CH2CH3	-CH2OCH2CH3
		$R_1$		-CH2C(CH3)2NHS(O)2CH3	-CH2C(CH3)2NHS(O)2CH3	-CH2C(CH3)2NHS(O)2CH3	-CH2C(CH3)2NHS(O)2CH3	-CH2C(CH3)2NHS(O)2CH3	-CH2C(CH3)2NHS(O)2CH3	-CH2C(CH3)2NHS(O)2CH3	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-(CH₂)₄NHS(O)₂CH₃	-(CH <sub>2</sub> )4NHS(0)2CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>4</sub> NHS(0) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>4</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> )4NHS(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> )4NHS(O) <sub>2</sub> CH <sub>3</sub>
		Compound		Example 7	Example 6	Analog 1	Analog 2	Analog 3	Analog 4	Analog 5	Analog 47	Example 368	Example 3	Análog 48	Analog 49	Analog 50	Analog 51

Analog 52	-(CH <sub>2</sub> )4NHS(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.37	7594	1931	13
Example 115	-(CH <sub>2</sub> ),NHC(O)-√	-(CH <sub>2</sub> ) <sub>2</sub> OH	1.11	>30	8361	*	1
Analog 53	°(CH2)4NHC(O)-√	-CH <sub>3</sub>	0.12	10	1538	1400	
Analog 54	-(CH <sub>2</sub> ) <sub>4</sub> NHC(O)-NO	-CH2CH3	0.37	3.33	4975	2570	
Analog 55	-(CH <sub>2</sub> ) <sub>4</sub> NHC(O)-NO	-CH2CH2CH3	0.12	1.11	11255	1298	3
Analog 56	-(CH <sub>2</sub> )4NHC(O)-NO	-CH2OCH2CH3	0.12	1.11	3433	1580	2
Analog 57	-(CH <sub>2</sub> )4NHC(O)-NO	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.04	6888	3494	∞
Example 122	-(CH <sub>2</sub> ) <sub>3</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OH	3.33	>30	9651	*	3
Analog 58	-(CH <sub>2</sub> ) <sub>3</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-CH3	1.11	30	2778	*	=
Analog 59	-(CH <sub>2</sub> ) <sub>3</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-CH2CH3	1.11	30	1912	238	2
Analog 60	-(CH <sub>2</sub> ) <sub>3</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-СН2СН2СН3	1.11	10	2148	109	3
Analog 61	-(CH <sub>2</sub> ) <sub>3</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-CH2OCH2CH3	0.37	10	1338	463	6
Analog 62	-(CH <sub>2</sub> ) <sub>3</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	1.11	3995	954	6
Example 131	-CH2C(CH3)2CH2NHS(O)2CH3	-(CH <sub>2</sub> ) <sub>2</sub> OH	0.37	>30	8361	*	-
Analog 63	-CH2C(CH3)2CH2NHS(O)2CH3	-CH <sub>3</sub>	0.37	10	6101	805	2
Analog 64	-CH2C(CH3)2CH2NHS(O)2CH3	-СН2СН3	0.12	3.33	1431	1453	3

Analog 65	-CH2C(CH3)2CH2NHS(O)2CH3	-СН2СН2СН3	0.12	10	1711	1929	2
Analog 66	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> CH <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-CH2OCH2CH3	0.12	0.37	561	3768	2
Analog 67	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> CH <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.04	1805	5467	10
Example 36	-(CH <sub>2</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OH	10	>30	3316	*	
Analog 68	-(CH <sub>2</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-CH3	0.12	10	1610	820	3
Analog 69	-(CH <sub>2</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-CH2CH3	0.12	10	3800	2401	9
Analog 70	-(CH <sub>2</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-CH2CH2CH3	30	10	2003	11432	2
Analog 71	-(CH <sub>2</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-CH2OCH2CH3	0.12	3.33	1465	4918	6
Analog 72	-(CH <sub>2</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.04	5858	8547	9
Example 125	-(CH <sub>2</sub> ) <sub>5</sub> S(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OH	0.37	>30	8361	*	
Analog 73	-(CH <sub>2</sub> ) <sub>5</sub> S(O) <sub>2</sub> CH <sub>3</sub>	-CH <sub>3</sub>	0.37	3.33	1294	771	21
Analog 74	-(CH <sub>2</sub> ) <sub>5</sub> S(O) <sub>2</sub> CH <sub>3</sub>	-CH2CH3	0.12	1.11	1062	1545	7
Analog 75	-(CH <sub>2</sub> ) <sub>5</sub> S(O) <sub>2</sub> CH <sub>3</sub>	-CH2CH2CH3	0.12	1.11	828	848	3
Analog 76	-(CH <sub>2</sub> ) <sub>5</sub> S(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	1.11	2695	6919	2
Example 133	-(CH <sub>2</sub> ) <sub>2</sub> O(CH <sub>2</sub> ) <sub>2</sub> N(CH <sub>3</sub> )S(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OH	0.37	>30	8361	*	
Analog 77	-(CH <sub>2</sub> ) <sub>2</sub> O(CH <sub>2</sub> ) <sub>2</sub> N(CH <sub>3</sub> )S(O) <sub>2</sub> CH <sub>3</sub>	-CH <sub>3</sub>	0.12	1.11	1001	3571	
Analog 78	-(CH <sub>2</sub> ) <sub>2</sub> O(CH <sub>2</sub> ) <sub>2</sub> N(CH <sub>3</sub> )S(O) <sub>2</sub> CH <sub>3</sub>	-CH2CH3	0.12	1.11	1803	2525	
Analog 79	-(CH <sub>2</sub> ) <sub>2</sub> O(CH <sub>2</sub> ) <sub>2</sub> N(CH <sub>3</sub> )S(O) <sub>2</sub> CH <sub>3</sub>	-CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	0.37	3.33	1055	1312	2
Analog 80	-(CH <sub>2</sub> ) <sub>2</sub> O(CH <sub>2</sub> ) <sub>2</sub> N(CH <sub>3</sub> )S(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.37	1630	2191	4
Example 99	-(CH <sub>2</sub> )3NHC(O)NHCH(CH <sub>3</sub> )2	-(CH <sub>2</sub> ) <sub>2</sub> OH	0.37	>30	21829	  *	-
Analog 81	-(CH <sub>2</sub> ) <sub>3</sub> NHC(O)NHCH(CH <sub>3</sub> ) <sub>2</sub>	-CH3	3.33	10	1134	490	1

Analog 82	-(CH <sub>2</sub> ) <sub>3</sub> NHC(O)NHCH(CH <sub>3</sub> ) <sub>2</sub>	-CH2CH2CH3	0.12	1.11	6571	3740	2
Analog 83	-(CH2)3NHC(O)NHCH(CH3)2	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.12	1.11	1289	1259	-
Example 120	-(CH <sub>2</sub> ) <sub>3</sub> NH <sub>2</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OH	3.33	>30	5636	*	-
Analog 84	-(CH2)3NH2	-CH <sub>3</sub>	3.33	>30	421	*	
Analog 85	-(CH <sub>2</sub> ) <sub>3</sub> NH <sub>2</sub>	-CH2OCH2CH3	0.12	30	1325	411	
Analog 86	-(CH <sub>2</sub> ) <sub>3</sub> NH <sub>2</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.04	1.11	3433	1674	
Example 128	-(CH <sub>2</sub> ),NHC(O)-~	-(СН <sub>2</sub> )1ОН	30	>30	75	*	3
Analog 87	-(CH <sub>2</sub> ) <sub>3</sub> NHC(O)-¬	-CH <sub>3</sub>	0.37	30	4843	463	2
Analog 88	-(CH <sub>2</sub> ) <sub>3</sub> NHC(0)-¬	-сн2осн2сн3	0.12	1.11	0299	1379	2
Analog 89	-(CH <sub>2</sub> ) <sub>3</sub> NHC(0)-¬O	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.014	5915	6169	2
Example 130	$-CH_2C(CH_3)_2CH_2NHC(O)NH-$	-(СН <sub>2</sub> )2ОН	0.014	3.33	8361	2001	
Analog 90	$-CH_2C(CH_3)_2CH_2NHC(O)NH$	-CH2CH3	0.014	0.12	922	2098	2
Analog 91	$-CH_2C(CH_3)_2CH_2NHC(O)NH$	-CH2OCH2CH3	0.014	0.04	1133	3618	2
Analog 92	$-CH_2C(CH_3)_2CH_2$ NHC(O)NH- $\bigcirc$	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.04	570	6449	2

Example 5	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> NHC(O)—	-сң2он	0.37	10	17274	1130	-
Analog 93	$-CH_2C(CH_3)_2$ NHC(0)— $\left\langle \right\rangle$	-СН2ОСН2СН3	0.37	0.37	1052	12173	13
Analog 94	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> NHC(0)—	-CH2OCH3	1.11	3.33	2518	9721	
Example 124	$-CH_2C(CH_3)_2CH_2NHC(O)$	-(CH <sub>2</sub> ) <sub>2</sub> OH	0.12	3.33	3980	1446	
Analog 95	-CH2C(CH3)2CH2NHC(O)-	-CH2OCH2CH3	0.04	0.37	832	1820	5
Analog 96	$-CH_2C(CH_3)_2CH_2NHC(O)$	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.014	2133	1812	
Example 126	-(CH <sub>2</sub> ) <sub>3</sub> NHC(O)NH(CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OH	1.11	>30	8361	*	-
Analog 97	-(CH <sub>2</sub> )3NHC(O)NH(CH <sub>2</sub> )3CH <sub>3</sub>	-CH2OCH2CH3	0.37	3.33	827	963	5
Analog 98	-(CH <sub>2</sub> )3NHC(O)NH(CH <sub>2</sub> )3CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.04	5915	6919	2
Example 129	-CH2C(CH3)2CH2NH2	-(CH <sub>2</sub> ) <sub>2</sub> OH	0.37	30	2702	85	
Analog 99	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> CH <sub>2</sub> NH <sub>2</sub>	-CH2CH3	0.04	0.37	405	13846	]
Analog 100	-CH2C(CH3)2CH2NH2	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.04	571	17626	
Example 132	-(CH <sub>2</sub> ) <sub>3</sub> NHC(O)-	-(CH <sub>2</sub> ) <sub>2</sub> OH	0.37	>30	8361	*	
Analog 101	$-(CH_2)_3$ NHC(0) $-$	-CH <sub>3</sub>	1.11	3.33	571	156	3

Analog 102	$-(CH_2)_3$ NHC(O) $-$	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	1.11	1504	3080	2
Example 137	-(CH <sub>2</sub> ) <sub>2</sub> NHC(0)NHCH <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OH	30	30	801	73	
Analog 103	-(CH <sub>2</sub> ) <sub>2</sub> NHC(O)NHCH <sub>2</sub> CH <sub>3</sub>	-CH <sub>2</sub> CH <sub>3</sub>	3.33	10	1031	3250	2
Analog 104	-(CH <sub>2</sub> ) <sub>2</sub> NHC(O)NHCH <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.12	2587	7719	4
Example 138	-(CH <sub>2</sub> ) <sub>2</sub> NHC(O)CH <sub>2</sub> CH(CH <sub>3</sub> ) <sub>2</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OH	3.33	>30	36	*	,
Analog 105	-(CH <sub>2</sub> ) <sub>2</sub> NHC(O)CH <sub>2</sub> CH(CH <sub>3</sub> ) <sub>2</sub>	-CH2CH3	3.33	30	851	587	2
Analog 106	-(CH <sub>2</sub> ) <sub>2</sub> NHC(O)CH <sub>2</sub> CH(CH <sub>3</sub> ) <sub>2</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.12	3.33	1204	5694	2
Example 142	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> NHC(O)NHCH(CH <sub>3</sub> ) <sub>2</sub>	-СН2ОН	1.11	>30	1554	*	-
Analog 107	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> NHC(O)NHCH(CH <sub>3</sub> ) <sub>2</sub>	-CH2CH2CH3	1.11	3.33	1428	6363	3
Analog 108	-CH2C(CH3)2NHC(O)NHCH(CH3)2	-CH2OCH2CH3	0.37	1.11	996	10587	4
Example 1	$-(CH_2)_3NHS(O)_2$ $\longrightarrow$ $-CH_3$	-(CH <sub>2</sub> ) <sub>2</sub> OH	0.37	10	1072	143	-
Analog 109	$-(CH_2)_3NHS(O)_2-$	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.04	0.37	638	6169	2
Example 2	$-(CH_2)_3NHC(0)$	-(СН <sub>2</sub> ),ОН	3.33	3.33	507	45	-
Analog 110	-(CH <sub>2</sub> ) <sub>3</sub> NHC(O) -	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.12	1.11	647	6169	2
Example 4	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> NH <sub>2</sub>	-CH <sub>2</sub> OH	0.37	3.33	1893	41	2
Analog 111	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> NH <sub>2</sub>	-CH2OCH2CH3	0.12	0.37	959	11475	7

Analog 112  CCH <sub>2</sub> ) <sub>4</sub> NHC(O) \\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	Example 111	$-(CH_2)_4$ NHC(O) $-\langle -\rangle$	-(СН <sub>2</sub> )2ОН	0.12	1.11	7753	983	_
-(CH <sub>2</sub> ) <sub>4</sub> NHS(O) <sub>2</sub> $\leftarrow$ - (CH <sub>2</sub> ) <sub>2</sub> OH 1.11  -(CH <sub>2</sub> ) <sub>4</sub> NHS(O) <sub>2</sub> $\leftarrow$ - (CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014  -(CH <sub>2</sub> ) <sub>4</sub> NH2  -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\leftarrow$ - (CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.04  -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\leftarrow$ - (CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014  -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\leftarrow$ - (CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014  -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\leftarrow$ - (CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014  -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\leftarrow$ - (CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014	og 112	1 4 ~	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.04	2127	1462	7
-(CH <sub>2</sub> ) <sub>4</sub> NHS(O) <sub>2</sub> $\leftarrow$ -(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014 -(CH <sub>2</sub> ) <sub>4</sub> NH <sub>2</sub> -(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.04 -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\leftarrow$ -(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014 -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\leftarrow$ -(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014 -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\leftarrow$ -(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014 -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\leftarrow$ -(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014	nple 112		-(CH <sub>2</sub> ) <sub>2</sub> OH	1.11	30	8361	76	
-(CH <sub>2</sub> ) <sub>4</sub> NH <sub>2</sub> -(CH <sub>2</sub> ) <sub>2</sub> OH 30 -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\stackrel{N}{\leftarrow}$ -(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.04 -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\stackrel{N}{\leftarrow}$ -(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014 -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\stackrel{N}{\leftarrow}$ -(CH <sub>2</sub> ) <sub>2</sub> OH 0.37 -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\stackrel{N}{\leftarrow}$ -(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014	log 113		-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.04	6032	3786	4
-(CH <sub>2</sub> ) <sub>4</sub> NH <sub>2</sub> -(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.04 -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\stackrel{N}{\leftarrow}$ -(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014 -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\stackrel{N}{\leftarrow}$ -(CH <sub>2</sub> ) <sub>2</sub> OH 0.37 -(CH <sub>2</sub> ) <sub>4</sub> NHC(O) $\stackrel{N}{\leftarrow}$ -(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub> 0.014	nple 114	-(CH <sub>2</sub> ) <sub>4</sub> NH <sub>2</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OH	30	>30	23	*	
$-(CH_2)_4NHC(O) - \begin{pmatrix} -N \\ -(CH_2)_2OH \\ -(CH_2)_4NHC(O) - \begin{pmatrix} -N \\ -(CH_2)_2OH \end{pmatrix} - (CH_2)_2OH \\ -(CH_2)_4NHC(O) - \begin{pmatrix} -N \\ -(CH_2)_2OH \\ -(CH_2)_4NHC(O) - \begin{pmatrix} -N \\ -N \\ -(CH_2)_2OCH_3 \end{pmatrix} 0.014$	log 114	-(CH <sub>2</sub> )4NH <sub>2</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.04	0.37	127231	724	1
-(CH <sub>2</sub> ) <sub>4</sub> NHC(O)—(N - O O O O O O O O O O O O O O O O O O	nple 116	$-(CH_2)_4$ NHC(O) $-$	-(CH <sub>2</sub> ) <sub>2</sub> OH	0.37	30	8361	1112	
7 $-(CH_2)_4NHC(O)$ $\stackrel{N}{\longleftarrow}$ $-(CH_2)_2OH$ 0.37 $-(CH_2)_4NHC(O)$ $\stackrel{N}{\longleftarrow}$ $-(CH_2)_2OCH_3$ 0.014	log 115	$-(CH_2)_4$ NHC(O) $-$	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.04	7545	9340	2
$-(CH_2)_4NHC(0)$ $-(CH_2)_2OCH_3$ 0.014	nple 117		-(CH <sub>2</sub> ) <sub>2</sub> OH	0.37	3.33	5520	1938	
	og 116	$-(CH_2)_4NHC(0)$	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.04	1129	7261	m

Example 118	$-(CH_2)_8$ NHC(O)NH $-$	-(СН <sub>2</sub> )2ОН	0.37	>30	5177	*	
Analog 117	$-(CH_2)_8$ NHC(O)NH $-(CH_2)_8$	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.12	1257	1372	
Example 119	-(CH <sub>2</sub> ) <sub>8</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OH	0.04	3.33	8361	693	-
Analog 118	-(CH <sub>2</sub> ) <sub>8</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.014	1914	1853	2
Example 121	$-(CH_2)_6$ NHC(O) $-$	-(CH <sub>2</sub> ) <sub>2</sub> OH	0.37	3.33	2441	180	1
Analog 119	-(CH <sub>2</sub> ) <sub>8</sub> NHC(O)—	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.014	1584	1995	-
Example 134	-(CH <sub>2</sub> ) <sub>2</sub> O(CH <sub>2</sub> ) <sub>2</sub> N(CH <sub>3</sub> )C(O)—	-(СН2)2ОН	3.33	30	8361	315	-
Analog 120	$-(CH_2)_2O(CH_2)_2N(CH_3)C(O)$	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.04	0.37	1394	3317	
Example 135	$-(CH_2)_2O(CH_2)_2N(CH_3)C(O)$	-(CH <sub>2</sub> ) <sub>2</sub> OH	3.33	30	2464	146	
Analog 121	$-(CH_2)_2O(CH_2)_2N(CH_3)C(O)$	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.37	1.11	1234	4849	2
Example 140	-(CH <sub>2</sub> ) <sub>2</sub> O(CH <sub>2</sub> ) <sub>2</sub> NHC(O)(CH <sub>2</sub> ) <sub>14</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OH	1.11	>30	673	*	-
Analog 121	-(CH <sub>2</sub> ) <sub>2</sub> O(CH <sub>2</sub> ) <sub>2</sub> NHC(O)(CH <sub>2</sub> ) <sub>14</sub> CH <sub>3</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	0.014	0.014	2556	11033	6
Example 141	-(CH <sub>2</sub> ) <sub>3</sub> NHC(O)CH(CH <sub>3</sub> ) <sub>2</sub>	-(CH <sub>2</sub> ) <sub>2</sub> OH	0.04	30	14046	243	-
Analog 123	-(CH <sub>2</sub> ) <sub>3</sub> NHC(0)CH(CH <sub>3</sub> ) <sub>2</sub>	-CH3	1.11	10	3011	405	2
Example 364	-CH2C(CH3)2CH2S(O)2CH3	-сн,он	1.11	30	5343	164	

				200	, 00.	0613	_
Analog 124 -CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> (	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> CH <sub>2</sub> S(O) <sub>2</sub> CH <sub>3</sub>	-CH2OCH2CH3	0.12	0.37	1924	9513	4
				000	1400	77	_
Example 365 -(CH <sub>2</sub> ) <sub>2</sub> NHC	-(CH <sub>2</sub> ) <sub>2</sub> NHC(O)NHCH(CH <sub>3</sub> ) <sub>2</sub>	-CH <sub>2</sub> OH	0.37	5.33	1488	/4	1
				,	3000	7510	,
Analog 125 -(CH <sub>2</sub> ) <sub>2</sub> NHC(C	-(CH <sub>2</sub> ) <sub>2</sub> NHC(O)NHCH(CH <sub>3</sub> ) <sub>2</sub>	-CH2OCH2CH3	0.37	01	7042	7317	,

\*TNF below experimental background of 40 pg/mL

Analogs 1-5, 47-52, 58-74, 109, 113, and 118 are either specifically exemplified in or are readily prepared using the synthetic methods disclosed in U.S. Patent Nos. 6,331,539 and 6,677,349.

Analogs 53-57, 81-83, 87-91, 97, 98, 103, 104, 107, and 108 are either specifically exemplified in or are readily prepared using the synthetic methods disclosed in U.S. Patent Nos. 6,541,485 and 6,573,273.

Analogs 73-76 and 124 are either specifically exemplified in or are readily prepared using the synthetic methods disclosed in U.S. Patent No. 6,664,264.

Analogs 77-80 are either specifically exemplified in or are readily prepared using the synthetic methods disclosed in U.S. Patent No. 6,683,088.

Analogs 84-86, 99, 100, 111, and 114 are either specifically exemplified in or are readily prepared using the synthetic methods Analogs 93-96, 101, 102, 105, 106, 110, 112, 115, 116, 119, and 123 are either specifically exemplified in or are readily prepared using the synthetic methods disclosed in U.S. Patent Nos. 6,451,810 and 6,756,382. disclosed in U.S. Patent Nos. 6,069,149 and 6,677,349.

Analogs 120-122 are either specifically exemplified in or are readily prepared using the synthetic methods disclosed in U.S. Patent No. 6,664,265. 15

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Compounds of the invention and in some instances, close analogs (Table 13 below), were tested for their ability to induce cytokine biosynthesis using the test method described above. The minimum effective concentration for the induction of IFN- $\alpha$ , minimum effective concentration for the induction of TNF- $\alpha$ , the maximal response for IFN- $\alpha$ , and the maximal response for TNF- $\alpha$  are shown in Table 12 below where # is the number of separate experiments in which the compound was tested. When a compound was tested in more than one experiment the values shown are the median values.

PCT/US2006/006222

	<del></del>	<del></del>		Τ-	<del></del>	<del></del>	1	Τ-	1	<del></del>	1	т	Τ_	<del></del>	T
		#	Γ	m	2	2	7		<u> </u>	-	<del> -</del>	<u></u>	-	<u> </u>	ļ
		Maximal Response (pg/mL)	TNF	1316	331	7275	7503	6219	3863	5526	798	8691	170	9234	1922
		Maximal (pg/	IFN	2290	2063	1674	3142	1952	2150	2484	1467	1647	1914	2465	1833
		fective n (μM)	TNF	10	30	1.11	0.37	3.33	3.33	1.11	10	0.014	30	0.37	3.33
12	Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z	Minimum Effective Concentration (μM)	IFN	1.11	3.33	0.12	0.04	0.37	0.37	0.12	1.11	0.014	10	0.04	1.11
Table 12	Z= Z	R		-СН2ОН	-(CH <sub>2</sub> ) <sub>2</sub> OH	-CH2OCH2CH3	-(CH <sub>2</sub> ) <sub>2</sub> OCH <sub>3</sub>	-CH <sub>3</sub>	-CH <sub>2</sub> CH <sub>3</sub>	-CH2CH2CH3	-СН2ОН	-СН2ОСН2СН3	-СН2ОН	-CH2OCH2CH3	-СН2ОН
		$R_{\rm l}$		-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> OH	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> OH	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> OH	-СН2С(СН3)2ОН	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> OH	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> OH	-СН2С(СН3)2ОН	-CH <sub>2</sub> C(CH) <sub>3</sub> NH(CO)-	-CH <sub>2</sub> C(CH) <sub>3</sub> NH(CO)-	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	-CH <sub>2</sub> CF(CH <sub>3</sub> ) <sub>2</sub>
		Compound		Example 148	Example 149	Analog 6	Analog 7	Analog 126	Analog 127	Analog 128	Example 143	Analog 129	Example 144	Analog 130	Example 551

					, ,		
-1	*	49	>30	30	-(CH <sub>2</sub> ) <sub>2</sub> OH	Example 147   -(CH <sub>2</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	Example 147
	*	1520	06<.	10	-СН2ОН	Example 146 -(CH <sub>2</sub> ) <sub>2</sub> NHS(O) <sub>2</sub> CH <sub>3</sub>	Example 146
2	7	3008	30	1.11	-(CH <sub>2</sub> ) <sub>2</sub> OH	Example 145 -(CH <sub>2</sub> )4NHS(O) <sub>2</sub> CH <sub>3</sub>	Example 145
12	669	1117	1.11	0.37	-CH3	-CH2CH(CH3)2	Analog 133
2	3160	1524	1.11	0.12	-CH2OCH2CH3	-CH2CH(CH3)2	Analog 132
1	363	1592	30	1.11	-СН2ОН	Example 633 -CH2CH(CH3)2	Example 633
2	1679	2120	10	0.37	-CH3	Analog 131   -CH <sub>2</sub> CF(CH <sub>3</sub> ) <sub>2</sub>	Analog 131
	84	1646	10	1.11	-(CH <sub>2</sub> ) <sub>2</sub> OH	Example 552 -CH <sub>2</sub> CF(CH <sub>3</sub> ) <sub>2</sub>	Example 552

\*Below the experimental background of 40 pg/mL

	Table 13	
Analog	Chemical Name	Reference
6	1-(4-amino-2-ethoxymethyl-1H-imidazo[4,5-	Example 148 Part E
}	c][1,5]naphthyridin-1-yl)-2-methylpropan-2-ol	
7	1-[4-amino-2-(2-methoxyethyl)-1H-imidazo[4,5-	Example 149 Part J
_	c][1,5]naphthyridin-1-yl]-2-methylpropan-2-ol	
126	1-(4-amino-2-methyl-1H-imidazo[4,5-	U.S. Patent No.
	c][1,5]naphthyridin-1-yl)-2-methylpropan-2-ol	6,194,425**
127	1-(4-amino-2-ethyl-1 <i>H</i> -imidazo[4,5-	U.S. Patent No.
	c][1,5]naphthyridin-1-yl)-2-methylpropan-2-ol	6,194,425**
128	1-(4-amino-2-propyl-1H-imidazo[4,5-	U.S. Patent No.
	c][1,5]naphthyridin-1-yl)-2-methylpropan-2-ol	6,194,425**
129	N-[2-(4-amino-2-ethoxymethyl-1H-imidazo[4,5-	Example 143 Part H
	c][1,5]naphthyridin-1-yl)-1,1-	
	dimethylethyl]cyclohexanecarboxamide	
130	N-[2-(4-amino-2-ethoxoxymethyl-1H-imidazo[4,5-	Example 144 Part A
	c][1,5]naphthyridin-1-yl)-1,1-	
	dimethylethyl]methanesulfonamide	
131	1-(2-fluoro-2-methylpropyl)-2-methyl-1 <i>H</i> -imidazo[4,5-	U.S. Patent No.
	c][1,5]naphthyridin-4-amine	6,194,425**
132	2-ethoxymethyl-1-(2-methylpropyl)-1 <i>H</i> -imidazo[4,5-	U.S. Patent No.
	c][1,5]naphthyridin-4-amine	6,194,425**
133	2-methyl-1-(2-methylpropyl)-1 <i>H</i> -imidazo[4,5-	U.S. Patent No.
	c][1,5]naphthyridin-4-amine	6,194,425
		Example 36

<sup>\*\*</sup>Although not a working example, the compound is readily prepared using the disclosed synthetic methods.

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The complete disclosures of the patents, patent documents, and publications cited herein are incorporated by reference in their entirety as if each were individually incorporated. Various modifications and alterations to this invention will become apparent to those skilled in the art without departing from the scope and spirit of this invention. It should be understood that this invention is not intended to be unduly limited by the illustrative embodiments and examples set forth herein and that such examples and embodiments are presented by way of example only with the scope of the invention intended to be limited only by the claims set forth herein as follows.

## WHAT IS CLAIMED IS:

1. A method of preferentially inducing the biosynthesis of IFN- $\alpha$  in an animal comprising administering an effective amount of a compound of Formula I:

$$R_{B}$$
 $R_{A}$ 
 $R_{1}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{1}$ 

5

wherein:

n is 1 or 2;

R<sub>A</sub> and R<sub>B</sub> are each independently selected from the group consisting of:

hydrogen,

10

halogen,

alkyl,

alkenyl,

alkoxy,

alkylthio and

15

 $-N(R_9)_2;$ 

or when taken together, R<sub>A</sub> and R<sub>B</sub> form a fused aryl ring or heteroaryl ring containing one heteroatom selected from the group consisting of N and S wherein the aryl or heteroaryl ring is unsubstituted or substituted by one or more R groups, or substituted by one R<sub>3</sub> group, or substituted by one R<sub>3</sub> group and one R group;

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or when taken together, R<sub>A</sub> and R<sub>B</sub> form a fused 5 to 7 membered saturated ring, optionally containing one heteroatom selected from the group consisting of N and S, and unsubstituted or substituted at a carbon atom by one or more R groups;

R is selected from the group consisting of:

halogen,

25

hydroxy,

alkyl,

alkenyl,

haloalkyl,

alkoxy,

```
alkylthio, and
                                 -N(R_9)_2;
                       R_1 is selected from the group consisting of:
                                 -R_4
 5
                                 -X-R<sub>4</sub>,
                                 -X-Y-R<sub>4</sub>,
                                 -X-Y-X-Y-R<sub>4</sub>, and
                                 -X-R<sub>5</sub>;
                       R<sub>3</sub> is selected from the group consisting of:
10
                                 -Z-R<sub>4</sub>,
                                 -Z-X-R<sub>4</sub>,
                                 -Z-X-Y-R<sub>4</sub>,
                                 -Z-X-Y-X-Y-R4, and
                                 -Z-X-R<sub>5</sub>;
```

X is selected from the group consisting of alkylene, alkenylene, alkynylene, arylene, heteroarylene, and heterocyclylene wherein the alkylene, alkenylene, and alkynylene groups can be optionally interrupted or terminated by arylene, heteroarylene or heterocyclylene and optionally interrupted by one or more -O- groups;

Y is selected from the group consisting of:

```
20
                              -O-,
                              -S(O)_{0-2}-,
                              -S(O)_2-N(R_8)-,
                              -C(R_6)-,
                              -C(R_6)-O-,
25
                             -O-C(R_6)-,
                              -O-C(O)-O-,
                              -N(R_8)-Q-,
                              -C(R_6)-N(R_8)-,
                             -O-C(R<sub>6</sub>)-N(R<sub>8</sub>)-,
30
                              -C(R_6)-N(OR_9)-
                             -O-N(R<sub>8</sub>)-Q-,
                              -O-N=C(R<sub>4</sub>)-,
```

Z is a bond or -O-;

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R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl, heteroaryloxyalkylenyl, alkylarylenyl, and heterocyclyl wherein the alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl, heteroaryloxyalkylenyl, alkylheteroarylenyl, and heterocyclyl groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, hydroxyalkyl, haloalkyl, haloalkoxy, halogen, nitro, hydroxy, mercapto, cyano, aryl, aryloxy, arylalkyleneoxy, heteroaryl, heteroaryloxy, heteroarylalkyleneoxy, heterocyclyl, amino, alkylamino, dialkylamino, (dialkylamino)alkyleneoxy, and in the case of alkyl, alkenyl, alkynyl, and heterocyclyl, oxo;

R<sub>5</sub> is selected from the group consisting of:

 $R_6$  is selected from the group consisting of =0 and =S;

R<sub>7</sub> is C<sub>2-7</sub> alkylene;

R<sub>8</sub> is selected from the group consisting of hydrogen, alkyl, alkoxyalkylenyl, hydroxyalkylenyl, arylalkylenyl, and heteroarylalkylenyl;

R<sub>9</sub> is selected from the group consisting of hydrogen and alkyl;

R<sub>10</sub> is C<sub>3-8</sub> alkylene;

A is selected from the group consisting of -O-, -C(O)-, -CH<sub>2</sub>-, -S(O)<sub>0-2</sub>-, and  $\mathbb{R}$ 

10  $-N(Q-R_4)-;$ 

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A' is selected from the group consisting of -O-, -S(O)<sub>0-2</sub>-, -N(-Q-R<sub>4</sub>)-, and -CH<sub>2</sub>-; Q is selected from the group consisting of a bond, -C(R<sub>6</sub>)-, -C(R<sub>6</sub>)-C(R<sub>6</sub>)-, -S(O)<sub>2</sub>-, -C(R<sub>6</sub>)-N(R<sub>8</sub>)-W-, -S(O)<sub>2</sub>-N(R<sub>8</sub>)-, -C(R<sub>6</sub>)-O-, -C(R<sub>6</sub>)-S-, and -C(R<sub>6</sub>)-N(OR<sub>9</sub>)-;

V is selected from the group consisting of  $-C(R_6)$ -,  $-O-C(R_6)$ -,  $-N(R_8)-C(R_6)$ -, and  $-S(O)_2$ -;

W is selected from the group consisting of a bond, -C(O)-, and  $-S(O)_2$ -; and a and b are independently integers from 1 to 6 with the proviso that a + b is  $\leq 7$ ; or a pharmaceutically acceptable salt thereof to the animal.

20 2. A method of preferentially inducing the biosynthesis of IFN-α in an animal comprising administering an effective amount of a compound of Formula II:

$$R_{B}$$
 $R_{A}$ 
 $R_{1}$ 
 $R_{1}$ 
 $R_{1}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{3}$ 
 $R_{4}$ 
 $R_{1}$ 

II

wherein:

25 G<sub>1</sub> is selected from the group consisting of:

```
-C(O)-R',
                             α-aminoacyl,
                             α-aminoacyl-α-aminoacyl,
                              -C(O)-O-R',
 5
                              -C(O)-N(R")R',
                              -C(=NY')-R'
                              -CH(OH)-C(O)-OY',
                              -CH(OC<sub>1-4</sub> alkyl)Y_0,
                              -CH<sub>2</sub>Y<sub>1</sub>, and
10
                              -CH(CH<sub>3</sub>)Y_1;
                    R' and R" are independently selected from the group consisting of C<sub>1-10</sub> alkyl,
          C<sub>3-7</sub> cycloalkyl, phenyl, benzyl, and 2-phenylethyl, each of which may be unsubstituted or
          substituted by one or more substituents independently selected from the group consisting
          of halogen, hydroxy, nitro, cyano, carboxy, C<sub>1-6</sub> alkyl, C<sub>1-4</sub> alkoxy, aryl, heteroaryl, aryl-
15
          C<sub>1-4</sub> alkylenyl, heteroaryl-C<sub>1-4</sub> alkylenyl, halo-C<sub>1-4</sub> alkylenyl, halo-C<sub>1-4</sub> alkoxy,
          -O-C(O)-CH<sub>3</sub>, -C(O)-O-CH<sub>3</sub>, -C(O)-NH<sub>2</sub>, -O-CH<sub>2</sub>-C(O)-NH<sub>2</sub>, -NH<sub>2</sub>, and -S(O)<sub>2</sub>-NH<sub>2</sub>,
          with the proviso that R" can also be hydrogen;
                    \alpha\text{-aminoacyl} is an \alpha\text{-aminoacyl} group derived from an \alpha\text{-amino} acid selected from
          the group consisting of racemic, D-, and L-amino acids;
20
                    Y' is selected from the group consisting of hydrogen, C<sub>1-6</sub> alkyl, and benzyl;
                    Y<sub>0</sub> is selected from the group consisting of C<sub>1-6</sub> alkyl, carboxy-C<sub>1-6</sub> alkylenyl,
          amino-C<sub>1-4</sub> alkylenyl, mono-N-C<sub>1-6</sub> alkylamino-C<sub>1-4</sub> alkylenyl, and
          di-N, N-C<sub>1-6</sub> alkylamino-C<sub>1-4</sub> alkylenyl;
                    Y<sub>1</sub> is selected from the group consisting of mono-N-C<sub>1-6</sub> alkylamino,
25
          di-N,N-C<sub>1-6</sub> alkylamino, morpholin-4-yl, piperidin-1-yl, pyrrolidin-1-yl, and
          4-C<sub>1-4</sub> alkylpiperazin-1-yl;
                    n is 1 or 2;
                    R<sub>A</sub> and R<sub>B</sub> are each independently selected from the group consisting of:
                              hydrogen,
30
                              halogen,
                              alkyl,
                              alkenyl,
```

```
alkoxy,
alkylthio and
-N(R_9)_2;
```

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or when taken together, R<sub>A</sub> and R<sub>B</sub> form a fused aryl ring or heteroaryl ring containing one heteroatom selected from the group consisting of N and S wherein the aryl or heteroaryl ring is unsubstituted or substituted by one or more R groups, or substituted by one R<sub>3</sub> group, or substituted by one R<sub>3</sub> group and one R group;

or when taken together, R<sub>A</sub> and R<sub>B</sub> form a fused 5 to 7 membered saturated ring, optionally containing one heteroatom selected from the group consisting of N and S, and unsubstituted or substituted by one or more R groups;

R is selected from the group consisting of:

```
halogen,
hydroxy,
alkyl,

15 alkenyl,
haloalkyl,
alkoxy,
alkylthio, and
-N(R<sub>9</sub>)<sub>2</sub>;

20 R<sub>1</sub> is selected from the group consisting of:
-R<sub>4</sub>,
-X-R<sub>4</sub>,
-X-Y-R<sub>4</sub>,
-X-Y-R<sub>4</sub>, and
```

R<sub>3</sub> is selected from the group consisting of:

-Z-R<sub>4</sub>, -Z-X-R<sub>4</sub>, -Z-X-Y-R<sub>4</sub>, 30 -Z-X-Y-X-Y-R<sub>4</sub>, and -Z-X-R<sub>5</sub>;

-X-R<sub>5</sub>;

X is selected from the group consisting of alkylene, alkenylene, alkynylene, arylene, heteroarylene, and heterocyclylene wherein the alkylene, alkenylene, and alkynylene groups can be optionally interrupted or terminated by arylene, heteroarylene or heterocyclylene and optionally interrupted by one or more -O- groups;

Y is selected from the group consisting of:

$$-O-,$$

$$-S(O)_{0-2}-,$$

$$-S(O)_{2}-N(R_{8})-,$$

$$-C(R_{6})-,$$

$$-C(R_{6})-O-,$$

$$-O-C(R_{6})-,$$

$$-O-C(O)-O-,$$

$$-N(R_{8})-Q-,$$

$$-C(R_{6})-N(R_{8})-,$$

$$-C(R_{6})-N(OR_{9})-,$$

$$-O-N(R_{8})-Q-,$$

$$-O-N=C(R_{4})-,$$

$$-C(=N-O-R_{8})-,$$

$$-CH(-N(-O-R_{8})-Q-R_{4})-,$$

$$-N-C(R_{6})-N-W-$$

$$R_{7}$$

$$-N-C(R_{6})-N-W-$$

$$R_{7}$$

$$-N-Q-$$

$$-(R_{10})^{N-C(R_{6})-N}$$

Z is a bond or -O-;

R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl,

heteroaryloxyalkylenyl, alkylheteroarylenyl, and heterocyclyl wherein the alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl, heteroaryloxyalkylenyl, alkylheteroarylenyl, and heterocyclyl groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, hydroxyalkyl, haloalkyl, haloalkoxy, halogen, nitro, hydroxy, mercapto, cyano, aryl, aryloxy, arylalkyleneoxy, heteroaryl, heteroaryloxy, heteroarylalkyleneoxy, heterocyclyl, amino, alkylamino, dialkylamino, (dialkylamino)alkyleneoxy, and in the case of alkyl, alkenyl, alkynyl, and heterocyclyl, oxo;

R<sub>5</sub> is selected from the group consisting of

 $R_6$  is selected from the group consisting of =0 and =S;

 $R_7$  is  $C_{2-7}$  alkylene;

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R<sub>8</sub> is selected from the group consisting of hydrogen, alkyl, alkoxyalkylenyl, hydroxyalkylenyl, arylalkylenyl, and heteroarylalkylenyl;

R<sub>0</sub> is selected from the group consisting of hydrogen and alkyl;

R<sub>10</sub> is C<sub>3-8</sub> alkylene;

A is selected from the group consisting of -O-, -C(O)-, -CH<sub>2</sub>-, -S(O)<sub>0-2</sub>-, and -N(Q-R<sub>4</sub>)-;

A' is selected from the group consisting of -O-, -S(O)<sub>0-2</sub>-, -N(-Q-R<sub>4</sub>)-, and -CH<sub>2</sub>-;

Q is selected from the group consisting of a bond,  $-C(R_6)$ -,  $-C(R_6)$ - $C(R_6)$ -,  $-S(O)_2$ -,  $-C(R_6)$ - $N(R_8)$ -W-,  $-S(O)_2$ - $N(R_8)$ -,  $-C(R_6)$ -O-,  $-C(R_6)$ -S-, and  $-C(R_6)$ - $N(OR_9)$ -;

V is selected from the group consisting of -C(R<sub>6</sub>)-, -O-C(R<sub>6</sub>)-, -N(R<sub>8</sub>)-C(R<sub>6</sub>)-, and -S(O)<sub>2</sub>-;

W is selected from the group consisting of a bond, -C(O)-, and  $-S(O)_2$ -; and a and b are independently integers from 1 to 6 with the proviso that a + b is  $\leq 7$ ; or a pharmaceutically acceptable salt thereof to the animal.

3. A method of preferentially inducing the biosynthesis of IFN-α in an animal comprising administering an effective amount of a compound of Formula III:

$$R_B$$
 $R_A$ 
 $R_1$ 
 $N$ 
 $(CH_2)_n$ -O-G<sub>2</sub>

Ш

wherein:

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G<sub>2</sub> is selected from the group consisting of:

 $-X_2-C(O)-R'$ ,

α-aminoacyl,

α-aminoacyl-α-aminoacyl,

 $-X_2-C(O)-O-R'$ , and

-C(O)-N(R'')R';

X<sub>2</sub> is selected from the group consisting of a bond; -CH<sub>2</sub>-O-; -CH(CH<sub>3</sub>)-O-; -C(CH<sub>3</sub>)<sub>2</sub>-O-; and, in the case of -X<sub>2</sub>-C(O)-O-R', -CH<sub>2</sub>-NH-;

R' and R" are independently selected from the group consisting of C<sub>1-10</sub> alkyl, C<sub>3-7</sub> cycloalkyl, phenyl, benzyl, and 2-phenylethyl, each of which may be unsubstituted or substituted by one or more substituents independently selected from the group consisting of halogen, hydroxy, nitro, cyano, carboxy, C<sub>1-6</sub> alkyl, C<sub>1-4</sub> alkoxy, aryl, heteroaryl, aryl-C<sub>1-4</sub> alkylenyl, heteroaryl-C<sub>1-4</sub> alkylenyl, halo-C<sub>1-4</sub> alkylenyl, halo-C<sub>1-4</sub> alkoxy, -O-C(O)-CH<sub>3</sub>, -C(O)-O-CH<sub>3</sub>, -C(O)-NH<sub>2</sub>, -O-CH<sub>2</sub>-C(O)-NH<sub>2</sub>, -NH<sub>2</sub>, and -S(O)<sub>2</sub>-NH<sub>2</sub>, with the proviso that R" can also be hydrogen;

 $\alpha$ -aminoacyl is an  $\alpha$ -aminoacyl group derived from an  $\alpha$ -amino acid selected from the group consisting of racemic, D-, and L-amino acids;

```
n is 1 or 2;
```

R<sub>A</sub> and R<sub>B</sub> are each independently selected from the group consisting of:

5 hydrogen,
halogen,
alkyl,
alkenyl,

alkoxy,

10 alkylthio and

15

 $-N(R_9)_2;$ 

or when taken together, R<sub>A</sub> and R<sub>B</sub> form a fused aryl ring or heteroaryl ring containing one heteroatom selected from the group consisting of N and S wherein the aryl or heteroaryl ring is unsubstituted or substituted by one or more R groups, or substituted by one R<sub>3</sub> group, or substituted by one R<sub>3</sub> group and one R group;

or when taken together, R<sub>A</sub> and R<sub>B</sub> form a fused 5 to 7 membered saturated ring, optionally containing one heteroatom selected from the group consisting of N and S, and unsubstituted or substituted by one or more R groups;

R is selected from the group consisting of:

20 halogen,
hydroxy,
alkyl,
alkenyl,
haloalkyl,
25 alkoxy,
alkylthio, and
-N(R<sub>9</sub>)<sub>2</sub>;
R<sub>1</sub> is selected from the

R<sub>1</sub> is selected from the group consisting of:

-R<sub>4</sub>,
-X-R<sub>4</sub>,
-X-Y-R<sub>4</sub>,
-X-Y-X-Y-R<sub>4</sub>, and

R<sub>3</sub> is selected from the group consisting of:

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$$-Z-X-R5$$
;

X is selected from the group consisting of alkylene, alkenylene, alkynylene, arylene, heteroarylene, and heterocyclylene wherein the alkylene, alkenylene, and alkynylene groups can be optionally interrupted or terminated by arylene, heteroarylene or heterocyclylene and optionally interrupted by one or more -O- groups;

Y is selected from the group consisting of:

15 
$$-S(O)_2-N(R_8)-$$
,

$$-C(R_6)-,$$

$$-C(R_6)-O-,$$

$$-O-C(R_6)-$$
,

$$-N(R_8)-Q_{-}$$

$$-C(R_6)-N(R_8)-,$$

$$-O-C(R_6)-N(R_8)-$$
,

$$-C(R_6)-N(OR_9)-$$

$$-O-N(R_8)-Q-,$$

$$-O-N=C(R_4)-$$

$$-N - R_7 - N - Q - R_7$$
,

 $-V - N R_{10}$ , and

 $-V - N R_{10}$ ,  $-V - N R_{10}$ 

Z is a bond or -O-;

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R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl, heteroarylalkylenyl, alkylarylenyl, and heterocyclyl wherein the alkyl, alkenyl, alkynyl, aryl, arylalkylenyl, aryloxyalkylenyl, alkylarylenyl, heteroaryl, heteroarylalkylenyl, heteroaryloxyalkylenyl, alkylheteroarylenyl, and heterocyclyl groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, hydroxyalkyl, haloalkyl, haloalkoxy, halogen, nitro, hydroxy, mercapto, cyano, aryl, aryloxy, arylalkyleneoxy, heteroaryl, heteroaryloxy, heteroarylalkyleneoxy, heterocyclyl, amino, alkylamino, dialkylamino, (dialkylamino)alkyleneoxy, and in the case of alkyl, alkenyl, alkynyl, and heterocyclyl, oxo;

R<sub>5</sub> is selected from the group consisting of

R<sub>6</sub> is selected from the group consisting of =O and =S;

 $R_7$  is  $C_{2-7}$  alkylene;

R<sub>8</sub> is selected from the group consisting of hydrogen, alkyl, alkoxyalkylenyl, hydroxyalkylenyl, arylalkylenyl, and heteroarylalkylenyl;

R<sub>9</sub> is selected from the group consisting of hydrogen and alkyl;

R<sub>10</sub> is C<sub>3-8</sub> alkylene;

A is selected from the group consisting of -O-, -C(O)-, -CH<sub>2</sub>-, -S(O)<sub>0-2</sub>-, and -N(Q-R<sub>4</sub>)-;

A' is selected from the group consisting of -O-,  $-S(O)_{0-2}$ -,  $-N(-Q-R_4)$ -, and  $-CH_2$ -; Q is selected from the group consisting of a bond,  $-C(R_6)$ -,  $-C(R_6)$ -,  $-S(O)_2$ -,

 $-C(R_6)-N(R_8)-W-$ ,  $-S(O)_2-N(R_8)-$ ,  $-C(R_6)-O-$ ,  $-C(R_6)-S-$ , and  $-C(R_6)-N(OR_9)-$ ;

V is selected from the group consisting of -C(R<sub>6</sub>)-, -O-C(R<sub>6</sub>)-, -N(R<sub>8</sub>)-C(R<sub>6</sub>)-, and -S(O)<sub>2</sub>-;

W is selected from the group consisting of a bond, -C(O)-, and  $-S(O)_2$ -; and a and b are independently integers from 1 to 6 with the proviso that a + b is  $\leq 7$ ; or a pharmaceutically acceptable salt thereof to the animal.

- 4. The method of any one of claims 1, 2 and 3, wherein n is 1.
- 15 5. The method of any one of claims 1, 2, and 3, wherein n is 2.
  - 6. The method of any one of claims 1 through 5 wherein R<sub>A</sub> and R<sub>B</sub> form a fused benzene ring which is unsubstituted or substituted by one or more R groups, or substituted by one R<sub>3</sub> group, or substituted by one R<sub>3</sub> group and one R group.

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- 7. The method of any one of claims 1 through 5 wherein R<sub>A</sub> and R<sub>B</sub> form a fused pyridine ring which is unsubstituted or substituted by one or more R groups, or substituted by one R<sub>3</sub> group, or substituted by one R<sub>3</sub> group and one R group.
- 25 8. The method of any one of claims 1 through 5 wherein R<sub>A</sub> and R<sub>B</sub> form a fused 5 to 7 membered saturated ring, optionally containing one heteroatom selected from the group consisting of N and S, wherein the ring is unsubstituted or substituted by one or more R groups.
- The method of any one of claims 1 through 5 wherein R<sub>A</sub> and R<sub>B</sub> are each independently selected from the group consisting of hydrogen, halogen, alkyl, alkenyl, alkoxy, alkylthio, and -N(R<sub>9</sub>)<sub>2</sub>.

- 10. The method of claim 9 wherein R<sub>A</sub> and R<sub>B</sub> are each methyl.
- 11. The method of any one of claims 1 through 10 wherein R<sub>1</sub> is selected from the group consisting of:

$$-X-Y-X^1-Y^1-R_4$$
, and

X is alkylene that is optionally interrupted or terminated by heterocyclylene and optionally interrupted by one -O- group;

Y is selected from the group consisting of:

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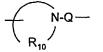
$$-S(O)_2-$$
,

$$-S(O)_2-N(R_8)-,$$

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$$-N(R_8)-Q-$$
,

$$-C(O)-N(R_8)-,$$



and

$$-N-R_7-N-Q-$$

X<sup>1</sup> is selected from the group consisting of alkylene and arylene;

25 Y<sup>1</sup> is selected from the group consisting of:

$$-S(O)_2-N(R_8)$$
-, and  $-N(R_8)-C(O)$ -;

R<sub>4</sub> is selected from the group consisting of hydrogen, alkyl, aryl, heterocyclyl, heteroaryl, heteroarylalkylenyl, alkynyl, arylalkylenyl, and arylalkenylenyl, wherein the alkyl, aryl, arylalkylenyl, heterocyclyl, heteroaryl, and heteroarylalkylenyl groups can be unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, alkoxy, haloalkyl, haloalkoxy, halogen, hydroxy, cyano, aryl, aryloxy, heteroaryl, heterocyclyl, amino, dialkylamino, and in the case of alkyl and heterocyclyl, oxo;

R<sub>5</sub> is selected from the group consisting of:

$$-N-C(O)$$
  $-N-S(O)_2$   $-N(R_8)-C(O)-N$   $A$   $(CH_2)_b$  ;

 $R_6$  is selected from the group consisting of =0 and =S;

 $R_7$  is  $C_{2-7}$  alkylene;

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R<sub>8</sub> is selected from the group consisting of hydrogen, alkyl, alkoxyalkylenyl, hydroxyalkylenyl, arylalkylenyl, and heteroarylalkylenyl;

R<sub>10</sub> is C<sub>3-8</sub> alkylene;

A is selected from the group consisting of  $-O_-$ ,  $-C(O)_-$ , and  $-N(R_4)_-$ ;

Q is selected from the group consisting of a bond,  $-C(R_6)$ -,  $-S(O)_2$ -,

$$-C(R_6)-N(R_8)-W-$$
,  $-S(O)_2-N(R_8)-$ ,  $-C(O)-O-$ , and  $-C(O)-S-$ ;

W is selected from the group consisting of a bond and -C(O)-; and a and b are independently integers from 1 to 6 with the proviso that a + b is  $\leq 7$ .

12. The method of any one of claims 1 through 11 wherein  $R_1$  is selected from the group consisting of  $C_{1-5}$  alkyl,  $C_{2-5}$  alkynyl, aryl $C_{1-4}$  alkylenyl, cycloalkyl $C_{1-4}$  alkylenyl,  $C_{1-4}$  alkylenyl, aryl- $S(O)_2$ - $C_{1-4}$  alkylenyl, aryl- $S(O)_2$ - $C_{1-4}$  alkylenyl,

 $C_{1-4}$  alkyl- $S(O)_2$ - $C_{1-4}$  alkylenyl-O- $C_{1-4}$  alkylenyl,  $C_{1-4}$  alkyl- $S(O)_2$ -NH- $C_{1-4}$  alkylenyl, hydroxy $C_{1-4}$  alkylenyl, halo $C_{1-4}$  alkylenyl, amino $C_{1-4}$  alkylenyl,  $C_{1-4}$  alkyl-C(O)-O- $C_{1-4}$  alkylenyl,  $C_{1-6}$  alkyl-C(O)-NH- $C_{1-4}$  alkylenyl,

aryl-C(O)-NH-C<sub>1-4</sub> alkylenyl wherein aryl is unsubstituted or substituted with one or two halogen groups, heteroaryl-C(O)-NH-C<sub>1-4</sub> alkylenyl,

di(C<sub>1-4</sub> alkyl)amino-S(O)<sub>2</sub>-NH-C<sub>1-4</sub> alkylenyl, aryl-S(O)<sub>2</sub>-NH-C<sub>1-4</sub> alkylenyl, aryl-NH-C(O)-NH-C<sub>1-4</sub> alkylenyl, heteroaryl-NH-C(S)-NH-C<sub>1-4</sub> alkylenyl, di(C<sub>1-4</sub> alkyl)amino-C(O)-NH-C<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkylamino-C(O)-NH-C<sub>1-4</sub> alkylenyl, di(C<sub>1-4</sub> alkyl)amino-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl, C<sub>1-4</sub> alkylenyl, amino-S(O)<sub>2</sub>-C<sub>1-4</sub> alkylenyl, heteroarylC<sub>1-4</sub> alkylenyl wherein heteroaryl is unsubstituted or

substituted by a substituent selected from the group consisting of aryl, heteroaryl, and alkyl, and heterocyclylC<sub>1-4</sub> alkylenyl wherein heterocyclyl is unsubstituted or substituted by one, two, or three substituents selected from the group consisting of alkyl, aryl, heteroaryl, and oxo.

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- 13. The method of any one of claims 1 through 12 wherein R<sub>1</sub> is selected from the group consisting of methyl, ethyl, propyl, 2-methylpropyl, 2,2-dimethylpropyl, butyl, pent-4-ynyl, 2-phenylethyl, 2-hydroxy-2-methylpropyl, 2-fluoro-2-methylpropyl, 2,3-dihydroxypropyl, 4-hydroxybutyl, 2-amino-2-methylpropyl, 2-aminoethyl, 4-aminobutyl,
- 2-(methylsulfonyl)ethyl, 2-(propylsulfonyl)ethyl, 4-(methylsulfonyl)butyl, 2,2-dimethyl-3-(methylsulfonyl)propyl, 3-(phenylsulfonyl)propyl, 2-methyl-2-[2-(methylsulfonyl)ethoxy]propyl, 4-acetoxybutyl, 2-[(methylsulfonyl)amino]ethyl, 4-[(methylsulfonyl)amino]butyl, 2-methyl-2-[(methylsulfonyl)amino]propyl, 2-{[(1-methylsulfonyl)amino}ethyl, 2-(benzenesulfonylamino)ethyl, 2-
- 20 (dimethylaminosulfonylamino)ethyl, 4-(aminosulfonyl)butyl, 4-[(methylamino)sulfonyl]butyl, 4-[(dimethylamino)sulfonyl]butyl, 2-[(cyclohexylcarbonyl)amino]-2-methylpropyl, 2-[(cyclopropylcarbonyl)amino]ethyl, 4-[(cyclopropylcarbonyl)amino]butyl, 2-[(cyclopropylcarbonyl)amino]-2-methylpropyl, 2methyl-2-{[(1-methylethyl)carbonyl]amino}propyl, 2-methyl-2-
- [(ethylcarbonyl)amino]propyl, 2-methyl-2-[(pyridin-3-ylcarbonyl)amino]propyl, 2-methyl-2-[(pyridin-4-ylcarbonyl)amino]propyl, 2-(acetylamino)-2-methylpropyl, 2-(benzoylamino)ethyl, 2-(benzoylamino)-2-methylpropyl, 2-[(4-fluorobenzoyl)amino]-2-methylpropyl, 2-[(3,4-difluorobenzoyl)amino]-2-methylpropyl, 2-[(pyridin-3-ylcarbonyl)amino]ethyl, 2-{[(1-methylethyl)carbonyl]amino}ethyl, 4-{[(1-methylethyl)carbonyl]amino}butyl,
  - 2-methyl-2-({[(1-methylethyl)amino]carbonyl}amino)propyl,

2-({[(1-methylethyl)amino]carbonyl}amino)ethyl, 4-(4-pyridin-2-ylpiperazin-1-yl)butyl, tetrahydro-2*H*-pyran-4-ylmethyl, 4-(1,1-dioxidoisothiazolidin-2-yl)butyl, (2,2-dimethyl-1,3-dioxolan-4-yl)methyl, 3-(3-methylisoxazol-5-yl)propyl, 3-(3-isopropylisoxazol-5-yl)propyl, 3-(3-pyridin-3-ylisoxazol-5-yl)propyl, 4-(3,5,5-trimethyl-1,2,4-oxadiazol-4(5*H*)-yl)butyl, 4-(3-methyl-1-oxa-2,4-diazaspiro[4.4]non-2-en-4-yl)butyl, 2-{[(pyridin-3-ylamino)carbonothioyl]amino}ethyl, 2-{[(dimethylamino)carbonyl]amino}ethyl, and 2-{[(phenylamino)carbonyl]amino}ethyl.

14. The method of any one of claims 1 through 11 wherein R<sub>1</sub> is selected from the group consisting of alkyl, aminoalkyl, dihydroxyalkyl, haloalkyl, and hydroxyalkyl.

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- 15. The method of claim 14 wherein R<sub>1</sub> is selected from the group consisting of methyl, ethyl, propyl, 2-methylpropyl, 2-amino-2-methylpropyl, 3-amino-2,2-dimethylpropyl, 2,3-dihydroxypropyl, 2-fluoro-2-methylpropyl, and 2-hydroxy-2-methylpropyl.
- 16. The compound or salt of claim 14 wherein R<sub>1</sub> is selected from the group consisting of (1-hydroxycyclobutyl)methyl, (1-hydroxycyclopentyl)methyl, and (1-hydroxycyclohexyl)methyl.
- 17. The method of any one of claims 1 through 11 wherein R<sub>1</sub> is heterocyclylalkylenyl wherein heterocyclyl is unsubstituted or substituted by one or more substituents independently selected from the group consisting of alkyl, aryl, heteroaryl, hydroxy, and oxo.
- 18. The method of claim 17 wherein heterocyclyl is selected from the group consisting of 1,3-dioxolanyl, tetrahydropyranyl, tetrahydrofuranyl, pyrrolidinyl, piperidinyl, and morpholinyl, each of which is unsubstituted or substituted by one, two, or three substituents selected from the group consisting of alkyl, aryl, heteroaryl, and oxo.
- 19. The method of claim 18 wherein heterocyclyl is selected from the group consisting of 1,3-dioxolanyl, tetrahydropyranyl, tetrahydrofuranyl, pyrrolidinyl, piperidinyl, and

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morpholinyl, each of which is unsubstituted or substituted by one, two, or three substituents selected from the group consisting of alkyl, aryl, heteroaryl, and oxo; and alkylenyl is  $C_{1-4}$  alkylenyl.

- 5 20. The method of claim 19 wherein R<sub>1</sub> is selected from the group consisting of tetrahydro-2*H*-pyran-4-ylmethyl and (2,2-dimethyl-1,3-dioxolan-4-yl)methyl.
  - 21. The compound or salt of claim 17 wherein R<sub>1</sub> is (4-hydroxytetrahydro-2*H*-pyran-4-yl)methyl.
- 22. The method of any one of claims 1 through 11 wherein R<sub>1</sub> is -X-Y-R<sub>4</sub> wherein X is C<sub>1-6</sub> alkylene which may be interrupted by one -O- group; Y is selected from the group consisting of -N(R<sub>8</sub>)-C(O)-, -N(R<sub>8</sub>)-S(O)<sub>2</sub>-, -N(R<sub>8</sub>)-C(O)-N(R<sub>8</sub>)-, and -S(O)<sub>2</sub> wherein R<sub>8</sub> is selected from hydrogen and methyl; and R<sub>4</sub> is selected from the group consisting of C<sub>1-6</sub> alkyl, isoquinolinyl, N-methylimidazolyl, pyridinyl, quinolinyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of chloro, cyano, fluoro, hydroxy, and methyl.
- 23. The method of claim 22 wherein R<sub>1</sub> is selected from the group consisting of 2[(cyclopropylcarbonyl)amino]ethyl, 4-[(cyclopropylcarbonyl)amino]butyl, 2[(cyclohexylcarbonyl)amino]-2-methylpropyl, 2-{[(1-methylethyl)carbonyl]amino}ethyl,
  4-{[(1-methylethyl)carbonyl]amino}butyl, 2-methyl-2-{[(1methylethyl)carbonyl]amino}propyl, 2-[(methylsulfonyl)amino]ethyl, 4[(methylsulfonyl)amino]butyl, 2-methyl-2-[(methylsulfonyl)amino]propyl, 2-methyl-2-[2(methylsulfonyl)ethoxy]propyl, and 2,2-dimethyl-3-(methylsulfonyl)propyl.
  - 24. The method of any one of claims 1 through 11 wherein  $R_1$  is -X-Y- $R_4$  wherein X is  $C_{1-6}$  alkylene which may be interrupted by an -O- group; Y is selected from the group consisting of -N( $R_8$ )-C(O)-, -N( $R_8$ )-S(O)<sub>2</sub>-, -N( $R_8$ )-C(O)-N( $R_8$ )-,

 $-S(O)_2$ -,  $R_{10}$  is pentylene,  $R_8$  is hydrogen or methyl; and  $R_4$  is selected from the group consisting of  $C_{1-6}$  alkyl, hydroxy $C_{1-6}$  alkyl, isoquinolinyl, *N*-methylimidazolyl, pyridinyl, quinolinyl, benzyl, 1-phenylethyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of chloro, cyano, fluoro, hydroxy, and methyl.

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25. The method of any one of claims 1 through 11 wherein R<sub>1</sub> is -X-R<sub>5</sub> wherein X is

$$-N-S(O)_2 -N(R_8)-C(O)-N A A$$
 C<sub>1-6</sub> alkylene, and R<sub>5</sub> is

- 26. The method of claim 25 wherein R<sub>1</sub> is selected from the group consisting of 4-(1,1-10 dioxidoisothiazolidin-2-yl)butyl, 4-[(4-morpholinecarbonyl)amino]butyl, and 2-[(4-morpholinecarbonyl)amino]ethyl.
  - 27. The method of any one of claims 1 through 7 or 11 through 26 except as dependent on claim 8, 9, or 10 wherein R<sub>3</sub> is selected from the group consisting of aryl, arylalkyleneoxy, heteroaryl, and heteroarylalkyleneoxy, wherein aryl, arylalkyleneoxy, heteroaryl, and heteroarylalkyleneoxy, are unsubstituted or substituted with one or more substituents selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and
- hydroxyalkyl.
- 28. The method of claim 27 wherein R<sub>3</sub> is phenyl, pyridin-2-yl, pyridin-3-yl, pyridin-4-yl, quinolin-3-yl, or thiazol-4-ylmethoxy any of which may be unsubstituted or substituted by one or more substituents selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl.
- 25 29. The method of claim 28 wherein R<sub>3</sub> is selected from the group consisting of pyridin-3-yl, pyridin-4-yl, 6-fluoropyridin-3-yl, 5-(hydroxymethyl)pyridin-3-yl, 2-ethoxyphenyl, quinolin-3-yl, and thiazol-4-ylmethoxy.
- 30. The compound or salt of any one of claims 1 through 7 or 11 through 26 except as dependent on claim 8, 9, or 10 wherein R<sub>3</sub> is thien-3-yl, phenyl, pyridin-2-yl, pyridin-3-yl,

pyridin-4-yl, or quinolin-3-yl any of which may be unsubstituted or substituted by one or more substituents selected from the group consisting of alkyl, alkoxy, halogen, cyano, hydroxy, and hydroxyalkyl.

- The method of any one of claims 1 through 7, or 11 through 26 except as dependent on claim 8, 9, or 10 wherein R<sub>3</sub> is -Z-X-Y-R<sub>4</sub> wherein X is phenylene, Y is selected from the group consisting of -C(O)-, -C(O)-N(R<sub>8</sub>)-, -N(R<sub>8</sub>)-C(O)-, -N(R<sub>8</sub>)-S(O)<sub>2</sub>-, and -N(R<sub>8</sub>)-C(O)-N(R<sub>8</sub>)- wherein R<sub>8</sub> is selected from hydrogen and methyl; Z is a bond; and R<sub>4</sub> is selected from the group consisting of C<sub>1-6</sub> alkyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl.
- 32. The method of any one of claims 1 through 7, or 11 through 26 except as dependent on claim 8, 9, or 10 wherein R<sub>3</sub> is -Z-X-Y-R<sub>4</sub> wherein Z is a bond, X is
  15 phenylene, Y is selected from the group consisting of -C(O)-, -C(O)-N(R<sub>8</sub>)-, -N(R<sub>8</sub>)-C(O)-, -N(R<sub>8</sub>)-S(O)<sub>2</sub>-, and -N(R<sub>8</sub>)-C(O)-N(R<sub>8</sub>)- wherein R<sub>8</sub> is selected from hydrogen and methyl; and R<sub>4</sub> is selected from the group consisting of C<sub>1-6</sub> alkyl, phenyl, and phenyl substituted by a substituent selected from the group consisting of alkyl, alkoxy, halogen, hydroxy, and hydroxyalkyl; with the proviso that when Y is -C(O)-N(R<sub>8</sub>)- or -N(R<sub>8</sub>)-C(O)-N(R<sub>8</sub>)- then R<sub>4</sub> can also be hydrogen; and with the further proviso that when Y is -C(O)- or -N(R<sub>8</sub>)-C(O)- then R<sub>4</sub> can also be morpholin-4-yl, piperidin-1-yl, or pyrrolidin-1-yl.
- 33. The method of claim 32 wherein R<sub>3</sub> is 3-(methylsulfonylamino)phenyl, 3(pyrrolidin-1-ylcarbonyl)phenyl, or 3-(morpholin-4-ylcarbonyl)phenyl.
  - 34. The method of any one of claims 1 through 8, or 11 through 33 except as dependent on claim 9 or 10 wherein R is not present.
- 35. The method of any one of claims 1 through 7, or 11 through 26 except as dependent on claim 8, 9, or 10 wherein neither R<sub>3</sub> nor R is present.

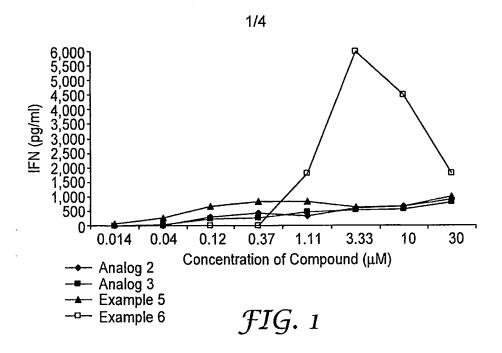
36. The method of any one of claims 1 through 8, or 11 through 26 except as dependent on claim 9 or 10 wherein R is selected from the group consisting of hydroxy and methoxy.

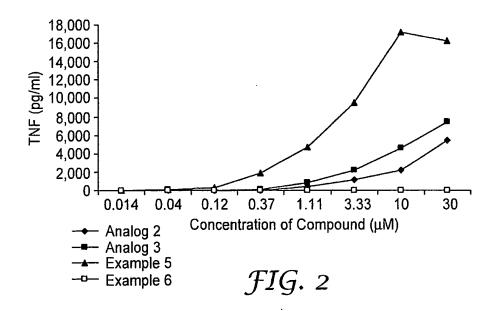
- 5 37. The method of claim 36 as dependent on any one of claims 1 through 7, or 11 through 26 except as dependent on claim 8, 9, or 10 wherein R<sub>3</sub> is not present.
  - 38. The method of any one of claims 1 through 37 wherein an effective amount of the compound or salt is administered as a pharmaceutical composition comprising a therapeutically effective amount of the compound or salt and a pharmaceutically acceptable carrier.

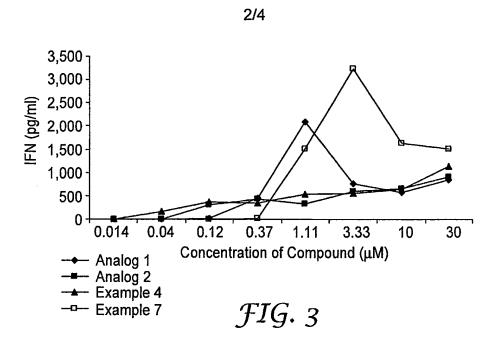
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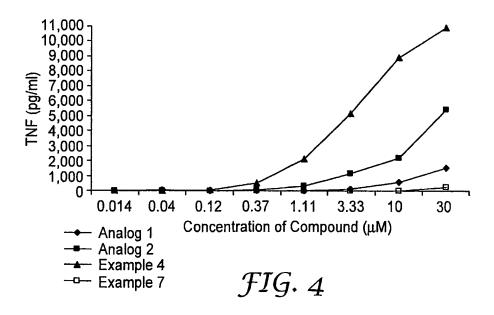
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- 39. A method of treating a viral disease in an animal in need thereof comprising preferentially inducing the biosynthesis of IFN-α in the animal according to the method of any one of claims 1 through 38.
- 40. A method of treating a neoplastic disease in an animal in need thereof comprising preferentially inducing the biosynthesis of IFN-α in the animal according to the method of any one of claims 1 through 38.
- 41. The method of anyone of claims 1 through 40 wherein the compound or salt is administered systemically.

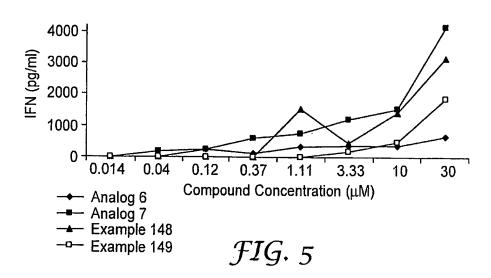


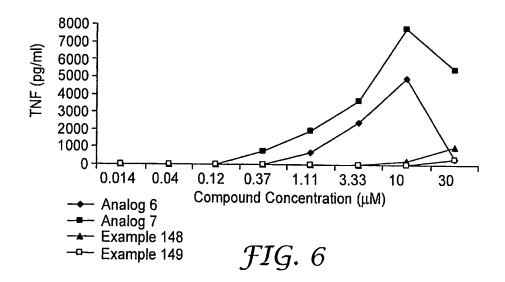






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